

Temporal and spatial variability of the stable isotopic composition of atmospheric molecular hydrogen: observations at six EUROHYDROS stations

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Abstract. Despite the potential of isotope measurements to improve our understanding of the global atmospheric molecular hydrogen (H_2) cycle, few H_2 isotope data have been published so far. Now, within the EUROpean network for atmospheric HYDRogen Observations and Studies project (EUROHYDROS), weekly to monthly air samples from six locations in a global sampling network have been analysed for H_2 mixing ratio ($m(H_2)$) and the stable isotopic composition of the H_2 ($\delta(D,H_2)$, hereafter referred to as δD). The time series thus obtained now cover one to five years for all stations. This is the largest set of ground station observations of δD so far. Annual average δD values are higher at the Southern Hemisphere (SH) than at the Northern Hemisphere (NH) stations; the maximum is observed at Neumayer (Antarctica), and the minimum at the non-arctic NH stations. The maximum seasonal differences in δD range from $\approx 18\%$ at Neumayer to $\approx 45\%$ at Schauinsland (Southern Germany); in general, seasonal variability is largest at the NH stations. The timing of minima and maxima differs per station as well. In Alert (Arctic Canada), the variations in δD and $m(H_2)$ can be approximated as simple harmonic functions with a ≈ 5 -month relative phase shift. This out-of-phase seasonal behaviour of δD and $m(H_2)$ can also be detected, but delayed and with a ≈ 6 -month relative phase shift, at Mace Head and Cape Verde. However, no seasonal δD cycle could be observed at Schauinsland, which likely reflects the larger influence of local sources and sinks at this continental station.

At the two SH stations, no seasonal cycle could be detected in the δD data. If it is assumed that the sink processes are the main drivers of the observed seasonality in $m(H_2)$ and δD on the NH, the relative seasonal variations can be used to estimate the relative sink strength of the two major sinks, deposition to soils and atmospheric oxidation by the hydroxyl (OH) radical. For the NH coastal and marine stations this analysis suggests that the relative contribution of soil uptake to the total annual H_2 removal increases with latitude.

1 Introduction

Molecular hydrogen (H_2) is present in the atmosphere with a typical mixing ratio ($m(H_2)$) of more than 500 ppb (nmole mole^{-1}) (Glueckauf and Kitt, 1957; Schmidt and Seiler, 1970; Schmidt, 1974; Ehhalt et al., 1977). Over the last decades, several studies examined the magnitude of the source and sink terms in the global H_2 budget (Novelli et al., 1999; Hauglustaine and Ehhalt, 2002; Sanderson et al., 2003; Rhee et al., 2006b; Price et al., 2007; Xiao et al., 2007; Ehhalt and Rohrer, 2009; Pieterse et al., 2011, Yashiro et al., 2011, see Table 1 for the budgets that included isotopes). These show that the largest H_2 sources are the atmospheric oxidation of methane and other hydrocarbons, and combustion processes. Production by nitrogen-fixing bacteria, on land or in the oceans, constitutes a smaller source. About three quarters of the H_2 thus produced is taken up by soil; the other quarter is oxidized by the hydroxyl radical (OH). However, large quantitative uncertainties still exist in the global H_2 budget. For example, the estimates for the sink strength



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Table 1. The global budget of atmospheric H_2 , with source and sink strengths and isotopic signatures used by different authors.

Sources	Rhee et al. (2006)		Price et al. (2007)		Pieterse et al. (2011)	
	Source strength (Tg H_2 a ⁻¹)	Source signature (‰)	Source strength (Tg H_2 a ⁻¹)	Source signature (‰)	Source strength (Tg H_2 a ⁻¹)	Source signature (‰)
Fossil fuel burning	15 ± 6	-270	18.3	-196	17.0 ⁺³ ₋₆	-196
Biomass burning	16 ± 3	-90	10.1	-290	15.0 ⁺⁵ ₋₅	-260
Biofuel			4.4	-290		
Ocean	6 ± 5	-700	6	-628	5.0 ⁺¹ ₋₂	-628
Land N ₂ fixation	6 ± 5	-700	0		3.0 ⁺³ ₋₃	-628
Methane oxidation	64 ± 12	+190	24.5	+162	37.3	+116
VOC oxidation	(incl. VOC)		9.8	+162		
Sinks	Sink strength (Tg H_2 a ⁻¹)	Sink fractionation factor ($\alpha = k_{HD}/k_{H_2}$)	Sink strength (Tg H_2 a ⁻¹)	Sink fractionation factor ($\alpha = k_{HD}/k_{H_2}$)	Sink strength (Tg H_2 a ⁻¹)	Sink fractionation factor ($\alpha = k_{HD}/k_{H_2}$)
Uptake by soils	88 ± 11	0.943 ± 0.007	55	0.943	55.8	0.925
Oxidation by OH	19 ± 3	0.58 ± 0.07	18	0.568	22.1	0.542

of soil uptake – the largest term in the budget – vary widely between these studies.

Research into the atmospheric H_2 budget has increased in recent years, because H_2 may become an important energy carrier in the future. In this case, emissions of H_2 to the atmosphere are likely to rise as a result of the inevitable leakage during production, storage and distribution of H_2 (Schultz et al., 2003). The associated rise of atmospheric H_2 levels is expected to affect the oxidative capacity of the atmosphere, with implications for the atmospheric lifetime of many species, including the strong greenhouse gas methane. Higher concentrations of H_2 will also affect stratospheric ozone levels, although estimates of the impact vary significantly among authors (Tromp et al., 2003; Warwick et al., 2004; Feck et al., 2008).

Due to the large relative mass difference between deuterated hydrogen (HD) and non-deuterated hydrogen (HH), particularly large isotope effects occur in the chemical processes that produce or remove H_2 . These result, for example, in very different isotopic signatures for H_2 produced by combustion processes, by oxidation sources or by biological processes (see Table 1). For this reason, determination of the isotopic composition is a promising addition to $m(H_2)$ observations to distinguish between different H_2 source and sink processes and to constrain the terms in the global budget. In this paper, the following definition of δD is used to indicate the isotopic composition of the hydrogen:

$$\delta D = \delta(D, H_2) = \left(\frac{R_{\text{Sample}}}{R_{\text{VSMOW}}} - 1 \right) \cdot 1000 \text{‰} \quad (1)$$

where R_{Sample} is the ratio of the number density of “heavy” deuterium atoms (D) to the number density of “light” hydrogen atoms (H) in the H_2 of the sample, and R_{VSMOW} is

the ratio of the number density of deuterium atoms to the number density of H atoms in Vienna Standard Mean Ocean Water, which is (155.75 ± 0.08) ppm (De Wit et al., 1980; Gonfiantini et al., 1993). Note that the ‰-sign is explicitly included in this formula.

Following up the initial work by (Ehhalt, 1966; Gerst and Quay, 2000, 2001), new analytical techniques that have become available recently have significantly simplified H_2 isotope analysis so that many more data have become available in the past few years (Rahn et al., 2002b; Rhee et al., 2004). This has led to new constraints on the isotopic signatures of the most important sources and sinks. The isotopic composition of H_2 from CH_4 oxidation was first examined by measurements in the stratosphere (Rahn et al., 2003; Röckmann et al., 2003; Rhee et al., 2006a), but also individual steps in the oxidation sequence have been investigated, especially photolysis of formaldehyde (HCHO) (Feilberg et al., 2007; Mar et al., 2007; Nilsson et al., 2007; Rhee et al., 2008; Röckmann et al., 2010b). Furthermore, more information on the main surface sources, biomass burning and fossil fuel combustion (Röckmann et al., 2010a; Vollmer et al., 2010) and the largest sink, uptake in soil (Rahn et al., 2002a; Gerst and Quay, 2001; Rice et al., 2011), has become available.

At the same time, H_2 and δD have been incorporated into chemical transport models (Price et al., 2007; Pieterse et al., 2009, 2011). Until now, δD observations to validate these model results have been scarce. Data have been published from Pacific Ocean transects, the Cheeka Peak observatory (Washington, US), the Point Barrow observatory (Alaska, US) and the CARIBIC aircraft sampling program (Gerst and Quay, 2000; Rhee et al., 2006b; Rice et al., 2010). These datasets have limited spatial coverage and temporal resolution, most notably in the higher latitudes of the NH. Here

we present observations from six ground stations covering high latitudes of both the NH and SH and all seasons, that contribute to the closing of this observational gap. The interpretation of the isotope record we present here is semi-quantitative. A more rigorous quantification of the terms in the global budget will require the use of the aforementioned global chemistry models. Our data have already been used with the global chemistry transport model TM5 (Pieterse et al., 2011).

2 Experimental

Air samples were collected by the Institut für Umweltphysik of the University of Heidelberg (UHEI-IUP), the Max Planck Institute for Biogeochemistry in Jena (MPI-BGC) and the Laboratoire des Sciences du Climat et de l'Environnement in Gif sur Yvette (LSCE) at the six stations listed in Table 2 and depicted in Fig. 1. These six stations are part of the EUROpean network for atmospheric HYDRogen Observations and Studies (EUROHYDROS) (Engel, 2009). 11 or 21 borosilicate 3.3 glass flasks with Kel-F (PCTFE) O-ring-sealed stopcocks (Normag) were used for most of the samples. This type of flask is known to be stable for a number of trace gases (Rothe et al., 2004). Storage tests performed at the MPI-BGC indicated that – except for a few individual outliers – $m(H_2)$ is also stable in these flasks. Only for Amsterdam Island, an older type of flask with Teflon PFA O-rings was sometimes used until May 2007. Most flasks were covered with a black shrink hose, and all flasks were stored in closed metal boxes when not in use, minimizing photochemical H_2 production after sampling.

The flasks used for sampling at Alert, Schauinsland and Neumayer (operated by UHEI-IUP) were conditioned by flushing with dry air (dewpoint of -76°C) over two hours on two consecutive days. During sample collection the flask air was dried cryogenically, to a dewpoint of -70°C for Alert and Neumayer and a dewpoint of -40°C for Schauinsland. The flasks were flushed for at least 15 min and then pressurized to ≈ 2 bar (absolute pressure) with KNF Neuberger pumps (type N86KNDC with EPDM membrane).

The flasks used for sampling at Mace Head and Amsterdam Island (operated by LSCE) were evacuated and filled with dry air between each sampling round. At the measurement sites, they were flushed for 10 min and then filled to ≈ 2 bar (absolute pressure). For these flasks, more information can be found in (Yver, 2010).

The flasks used for sampling at Cape Verde (operated by MPI-BGC) were conditioned by evacuating at 70°C for three days, followed by flushing with 301 of dried ambient air before the first use, and reconditioned between each sampling round by flushing with dried ambient air at 1.6 bar. They were pressurized to 1.6 bar before shipping to Cape Verde, and flushed with the sample air before pressurizing with a membrane pump. The sample air was dried with $Mg(ClO_4)_2$.

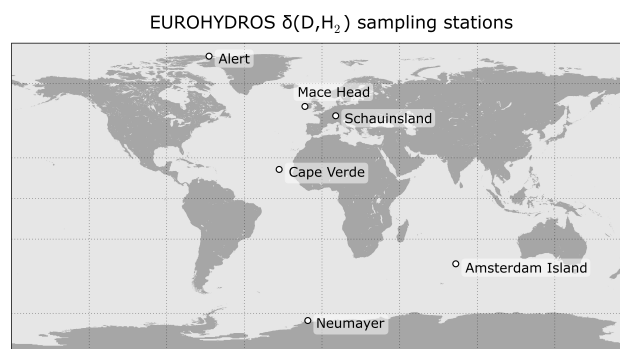


Fig. 1. The six EUROHYDROS stations that were used for δD observations.

The samples were first shipped to the institute that operated the station, where mixing ratios of H_2 and other trace gases were measured. H_2 mixing ratios were determined with reduction gas analysers, either the RGA-3 (Trace Analytics Inc.) or the Peak Performer 1 RCP (Peak Laboratories) (Hammer and Levin, 2009; Yver et al., 2009; Jordan and Steinberg, 2011). Reduction gas analysers separate reduced gases such as H_2 from the air matrix by gas chromatography, and then quantify them via a redox reaction with mercuric oxide and detection of the resulting mercury vapour by UV absorption. We give mixing ratios in units of “parts per billion” (ppb, equivalent to the SI-unit “ nmole mole^{-1} ”). All laboratories used laboratory working standards that are now on the MPI2009 scale, developed at the MPI-BGC (Jordan and Steinberg, 2011). The samples were then shipped to the isotope laboratory at the Institute for Marine and Atmospheric Research (IMAU) of Utrecht University. Typically, several months passed between sample collection and isotope analysis. A total of 480 flasks were analysed; the number of samples per station is listed in Table 2.

In Utrecht, a gas chromatography isotope-ratio mass spectrometry (GC-IRMS) system as described in (Rhee et al., 2004) is used to separate H_2 from the air matrix and to determine its isotopic composition in a four-step procedure, as follows:

- A ≈ 750 ml glass sample volume is filled with sample air until pressure in the volume has reached ≈ 700 mbar. This sample air is then exposed to a Cold Head cooled by a liquid helium compressor to ≈ 40 K, so that all except the most volatile gases (H_2 , He and Ne) condense.
- The remaining volatile gases are flushed with He carrier gas to a pre-concentration trap, consisting of a 1/8 inch (3.2 mm) stainless steel tube filled with 5 Å molecular sieve, immersed in liquid nitrogen, which is cooled down to the triple point of nitrogen (63 K) by pumping off the gas phase.

Table 2. The locations and operating institutes of the six EUROHYDROS flask sampling stations discussed here, and the length, average values and maximum variations of their $m(H_2)$ and δD sample records.

Station	Coordinates	Operating Institute	Samples	Record length (months)	Average $m(H_2)$ (ppb)	Max. $m(H_2)$ variation	Average δD (% vs. VSMOW)	Max. δD variation
Alert	82°28' N 62°30' W	UHEI-IUP	194	47	496.9 ± 1.2	131	124.1 ± 0.5	41
Mace Head	53°20' N 9°54' W	LSCE	71	41	518.3 ± 1.4	72	121.2 ± 0.7	34
Schauinsland	47°55' N 7°55' E	UHEI-IUP	127	34	523.7 ± 0.8	67	117.1 ± 0.9	45
Cape Verde	16°52' N 24°52' W	MPI-BGC	42	12	538.3 ± 1.3	57	119.9 ± 1.1	36
Amsterdam Island	37°48' S 77°32' E	LSCE	63	38	549.5 ± 0.6	28	143.8 ± 1.0	32
Neumayer	70°39' S 8°15' W	UHEI-IUP	110	59	551.1 ± 0.4	30	149.8 ± 0.4	18

- After pre-concentration the trap is lifted from the liquid nitrogen and the trapped gases are flushed to a cryo-focus trap. This trap consists of a molsieve 5 Å capillary column, jacketed in a stainless steel tube and immersed in liquid nitrogen at ambient pressure (77 K). After focussing, the trap is lifted from the liquid nitrogen and the gases are injected into a 5 Å molsieve Gas Chromatography column kept in an oven at 50 °C, where the H_2 is separated from any potential contaminants.
- The purified H_2 is then injected through an open split system into the IRMS (ThermoFinnigan Delta plus XL) for determination of the D/H ratio.

In the IRMS chromatogram, The sample peak was typically bracketed by seven H_2 laboratory working gas peaks of pure H_2 before and two H_2 working gas peaks after. By the same procedure as for the samples, comparable quantities of air from laboratory reference air bottles were typically measured twice a day. The H_2 mixing ratio in the laboratory reference bottle that was in use from March 2007 until February 2010 was determined by UHEI-IUP to be (546.2 ± 2.5) ppb, later confirmed by MPI-BGC to be (545.0 ± 0.5) ppb. Its δD ((+73.0 ± 1.8) ‰, where the error bar indicates one standard error from 5 different determinations) was calculated using mixtures of synthetic air with H_2 of known isotopic composition ((−9.5 ± 0.5) ‰ and (+205 ± 2) ‰, certified by Messer Griesheim, Germany) that were measured on the GC-IRMS system on the same days as the laboratory reference bottle. When this laboratory reference air bottle became exhausted, it was replaced with two other mixtures of synthetic air and H_2 ((580.78 ± 0.03) ppb and (244.3 ± 0.8) ppb, as determined by BGC Jena) that were regularly measured from December 2008 onward. From measurements during the overlap period, the isotopic composition of these new reference gases was calibrated versus the old reference air bottle ((+207.0 ± 0.3) ‰ and (+198.2 ± 0.5) ‰). Inspection of these measurements also allowed a robust assessment of the reproducibility of our system yielding a standard deviation of 4.5 ‰ in δD . It should be borne in mind that the error estimates given here are estimates of the random error due to

measurement scatter only. The determination of $\delta(D,H_2)$ of the laboratory reference air bottle comprises additional systematic uncertainties, e.g. the correctness of the initially assigned $\delta(D,H_2)$ values of the commercial calibration gases, changes of these values in the process of creating calibration mixtures with near-ambient $m(H_2)$, and the calibration measurements themselves. This may lead to an additional systematic error of a few ‰.

The measurements of the laboratory reference bottles were used to construct 5-day moving average values of measured δD values and the deviation of the measured from the assigned δD value for each measurement day. The moving average of the laboratory reference bottle measurements was replaced by the average of an adjusted selection of these measurements for those days around which a relatively sudden shift in the measured values seemed to occur. An empirical scale contraction factor (1.0613) was determined from the measurements of the mixtures of synthetic air with H_2 of known isotopic composition. This factor was used to determine the corrected δD value for the sample measurements by inter- or extrapolation from the 5-day moving average values of the reference gas results for that day. (It was assumed that the GC-IRMS follows a linear response curve.)

A blank measurement was usually performed once a day as well. The blank peak area was usually less than 4 % of a typical sample peak area. Due to the small peak area, the δD value of the blank could not be determined reliably from these measurements and could therefore not be taken into account in the calculation of the sample δD values. The contribution of the blank signal to the random scatter is included in the 4.5 ‰ reproducibility that is obtained for repeated measurements of the laboratory reference air bottle. However, the blank may cause an unquantified systematic error.

The laboratory reference bottle and blank measurements were used to calculate $m(H_2)$ for the samples as well. Mixing ratios determined by GC-IRMS measurement typically deviated less than 4 % from the determinations by RGA-3 or Peak Performer. They were used for quality assessment of the data, but are not considered further in this paper.

Despite these calibration measures, δD measurements from June 2010 onwards seemed to have a positive offset with respect to the previously measured samples and an increased scatter. By comparing the respective samples from stations with long data series (Alert and Neumayer) with measurements from previous years, the offset was empirically determined to be on average 9.5 ‰. Unfortunately, it could not be determined whether the offset was caused by simultaneous drift in the two laboratory reference bottles in use at that time or by the replacement of parts of the setup. At present the system is undergoing substantial rebuilding and further automation in order to be able to accommodate more reference gas measurements in the future.

The affected measurements of Alert and Neumayer samples still showed the same seasonal patterns as the previous measurements, and were therefore adjusted to fall on the same scale as the previous data by subtracting the empirically determined 9.5 ‰ offset. The respective data are indicated with open symbols in Fig. 2. Cape Verde samples that were also measured during this period showed a more erratic behaviour and were therefore omitted from the time series presented here. The Schauinsland samples collected after September 2009 showed strong pollution signatures (high mixing ratios, as determined by both RGA-3 and GC-IRMS measurements, and very low δD values), which coincided with CO_2 contaminations. As a leak or contamination in the sampling system was suspected, these data were also omitted.

Some of the $m(H_2)$ and δD data points seem to fall outside of the normal range of variation at the respective sampling station. For several of these outliers, we have looked into back-trajectory calculations (see Sect. 3.1.2) to determine if the origin of these air parcels was unusual, but no clear conclusions could be derived from this analysis. Therefore, we have selected outliers based only on visual inspection of the time series. These data points are indicated with open stars in Fig. 2. The data indicated with open symbols in Fig. 2 (the outliers and the samples measured after June 2010) are not used in further calculations.

3 Results and discussion

3.1 Time series

The time series of $m(H_2)$ and δD for all six stations are shown in Fig. 2, and will be discussed individually in the following subsections. Figure 3 shows the seasonal averages from which an annual average was calculated for each station. Using the seasonal averages rather than the raw data for calculation of the mean has the advantage that each of the seasons (DJF, MAM, JJA, SON) has equal weight, which avoids bias from having more samples from some seasons than from others.

Least-squares harmonic fits were applied to the $m(H_2)$ and δD data of the three stations that showed the clearest seasonal cycles in δD . We use the function:

$$y = a \cos(2\pi(x - \varphi)) + m \quad (2)$$

where x is the time in years and a (amplitude), φ (phase) and m (average value) are the fitting parameters. A term for a possible temporal trend is not included in the fit function because our time series of H_2 and δD are still relatively short. The results (listed in Table 3) were used to construct the ellipses in the phase diagrams in Fig. 6.

3.1.1 Alert

Alert is the station with the largest number of analysed samples (Table 2) and about four full seasonal cycles of data. The annual average values are $m(H_2) = (496.9 \pm 1.2)$ ppb, and $\delta D = (124.1 \pm 0.5)$ ‰, where the error bar indicates the standard error of the mean (the standard deviation as a result of the variance in the measurements, divided by the square root of the number of measurements). A difference of 131 ppb is found between the highest and lowest $m(H_2)$ value, as well as a difference of 41 ‰ between the highest and lowest δD (excluding the measurements affected by the 9.5 ‰ offset and outliers for δD).

A distinct seasonal cycle is observed for both $m(H_2)$ and δD (Fig. 2). The two quantities clearly increase and decrease out-of-phase. This out-of-phase timing results from the seasonal variations of the mainly deuterium-depleted (combustion) sources and the deuterium-enriching removal processes. Photochemically produced H_2 can contribute to $m(H_2)$ changes, but contributes little to changes in δD , since its δD value is relatively close to ambient δD values. So, in winter, H_2 accumulates from depleted sources, leading to an mH_2 peak and a δD minimum in spring, while during summer, the removal processes are strong and the remaining H_2 becomes enriched, leading to an mH_2 minimum and δD maximum in autumn. The detailed study of the global H_2 isotope budget with the TM5 model (Pieterse et al., 2011) shows that the soil sink contributes most to the seasonal cycle of the mixing ratio, while the smaller photochemical sink has a slightly larger effect on the isotopic composition, due to the larger isotope fractionation. The emissions from combustion sources are not strongly seasonal. Photochemical production does have a seasonal cycle, but contributes little to the variations in δD .

The phase of the least-square harmonic fits to the Alert $m(H_2)$ data (Table 3) is similar to the phase reported by Rhee et al. (2006b) for the cycle of NH near-tropopause averages of $m(H_2)$ obtained from three flights with a passenger aircraft, whereas the phase found for the δD cycle is slightly different (the Alert δD -cycle is slightly ahead of the (Rhee et al., 2006b) cycle). The average mixing ratio is lower. The most striking difference is in the amplitudes; the amplitudes of the seasonal cycles found at Alert are larger than for the

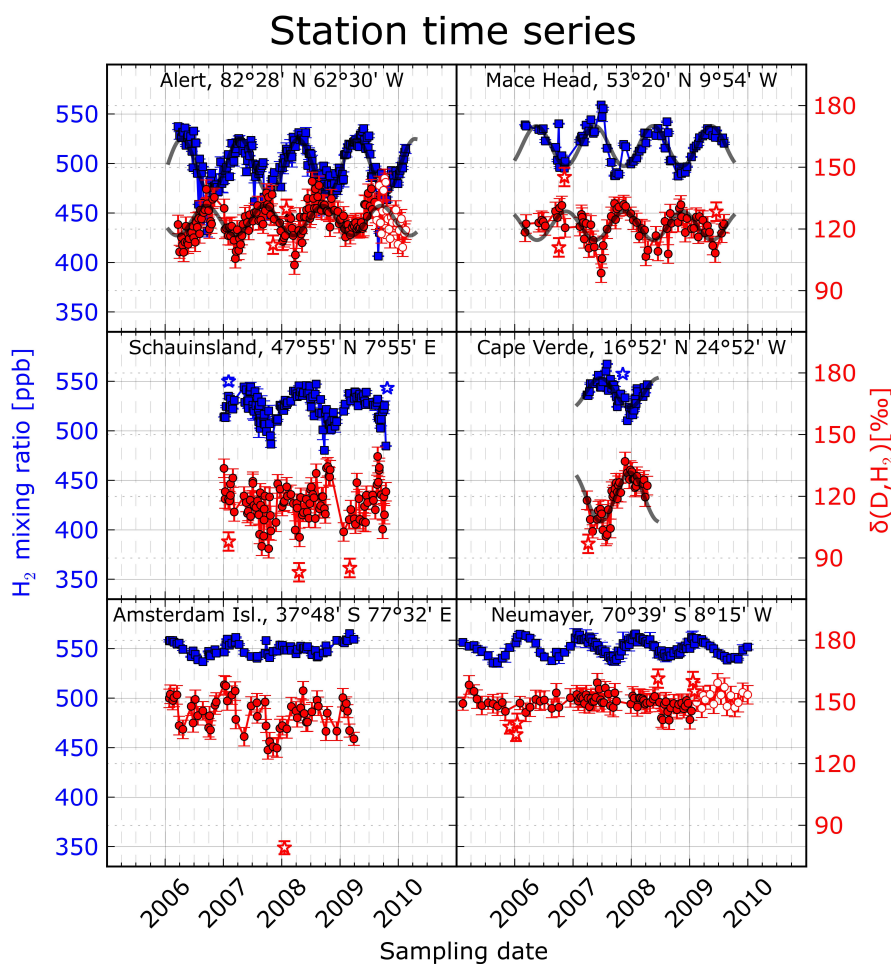


Fig. 2. $m(H_2)$ (blue squares, measured by UHEI-IUP, LSCE and MPI-BGC) and δD (red circles) measured by IMAU on samples from the six stations. Solid grey lines represent a harmonic best fit to the data from Alert, Mace Head and Cape Verde. Error bars indicate one standard deviation for $m(H_2)$ determined from successive measurements, and one standard error for δD (the standard deviation of 4.5 ‰, as determined from repeated laboratory standard bottle measurements, divided by the square root of the number of repeat measurements of the same flask). Open red circles indicate the δD -values where an empirical value of 9.5 ‰ was subtracted to adjust for the bias that affected the GC-IRMS system from June 2010 onward. Open blue and red stars indicate other unexplained outliers in $m(H_2)$ and δD , respectively. None of the open-symbol data are used in any further calculations. Some outliers at Amsterdam Island are off the scale.

Rhee et al. (2006b) NH near-tropopause averages, even more so since on visual inspection the harmonic fit seems to underestimate the amplitudes. The larger amplitudes can likely be attributed to the larger seasonality in the sinks at higher NH latitudes, due to the larger relative soil surface and the seasonal snow cover variation that modulates the soil sink. As we will show below, for the high-latitude NH stations, soil uptake dominates the seasonal cycle more than at lower NH latitudes. It seemingly also dominates the Alert seasonal cycle more than the NH high tropospheric average cycle. In a modelling study that focused on H_2 uptake by soil, Yashiro et al. (2011) found that the soil uptake flux for the latitude range north of 45° N has a much larger seasonal variation than for the 15° N to 45° N latitude range, which is in accordance with this finding. However, it should be noted that the

analysis of Rhee et al. (2006b) is based on fitting a harmonic function to three data points only, whereas the Alert data series clearly captures the full seasonal evolution of the isotope signal.

Gerst and Quay (2000) published six δD measurements on samples from another high northern latitude station, Point Barrow, Alaska, collected in the period from February to August 1997. Unexpectedly, the δD values of $(+92 \pm 10)$ ‰, (one standard deviation) at Point Barrow are significantly lower than what we find at Alert and other NH stations (see below). The reason for this discrepancy is not understood, and the TM5 model (Pieterse et al., 2011) does not predict such a large difference either. It should be noted, however, that in-cylinder growth of H_2 is an issue in the early data from Gerst and Quay (2000). Specifically, four other samples from

Table 3. The obtained fitting parameters for a least-square fit with a harmonic function to the data from Alert, Mace Head and Cape Verde ($y = a \cos(2\pi(x - \phi)) + m$).

	Alert		Mace Head		Cape Verde		(Rhee et al., 2006b) NH hemisphere average	
	$m(H_2)$ (ppb)	δD (‰)	$m(H_2)$ (ppb)	δD (‰)	$m(H_2)$ (ppb)	δD (‰)	$m(H_2)$	δD
Amplitude (a)	27.8 ± 1.4	7.3 ± 0.7	20.3 ± 1.9	7.0 ± 1.0	15.0 ± 1.7	12.0 ± 1.2	18.2 ± 1.6	4.0 ± 0.9
Phase (ϕ)	0.24 ± 0.01	0.66 ± 0.02	0.33 ± 0.01	0.84 ± 0.02	0.43 ± 0.02	0.94 ± 0.02	0.28 ± 0.01	0.85 ± 0.05
Average (m)	497.0 ± 1.0	124.1 ± 0.5	517.6 ± 1.3	121.6 ± 0.7	538.4 ± 1.1	119.7 ± 0.8	543.4 ± 0.8	128.3 ± 0.7

Point Barrow in the study from Gerst and Quay (2000) were considered contaminated since their mixing ratios differed by more than 3σ from a multi-year average of NOAA-CMDL data. All samples were measured more than one year after collection and the discrepancy to the new values from Alert may indicate that the six samples considered reliable may have been contaminated too. The isotopic composition of the cylinder grown contamination was determined by (Gerst and Quay, 2000) to be $\delta D = -614$ ‰, and thus only a few % of contamination from this source could lead to the observed depletions.

3.1.2 Mace Head

The time series for Mace Head span more than three years now. The maximum amplitudes are 72 ppb and 34‰ for $m(H_2)$ and δD , respectively and the annual averages are (518.3 ± 1.4) ppb and (121.2 ± 0.7) ‰. $m(H_2)$ shows a clear seasonal cycle; it increases steadily from autumn until summer, and then decreases rapidly. The δD time series in 2007 and 2008 shows a minimum in late spring or early summer, indicating a seasonal cycle in δD at Mace Head, which is also predicted by the TM5 model (Pieterse et al., 2011). In general, the δD cycle moves in antiphase to the $m(H_2)$ cycle, as discussed above for Alert. However, it can be seen from the seasonal averages (Fig. 3), as well as from the phase obtained from the harmonic fit (Table 3), that the timing of the minimum is delayed with respect to Alert. The phase obtained from the harmonic fit to the Mace Head δD data is in fact closer to the phase found by Rhee et al. (2006b) than the phase obtained from the Alert cycle. Still, as at Alert, the amplitude is larger than that from Rhee et al. (2006b). Mace Head can still be considered to be in the higher latitude part of the NH (especially since it receives much air from regions to the North-West, see below), and this large seasonal variation can thus be attributed to the large seasonal variation in the soil uptake flux at high NH latitudes as well. That this isotopic seasonal cycle is not visible in the first year of the series (2006) may be due to the small number of samples in this year. In 2009, there is one high δD outlier right in the seasonal minimum. The δD seasonal cycles vary con-

siderably between 2007 and 2009, but the same is true for $m(H_2)$, so what causes the interannual variations in $m(H_2)$ likely causes the interannual variations in δD .

To investigate the origin of the interannual variations, backward trajectories were calculated with the NOAA HYSPLIT model to investigate if the rather high $m(H_2)$ and low δD values in summer 2007 might have been a result of synoptic conditions that brought more polluted air than usual to the station. Such an effect was not found. The trajectories showed that most air masses that arrive at Mace Head come from either the north Atlantic or the NH northern temperate to boreal regions to the west, and air masses that were sampled in summer 2007 were no exception. Grant et al. (2010) also showed that Mace Head rarely receives air masses from lower latitudes, supporting our limited trajectory study. Therefore, transport of H_2 -rich and δD -depleted air from lower latitudes does not seem a very likely explanation. It is more likely that the full interannual variability in H_2 cannot be seen at this station due to the limited time period over which samples have been taken so far.

3.1.3 Schauinsland

In the time series from Schauinsland station, the maximum differences are 67 ppb and 45‰, and the seasonally weighted average values are (523.7 ± 0.8) ppb and (117.1 ± 0.9) ‰. This is relatively similar to the values from Mace Head, but the seasonal cycles show differences. At Schauinsland, a clear seasonal cycle is visible in the $m(H_2)$ data, but not in the δD data. In 2007, the δD signal seems to trend downward, while in 2008 and 2009 it appears to trend upward. These trends are not statistically significant, though, but they underline that no clear seasonal cycle in δD is present at Schauinsland.

Unlike Alert and Mace Head, Schauinsland is not a coastal but a continental location, close to anthropogenic sources, especially in the Rhine valley, and in the middle of a large land area where deposition to soil occurs. Possibly, significant nitrogen fixation by soil microbes takes place in the surrounding region as well. H_2 produced by nitrogen fixers has a very depleted source signature (Table 1), so only a small

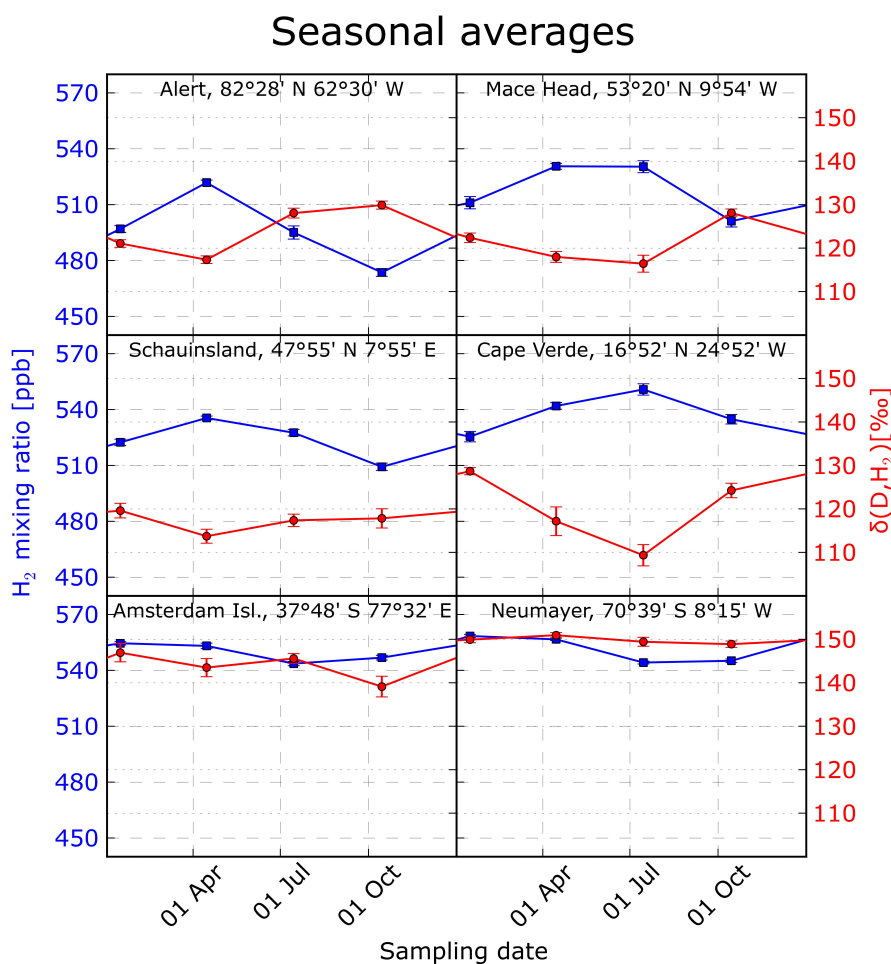


Fig. 3. Seasonal averages of $m(H_2)$ (UHEI, LSCE and MPI-BGC) and δD (IMAU) plotted versus time of year. Error bars indicate one standard error, which is calculated purely from variance in the values obtained for one season.

amount of this H₂ may have a significant effect on the isotopic signature of the total ambient H₂. Although this was not investigated in more detail, interannual variations in microbial production may thus contribute to the difference in seasonal patterns between years.

The station is usually above the boundary layer at night, and within the boundary layer during the day, particularly in summer (Schmidt et al., 1996). The upper edge of the boundary layer generally passes the station altitude after 10:00 a.m. (CET), so samples collected after this time are potentially locally influenced; they may contain boundary layer air that has been in contact with local (anthropogenic) sources and the surface. Most of the samples analysed for this record were sampled before 10:00 a.m.; about 10% was collected later in the day. However, excluding these samples from the time series did not yield a more distinct seasonal cycle. This indicates that the deviations in the isotopic composition are not a product of local factors only. It should be noted that also the TM5 model predicts a considerably

smaller seasonal cycle in δD for Schauinsland compared to Mace Head (Pieterse et al., 2011), so a longer time series may be needed to clearly distinguish the cycle from the data scatter.

3.1.4 Cape Verde

Although the Cape Verde time series is the shortest (little more than one year), the data indicate a clearer seasonal cycle than at Mace Head. As at Alert and Mace Head, $m(H_2)$ and δD vary out-of-phase. The timing of the minima and maxima is closer to what is observed at Mace Head than at Alert. The mixing ratio peaked at 567 ppb in August and then decreased by 57 ppb to 510 ppb at the end of the year. From April until August the δD value was rather constant (minimum value of 100 ‰ in July) and after August it increased until the end of the year (maximum observed value 137 ‰ in November). Seasonally weighted annual averages of (538.3 ± 1.3) ppb and (119.9 ± 1.1) ‰ were observed.

The amplitude found with the harmonic fit to the data is remarkably large, larger than for Mace Head and Alert. This contrasts with the TM5 model results (Pieterse et al., 2011) where the seasonal variation in δD at Cape Verde is somewhat smaller than for Mace Head and Alert. However, in Alert and Mace Head, the fit seems to underestimate the amplitudes of the seasonal variation due to differences between years. The amplitude obtained from the fit is apparently dependent on the presence of interannual variations that cannot be detected for Cape Verde (because of the limited length of the measurement record). Therefore, it seems more reasonable in the case of Cape Verde to compare the δD variation in the original data with the other stations (Fig. 2 and Table 2). When these are compared, the seasonal δD variation does not seem exceptionally large in Cape Verde.

The cycles shift to later in the year when moving from Alert to Mace Head to Cape Verde. This can also be seen in the phases obtained from the cosine fits (Table 3). It is possible that seasonal variation in the sources (e.g. tropical biomass burning) or transport (e.g. interhemispheric exchange) contribute to these shifts. Due to the dominant NNW wind direction (trade winds), Cape Verde receives much air from Mauretania and surrounding countries. The biomass burning season in this region is in winter, but we do not see the expected $m(H_2)$ increase and concomitant δD decrease in this season. Most interhemispheric mixing occurs in summer, when Cape Verde is close to the edge of the ITCZ. Since δD values are higher in the SH (Gerst and Quay, 2000, Rice et al., 2010, see also Fig. 4b below), increased transport from the SH should result in higher δD values, but this is not observed. Therefore, we cannot identify individual source or transport signatures that cause the cycles in Cape Verde to be delayed with respect to the Alert cycles.

3.1.5 Amsterdam Island

The time series from Amsterdam Island does not show a very distinct seasonal cycle. H_2 mixing ratios only vary between 537 and 565 ppb, with seasonal maxima in austral summer, and no clear seasonal signal can be identified in the δD data. This may be due to the remoteness of this location; it is far removed from densely populated areas where anthropogenic H_2 is produced, and far away from large land areas where deposition to soil can occur. Other $m(H_2)$ time series from locations in the Indian Ocean, such as the time series from the Seychelles (in Novelli et al., 1999), also show very small seasonal cycles.

Yashiro et al. (2011) concluded from the timing of the $m(H_2)$ seasonal cycle that photochemical production is a dominant driver of the seasonality in $m(H_2)$ south of 30° S, which is in accordance with the conclusions that Rhee et al. (2006b) drew from δD observations. Since the δD value of photochemically produced H_2 is close to ambient values, this does not cause a large seasonality in δD . Nonetheless, in austral summer photochemical H_2 destruction will occur, with a

D-enriching effect on the reservoir. That we see no seasonal cycle in δD implies that a D-depleting process compensates this enrichment. Transport of depleted H_2 from lower latitudes (from tropical biomass burning in SH spring and interhemispheric exchange) is a likely explanation. The budget calculations made by Pieterse et al. (2011) suggest that transport of depleted H_2 from lower latitudes is a significant contributor to the seasonal δD changes in the extratropical SH.

Some unexplained extreme outliers do occur in the measurements. Excluding these outliers, the maximal variability in mixing ratio is 28 ppb, and in the isotopic composition 32 ‰. The yearly average values are (549.5 ± 0.6) ppb and (143.8 ± 1.0) ‰.

3.1.6 Neumayer

At present, the Neumayer time series covers 5 full seasonal cycles, and average values are (551.1 ± 0.4) ppb and (149.8 ± 0.4) ‰. A clear seasonal variation can be distinguished in the mixing ratio data, but not in the isotope data. The mixing ratios vary by at most 30 ppb over the year, and the isotope values by at most 18 ‰, making Neumayer the station with the lowest variability in δD . A relatively sharp “dip” appears in the δD values at the end of 2005. This feature was not observed in the other years, so we suspect that this is an artefact rather than a real atmospheric signal and flagged these data points as outliers.

The variation in δD at Neumayer is remarkably small, considering that $m(H_2)$ does show a clear cycle. As in Amsterdam Island, there is an $m(H_2)$ maximum in austral summer, likely an effect of photochemical production. That no accompanying change in δD is observed can probably be attributed to the same processes that cause the lack of seasonality in δD at Amsterdam Island: a maximum in photochemical production produces H_2 without affecting the δD value much, and the D-enriching effect from photochemical destruction of H_2 in summer is balanced by a D-depleting process like transport of depleted H_2 from lower latitudes. The timing of these processes is, however, somewhat problematic. According to the TM5 model calculations (Pieterse et al., 2011), horizontal transport has the largest depleting effect on the H_2 deuterium content south of 30° S in the months July, August and September. This is too early to compensate for the expected enriching effect of photochemical H_2 destruction in austral summer. The lack of seasonal variation at Neumayer, and to some extent Amsterdam Island, is therefore not fully explained and requires further study.

3.1.7 Seasonal mean variations

Averages per season (December January February (DJF), March April May (MAM), June July August (JJA) and September October November (SON)) were calculated for each station and are shown in Fig. 3. This averaging

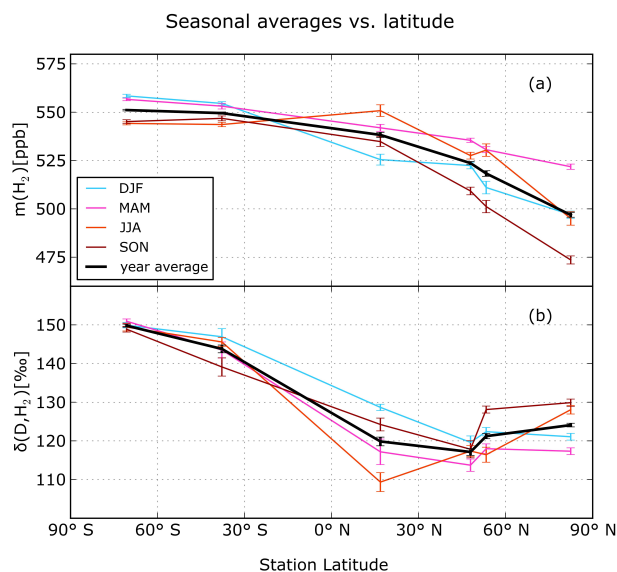


Fig. 4. Seasonal averages and seasonally weighted yearly averages of $m(H_2)$ (a) and δD (b).

eliminates the scatter from analytical uncertainty and short-term natural variability and allows the general shape of the seasonal cycles to be compared between the stations. The anticorrelated seasonal cycles of δD and $m(H_2)$ at Alert can be seen very clearly. There is also an anticorrelation at Mace Head and Cape Verde, but the timing of the minima and maxima is different from the timing at Alert. Although it was not visible in the raw data in Fig. 2, the seasonal averages of the Schauinsland data seem to show a weak anticorrelation, which is primarily caused by the δD value having a minimum in spring, when $m(H_2)$ shows a maximum. In comparison to the NH stations, the two SH stations have much smaller seasonal cycles in $m(H_2)$ and δD . The δD cycle at Neumayer does not seem well-correlated to the δD cycle at Amsterdam Island.

3.2 Latitudinal gradients

Figure 4 shows the latitudinal dependence of the seasonal averages. In all seasons, both mixing ratios and δD values are higher in the SH than in the NH. The lowest average mixing ratios are found for the highest-latitude NH station (Alert) in all seasons. The minimum in δD is, however, not found at Alert for any season, but either at one of the mid-latitude stations, or, in summer (JJA), at Cape Verde. As anthropogenic H_2 emissions originate mainly from temperate latitudes (Hauglustaine and Ehhalt, 2002; Price et al., 2007), and H_2 from anthropogenic source regions is usually depleted with respect to atmospheric H_2 , the midlatitude minimum can be attributed to anthropogenic influence. This is also evident in the spatial δD distribution modelled in TMS (Pieterse et al., 2011).

Gerst and Quay (2000) concluded from their dataset, obtained from a Pacific Transect at the end of 1998 and two stations in North America, that there was a poleward decrease in δD . However, at their northernmost sampling point (Point Barrow, 71° N), the δD values seem to be exceptionally low. Whereas the rest of the data agree reasonably well with our measurements, their measurements from Point Barrow are more than 20 ‰ lower than our data from Alert. The potential problems with these Point Barrow measurements have been discussed above. We note that without this sampling point, also the data from Gerst and Quay (2000) show a δD minimum at temperate northern latitudes.

From Fig. 4, it is clear that the latitudinal gradient varies with season. As the seasonal variation in both $m(H_2)$ and δD is much larger in the NH than in the SH, the size of the latitudinal gradient is for the largest part determined by the variation in the NH. Hence, the smallest pole-to-pole δD difference (19 ‰ between Neumayer and Alert) is found when δD is at its maximum in Alert, i.e., in SON. During this season, $m(H_2)$ is lowest in Alert, and therefore the $m(H_2)$ difference is largest (72 ppb). At that time, soil uptake in Alert is past its summer peak, so that H_2 mixing ratios are low and the remaining H_2 is isotopically enriched. Reversely, the largest δD pole-to-pole difference (34 ‰) is found when δD is at its minimum in Alert and $m(H_2)$ is at its maximum, namely in MAM, when over winter H_2 from depleted sources has accumulated without being absorbed by the snow-covered soil. This is accompanied by the smallest difference in $m(H_2)$ (35 ppb).

The interhemispheric difference reported by Gerst and Quay (2000) was about 15 ‰, and Rice et al. (2010) found a similar gradient, (16 ± 12) ‰. The interpolar difference from the EUROHYDROS stations is clearly larger, but we miss a station at low southern latitudes to be able to make a reliable southern hemispheric average and to calculate a reliable difference between the two hemispheres.

3.3 Latitude dependence of the apparent fractionation factor

If it is assumed that the seasonal cycles of H_2 and its isotopic composition are primarily determined by the sink processes, the fractionation factor α , i.e., the ratio of the removal rate of the heavy isotope species (k_{HD}) to the removal rate of the light isotope species (k_{HH}), can be calculated from the time series as for a single stage Rayleigh fractionation process (see, Rhee et al., 2006b). In such a Rayleigh removal process, $m(H_2)$ and δD evolve as

$$(\alpha - 1) \ln \left[\frac{m(H_2)}{m(H_2, \max)} \right] = \ln(\delta D + 1) + c \quad (3)$$

where $m(H_2, \max)$ is the maximum value for $m(H_2)$ found in the series and c is a constant. The real situation at the measurement stations is more complex than this simple Rayleigh fractionation model, as will be discussed below. Still, an

“apparent” fractionation factor was calculated for each station from the slope of a linear fit to the data in a Rayleigh fractionation plot (Fig. 5). This apparent fractionation factor does not quantify the inherent isotope fractionation in one single reaction, but is calculated from atmospheric data collected in a complex situation that involves more than one sink reaction, as well as other processes such as mixing, and we denote this apparent fractionation factor by α_{app} . To make the straight-line fits, a Weighted Total Least-Squares (WTLS) fitting algorithm (Krystek and Anton, 2007) was used. This algorithm takes errors in both the x- and the y-direction into account. The squared correlation coefficient (R^2) and the F-test p -value were calculated for each station in the same way as they would be calculated for an “ordinary” least-squares fit of a straight line. The fits to the Amsterdam Island and Neumayer data yielded p -values above 0.05, suggesting a less than 95 % significance of the correlation. The fit to the Schauinsland data did yield a p -value that was slightly smaller than 0.05, but the R^2 -value for this fit is very low, which indicates that only a very small part of the variation present in the dataset is described by the fit. This means that the simple Rayleigh fractionation model may not be adequate for evaluation of the time series from Schauinsland and the SH. This is not surprising, considering that no clear isotope seasonal cycles were found for these stations.

Of course, the isotopic composition of the H_2 at the measurement stations depends not only on the fractionation in the removal processes, but also on the isotopic composition of the sources that influence the sites and their variation over the year. Also the relative contribution of the two sink processes (atmospheric OH oxidation and uptake by soils) to the total sink could conceivably vary with season. In this light, it is illustrative to look at the phase diagrams (δD plotted against $m(H_2)$) of the monthly means that can be constructed from the three stations that show clear cycles (Fig. 6). For Alert, it is clear that these points do not fall on a single straight fractionation line, but on the ellipse that is traced by the two harmonic functions that were fitted to the $m(H_2)$ and δD time series. This shows that in the fractionation plots in Fig. 5, not all of the spread around the fractionation lines is random scatter; some of this spread is caused by the phase difference between the $m(H_2)$ and δD seasonal cycles. The largest part of this phase difference is probably caused by seasonal variation in the sources and sinks.

Allan et al. (2001) described such phase ellipses for model results of methane mixing ratio and the carbon isotopic composition of methane ($\delta(^{13}C, CH_4)$). In their simulations with only one source and one sink process, the monthly means fell on a straight line. This line broadened into an ellipse when different sources with different seasonal cycles were combined, with the largest broadening effect for the sources of which the isotopic signature differed the most from the isotopic signature of the mean source. However, the differing phases of the different sources did not appreciably affect the slope of the major axis of the ellipse: this slope depended

robustly on the fractionation factor assumed for the sink, suggesting that even if the phase diagram shows a broadened ellipse, this slope can provide a good estimate of the fractionation in the sink. The ellipse major axis coincided largely with a relationship the authors termed the “KIE line” after the Kinetic Isotope Effect, which is described by:

$$\Delta\delta(^{13}C, CH_4) = \epsilon(1 + \delta_0(^{13}C, CH_4)) \frac{\Delta m(CH_4)}{m_0(CH_4)} \quad (4)$$

with subscript zero indicating a mean value, Δ indicating the difference of a value from the mean, $m(CH_4)$ indicating methane mixing ratio, and $\epsilon = \alpha - 1$.

In the case of H_2 , seasonal variation in the sources and sinks contributes to the ellipse eccentricity, but Rhee et al. (2006b) showed that as long as the H_2 system is in isotopic equilibrium on an annual basis, the isotopic fractionation does not depend on seasonal changes in source emissions. Since there is no evidence for a tropospheric trend in either $m(H_2)$ or δD , it is reasonable to assume isotopic equilibrium for the years over which the data presented here were collected.

A factor that could still affect the analysis is a seasonally varying contribution of the two main sinks. However, especially in Alert, any variations in the relative contributions of the two sinks over the year are expected to be small, because even if soil uptake and atmospheric OH oxidation do not follow the exact same seasonal pattern, their patterns are expected to share the main characteristics, with a minimum in winter, and a peak in summer. Hence, although the Rayleigh fractionation model does not represent the full complexity of reality at the stations, the apparent fractionation constants determined with this model provide a good estimate of the total sink fractionation.

For Alert, Mace Head and Cape Verde, KIE lines were calculated with the hydrogen α_{app} obtained for the different stations, as well as with the fractionation constants (Rhee et al., 2006b) for the two H_2 sink processes and added to Fig. 6. Clearly, for Alert and Mace Head the KIE line of the fit lies closest to the KIE line of soil uptake, whereas in Cape Verde it lies closest to atmospheric OH oxidation. For these three stations, α_{app} is plotted versus station latitude in Fig. 7a. The apparent fractionation factor can be interpreted as a combination of the fractionation factors of the main sink processes. For Alert and Mace Head, the α_{app} values (0.905 ± 0.008 and 0.884 ± 0.014 , respectively) are close to the fractionation factor assigned to soil uptake ($\alpha = 0.943 \pm 0.007$). Rhee et al. (2006b) also concluded from a similar analysis on a small set of upper tropospheric air samples that the seasonal cycle of H_2 in the NH is largely driven by soil uptake. Closer to the equator, at Cape Verde, α_{app} (0.704 ± 0.029) is lower and closer to the value for oxidation by OH ($\alpha = 0.58 \pm 0.07$). This indicates that the relative importance of the uptake by soil with respect to the destruction by OH increases with latitude.

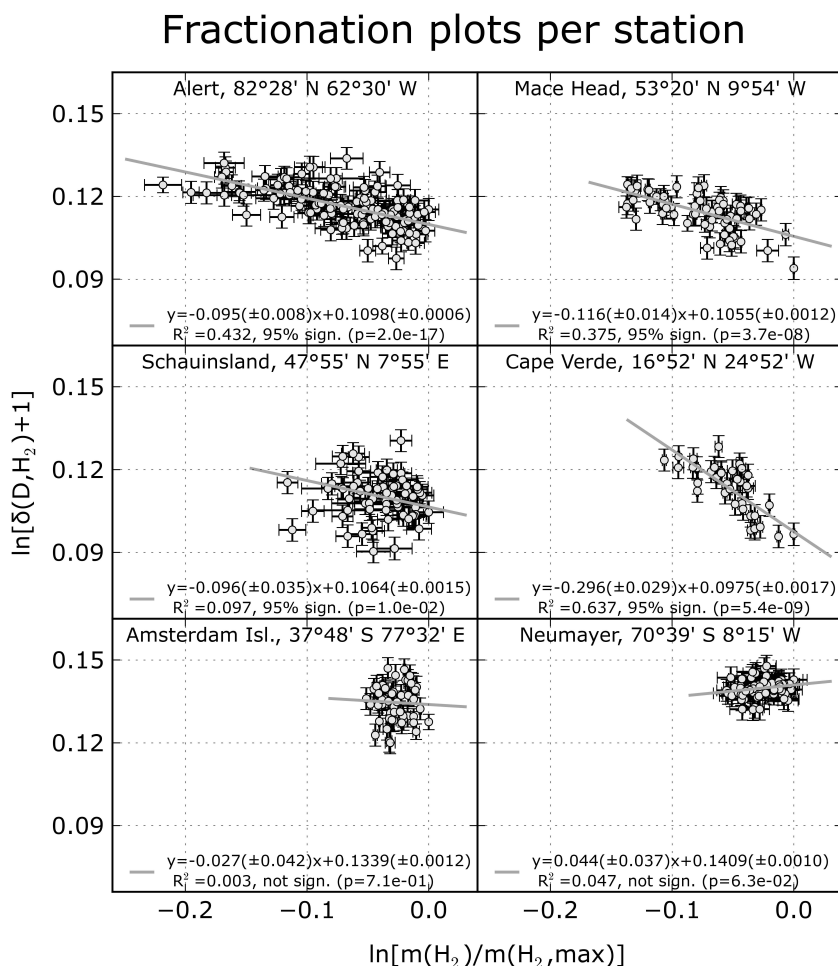


Fig. 5. Fractionation plots of the datasets from the different stations. Plotted are $\ln(\delta D + 1)$ vs. $\ln(m(H_2)/m(H_2, \max))$ with $m(H_2, \max)$ the maximum mixing ratio found. Grey lines are the fitted lines obtained from the Weighted Total Least-Squares fitting routine. Error bars indicate one standard deviation. Indicated errors in the fit parameters indicate one standard error. R^2 is the squared correlation coefficient and p is the p -value for the F statistic, both calculated as for an “ordinary” least squares fitted line.

Figure 7b shows the relative contribution of both sinks processes to the total destruction of H_2 , as can be calculated from these isotope data if α_{app} is assumed to be a simple mass weighted average of the fractionation factors of soil removal and reaction with OH. This calculation yields that OH destroys more than half (66 ± 15 %) of the H_2 at Cape Verde, but that at Alert, ≈ 90 % of the H_2 is destroyed by soil uptake. This finding of a latitudinal trend in the data presented here depends heavily on the Cape Verde station, where the formal application of the Rayleigh fractionation model yields a very low α_{app} and therefore a large contribution of OH oxidation to the total sinks. As discussed in Sect. 3.1.4, the $m(H_2)$ and δD cycles at Cape Verde are shifted in time with respect to Alert and Mace Head, which indicates that the way the source, sink and transport processes drive the seasonal variations is somewhat different between these stations. But although α_{app} may be underestimated in Cape Verde, the gen-

eral trend towards a higher OH fraction with decreasing latitude is in accordance with expectation and model results. In the TM5 model study, H_2 destruction is clearly dominated by the soil sink in the high northern latitude band ($30^\circ N$ – $90^\circ N$), whereas the two sinks are of comparable magnitude in the tropical latitude band ($30^\circ N$ – $30^\circ S$) (Fig. 8 in Pieterse et al., 2011). Therefore, our findings indicate that the NH average α_{app} of 0.90 ± 0.02 found by Rhee et al. (2006b) for a limited number of samples in the tropopause region and the associated (87 ± 7 %) relative contribution of the soil sink may be overestimates. This also means that their H_2 lifetime estimate of (1.4 ± 0.2) years may be too short. To investigate this further, it would be useful to analyse samples from subtropical northern latitudes, for example the Iberian Peninsula, the Canary Islands or Morocco.

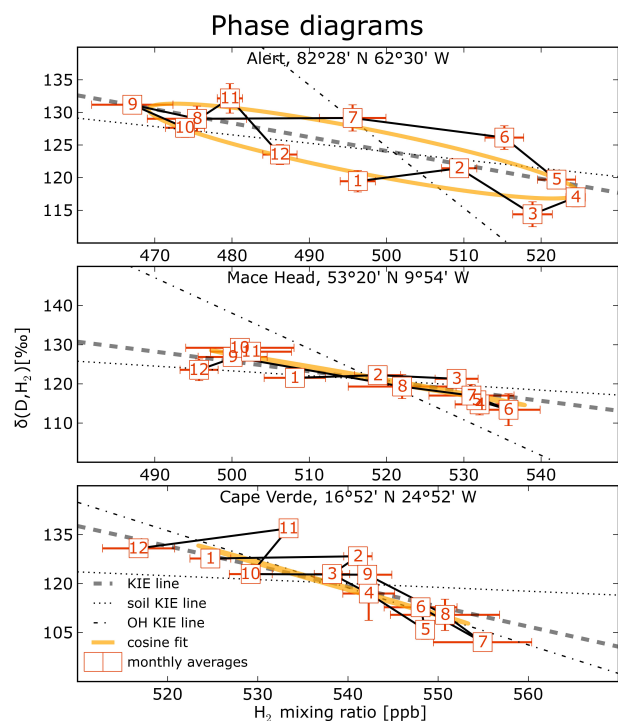


Fig. 6. Phase diagrams (δD vs. $m(H_2)$) of the monthly averages (labelled with month number) of the three stations that show clear seasonal cycles. The orange ellipses are derived from the sinusoidal fits to the time series. The thick grey dashed line is the “KIE” line for the apparent fractionation factor determined from the fractionation plot for the respective station as defined in (Allan et al., 2001). The dotted and dash-dotted lines are the KIE lines for the two different sink processes, with values for α from (Rhee et al., 2006b). Note that the determined KIE line at Alert and Mace Head is closest to the soil uptake KIE line, while the determined KIE line at Cape Verde is closer to the OH oxidation KIE line.

4 Conclusions

Air samples from six EUROHYDROS stations have been analysed regularly for $m(H_2)$ and δD , which allows analysis of the temporal and latitudinal distribution of H_2 and its isotopic composition. These data greatly expand the existing δD dataset, and can be used to constrain the global and regional H_2 budgets with the help of global and regional models.

The out-of-phase behaviour of $m(H_2)$ and δD that was proposed for the NH (Rhee et al., 2006b) is seen at Mace Head and Cape Verde, and particularly clearly at Alert, but not at Schauinsland, which is a continental station and closer to anthropogenic source regions. The SH stations show very little variation in δD , even when $m(H_2)$ exhibits a small seasonal cycle. The δD data show a clear latitudinal gradient with higher values in the SH than in the NH. This gradient varies with season, mainly driven by the seasonality in the NH. The observed pole-to-pole differences are larger than observed on ship cruises (Gerst and Quay, 2000; Rice et al., 2010). The

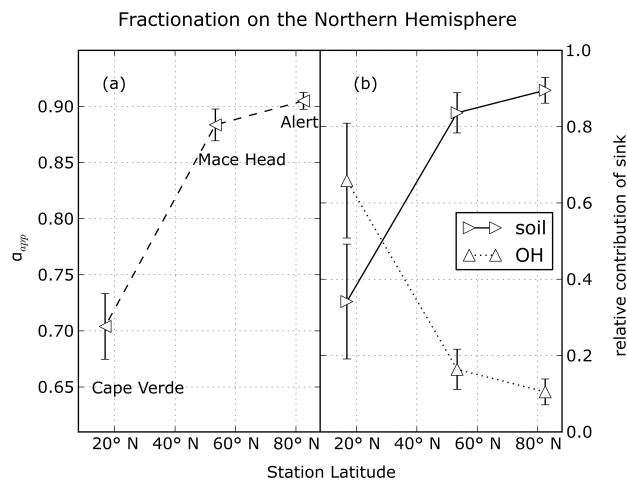


Fig. 7. (a) Apparent fractionation factors for the NH, calculated from a least-squares fit to a Rayleigh plot, plotted as a function of station latitude. (b) Relative contributions of uptake by soils and atmospheric oxidation by OH to the total sinks, calculated from the apparent fractionation factors and the values for the fractionation processes used by Rhee et al. (2006b). Error bars indicate one standard error.

lowest annual average δD value is not found at the highest NH latitude station, but at lower NH latitudes, where the influence of the anthropogenic fossil fuel combustion source is strongest. If it is assumed that the seasonality of the $m(H_2)$ and δD time series is mainly determined by the removal processes, the relative changes of $m(H_2)$ and δD can be used to provide information on the relative sink strengths of soil deposition and atmospheric oxidation. The data then indicate that at high NH latitudes, the removal processes are almost completely dominated by deposition to soil, whereas in the tropics, soil deposition and atmospheric oxidation by OH are of similar importance. Our findings imply that the previously reported average value of α_{app} for the NH and the large relative contribution of the soil sink derived from it may be overestimates.

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For the trajectory study of the Mace Head data we used the NOAA Air Resources Laboratory (ARL) Hybrid Single Particle Lagrangian Integrated Trajectory Model (HYSPLIT): http://www.arl.noaa.gov/HYSPLIT_info.php. Graphics for this article were generated using Python 2.6 with the matplotlib and pylab packages.

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