# Variations in <sup>13</sup>C discrimination during CO<sub>2</sub> exchange by *Picea sitchensis* branches in the field

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### ABSTRACT

We report diurnal variations in <sup>13</sup>C discrimination ( $^{13}\Delta$ ) of Picea sitchensis (Bong.) Carr. branches measured in the field using a branch chamber technique. The observations were compared to predicted <sup>13</sup> based on concurrent measurements of branch gas exchange. Observed <sup>13</sup>A values were described well by the classical model of <sup>13</sup>A including isotope effects during photorespiration, day respiration and CO<sub>2</sub> transfer through a series of resistances to the sites of carboxvlation. A simplified linear of model <sup>13</sup>∆ did not capture the observed diurnal variability. At dawn and dusk, we measured very high  ${}^{13}\Delta$  values that were not predicted by either of the said models. Exploring the sensitivity of  $^{13}\Delta$  to possible respiratory isotope effects, we conclude that isotopic disequilibria between the gross fluxes of photosynthesis and day respiration can explain the high observed <sup>13</sup> values during net photosynthetic gas exchange. Based on the classical model, a revised formulation incorporating an isotopically distinct substrate for day respiration was able to account well for the high observed dawn and dusk  $^{13}\Delta$ values.

*Key-words*: branch chambers; carbon isotopes; day respiration; photosynthetic discrimination.

# INTRODUCTION

Nearly 30% of the entire atmospheric pool of CO<sub>2</sub> (750 Gt) is exchanged across leaf surfaces each year. The  $\delta^{13}$ C enrichment that accompanies the summer drawdown of CO<sub>2</sub> in the atmosphere can be ascribed to photosynthetic discrimination against <sup>13</sup>CO<sub>2</sub> by the terrestrial biosphere. Globally, photosynthetic uptake of CO<sub>2</sub> discriminates by 15 to 18‰ against <sup>13</sup>C (weighted mean of C<sub>3</sub> and C<sub>4</sub> vegetation, Lloyd & Farquhar 1994). For plants with the C<sub>3</sub> photosynthetic pathway, the photosynthetic discrimination against <sup>13</sup>C (<sup>13</sup>Δ)

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has been related to the lower reactivity of <sup>13</sup>C during fixation by photosynthetic enzymes, as well as to differential diffusivities of <sup>13</sup>CO<sub>2</sub> and <sup>12</sup>CO<sub>2</sub> in air (Farquhar, O'Leary & Berry 1982; O'Leary 1984). The magnitude of photosynthetic <sup>13</sup>C discrimination is sensitive to environmental variables such as vapour pressure deficit, photon flux density and air temperature. The  $\delta^{13}$ C composition of CO<sub>2</sub> thus provides a natural tracer for component processes and their regulation by environmental conditions.

The  $\delta^{13}$ C signature of photosynthetic fluxes often responds differently to changes in environmental conditions than that of concurrent plant and soil respiratory fluxes, creating transient isotopic disequilibria (Lloyd et al. 1996). This is employed at the ecosystem and global scales to interpret field data and to simulate isotopic gas exchange between the terrestrial biosphere and the atmosphere with numerical models. For example, time series of the  $\delta^{13}$ C of atmospheric  $CO_2$  are tools in inversion studies for tracking changes in CO<sub>2</sub> fluxes of the terrestrial biosphere and ocean in response to climate variability (Tans, Berry & Keeling 1993; Francey et al. 1995). Plant physiology models incorporating carbon isotope ratios are important for predicting the influence of different biomes on the global patterns of  $\delta^{13}$ C in atmospheric CO<sub>2</sub>, providing a tool for constraining largescale fluxes (Lloyd & Farquhar 1994; Randerson et al. 2002). At the ecosystem scale, partitioning may be possible at sites measuring eddy flux and CO<sub>2</sub> concentration profiles with observations of the  $\delta^{13}$ C composition of CO<sub>2</sub> (Lloyd *et al.* 1996; Yakir & Wang 1996; Bowling, Tans & Monson 2001; Ogée et al. 2003, 2004).

But field measurements of photosynthetic <sup>13</sup>C discrimination have only been reported for a few ecosystems (Harwood 1997; Harwood *et al.* 1998). Thus, a gap exists between studies of photosynthetic <sup>13</sup>C discrimination at the leaf scale (generally conducted under steady-state conditions in the laboratory) on the one hand, and the interpretation of field data and modelling at the ecosystem and global scale on the other. We try to link these scales by exploring the natural variability of <sup>13</sup>C discrimination at the largest scale possible for direct measurements: whole branches enclosed in chambers. Through this choice of scale and method of observation, we have a useful platform for making observations under field conditions relevant for the

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interpretation of isotopic data at increasing temporal resolution (Bowling *et al.* 2003, 2005; Griffis *et al.* 2004; Griffis, Baker & Zhang 2005).

Here, we present daily variations in environmental conditions, CO<sub>2</sub> and water fluxes of branches in a *Picea sitchensis* (Bong.) Carr. plantation in central Scotland alongside concurrent measurements of discrimination against <sup>13</sup>C (<sup>13</sup> $\Delta_{obs}$ ) in the field. This paper demonstrates how trace gas measurements during 5 min closure intervals of branch chambers can be used to: (1) obtain diurnal data sets of <sup>13</sup> $\Delta_{obs}$  from flask observations in the field; (2) derive estimates of <sup>13</sup>C discrimination from gas-exchange and microclimate data that are directly comparable to the flask observations; and (3) derive <sup>13</sup> $\Delta$  estimates that correspond to those expected in the absence of chambers. We also explore the sensitivity of net <sup>13</sup>C discrimination to isotope effects during internal CO<sub>2</sub> transfer, photorespiration and day respiration.

### MATERIALS AND METHODS

#### Study site and gas-exchange measurements

The study was made in Griffin Forest, an even-aged plantation of Queen Charlotte Islands provenance *P sitchensis* (Bong.) Carr. located near Aberfeldy, Perthshire, UK (56°37″N, 3°48″W). A description of the site, a long-term monitoring station within the CARBOEUROPE-IP network can be found at http://www.carboeurope.org.

Three branch chambers constructed from extruded acrylic and polypropylene film (34  $\mu$ m thick, ICI Propafilm, Dumfries, UK) were installed on two trees of 13.6 and 9.5 m height within a 0.01 ha plot from 18 May 2001 to 22 July 2001. Bud burst was partially occurring in the canopy in May, while in July shoot expansion was complete. Branches were chosen as representative of sun-lit branches visible from the tower. Two branches were used in the upper canopy at 10.5 m (chamber 1) and 9.4 m (chamber 3) and one in the middle canopy at 8.1 m (chamber 4) in height. During the July campaign, chamber 4 was sealed without a branch inside and designated as a control. Chambers were positioned to the south of the tower to minimize shading effects. They were held in position by metal booms projecting from the canopy access tower and nylon cord, allowing the chambers to move with the branches. The chambers are fully described by Rayment & Jarvis (1999) and Wingate (2003). The experimental set-up has been used to investigate branch O2:CO2 exchange and 18O discrimination (Seibt et al. 2004, 2006). The latter also employed the same simulation approach as the present study.

During the day, the bags were closed for 5 min, after which they were opened and ventilated for 15 min, i.e. measurements were made on each branch every 20 min. At night, they were closed for 10 min and ventilated for 30 min, i.e. the measurement cycle changed to every 40 min. During closure periods, relative humidity (h) and air temperature ( $T_a$ ) were monitored within each bag [Vaisala HMP 35A; Vaisala (UK) Ltd, Cambridge, UK]. Needle temperature  $(T_1)$  was measured in three locations per chamber (0.2 mm in diameter Cu-Con thermo-junction, referenced to air temperature). Photosynthetic photon flux density (Q) incident upon each branch was measured using a photosynthetic photon flux density sensor (SD101QV; Macam Ltd, Livingston, UK) attached vertically on the branch midway along its length. At the same time, the CO<sub>2</sub> mole fraction of chamber air was monitored with an infrared gas analyser (IRGA) (LI-6262; Li-Cor Inc., Lincoln, NE, USA). Outputs were recorded with a logger (CR10; Campbell Scientific Ltd, Shepshed, UK) that also served to initiate the measurement sequence. The CO<sub>2</sub> concentration in reference bottles of air was calibrated against gas mixtures prepared by mixing pure  $CO_2$  with  $CO_2$ -free air using three precision gas mixing pumps (G27/3F, SA18/3F and SA17/3F; H.Wösthoff GmbH, Bochum, Germany). H<sub>2</sub>O vapour span was set using a dew point generator (LI-610; Li-Cor Inc.). The chambers were kept open in a continuously ventilated state when not in use.

Rates of net CO<sub>2</sub> assimilation (A,  $\mu$ mol m<sup>-2</sup> s<sup>-1</sup>), transpiration (E, mol m<sup>-2</sup> s<sup>-1</sup>) and branch conductances to water vapour and CO<sub>2</sub> ( $g_s$ ,  $g_c$ , mol m<sup>-2</sup> s<sup>-1</sup>) were calculated and integrated over the full chamber closure periods (see Appendix I). To calculate the leaf surface CO<sub>2</sub> mole fraction ( $C_s$ ), the leaf boundary layer conductance for CO<sub>2</sub> inside the chamber ( $g_b$ ) was estimated as  $1.5 \pm 0.5$  mol m<sup>-2</sup> s<sup>-1</sup> from energy balance considerations. The intercellular CO<sub>2</sub> mole fraction incorporating ternary effects (Jarman 1974),  $C_{i,g}$  ( $\mu$ mol mol<sup>-1</sup>), was calculated following the notation of von Caemmerer & Farquhar (1981) as

$$C_{i,g} = \frac{\left(g_c - \frac{E}{2}\right)C_s - A}{\left(g_c + \frac{E}{2}\right)} \tag{1}$$

The subscript g denotes that for a heterogeneous system such an estimate is weighted according to the conductances and not that of A (Farquhar 1989) (see Appendix III).

#### Collection and analysis of organic material

During the May field campaign, needle and non-green twig samples were collected from the same or adjacent Sitka spruce trees. Samples were taken at the same heights and close to the chamber locations. All samples were stored in glass sample vials, transferred to the laboratory as rapidly as possible and stored at -20 °C until analysis. Samples were dried and ground with a ball mill to a fine homogeneous powder. Subsamples were weighed and their carbon isotope ratios determined on an elemental analyser (NA 1110 CN; CE Instruments, Rodano, Italy), coupled via open split interface to a mass spectrometer (ConFlo III and Delta<sup>+</sup>XL; both Finnigan MAT, Bremen, Germany). SD for dry matter  $\delta^{13}$ C was 0.05‰. The samples were analysed at the Max Planck Institute for Biogeochemistry, Jena, Germany.

#### Collection and analysis of flask samples

Pairs of air samples from branch chambers were collected at intervals of approximately 3 h over 24 h in spring (18/19 May) and summer (20 July) of 2001. Air was circulated from the branch chambers through sampling lines into a flask sampling system (separate from that used for gasexchange measurements), and back into the chamber. Within the sampling system, the air stream was passed through a magnesium perchlorate cylinder to remove water vapour, and pumped through two flasks in series. We used 1 dm<sup>3</sup> glass flasks with a valve (Glass Expansion, Melbourne, Australia) on each end, sealed with Teflon perfluoroalkoxy (PFA) O-rings (DuPont Fluoro products, Wilmington, DE, USA), and 1.3 dm3 flasks with two valves (Louwers, Hapert, the Netherlands) on the same end. Flask samples were taken at two points in the sequence of opening and closing of branch chambers. The first flask was collected 3 to 4 min before closure ('open sample') and reflects the  $CO_2$  mole fraction and isotopic signature of ambient canopy air. The second flask ('closed sample') was collected shortly before reopening of the chamber.

The CO<sub>2</sub> mole fractions in the air samples were determined using a gas chromatograph (HP 6890; Hewlett Packard, Palo Alto, CA, USA) linked to a methanizer and flame ionization detector. Subsamples of CO<sub>2</sub> were then extracted cryogenically from the dry air samples ('BGC-AirTrap', Werner, Rothe & Brand 2001), and their isotope ratio determined on a dual inlet isotope ratio mass spectrometer (Delta<sup>+</sup>XL; Finnigan MAT). The analytical precision was in the order of  $0.08 \,\mu$  mol mol<sup>-1</sup> for CO<sub>2</sub> mole fraction and 0.01‰ for  $\delta^{13}$ C, reported with respect to V-PDB-CO<sub>2</sub>. Uncertainties of flask data were estimated from the SDs of laboratory analysis of duplicate flasks, 0.13  $\mu$ mol mol<sup>-1</sup> for CO<sub>2</sub> mole fraction and 0.01‰ for  $\delta$ <sup>13</sup>C. Additional uncertainties arising from the sampling procedure, applying to samples collected at the end of closure periods, were estimated from control measurements in an empty chamber. They were  $1.3 \,\mu$ mol mol<sup>-1</sup> for CO<sub>2</sub> mole fraction and 0.2‰ for  $\delta^{13}$ C.

# Calculations of $\delta^{13}$ C signatures of CO<sub>2</sub> exchange from flask data

The observed values of <sup>13</sup>C discrimination during photosynthesis, <sup>13</sup> $\Delta_{obs}$  (‰) of foliage in the closed branch chamber were determined following Guy *et al.* (1989):

$${}^{13}\Delta_{obs} = \frac{\ln\frac{\Re_e}{\Re_o}}{\ln\frac{C_e}{C_o}} \left(1 + \frac{\ln\frac{\Re_e}{\Re_o}}{\ln\frac{C_e}{C_o}}\right)^{-1},$$
(2)

where  $C_o$  and  $C_e$  are the mole fractions ( $\mu$ mol mol<sup>-1</sup>) and  $\Re_o$ and  $\Re_e$  the <sup>13</sup>C/<sup>12</sup>C ratios of CO<sub>2</sub> at the beginning and end of closure periods, respectively. These correspond to the samples collected from the open and closed chambers for flask measurements, and to starting time values and those integrated over the closure periods for gas-exchange data (see Appendix I). We use Eqn 2 for both flask observations and <sup>13</sup> $\Delta$  predictions from gas-exchange data (although not strictly valid as <sup>13</sup> $\Delta$  is changing during the closure period) because it allows us to directly compare the two approaches. Estimated uncertainties for <sup>13</sup> $\Delta_{obs}$  were calculated using Gaussian error propagation. They were inversely related to net flux rates, usually < 1‰, but > 5‰ at dawn and dusk. Discrimination values were also largest at these times.

All calculations were performed in interactive data language (IDL) (Version 6.1; Research Systems Inc., Boulder, CO, USA). Correlation parameters between predicted and observed <sup>13</sup> $\Delta$  were obtained from least absolute deviation regressions using the flask observations as independent variable. Four data points were excluded due to lack of reliable gas-exchange measurements (18 May, 0710 h IRGA tubing detached, 0740 h relative humidity constant at 50%, 1740 h sensor artefacts from direct sunlight, 20 July, 0340 h air saturated).

## SIMULATIONS

#### Predictions of ${}^{13}\Delta$ during foliage gas exchange

We calculated  ${}^{13}\Delta$  values from chamber measurements of environmental conditions, air composition and fluxes (see Appendix I) based on the three equations described next. Summary of symbols and definitions are given in Table 1. The most comprehensive ('classical') model of net  ${}^{13}C$  discrimination ( ${}^{13}\Delta_{classical}$ ,  ${}^{\infty}_{0}$ ) for C<sub>3</sub> plants describes our present understanding of fractionation steps during net assimilation of CO<sub>2</sub> (Farquhar *et al.* 1982) as

$${}^{13}\Delta_{\text{classical}} = a_{\text{b}} \frac{C_{\text{a}} - C_{\text{s}}}{C_{\text{a}}} + a \frac{C_{\text{s}} - C_{\text{i}}}{C_{\text{a}}} + a_{\text{m}} \frac{C_{\text{i}} - C_{\text{c}}}{C_{\text{a}}} + b \frac{C_{\text{c}}}{C_{\text{a}}}$$
(3)  
$$- f \frac{\Gamma_{*}}{C_{\text{a}}} - e \frac{R_{\text{d}}}{kC_{\text{a}}}$$

where  $a_{\rm b}$  and a are the fractionations during diffusion of CO<sub>2</sub> in air through the leaf boundary layer and the stomata ( $\approx 2.9$  and 4.4‰, Craig 1953), and b is the combined fractionation during carboxylation by ribulose 1,5bisphosphate carboxylase/oxygenase (Rubisco) and phospho-enol-pyruvate carboxylase (PEPc) (assumed to be 29‰, Roeske & O'Leary 1984; Guy et al. 1989; Guy, Fogel & Berry 1993).  $a_{\rm m}$  is the sum of the fractionation factors during internal CO<sub>2</sub> transfer, combining an equilibrium fractionation of CO<sub>2</sub> entering solution ( $e_s = 1.1\%$  at 25 °C, Mook, Bommerson & Staverman 1974) and a diffusional fractionation of dissolved CO<sub>2</sub> in water ( $a_d = 0.7\%$ , O'Leary 1984). The respiratory fractionation factors are f for photorespiration ( $8 \pm 1$ %, Rooney 1988; Gillon & Griffiths 1997) and e for day respiration,  $R_d$ .  $C_c$  is the CO<sub>2</sub> mole fraction at the sites of carboxylation, k is the carboxylation efficiency and  $\Gamma_*$  is the CO<sub>2</sub> compensation point in the absence of day respiration.

Table 1.	List of	symbols	used	in	the	text	
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Symbols used		
A	Measured net rate of assimilation	$\mu$ mol m <sup>-2</sup> s <sup>-1</sup>
$A_{\max}$	Fitted maximum assimilation rate	$\mu$ mol m <sup>-2</sup> s <sup>-1</sup>
a	<sup>13</sup> C fractionation during diffusion through the stomata	4.4‰
a <sub>b</sub>	<sup>13</sup> C fractionation during diffusion through the boundary layer	2.9‰
a <sub>d</sub>	$^{13}$ C fractionation during diffusion through water	0.7‰
a <sub>m</sub>	<sup>13</sup> C fractionation during internal CO <sub>2</sub> transfer ( $e_s + a_d$ ) Not <sup>13</sup> C fractionation during aerbowilation by ribulase 1.5 hisphasenbeta carbovylase/ovygapase and	1.8‰ 29‰
Ь	Net <sup>13</sup> C fractionation during carboxylation by ribulose 1,5-bisphosphate carboxylase/oxygenase and Phospho-enol-pyruvate carboxylase	29/00
$\overline{b}$	Estimated net <sup>13</sup> C fractionation during carboxylation, internal $CO_2$ transfer and decarboxylation	27‰
$C_{\rm a}$	$CO_2$ mole fraction in ambient air	$\mu$ mol mol <sup>-1</sup>
$C_{c}$	$CO_2$ mole fraction at the sites of carboxylation	$\mu$ mol mol <sup>-1</sup>
C <sub>e</sub>	$CO_2$ mole fraction of chamber air at the end of closure period	$\mu$ mol mol <sup>-1</sup>
Ci	$CO_2$ mole fraction in the intercellular spaces	$\mu$ mol mol <sup>-1</sup>
C <sub>i, A</sub>	Assimilation weighted intercellular $CO_2$ mole fraction	$\mu$ mol mol <sup>-1</sup>
C <sub>i, g</sub>	Conductance weighted intercellular CO <sub>2</sub> mole fraction	$\mu$ mol mol <sup>-1</sup>
$C_{\rm o}$	$CO_2$ mole fraction of chamber air at the beginning of closure period	$\mu$ mol mol <sup>-1</sup>
$C_{\rm s}$	$CO_2$ mole fraction at the leaf surface	$\mu$ mol mol <sup>-1</sup>
$^{13}\Delta$	Net discrimination against <sup>13</sup> C during photosynthesis	
$^{13}\Delta_{obs}$	Observed net <sup>13</sup> C discrimination (Eqn 2)	
$^{13}\Delta_{classical}$	Predicted net <sup>13</sup> C discrimination (Eqn 3, or eqn B24 of Farquhar <i>et al.</i> 1982)	
$\Delta_{\text{pred}}$	Predicted net ${}^{13}$ C discrimination	
$^{13}\Delta_{\text{revised}}$	Predicted net <sup>13</sup> C discrimination (Eqn 5) Predicted net <sup>13</sup> C discrimination (Eqn 4, or eqn 10 of Farquhar <i>et al.</i> 1982)	
$^{13}\Delta_{\text{simple}}$	Carbon isotope signature of the respiratory flux (Eqn A2.10) defined as net respiratory	
$\Delta_{\rm R}$	discrimination in analogy to photosynthetic gas exchange	
$\delta^{13}C$	Carbon isotope composition ( $\Re_{\text{sample}} / \Re_{\text{standard}} -1$ ) × 1000	‰
$\delta^{13}C_a$	Carbon isotope composition of ambient air	%o
$\delta^{13}C_{assimilates}$	Carbon isotope composition of assimilates	%
$\delta^{13}C_e$	Carbon isotope composition of chamber air at the end of closure period	%o
$\delta^{13}C_{needle}$	Carbon isotope composition of bulk needle material	%
$\delta^{13}C_o$	Carbon isotope composition of chamber air at the beginning of closure period	‰
$\delta^{13}C_{plant}$	Carbon isotope composition of bulk plant material	‰
$\delta^{13}C_R$	Carbon isotope composition of respired CO <sub>2</sub>	‰
$\delta^{13}C_{substrate}$	Carbon isotope composition of substrate for dark respiration	‰
$\delta^{13}C_{wood}$	Carbon isotope composition of bulk twig material	%
$D_{\mathrm{a}}$	Vapour mole fraction deficit	mmol mol <sup>-1</sup>
Ε	Transpiration rate	$mol m^{-2} s^{-1}$
е	<sup>13</sup> C fractionation during decarboxylation	0 or -6‰
e <sub>s</sub>	$^{13}$ C fractionation during equilibration of CO <sub>2</sub> entering solution	1.1‰
e* c	Apparent discrimination against <sup>13</sup> C during decarboxylation (= $\Re_{\text{substrate}} / \Re_{\text{assimilates}} - 1$ )	%o
f	Discrimination against ${}^{13}CO_2$ during photorespiration Total branch conductance to water vapour	8‰ mol m <sup>-2</sup> s <sup>-1</sup>
g	Branch boundary layer conductance for CO2	$1.1 \text{ mol m}^{-2} \text{ s}^{-1}$
g <sub>b</sub>	Branch conductance to $CO_2$	mol m <sup>-2</sup> s <sup>-1</sup>
ge	Internal transfer conductance to $CO_2$	$0.16 \text{ mol m}^{-2} \text{ s}^{-1}$
gi gs	Branch conductance to water vapour	$mol m^{-2} s^{-1}$
$\Gamma_*$	Temperature dependent $CO_2$ compensation point (Brooks & Farquhar 1985)	$\mu$ mol mol <sup>-1</sup>
h h	Relative humidity	%
K	Carboxylation efficiency	mol $m^{-2} s^{-1}$
L <sub>A</sub>	Projected needle area	m <sup>2</sup>
Р	Atmospheric pressure	Pa
Q	Photosynthetic photon flux density	$\mu$ mol mol <sup>-1</sup>
$R_{\rm d}$	Measured rate of dark respiration	$\mu$ mol m <sup>-2</sup> s <sup>-1</sup>
$R_{d,n}$	Estimated rate of day respiration, T dependent	$\mu$ mol m <sup>-2</sup> s <sup>-1</sup>
R <sub>d,i</sub>	Estimated rate of day respiration, $T$ and $Q$ dependent	$\mu$ mol m <sup>-2</sup> s <sup>-1</sup>
R <sub>d 20</sub>	Fitted rate of dark respiration at 20 °C	$\mu$ mol m <sup>-2</sup> s <sup>-1</sup>
R <sub>a</sub>	Isotope $\binom{13}{12}$ ratio of ambient CO <sub>2</sub>	
Rassimilates	Isotope ( <sup>13</sup> C) ratio of assimilates	
R <sub>e</sub>	Isotope ( <sup>13</sup> C) ratio of chamber air at end of closure period	
R <sub>o</sub>	Isotope ( <sup>13</sup> C) ratio of chamber air at beginning of closure period	
R <sub>standard</sub>	Isotope ( <sup>13</sup> C) ratio of V–PDB–CO <sub>2</sub> standard	
$\Re_{\text{substrate}}$	Isotope ( <sup>13</sup> C) ratio of substrate used for dark respiration	°C
$T_{\rm a}$	Air temperature Air vapour mole fraction	mmol mol <sup>-1</sup>
W <sub>a</sub>	Initial air vapour mole fraction in chamber at the starting time $t_0$	mmol mol <sup>-1</sup>
W <sub>a0</sub> W <sub>a</sub> (t)	Air vapour mole fraction in chamber calculated from relative humidity	mmol mol <sup>-1</sup>
W <sub>a</sub> (h)	Air vapour mole fraction or corrected for the return flow of dry air during flask sampling	mmol mol <sup>-1</sup>
V <sub>a(corr)</sub> V <sub>i</sub>	Leaf vapour mole fraction	mmol mol <sup>-1</sup>
* 1	Molar volume of air enclosed in the chamber	mol

The most often applied ('simple') model of <sup>13</sup>C discrimination ( ${}^{13}\Delta_{\text{simple}}$ ) neglects any effects of photo- and day respiration on net discrimination, but implicitly accounts for transfer conductance by using a lower value for *b* (Farquhar *et al.* 1982; Farquhar & Richards 1984):

$${}^{13}\Delta_{\text{simple}} = a + (\overline{b} - a)\frac{C_{\text{i}}}{C_{\text{a}}},\tag{4}$$

where  $\overline{b}$  is the weighted fractionation during internal transfer of CO<sub>2</sub> and fixation by RuP<sub>2</sub> and PEP carboxylases (27‰), applied to C<sub>i</sub> (instead of C<sub>c</sub> as in Eqn 3).

Equation 3 is based on the assumption that recent assimilates form the substrates of photorespiration and day respiration. This is likely to be the case for photorespiration as it recycles freshly assimilated carbon (Fig. 3). However, day respiration may also utilize other substrates, such as older carbohydrates (Schnyder et al. 2003; Nogués et al. 2004). If the  $\delta^{13}$ C composition of these substrates is not the same as that of freshly assimilated carbon, the respiratory  $\delta^{13}$ C signature would differ from instantaneous  ${}^{13}\Delta$  by more than the fractionation factor (e). To take this effect into account, we incorporate an apparent fractionation for day respiration,  $e^*$ , defined as  $e^* = \Re_{\text{assimilates}} / \Re_{\text{substrate}} - 1$ , expressing the difference between the isotopic composition of the respiratory substrate ( $\delta^{13}C_{substrate}$ ) and photosynthetic assimilates  $(\delta^{13}C_{assimilate} \approx \delta^{13}C_a - {}^{13}\Delta)$  at a given time, i.e.  $e^* \approx \delta^{13}C_a - {}^{13}\Delta - \delta^{13}C_{substrate}$ . This leads to the 'revised' model, a modified version of the classical model (Eqn 3, omitting boundary layer effects for clarity):

$$a + (a_{\rm m} - a)\frac{C_{\rm i}}{C_{\rm a}} + (b - a_{\rm m})\frac{C_{\rm c}}{C_{\rm a}} - f\frac{\Gamma_{*}}{C_{\rm a}} - \frac{(e + \delta^{13}C_{\rm a} - \delta^{13}C_{\rm substrate})\frac{R_{\rm d}}{A + R_{\rm d}}\frac{C_{\rm i} - \Gamma_{*}}{C_{\rm a}}}{1 - \frac{R_{\rm d}}{A + R_{\rm d}}\frac{C_{\rm i} - \Gamma_{*}}{C_{\rm a}}}$$
(5)

Alternatively, Eqn 6 retains the apparent factor  $e^*$  so that it can be easily compared to Eqn 3:

$${}^{13}\Delta_{\text{revised}} = a_{\text{b}} \frac{C_{\text{a}} - C_{\text{s}}}{C_{\text{a}}} + a \frac{C_{\text{s}} - C_{\text{i}}}{C_{\text{a}}} + a_{\text{m}} \frac{C_{\text{i}} - C_{\text{c}}}{C_{\text{a}}} + b \frac{C_{\text{c}}}{C_{\text{a}}}$$
(6)  
$$- f \frac{\Gamma_{*}}{C_{\text{a}}} - (e + e^{*}) \frac{R_{\text{d}}}{kC_{\text{a}}}$$

The derivation of Eqns 5 and 6 are given in Appendix II (note that the <sup>13</sup> $\Delta$  term of  $e^*$  is included in the left hand side of Eqn 5). When  $\delta^{13}C_{\text{substrate}} = \delta^{13}C_{\text{assimilate}}$  (i.e.  $e^* = 0$ ), Eqn 5 is again the classical Farquhar *et al.* (1982) description (Eqn 3).

In the following, we use the time series from the July campaign as an example to illustrate the different model predictions. Using Eqns 3, 4 and 5,  ${}^{13}\Delta$  values were calculated and integrated over the chamber closure periods as described in Appendix I. Parameters required for the models are  $\delta^{13}C_a$  from flask observations (i.e.  $\delta^{13}C_o$ ),  $g_b$ ,  $g_c$ 

and  $C_a$  from chamber measurements (see Methods), fractionation factors ( $a_b$ , a,  $a_m$ , b, f, e), and  $g_i$ ,  $\Gamma_*$ ,  $R_d$  and  $\delta^{13}C_{substrate}$ . The internal transfer conductance to CO<sub>2</sub> ( $g_i$ ) was estimated from a linear regression of the difference between predicted  ${}^{13}\Delta$  (Eqn 5 with infinite  $g_i$ ) and  ${}^{13}\Delta_{obs}$  against  $A/C_a$  (von Caemmerer & Evans 1991) using data from both months (not shown). The value obtained ( $g_i = 0.16 \text{ mol m}^{-2} \text{ s}^{-1}$ ) was consistent with the relationship  $g_i = 0.019 \cdot A_{max}$  (Warren *et al.* 2003) and  $A_{max}$  of  $8.0 \pm 1.5 \,\mu\text{mol m}^{-2} \text{ s}^{-1}$  (obtained by fitting a non-rectangular hyperbola to the light response of branches).

The CO<sub>2</sub> compensation point in the absence of dark respiration,  $\Gamma_*$  (µmol mol<sup>-1</sup>), was calculated from needle temperature (Farguhar & von Caemmerer 1982; Brooks & Farquhar 1985). Mitochondrial respiration associated with the tricarboxylic acid (TCA) cycle continues in the light, although it does not necessarily proceed at the same rate as that observed in the dark (Villar, Held & Merino 1994; Atkin et al. 2000; Tcherkez et al. 2005). To account for the potential range of day respiration rates, we calculated temperature-dependent rates  $(R_d)$  using  $R_{20}$  derived from nocturnal measurements (see Results section) and observed needle temperature, and based on these, light inhibited rates  $(R_{di})$  as described by Lloyd *et al.* (1995). Day respiration was estimated to release 15% of gross photosynthetic CO<sub>2</sub> uptake during the day (only 4% for light-inhibited rate). During mornings and evenings,  $R_{\rm d}$ and  $R_{d,i}$  contributions were much larger (30 to 60%) while photorespiration was small (5 to 7%). Values of  $\delta^{13}C_R$ measured for nocturnal respiration (Table 2) were used as estimates of  $\delta^{13}C_{\text{substrate}}$  of day respiration in Eqn 5. We assumed that either 60 or 100% of day respiration was fed from this older substrate pool, with the remainder from newly assimilated carbon (Schnyder et al. 2003; Nogués et al. 2004). We used values of 0 and -6‰ for fractionation during day respiration (e) in Eqns 3 and 5 (see Discussion for a more detailed sensitivity analysis).

### RESULTS

# Diurnal patterns of branch microclimate and gas exchange

Daily time courses for environmental variables within branch chambers (at the beginning of chamber closure periods) are shown in Fig. 1. During May, direct beam irradiance dominated in the upper canopy, but in July, diffuse irradiance was persistent as indicated by low Q in the upper canopy (Fig. 1a,g). Marked diurnal patterns in air temperature ( $T_a$ ) and vapour mole fraction deficit ( $D_a$ ) were observed during both field campaigns (Fig. 1b,c,h,i). Temperatures were similar for both upper canopy chambers (1 and 3) during May and July, reaching 15 to 25 °C over the midday period (1000–1400 h), while  $T_a$  was lower in the middle canopy chamber (4) in May, ranging between 10 to 15 °C. Air vapour mole fraction deficits in May were higher than the values observed in the same chambers during July.

Samiling campaion				$\chi^{13}C_{2}$ + estimated	Assimilation wei	Assimilation weighted daily $\delta^{13}C_{assimilates}$ and $^{13}\Delta~(\%_{o})$	similates and $^{13}\Delta$ (‰)	
camping campaign, Canopy Location	Chamber	$\partial^{13} C_{needle} \pm 1SD(n)$ (%)	$\partial^{3}\mathrm{C}_{\mathrm{wood}} \pm 1\mathrm{SD}\left(n\right)\left(\%_{0} ight)$	uncertainty (‰)	Simple model	Classical	Revised	Best fit
May, Upper	1	$-29.0 \pm 0.1$ (4)	$-27.7 \pm 0.2$ (4)	-27.7 ± 4.7	-26.1 (19.5)	-24.4 (17.7)	-24.3 (17.6)	-24.3 (17.6)
	3			$-27.4 \pm 5.8$	-29.2 (22.7)	-29.9(23.5)	-30.4(24.0)	-30.9 (24.5)
May, Middle	4	$-30.1 \pm 0.3 (4)$		$-26.1 \pm 13.0$	-29.3 (22.8)	-28.0(21.5)	-28.3(21.8)	-28.6 (22.1)
May, Lower		$-31.1 \pm 0.1$ (4)	$-30.4 \pm 0.0$ (2)					
July, Upper	1			$-25.2 \pm 7.9$ $-26.2 \pm 1.8$	-25.5 (18.8)	-23.4 (16.7)	-23.6 (16.9)	-23.6 (16.9)
	3			$-29.9 \pm 8.5$	-27.9 (21.3)	-25.9 (19.2)	-26.0(19.3)	-26.0(19.4)

Daily time courses of gas exchange (beginning of closure periods) are also shown in Fig. 1. During May, assimilation rates (A) were highest before midday when Q was high and  $D_a$  was low, and were greater than in July (11 and 8  $\mu$ mol m<sup>-2</sup> s<sup>-1</sup>, respectively, Fig. 1d,j). From 0900 to 1040 h and 1230 to 1500 h on 18 May 2001, technical difficulties in chambers 1 and 3 prevented determination of A, although all other measurements remained unaffected.

Values of branch conductance to water vapour  $(g_s)$ ranged between 100 and 250 mmol m<sup>-2</sup> s<sup>-1</sup> in the middle canopy and 30 to 170 mmol m<sup>-2</sup> s<sup>-1</sup> in the upper canopy during May (Fig. 1e), but values were much lower in July, between 50 and 150 mmol m<sup>-2</sup> s<sup>-1</sup> (Fig. 1k). Maximum  $g_s$ values were observed in the morning, subsequently declining over the day. The ratio of intercellular to ambient CO<sub>2</sub> mole fraction ( $C_{i,g}/C_a$ , conductance-weighted) during May remained between 0.4 to 0.8 for the upper canopy and 0.7 to 0.9 for the middle canopy (Fig. 1f). During July,  $C_{i,g}/C_a$  was stable over most of the day and ranged between 0.5 and 0.8 for both upper branch chambers (Fig. 11). However, as Q approached 0 during dawn and dusk,  $C_{i,g}/C_a$  was typically 0.8 to 1. At dawn, the large  $C_{i,g}/C_a$  occurred when values of A were low but  $g_s$  high. At dusk, high  $C_{i,g}/C_a$  occurred primarily because of low A.

Nocturnal branch gas-exchange data for each chamber (Fig. 1d,j) were fitted with an Arrhenius type 'activation energy' response to needle temperature (e.g. Lloyd & Taylor 1994). The derived parameter,  $R_{20}$  (efflux rate at 20 °C) ranged from 2.5 to 3.3  $\mu$ mol m<sup>-2</sup> s<sup>-1</sup> in May, and 1.3 to 1.4  $\mu$ mol m<sup>-2</sup> s<sup>-1</sup> in July. Nocturnal branch respiration rates during May 2001 were larger than those observed in July 2001, even though temperatures were similar during the night.

# Diurnal <sup>13</sup>C discrimination

We determined 22 values of <sup>13</sup>C discrimination during photosynthesis,  ${}^{13}\Delta_{obs}$ , from flask sample pairs (Eqn 2). During both field campaigns, we found a pronounced systematic diurnal variability in  ${}^{13}\Delta_{obs}$  (Fig. 2). Highest  ${}^{13}\Delta_{obs}$  occurred at dawn and dusk, ranging between 30 to 35%. Around midday,  ${}^{13}\Delta_{obs}$  were lowest at around 15‰. The middle canopy showed systematically higher  ${}^{13}\Delta_{obs}$  than the upper canopy at similar times of the day. The highest daytime value (41‰) was measured in the mid canopy at noon in May. This and the high dusk and dawn values coincided with low net carbon and water vapour fluxes (Fig. 1). We also determined seven values for isotopic signatures of nocturnal foliage respiration,  $\delta^{13}C_R$ , from mass balances using flask samples pairs  $\left[\delta^{13}C_{R} = \left(\delta^{13}C_{e}C_{e} - \delta^{13}C_{o}C_{o}\right)\right]$  $(C_e - C_o)$ ]. Values of  $\delta^{13}C_R$  in the upper canopy were more negative than in the middle canopy (Fig. 2). The latter was associated with a large uncertainty. For the same branch,  $\delta^{13}C_R$  differed by nearly 2‰ between May and July. A relatively enriched  $\delta^{13}C_R$  value of -19‰ was measured shortly before dawn during the July field campaign (Fig. 2b).

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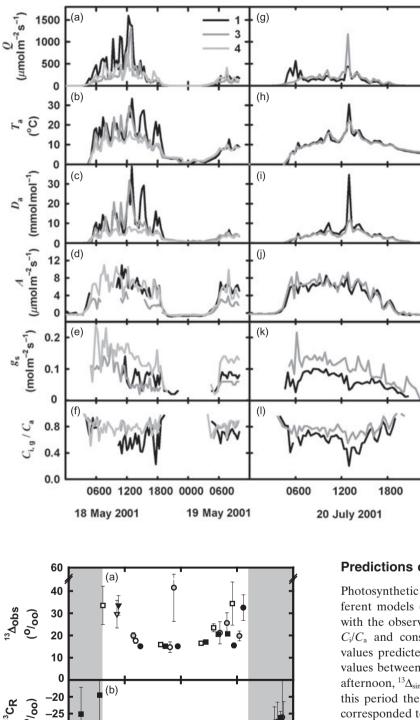
Table 2.

nocturnal branch respiration derived from mass balance, the

signature of

The  $\delta^{13}$ C of organic material collected in May 2001 at Griffin Forest, alongside the observed  $\delta^{13}$ C<sub>R</sub>

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photosynthetic photon flux density (Q), (b,h) air temperature  $(T_a)$ , (c,i) air humidity deficit  $(D_a)$ , (d,j) branch assimilation rate (A), (e, k) conductance to water vapour  $(g_s)$ , and (f,l)conductance-weighted ratios of intercellular to ambient CO2 mole fraction  $(C_{i,g}/C_a)$ , measured every 20 min in branch chambers 1 (black), 3 (dark grey) and 4 (light grey) during May (a-f) and July (g-l) 2001.

Figure 1. Daily variation of (a,g)

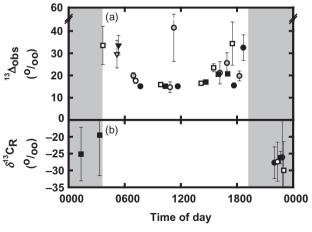
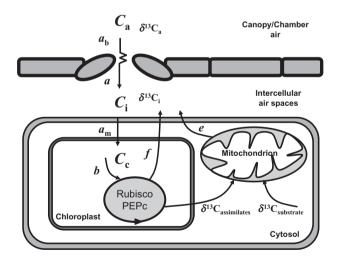


Figure 2. Diurnal variation in (a) observed net <sup>13</sup>C discrimination  $({}^{13}\Delta_{obs})$  and (b) the isotopic composition of dark respiration ( $\delta^{13}C_R$ ) collected in different branch chambers (1: black, 3: white, 4: grey) during 18 ( $\bigcirc$ ) and 19 ( $\bigtriangledown$ ) May 2001 and 20 July 2001 (□). Error bars indicate estimated uncertainties.

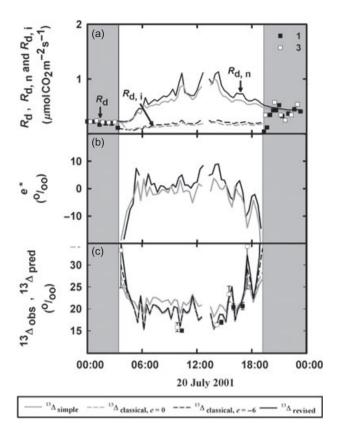
# Predictions of <sup>13</sup><sup>Δ</sup> during foliage gas exchange

Photosynthetic <sup>13</sup>C discrimination calculated using the different models (Eqns 3-5) is presented in Fig. 4c together with the observations. As  ${}^{13}\Delta_{simple}$  is linearly dependent on  $C_i/C_a$  and constrained by a and  $\overline{b}$  (Eqn 4), the largest values predicted at dusk and dawn were 27‰, with most values between 16 and 22‰ for the rest of the day. In the afternoon,  ${}^{13}\Delta_{simple}$  was close to  ${}^{13}\Delta_{obs}$ , indicating that during this period the actual drop between  $C_i$  and  $C_c$  may have corresponded to that implicitly included in  $\overline{b}$  of Eqn 4. But at dusk and dawn, the observations exceeded the  ${}^{13}\Delta_{simple}$ predictions consistently by up to 7‰ in both May (not shown) and July (Fig. 4c). Including additional effects in  $^{13}\Delta_{classical}$  (Eqn 3), in particular those of internal CO<sub>2</sub> transfer when A was large during the day, improved the fit between predictions and observations (Fig. 4c). The contribution of photorespiration to net <sup>13</sup> values was generally less than 1‰, smaller than the effect including the internal conductance term,  $g_i ~(\approx 2\%)$ . Using the largest negative fractionation reported for dark respiration so far (-6‰, Duranceau et al. 1999), the resulting contribution of day respiration increased  ${}^{13}\Delta_{classical}$  considerably but did not fully reconcile offsets between  ${}^{13}\Delta_{classical}$  and  ${}^{13}\Delta_{obs}$  at dusk and dawn. To illustrate the isotopic effect of day respiration on net  ${}^{13}\Delta$  from the revised model (Eqn 5), the apparent fractionation factor  $(e^*)$  is shown in Fig. 4b. During the day, the calculated  $\delta^{13}$ C signature of assimilates (e.g.  $\delta^{13}C_a - {}^{13}\Delta \approx -25\%$ ) was close to that of the respiratory substrate ( $\delta^{13}C_{substrate} \approx -26\%$ , estimated from measurements of dark respiration), yielding small positive  $e^*$ values ( $\approx 1$ %). Accordingly, day respiration had only minor effects on net  ${}^{13}\Delta_{revised}$  values (Eqn 5) during most of the day (Fig. 4c). On the other hand, assimilates were substantially depleted ( $\approx -36\%$ ) compared to the respiratory substrate (-26%) at dusk and dawn, yielding large negative  $e^*$  values ( $\approx -10\%$ ) at these times. The negative  $e^*$ together with high contributions of day respiration resulted in high  ${}^{13}\Delta_{revised}$  at dusk and dawn comparable to the observations (Fig. 4c).

Values of  ${}^{13}\Delta$  obtained from chamber integrations for the three modelling approaches ( ${}^{13}\Delta_{pred}$ ) were compared with flask observations ( ${}^{13}\Delta_{obs}$ ). The classical and revised models both improved the correlation between predicted and observed  ${}^{13}\Delta$  values (Table 3). During the day, the classical and revised models predicted almost identical  ${}^{13}\Delta$  values due to the generally small contribution of day respiration. But only the revised formulation showed reasonable agreement with the high  ${}^{13}\Delta_{obs}$  at dusk and dawn (Fig. 5). Assuming that day respiration was fully supplied by 'old'



**Figure 3.** Conceptual diagram showing the flow of carbon during photosynthetic uptake and the fractionation factors associated with each part of the pathway:  $a_b$  for diffusion through the boundary layer, *a* for diffusion through stomata,  $a_m$  for internal CO<sub>2</sub> transfer, *b* for ribulose 1,5-bisphosphate carboxylase/oxygenase/phospho-enol-pyruvate carboxylase (Rubisco/PEPc) reactions, *f* for photorespiration and *e* for day (mitochondrial) respiration. Also shown are the CO<sub>2</sub> mole fractions ( $C_a$ ,  $C_i$ ,  $C_c$ ) and the isotopic signatures of ambient CO<sub>2</sub>,  $\delta^{13}C_a$ , and intercellular CO<sub>2</sub>,  $\delta^{13}C_i$ , affected by the various fractionations during all CO<sub>2</sub> exchange processes, and those of newly assimilated carbohydrates,  $\delta^{13}C_{assimilates}$ , and other respiratory substrates,  $\delta^{13}C_{substrate}$ .

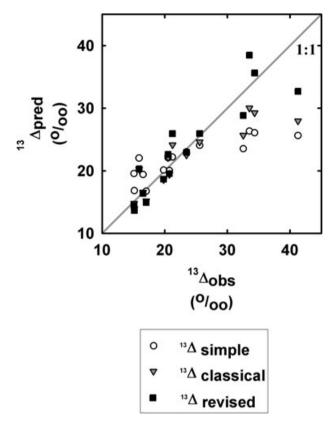


**Figure 4.** Daily variation in (a) observed dark respiration rates for each chamber (square symbols) ( $R_d$ ), estimated rate of day respiration dependent on temperature ( $R_{d:n}$  solid lines) and light-inhibited ( $R_{d:i}$ , dotted lines) for chambers 1 and 3 (black and grey lines, respectively), (b) the apparent fractionation factor,  $e^*$ for chambers 1 and 3 (black and grey lines) and (c) predicted <sup>13</sup> $\Delta$ for chamber 3 from the simple model ( $^{13}\Delta_{simple}$ , Eqn 4, solid grey line), classical model ( $^{13}\Delta_{classical}$ , Eqn 3, assuming e = 0 or e = -6%, dotted grey and black lines) and revised model ( $^{13}\Delta_{revised}$ , Eqn 5, solid black line) alongside net observed discrimination,  $^{13}\Delta_{obs}$ , for chambers 1 and 3 (black and white symbols) during 20 July 2001.

substrates with a  $\delta^{13}$ C signature equal to that measured at night, together with a fractionation factor (*e*) of -6‰, provided the highest correlation for our data set (Table 3).

# Daily foliage isofluxes and the $\delta^{13}$ C of organic matter

The aforementioned models of <sup>13</sup>C discrimination were applied to diurnal gas-exchange data (20 min time step) for the different branch chambers to yield diurnal variations of <sup>13</sup> $\Delta$  expected in the absence of chambers. The  $\delta^{13}C_a$  values, used in Eqn 5 and to compute  $\delta^{13}C_{assimilates}$ , were derived from a linear regression fitted to open branch chamber flask observations: May:  $\delta^{13}C_a = 8194/C_a - 30.1\%$  ( $r^2 = 0.91$ , n = 11, P < 0.0001) and July:  $\delta^{13}C_a = 6846/C_a - 26.5\%$  ( $r^2 = 0.98$ , n = 9, P < 0.0001), with daytime  $\delta^{13}C_a$  variations of less than 1% for both field campaigns. Assimilation weighted diurnal average <sup>13</sup> $\Delta$  and the corresponding  $\delta^{13}C_{assimilates}$  values were then calculated for each



**Figure 5.** Observed  $({}^{13}\Delta_{obs})$  and predicted  $({}^{13}\Delta_{pred})$  discrimination against  ${}^{13}$ C using the simple (Eqn 4), classical (Eqn 3) and revised models (Eqn 5) for all branch chambers during May and July 2001.

branch chamber and month (Table 2). The simple model predicted a 0.7–2.1‰ higher average <sup>13</sup>∆ than the classical and revised models, and a correspondingly depleted  $\delta^{13}C_{assimilates}$  of daily assimilated plant material. For all models,  $\delta^{13}C_{assimilates}$  were enriched relative to  $\delta^{13}C_{needle}$  for chambers 1 and 4, but similar for chamber 3. There were no consistent offsets between  $\delta^{13}C_{assimilates}$  and  $\delta^{13}C_R$  signatures of nocturnal respiration. For chamber 1,  $\delta^{13}C_{assimilates}$ were enriched relative to  $\delta^{13}C_R$ , but depleted for chambers 3 and 4.

The  $\delta^{13}$ C values of foliage ( $\delta^{13}$ C<sub>needle</sub>) and woody ( $\delta^{13}$ C<sub>wood</sub>) material in the upper canopy were typically

enriched by 2 to 3‰ relative to lower canopy values
(Table 2). This translates to a vertical gradient of 0.5‰
decrease in $\delta^{13}C_{needle}$ for every meter down the canopy
profile ( $r^2 = 0.9, n = 17, P < 0.0001$ ). We found that $\delta^{13}C_R$ was
similar to $\delta^{13}C_{wood}$ but enriched relative to $\delta^{13}C_{needle}$ (2 to
4‰). This is consistent with many other observations (see
Badeck et al. 2005 for review), however, we emphasize that
the organic material was not collected from the branches
measured for gas exchange, and that large uncertainties
were associated with measurements at low flux rates
(Table 2).

# DISCUSSION

Our study aimed to characterize the diurnal patterns of  $^{13}\Delta_{obs}$  during photosynthesis of *P. sitchensis* branches in the field (Fig. 2), and to explore how important internal CO<sub>2</sub> transfer, photorespiration and day respiration (see Fig. 3) were in shaping these patterns (Fig. 4). We measured  ${}^{13}\Delta_{obs}$ over several diurnal periods on whole branches of P. sitchensis. We found a range of 15 to 41‰, comparable to that reported for a leaf scale study on Piper aduncum in a tropical forest of Trinidad (Harwood et al. 1998). Despite the differences in ecosystem and plant type, a similar diurnal pattern was observed in both studies: high  ${}^{13}\Delta_{obs}$  values in the morning and low values around midday. Harwood et al. (1998) also documented large  ${}^{13}\Delta_{obs}$  values at dawn. The observations reported in our and previous studies (Gillon et al. 1997; Harwood et al. 1998) were not captured by the commonly used simple Farquhar et al. (1982) model because it expresses  ${}^{13}\Delta$  as a linear function of  $C_i/C_a$  alone (Eqn 4) and does not account explicitly for the various processes concurrently contributing to the net photosynthetic flux. In contrast,  ${}^{13}\Delta_{obs}$  values were described well during most of the day by the classical Farquhar et al. (1982) model (Eqn 3) that incorporates photorespiration, day respiration and the diffusion of CO<sub>2</sub> to the site of carboxylation through a series of resistances. But the surprisingly large  ${}^{13}\Delta_{obs}$  values at dawn and dusk systematically observed in our study (and by Harwood et al. 1998) could not be reconciled using Eqn 3.

It seems unlikely that the difference between predicted  $^{13}\Delta$  and  $^{13}\Delta_{obs}$  was due to erroneous assumptions for the parameters *f* and *g*<sub>i</sub>. Both tend to decrease  $^{13}\Delta$ , and neither has large effects at dawn and dusk. Secondly, any alteration

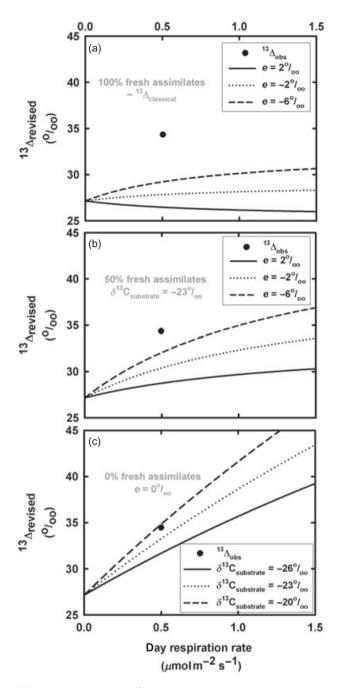
	Parameters and values assigned									
Model	b	f	е	$g_{\rm w}$	$R_{\rm d}$ %	<i>e</i> *	$R_{d,n}$ or $R_{d,i}$	Slope	Int	RMSE
Simple $(\Delta_{simple})$	27	_	_	_	0	_	_	0.28	16.3	6.1
Classical $(\Delta_{classical})$	29	8	0	0.16	0	_	$R_{\rm d,n}$	0.54	8.2	5.7
-Light inhibited	29	8	0	0.16	0	_	$R_{ m d,i}$	0.54	8.2	5.7
Classical $(\Delta_{classical})$	29	8	-6	0.16	0	_	$R_{\rm d,n}$	0.61	7.9	5.0
-Light inhibited	29	8	-6	0.16	0	-	$R_{\rm d,i}$	0.58	7.8	5.4
Revised ( $\Delta_{revised}$ )	29	8	-6	0.16	60	Var	$R_{\rm d,n}$	0.76	5.0	4.5
-Light inhibited	29	8	-6	0.16	60	Var	$R_{\rm d,i}$	0.62	7.1	5.2
Best Fit	29	8	-6	0.16	100	Var	$R_{\rm d,n}$	0.95	0.8	4.2

**Table 3.** Regression output parametersbetween predicted and observeddiscrimination using the simple (Eqn 4),classical (Eqn 3) and revised models(Eqn 5) for all branch chambers during Mayand July 2001

in either of these would also affect all other  ${}^{13}\Delta$  predictions, and even increasing  ${}^{13}\Delta$  to the limits of b by adopting an infinite  $g_i$  cannot explain all  ${}^{13}\Delta_{obs}$  values. To attribute the large dusk and dawn  ${}^{13}\Delta_{obs}$  primarily to fractionation by Rubisco, we would require a b value of at least 34‰. The value determined for Rubisco fractionation in vitro is  $\approx 30\%$  (Roeske & O'Leary 1984; Guy et al. 1987), with the additional contributions of PEP carboxylase typically decreasing the overall fractionation during carboxylation by 0.7 to 1.4‰. Apparent values of b were found between 27‰ for rice and radish and 32‰ for wheat (von Caemmerer & Evans 1991), hence, it is unknown how realistic a b of 34‰ is. On the other hand, changing b would increase  ${}^{13}\Delta$ predictions over the whole diurnal cycle and thus require a concurrent decrease in our estimate of  $g_i$  by one third (to 0.1 mol m<sup>-2</sup> s<sup>-1</sup>) to recover the midday  ${}^{13}\Delta_{obs}$  values. Transfer conductances of 0.1 to 0.2 mol m<sup>-2</sup> s<sup>-1</sup> have been observed for woody species (von Caemmerer & Evans 1991; Lloyd et al. 1992; Loreto et al. 1992; Epron et al. 1995; Warren et al. 2003). Temperature-related fluctuations in  $g_i$  (Bernacchi et al. 2002) were probably negligible because of the conservative temperature range during flask sampling. However, recalculations of the classical model incorporating a b of 34‰ and  $g_i$  of 0.1 mol m<sup>-2</sup> s<sup>-1</sup> still underestimated the dusk and dawn values by 3 to 5‰.

The assumption incorporated in the classical model (Eqn 3) is that day respiration utilizes recently synthesized carbohydrates, with a  $\delta^{13}$ C signature indistinguishable from that of the concurrent net assimilation flux. Hence, day respiration affects  $^{13}\Delta_{classical}$  (Eqn 3) only if there is fractionation during dark respiration range from 6 to -6% (Duranceau *et al.* 1999; Ghashghaie *et al.* 2003), but these might not be valid for day respiration (Tcherkez *et al.* 2004). However, we would need unrealistic values of *e* (up to -20%) compared to the effects of much lower *e* values shown in Fig. 6a to explain  $^{13}\Delta_{obs}$  purely based on fractionation during day respiration. Thus, recently synthesized carbohydrates may not constitute the only substrate for day respiration of *P. sitchensis* branches.

For dark respiration, older carbon pools may provide as much as 60% of the substrate (Schnyder et al. 2003; Nogués et al. 2004). Mitochondrial respiration is likely to utilize different substrates over time (Gleixner & Schmidt 1996; Schnyder et al. 2003; Tcherkez et al. 2003; Nogués et al. 2004; Prater, Mortazavi & Chanton 2006). Substrate shifts could occur between day and night in response to light-activated changes in metabolic pathways (Atkin et al. 2000; Tcherkez et al. 2005). In addition, these pathways may be affected by the non-statistical distribution of <sup>13</sup>C in carbohydrates (Gleixner & Schmidt 1996) or differ in isotope fractionation during the reactions themselves (Ghashghaie et al. 2003; Tcherkez et al. 2003). If day respiration utilizes carbon pools other than recently synthesized carbohydrates, its isotopic signature could differ from that of the concurrent assimilation flux (even for e = 0). This would create 'isotopic singularities', with extremely enriched or depleted  ${}^{13}\Delta$  and  $\delta^{13}C_R$  values as the net flux approaches zero. Such extreme



**Figure 6.** Sensitivity of <sup>13</sup> $\Delta_{revised}$  (Eqn 5) to parameter assumptions relating to day respiration assuming (a) different values for *e* and using only fresh assimilates  $\approx {}^{13}\Delta_{classical}$  (Eqn 3), (b) different values for *e* and using 50% fresh assimilates and 50% old substrate, and (c) different values for  $\delta^{13}C_{substrate}$ , using only old substrate and an *e* of 0‰. The example chosen here was the high dusk value in July 2001, with  ${}^{13}\Delta_{obs} = 34.4\%$ ,  $Q = 36 \,\mu\text{mol m}^{-2} \,\text{s}^{-1}$ ,  $g_s = 0.1 \,\text{mmol m}^{-2} \,\text{s}^{-1}$ ,  $A = 0.8 \,\mu\text{mol m}^{-2} \,\text{s}^{-1}$ ,  $R_d = 0.5 \,\mu\text{mol m}^{-2} \,\text{s}^{-1}$ ,  $C_a = 360 \,\mu\text{mol mol}^{-1}$ ,  $C_c = 341 \,\mu\text{mol mol}^{-1}$ ,  $\delta^{13}C_a = -7.5\%$ ,  $\delta^{13}C_{substrate} = -23\%$ .

values are the result of interpreting two opposing fluxes of similar magnitude with different isotopic signatures as a single net flux with a combined 'apparent' isotopic signature. Similar analyses of apparent isotopic signatures of net

Age class	$L_{\rm A}$ (% of total)	A (% of current)	<i>g</i> s (% of current)	Predicted ${}^{13}\Delta$ (‰)	A weighted ${}^{13}\Delta$ (‰)	Difference $g_s - A$ weighted ${}^{13}\Delta$ (‰)
Current	66	100	100	16.7	15.7	0.3
Current + 1	30	56	39	12.2		
Current + 2	4	54	40	13.1		
Branch total	$L_{\rm A}$ (cm <sup>2</sup> )	$\begin{array}{c} A \\ (\mu \mathrm{mol} \ \mathrm{m}^{-2} \ \mathrm{s}^{-1}) \end{array}$	$g_{\rm s} \pmod{{\rm m}^{-2}{\rm s}^{-1}}$	$g_{\rm s}$ weighted <sup>13</sup> $\Delta$ (‰)	<sup>13</sup> Δ <sub>obs</sub> (‰)	Difference $g_s$ weighted $-{}^{13}\Delta_{obs}$ (‰)
	3842	4.31	0.025	16.0	17.0	-1.0

**Table 4.** Effects of averaging over different needle age classes on the value predicted for  ${}^{13}\Delta$ 

Relative needle area ( $L_A$ ), assimilation rate (A) and stomatal conductance ( $g_c$ ) for three different needle age classes, and their total values for the whole branch, with <sup>13</sup> $\Delta$  values predicted for each age class separately and for the whole branch. The conductance-weighted age class average <sup>13</sup> $\Delta$  corresponds to that calculated from branch total values, i.e. from gas-exchange measurements. The assimilation-weighted age class average <sup>13</sup> $\Delta$  corresponds to that expected from isotope measurements.

fluxes have been presented at large scales for  ${}^{13}C$  (Miller & Tans 2003; Helliker *et al.* 2005), and at the leaf scale for  ${}^{13}C$  (Seibt 2003; Wingate 2003) and  ${}^{18}O$  (Cernusak *et al.* 2004).

Here, we incorporated the opposing flux of day respiration into the equation for net <sup>13</sup>C discrimination (Eqn 5) to account for the mixing and recycling of respired CO2 within the intercellular spaces in the same way as for photorespiration (see Fig. 3 & Appendix II). For example, using the fluxes associated with the large dusk  $^{13}\!\Delta_{obs}$  in July 2001, (34.4‰, measured net A of  $0.8 \,\mu\text{mol}\,\text{m}^{-2}\,\text{s}^{-1}$ , estimated  $R_{\rm d} \approx 0.5 \ \mu {\rm mol} \ {\rm m}^{-2} \ {\rm s}^{-1}$ , correspondingly higher gross photosynthetic rate), calculations without recycling (eqn 7 of Miller & Tans 2003) predicted  $\approx 1\%$  larger net <sup>13</sup> $\Delta$  than Eqn 5 for identical  $\delta^{13}$ C signatures of the two gross fluxes. Incorporating an isotopically different substrate for day respiration, Eqn 5 thus can explain the large net  ${}^{13}\Delta_{obs}$  values at dawn and dusk that could not be resolved by either of the other models (Eqns 3 & 4). Alternatively, the elevated  ${}^{13}\Delta_{obs}$ can be thought of as reflecting a residual isoflux composed of the rate of day respiration, the  $\delta^{13}$ C signature of the respiratory substrate and the fractionation. Very large absolute values of e are required if the substrate is assumed identical to newly assimilated carbohydrates. Smaller e values are needed the more the respiratory substrate differs from recent assimilates, and the larger the relative contribution of this substrate is (see Fig. 6). The respiratory isoflux thus contains too many unknowns to be decomposed without independent knowledge of its components. Ultimately, this can only be addressed through controlled laboratory and field measurements.

For the low  ${}^{13}\Delta$  values usually coinciding with high *A*, *Q* and *T*, there was little difference between the classical and revised formulations. The slightly higher values of the classical model fitted the observed data better than our revised description. As the day respiration term is the only difference between the two models, this might indicate that the revised version overestimates the day respiratory contribution at these times. This could be resolved easily by assuming stronger inhibition of day respiration at high light

and/or temperature levels (Atkin *et al.* 2000). Such a variable day respiration rate, contributing more at dawn and dusk and less during the rest of the day, would explain the observed diurnal pattern of <sup>13</sup> $\Delta$  including the high dusk and dawn values. This scenario does not rule out, but also does not depend on, additional shifts in respiratory substrates, such as two similar but opposite substrate switches, one at dawn, the other at dusk, to explain the high <sup>13</sup> $\Delta$  values observed at both times.

Lastly, we note that in our gas-exchange calculations, the foliage in the chambers was treated as a big leaf. However, enclosing whole branches meant that chamber measurements included the gas exchange of different aged needle cohorts, as well as contributions from woody material. Flask measurements provide assimilation-weighted  $^{13}\Delta$  values, whereas gas-exchange measurements give conductanceweighted values for different age classes. Potential age classrelated weighting effects on  ${}^{13}\Delta$  (see Appendix III) are illustrated in Table 4. In addition, we calculated rates of woody respiration based on temperature measurements (Bosc, de Grandcourt & Loustau 2003). Woody respiration was estimated to contribute about 10% to day respiration rates at low light, i.e. less than 5% of photosynthetic foliage fluxes (even less during the rest of the day), due to the small ratio of stem/twig to foliage area. Therefore, the contributions of woody tissue fluxes to net CO<sub>2</sub> exchange and  $\delta^{13}$ C fractionation observed in this study were considered negligible and ignored in the calculations detailed in Appendix I.

At larger spatial and temporal scales, the isotopic effects of day respiration can probably be neglected as the net flux of CO<sub>2</sub> during dusk and dawn periods represents such a small contribution to the daily flux weighted <sup>13</sup>Δ signal. However, considering the increasing temporal resolution of  $\delta^{13}$ C measurements (Bowling *et al.* 2001; Ogée *et al.* 2003, 2004; Knohl & Buchmann 2005), in particular, the high-resolution isotopic time series constructed using tunable diode laser absorption spectrometry (TDLAS) technology (Bowling *et al.* 2003; Griffis *et al.* 2005), we are increasingly likely to directly encounter the isotopic contributions from day respiration during leaf and ecosystem scale studies. We also hypothesize that these respiratory signals could be more pronounced during periods of high meristematic activity, for instance, during shoot elongation [see, e.g. the increased  $R_d$  values in May (Table 2) associated with bud swelling]. During tissue growth in temperate species, storage carbohydrates such as starch are re-mobilized to provide metabolic energy at times of high demand. In the case of evergreen species such as P. sitchensis, these starch reserves accumulate during favourable environmental conditions over the winter (Bradbury & Malcolm 1978), whereas for deciduous species, they are accumulated towards the end of the previous year's growing season. These reserves could then contribute their distinct isotopic signatures to day respiration and, hence, net <sup>13</sup>C discrimination. The revised model (Eqn 5) presented here, together with detailed studies of metabolic cycling and storage in plants, will enable us to explain the resulting field observations of net  ${}^{13}\Delta$ . At the same time, high-resolution field data on  $\delta^{13}$ C signatures of CO<sub>2</sub> fluxes during the light and dark will enable us to address some of the hypotheses and uncertainties highlighted in our study, and to make use of the potentially valuable information on C metabolism contained in photosynthetic isotope signals.

# CONCLUSIONS

Our study highlights isotopic disequilibria during net <sup>13</sup>C discrimination leading to pronounced diurnal variability of  $^{13}\Delta_{obs}$  in the field. The results of our study also emphasize the need for independent observations of transfer conductance to CO<sub>2</sub>, as well as studies on the potential variability of *b* among and within species. With better constraints on the most critical parameters ( $g_i$  and b), observations of  $^{13}\Delta$  and metabolite composition under controlled experimental conditions could allow us to indirectly probe respiratory signals during the light using stable carbon isotopes as tracers.

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# APPENDIX I: ANALYSIS OF CLOSED CHAMBER MEASUREMENTS

Gas exchange in closed chambers will lead to transient changes in micro-environmental conditions, air composition and fluxes during the closure period. Flask samples collected at the start and end of chamber closure periods integrate over such changes. Furthermore, the air flushed through the flasks and back into the chambers was dried, weakening the increase in chamber water vapour content. Here, we briefly describe how we take both into account when integrating chamber gas-exchange data (5 s time step) to obtain <sup>13</sup> $\Delta$  predictions that are directly comparable to flask observations (see also appendix II of Seibt *et al.* 2006). The data processing described in the following (for chamber 1 on 20 July 2001, 1440 to 1445 h) was implemented in an IDL program.

During the 5 min chamber closure, photon flux density was  $127 \pm 3 \,\mu$ mol m<sup>-2</sup> s<sup>-1</sup>, and air and leaf temperatures were  $14 \pm 0.7$  °C. We assumed that to a good approximation, stomatal conductance remained constant within this period (Ludlow & Jarvis 1971). Relative humidity data (*h*) were used to calculate the vapour mole fraction of the enclosed air  $[w_{a(h)}, Fig. 7a]$ . During flask sampling periods, the air vapour mole fraction was corrected for the flow of dry air (3 dm<sup>3</sup> min<sup>-1</sup>) returned from the flask sampling system  $[w_{a(corr)}]$ . Assuming saturated air at leaf temperature yielded a leaf vapour mole fraction (*w*<sub>i</sub>) of 16.4 mmol mol<sup>-1</sup>. The total conductance to water vapour (*g*, 0.051 mol m<sup>-2</sup> s<sup>-1</sup> for this period) was obtained from the increase in air vapour mole fraction  $[w_{a(corr)}: 9 \text{ to } 14 \text{ mmol mol}^{-1}, h: 56 \text{ to } 78\%]$  by fitting

$$w_{a} = w_{i} + (w_{a0} - w_{i})e^{-\frac{L_{a}}{V_{a}}g(t-t_{0})}$$
(A1.1)

to leaf and air vapour mole fraction data at time *t* during the closure periods, where  $w_{a0}$  is the initial air vapour mole fraction at the starting time  $t_0$ .  $L_a$  and  $V_a$  are the leaf area and air molar volume enclosed in the chamber and *g* represents the total conductance to water vapour combining boundary layer and stomatal components. The transpiration rate decreased from 0.4 to 0.1 mmol m<sup>-2</sup> s<sup>-1</sup> during the closure period. The CO<sub>2</sub> mole fraction of chamber air ( $C_a$ , Fig. 7b) decreased from 350 to 275  $\mu$ mol mol<sup>-1</sup>. The  $C_a$  data were fitted with a quadratic equation, its derivative yielding a linear approximation of *A* (most  $R^2 > 0.99$ ). Because  $C_a$  and *A* varied concurrently, the change in $C_i$  (Eqn 1, Fig. 7b) was weaker than that of  $C_a$ , leading to a small increase in  $C_i/C_a$  over the closure period (Fig. 7c).

Using the  $\delta^{I3}C_a$  values observed in the open chambers as starting point values, we then calculated  ${}^{13}\Delta$  (Eqns 3–5) and a new  $\delta^{I3}C_{a(pred)}$  (Eqn 2) for each 5 s time step. These  ${}^{13}\Delta$  and  $\delta^{I3}C_{a(pred)}$  values increased from 17 to 18‰ (Fig. 7c) and –7

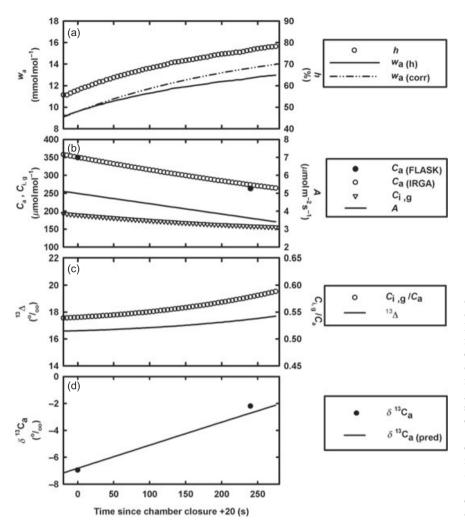


Figure 7. Changes in branch chamber 1 during the 5 min closure period at 1440 h on 20 July 2001: (a) observed relative humidity (h) and calculated air vapour mole fraction, uncorrected  $[w_{a(h)}]$  and corrected  $[w_{a(corr)}]$  for the recirculation of dried air by the sampling system; (b) CO<sub>2</sub> mole fraction observed by the infrared gas analyser (IRGA)  $[C_{a(IRGA)}]$  and flask sampling method  $[C_{a(FLASK)}]$ , and calculated intercellular CO2 mole fraction  $(C_i)$  and net assimilation rate (A); (c) calculated ratio of ambient to intercellular CO<sub>2</sub> mole fraction  $(C_{i,e}/C_a)$ and photosynthetic <sup>13</sup>C discrimination  $(^{13}\Delta)$ , and (d) flask observed ( $\delta^{13}C_a$ ) and calculated  $[\delta^{13}C_{a (pred)}]$  isotopic signature of chamber air.

to -2% (Fig. 7d) during the closure periods, respectively. Weighted averages of  $\delta^{13}C_a$  and  $C_a$  were calculated from the time steps to reflect the air mixture collected in the 'closed' flask sample. Integrated  $^{13}\Delta$  values were then obtained from  $\delta^{13}C_a$  and  $C_a$  at the start and their weighted averages at the end of chamber closure periods using Eqn 2. This approach is equivalent to the calculations using flask data, making both methods directly comparable. For the example used here, the integrated  $^{13}\Delta$  value of 16.9‰ was closer to the flask observed values of 17.0‰ than the starting point estimate of 16.6‰.

## APPENDIX II: THE EFFECT OF DAY RESPIRATION ON NET DISCRIMINATION

Currently, the most complete, 'classical' model for net <sup>13</sup>C discrimination, <sup>13</sup> $\Delta$ , during photosynthesis of C<sub>3</sub> plants (Farquhar *et al.* 1982) accounts for fractionation (*e*) during day respiration (*R*<sub>d</sub>) but assumes that current assimilates form its substrate (Eqn 3). In the following, we use a simplified version of Eqn 3 (see eqn B24 of Farquhar *et al.* 1982) omitting boundary layer and transfer conductances for clarity (the actual calculations always included all terms):

$${}^{13}\Delta_{\text{classical}} = a + (b-a)\frac{C_{\text{i}}}{C_{\text{a}}} - f\frac{\Gamma_{*}}{C_{\text{a}}} - e\frac{R_{\text{d}}}{kC_{\text{a}}}$$
(A2.1)

Here, we derive an equation for net photosynthetic <sup>13</sup>C discrimination that accounts for an isotopically distinct substrate used for day respiration. No specific assumption is made about the substrate and fractionation (*e*), which may come from various pools (sucrose, starch, lipids, etc.) with different metabolic pathways (Gleixner & Schmidt 1996; Schnyder *et al.* 2003; Tcherkez *et al.* 2003, 2004; Nogués *et al.* 2004). In the following, we use the notation of Farquhar *et al.* (1982), with B indicating equations from their appendix II. Using eqns B12, B12', B18, B18', B19 and B19', the <sup>13</sup> $\Delta$  of net assimilation (including photorespiration and daytime dark respiration) is given by

$$\frac{A'}{A} = (1^{-13}\Delta)\mathfrak{R}_{a} = \frac{g'}{g} \cdot \frac{g+kP}{g'+k'P} \cdot \frac{k'(C'_{a}-\Gamma'_{*})-R'_{d}}{k(C_{a}-\Gamma_{*})-R_{d}}, \qquad (A2.2)$$

where  $\Re_a = C'_a/C_a$  denotes the isotope ratio of atmospheric CO<sub>2</sub>, and *P* is atmospheric pressure. The fractionations

associated with diffusion (*a*, eqn B22), carboxylation (*b*, eqn B16) and photorespiration (*f*, eqn B5) are defined as in Farquhar *et al.* (1982). But the isotope ratio of recent assimilates (A'/A) in eqn B8  $[R'_d/R_d = (1-e)(A'/A)]$  is replaced by the isotope ratio of the substrate used for day respiration,  $\Re_{\text{substrate}}$ :

$$\frac{R'_{\rm d}}{R_{\rm d}} = (1 - e)\Re_{\rm substrate} \tag{A2.3}$$

Using the fractionation factors and  $\Gamma'_*/\Gamma_* \approx (1+b^{-13}\Delta - f) \cdot \Re_a$  (eqn B17) in Eqn A2.2 gives

$$(1 - {}^{13}\Delta) \cdot \mathfrak{R}_{a} = (1 - a) \cdot \frac{g + kP}{g(1 - a) + kP(1 - b)} \cdot$$

$$\frac{k(1 - b)(C_{a} - \Gamma_{*} - \Gamma_{*}(b - {}^{13}\Delta - f))}{\mathfrak{R}_{a} - R_{d}(1 - e)\mathfrak{R}_{substrate}}$$

$$\frac{\mathfrak{R}_{a} - R_{d}(1 - e)\mathfrak{R}_{substrate}}{k(C_{a} - \Gamma_{*}) - R_{d}}$$
(A2.4)

Rearranging and neglecting second-order terms lead to the following expression:

$$(1^{-13}\Delta)\cdot\mathfrak{R}_{a} \approx (1-a)\cdot\frac{g+kP}{g+kP-(ag+bkP)} \cdot (A2.5)$$

$$= \frac{[k(C_{a}-\Gamma_{*})-R_{d}]\mathfrak{R}_{a}-[k(bC_{a}-(^{13}\Delta+f)\Gamma_{*})]}{-R_{d}]\mathfrak{R}_{a}-R_{d}(1-e)\mathfrak{R}_{substrate}}$$

$$\approx (1-a)\cdot\left(1+\frac{ag+bkP}{g+kP}\right) \cdot \left\{\mathfrak{R}_{a}-\frac{[k(bC_{a}-(^{13}\Delta+f)\Gamma_{*})-R_{d}]}{k(C_{a}-\Gamma_{*})-R_{d}}\right\}$$

$$\approx \left(1-a+\frac{ag+bkP}{g+kP}\right) \cdot \left\{1-b\frac{kC_{a}}{k(C_{a}-\Gamma_{*})-R_{d}}\right\}$$

$$\approx \left(1-a+\frac{ag+bkP}{g+kP}\right) \cdot \left\{1-b\frac{kC_{a}}{k(C_{a}-\Gamma_{*})-R_{d}}\right\}$$

$$+\left(^{13}\Delta+f\right)\frac{k\Gamma_{*}}{k(C_{a}-\Gamma_{*})-R_{d}}$$

$$+e\frac{R_{d}}{k(C_{a}-\Gamma_{*})-R_{d}}\frac{\mathfrak{R}_{substrate}}{\mathfrak{R}_{a}} - \frac{R_{d}(1-e)}{k(C_{a}-\Gamma_{*})-R_{d}}$$

$$\approx \left\{1+\frac{kP}{g+kP}[b-a]-\frac{kC_{a}}{\mathfrak{R}_{a}}+\frac{R_{d}}{kC_{a}}\left(\frac{\mathfrak{R}_{substrate}}{\mathfrak{R}_{a}}+\frac{R_{d}}{kC_{a}}\left(\frac{\mathfrak{R}_{substrate}}{\mathfrak{R}_{a}}-1\right)\right]\right\}\cdot\mathfrak{R}_{a}$$

Substituting  $\frac{kP}{g+kP} = \frac{kC_a}{k(C_a - \Gamma_*) - R_d} \cdot \frac{C_a - C_i}{C_a}$  (eqns B12, B18 and B19), we get

$${}^{13}\Delta \approx \frac{kC_{\rm a}}{k\left(C_{\rm a} - \Gamma_{*}\right) - R_{\rm d}} \left[ a + (b - a)\frac{C_{\rm i}}{C_{\rm a}} - \left({}^{13}\Delta + f\right) \right]$$

$$\frac{\Gamma_{*}}{C_{\rm a}} - e\frac{R_{\rm d}}{kC_{\rm a}}\frac{\Re_{\rm substrate}}{\Re_{\rm a}} + \frac{R_{\rm d}}{kC_{\rm a}} \left(\frac{\Re_{\rm substrate}}{\Re_{\rm a}} - 1\right) \right]$$
(A2.6)

Because  ${}^{13}\Delta$  appears on both sides of Eqn A2.6, we rearrange (neglecting second-order terms) to get the net  ${}^{13}C$  discrimination for any dark respiratory substrate:

$${}^{13}\Delta_{\text{revised}} \approx \frac{kC_{\text{a}}}{k\left(C_{\text{a}} - \Gamma_{*}\right) - R_{\text{d}}} \left[a + (b - a)\frac{C_{\text{i}}}{C_{\text{a}}} - f\frac{\Gamma_{*}}{C_{\text{a}}} - (A2.7)\right]$$
$$\left(e + \delta^{13}\mathbf{C}_{\text{atm}} - \delta^{13}\mathbf{C}_{\text{substrate}}\right)\frac{R_{\text{d}}}{kC_{\text{a}}}$$

If the substrate effect of dark respiration on net <sup>13</sup>C discrimination is expressed as apparent fractionation,  $e^* \approx \delta^{13}C_{atm} - \delta^{13}C_{substrate} - {}^{13}\Delta$ , i.e. the isotopic difference between dark respiration and photosynthesis, we can substitute  $\Re_{substrate} = (1 - e^*)$  (A'/A) in Eqn A2.3 to give  $R'_d/R_d = [1 - (e + e^*)](A'/A)$ . Using this apparent factor  $e^*$ , Eqn A2.7 can be directly compared to Eqn A2.1:

$${}^{13}\Delta_{\text{revised}} = a + (b-a)\frac{C_{\text{i}}}{C_{\text{a}}} - f\frac{\Gamma_{*}}{C_{\text{a}}} - (e+e^{*})\frac{R_{\text{d}}}{kC_{\text{a}}}$$
(A2.8)

If  $\delta^{13}C_{\text{substrate}} = \delta^{13}C_{\text{atm}} - {}^{13}\Delta$  ( $e^* = 0$ ), the revised equation (A2.7, A2.8 & A2.9) is again the classical model of Farquhar *et al.* (1982) as in Eqn A2.1.

Alternatively, the carboxylation efficiency k can be substituted in Eqn A2.7 using  $k = (A + R_d) / (C_i - \Gamma_*)$  from appendix II, (Farquhar *et al.* 1982) to yield Eqn 5 of the main text, a useful version because it contains only variables that were measured in our study:

$$a + (b - a)\frac{C_{i}}{C_{a}} - f\frac{\Gamma_{*}}{C_{a}} - (e + \delta^{13}C_{atm} - \frac{\delta^{13}C_{substrate}}{A + R_{d}}\frac{R_{d}}{A + R_{d}}\frac{C_{i} - \Gamma^{*}}{C_{a}} - \frac{1 - \frac{R_{d}}{A + R_{d}}\frac{C_{i}\Gamma^{*}}{C_{a}}}{1 - \frac{R_{d}}{A + R_{d}}\frac{C_{i}\Gamma^{*}}{C_{a}}}$$
(A2.9)

Lastly, in the absence of light, k = 0 and Eqn A2.7 reduces to

$${}^{13}\Delta_{\rm R} \approx e - \delta^{13} \mathcal{C}_{\rm substrate} + \delta^{13} \mathcal{C}_{\rm atm}, \qquad (A2.10)$$

demonstrating that in the absence of carboxylation, the  $\delta^{13}$ C signature of the respiratory flux ( $^{13}\Delta_R$ ) with respect to the atmospheric composition is given by the respiratory substrate minus the fractionation factor (keeping the notation of Eqn A2.7, but *e* referring to dark respiration here). We therefore recommend the use of Eqn A2.7 instead of Eqn A2.1 because it is valid at any time of the day, even for dark respiration).

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# APPENDIX III: WEIGHTING OF GAS-EXCHANGE AND ISOTOPE MEASUREMENTS

The calculations of gas exchange in this study have effectively assumed the branch can be represented as a 'big leaf', despite the inclusion of multiple needle cohorts where uniform assimilation, stomatal and internal conductances are unlikely. The implications of such heterogeneity on the weighting of gas-exchange and isotope measurements have already been shown for individual leaves (Farguhar 1989; Lloyd et al. 1992) and plant canopies (Lloyd et al. 1996). Indeed, there are systematic differences in the gasexchange characteristics of differing needle age classes for P. sitchensis (Ludlow & Jarvis 1971; Watts, Neilson & Jarvis 1976; Leverenz et al. 1982; Barton 1997). The presence of different needle age classes on the same branch thus represents a similar situation to that of stomatal heterogeneity for single leaves. Branches in our study contained a maximum of three age classes, with each age class transpiring and assimilating at different rates. Carbon isotope discrimination is carried by the net flux of CO<sub>2</sub>, i.e. different age classes contribute to the average  ${}^{13}\Delta$  according to their respective assimilation rates. Deriving  $C_i/C_a$  from flask <sup>13</sup> $\Delta$ measurements thus provides an assimilation-weighted value of  $C_i$  ( $C_{i,A}$ ). In contrast, different age classes contribute to the average  $C_i/C_a$  derived from IRGA gas-exchange data according to their respective stomatal conductances, i.e. gas-exchange measurements give a conductanceweighted value of  $C_i$  ( $C_{i,g}$  see Eqn 1) (Farquhar 1989).

Potential age class-related weighting effects on  ${}^{13}\Delta$  are illustrated in Table 4. Values of  ${}^{13}\Delta$  were determined for each needle age class separately, based on variations of A and g<sub>s</sub> observed across age classes for *P. sitchensis* (Ludlow & Jarvis 1971). Assimilation and conductance weighted  $^{13}\Delta$ averages for the whole branch were then calculated according to the contributions of the different age classes to the total needle area measured on our branches. If differences between  ${}^{13}\Delta$  predicted from gas exchange and  ${}^{13}\Delta_{obs}$ determined from flask measurements were attributed to differences between  $C_{i,g}$  and  $C_{i,A}$  via stomatal heterogeneity, we would expect similar offsets between conductance-weighted  $^{13}\!\Delta$  and flask  $^{13}\!\Delta_{obs}$  values as between conductance-weighted and assimilation-weighted  $^{13}\Delta$ values. But calculated offsets had opposite signs indicating that the differences between predicted and observed  ${}^{13}\Delta$ are unlikely to result from foliage heterogeneity. Moreover, the shift expected from weighting was small (0.24‰) compared to the actual difference between predicted and observed  ${}^{13}\Delta$  values (1.1‰). Although estimating such shifts is limited by the fact that diverse (and even contrasting) patterns of changes in A and  $g_s$  across needle age classes have been observed in different studies (Ludlow & Jarvis 1971; Wang 1988), none of these patterns would seem sufficient to induce shifts of the observed direction and magnitude. Again, the more likely explanation is the contribution of further isotopic effects during net uptake such as day respiration, photorespiration and internal CO<sub>2</sub> transfer (see text).