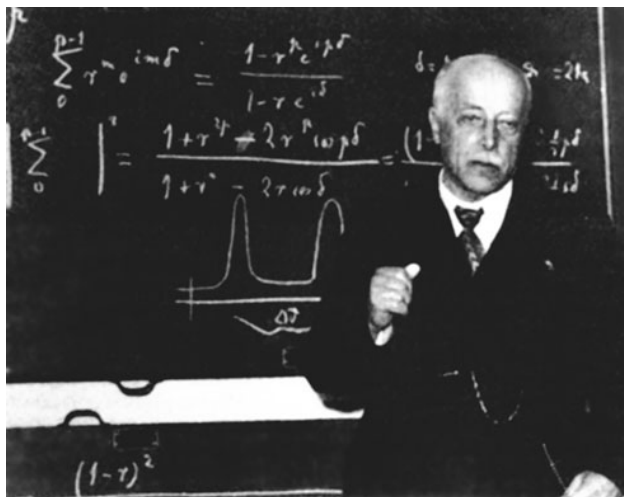


Figures for 2012 Publication:



Herbert Freundlich
1880 – 1941
(I: 1916 – 1933)

Fig. 1



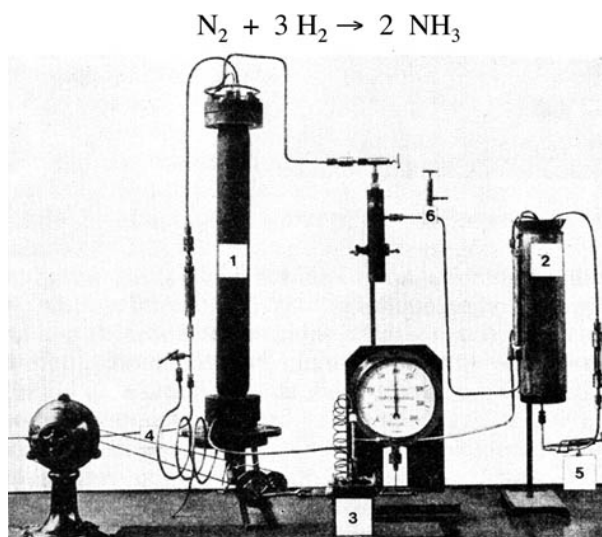
Max von Laue
1879 – 1960
(I: 1951–1960)

Fig. 2



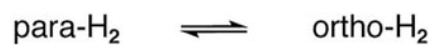
Fritz Haber
1868 – 1934
(I: 1911–1933)

Fig. 3



F. Haber & R. LeRossignol, 1909
Z. Elektrochem. **16** (1910), 244; **19** (1913), 53

Fig. 4



Nuclear spin

$I = 0$

$I = 1$

Rotational quantum numbers

$J = 0, 2, 4, \dots$

$J = 1, 3, 5, \dots$

Fig. 5

K.F. Bonhoeffer
1899 - 1957
(l: 1923-1930, 1948-1949)



P. Harteck
1902 - 1985
(l: 1928-1933, e: 1956-1985)

Fig. 6

Über Para- und Orthowasserstoff.

Von

K. F. Bonhoeffer und P. Harteck.

(Aus dem Kaiser Wilhelm-Institut für physikalische Chemie und Elektrochemie, Berlin-Dahlem.)

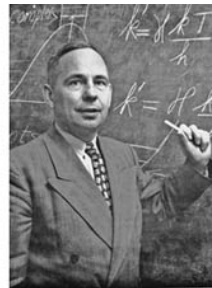
(Mit 5 Figuren im Text.)

(Eingegangen am 22. 5. 29.)

Z. phys. Chem. B 4 (1929), 113

Fig. 7

H. Eyring
1901 - 1981
(l: 1929-1930)



M. Polanyi
1891 - 1976
(l: 1923-1933)

Über einfache Gasreaktionen.

Von

H. Eyring und M. Polanyi.

(Aus dem Kaiser Wilhelm-Institut für physikalische Chemie und Elektrochemie, Berlin-Dahlem.)

(Mit 17 Figuren im Text.)

(Eingegangen am 7. 2. 31.)

Z. phys. Chem. B 12 (1931), 279

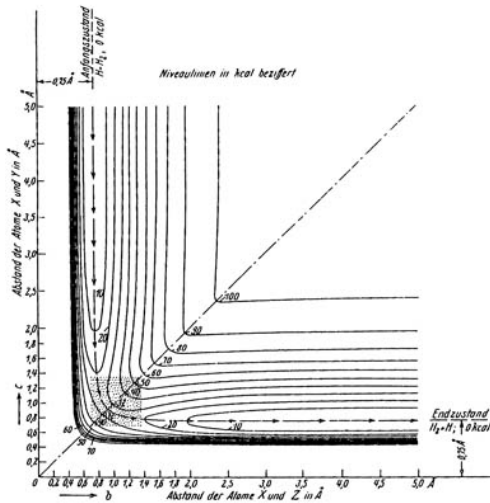
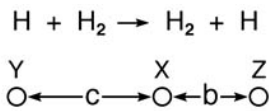
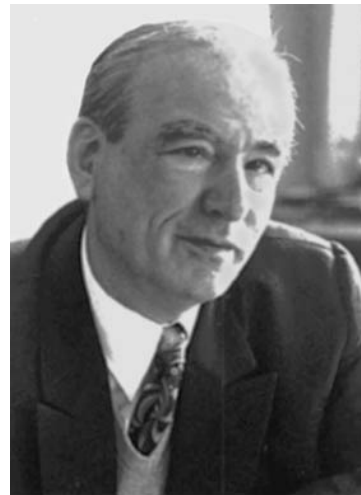


Fig. 8



E. Ruska
1906 - 1988
(l: 1949-1988)

Fig. 10



50 Years Dynamics of Chemical Reactions
Berlin, October 12–15, 1981

Fig. 9

Ag catalyst

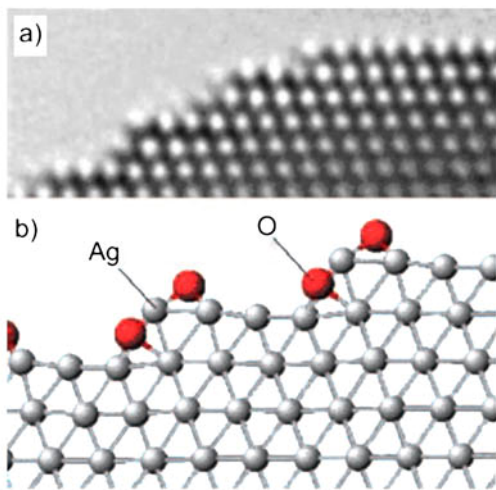


Fig. 11



E.W. Müller
1911 – 1977
(l: 1947-1951, e: 1957-1977)

Fig. 12

Zeitschrift für Physik, Bd. 131, S. 136–142 (1951).

Das Feldionenmikroskop.

Von
ERWIN W. MÜLLER.
Mit 3 Figuren im Text.

(Eingegangen am 27. August 1951.)

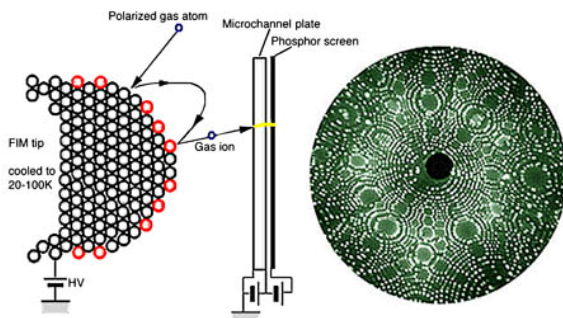
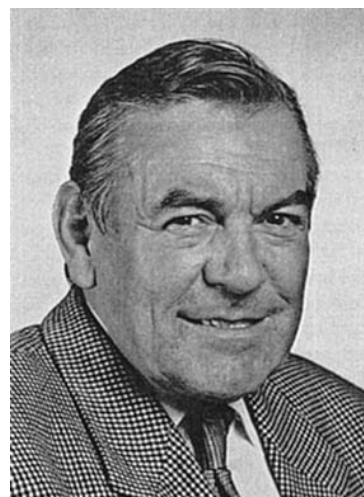


Fig. 13



J. Block
1929 – 1995
(l: 1966–1995)

Fig. 14

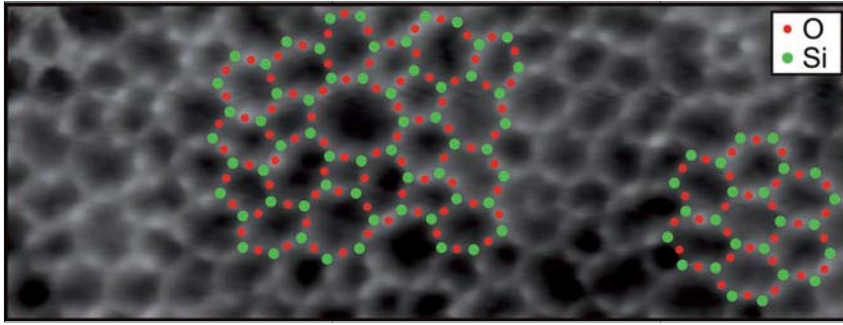


Fig. 15

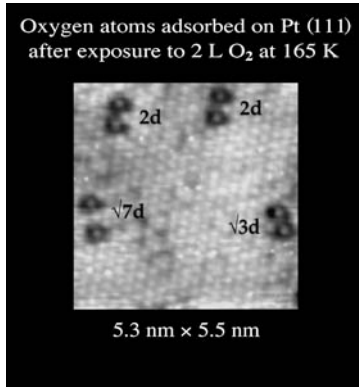


Fig. 16

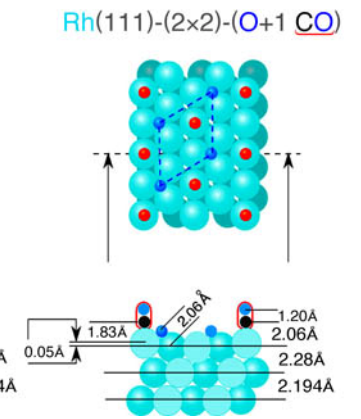
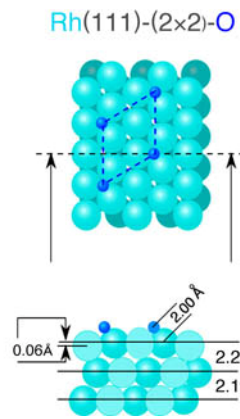
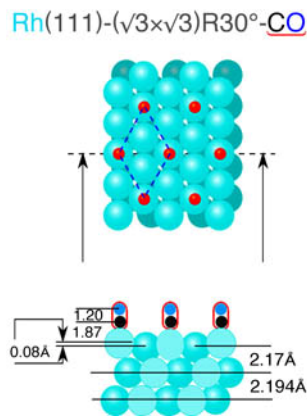


Fig. 17

CO^{cus} + O^{cus} → CO₂/RuO₂(110) : Reaction barrier

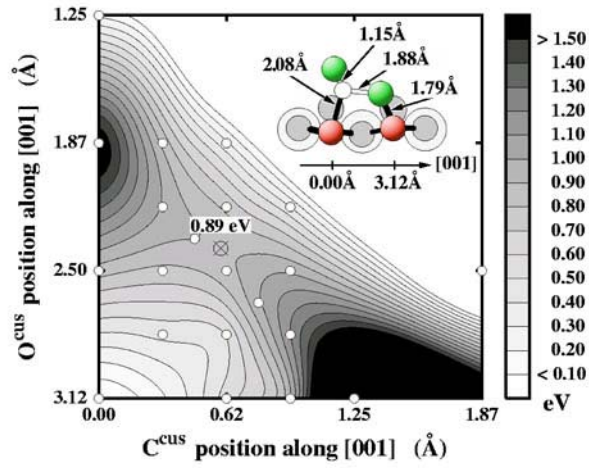
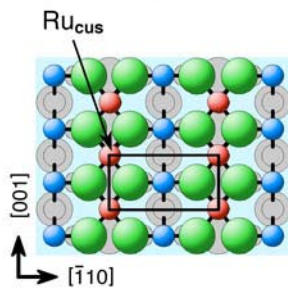


Fig. 18

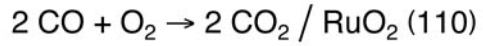


Fig. 19

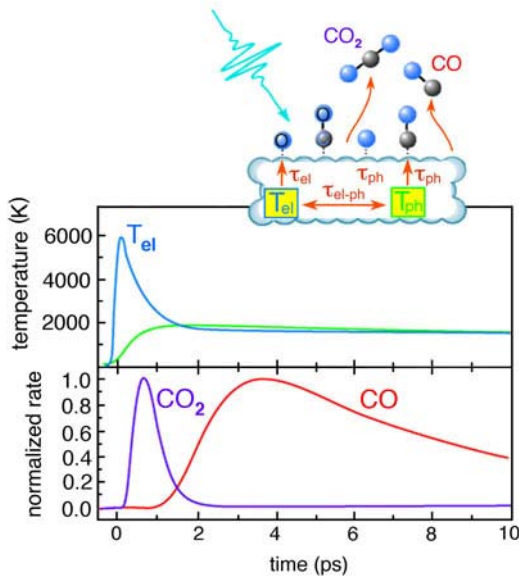
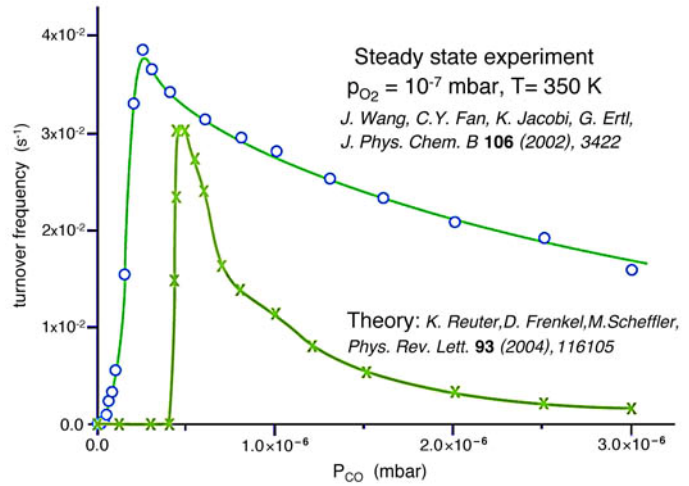


Fig. 20

Z. Elektrochem. **52** (1948), 149

Über periodische chemische Reaktionen

Das anodische Verhalten von Kupfer in Salzsäure.
K. F. Bonhoeffer und Heinz Gerischer.

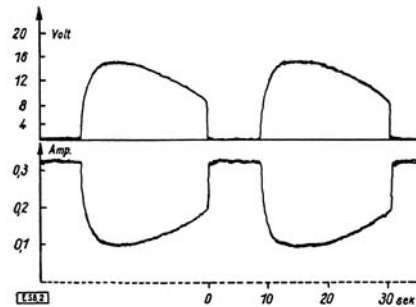


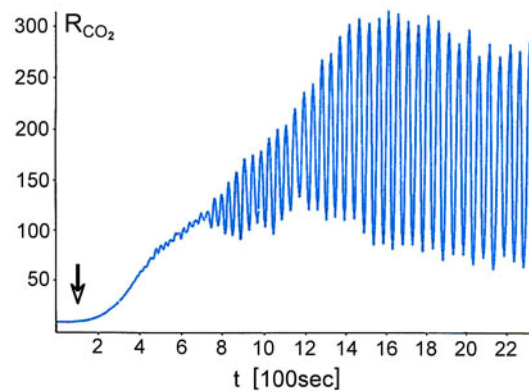
Fig. 21



H. Gerischer

1919 – 1994
(I: 1948–1949, 1969–1994)

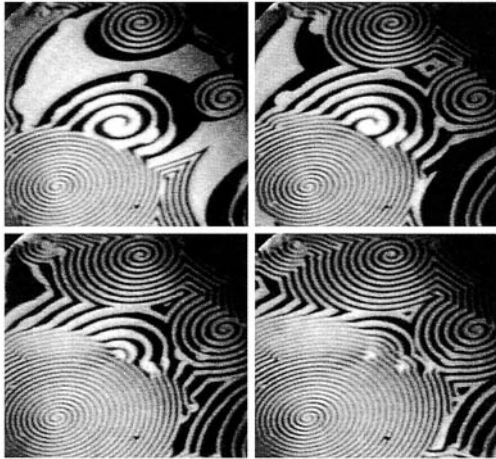
Fig. 22



$T = 470\text{K}; p_{\text{CO}} = 3 \times 10^{-5} \text{mbar}; p_{\text{O}_2} = 2.0 \rightarrow 2.7 \times 10^{-4} \text{mbar}$

Fig. 23

Spiral waves during CO-oxidation on Pt(110)



PEEM images with 500 μm diameter, real time steady-state conditions: $p_{\text{O}_2} = 4 \times 10^{-4}$ mbar, $p_{\text{CO}} = 4.3 \times 10^{-3}$ mbar, $T = 448$ K

Fig. 24

Science 331 (2011), 892

Quantum Reflection of He₂ Several Nanometers Above a Grating Surface

Bum Suk Zhao,* Gerard Meijer, Wieland Schöllkopf

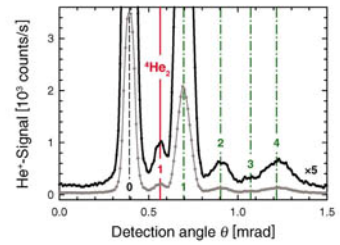
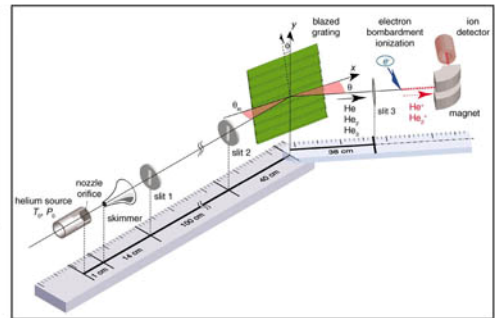
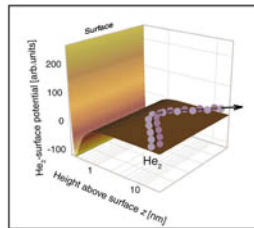


Fig. 25



**Anniversary March
(Fritz-Haber-Jubiläumsmarsch)**

Thomas Hennig

Klavier 1

ff

Allegro ma non troppo

Klavier 2

ff

f

Fig. 26