

Azamacrocyclic Ca²⁺ Sensitive Contrast Agents for MR Imaging

Ilgar Mamedov¹, Goran Angelovski¹, Jörg Henig², Hermann A. Mayer² and Nikos K. Logothetis¹,

¹Department for Physiology of Cognitive Processes, Max-Planck Institute for Biological Cybernetics, Tübingen, Germany, ²Institute for Inorganic Chemistry, University of Tübingen, Tübingen, Germany

Introduction. As calcium plays an important role in regulating a great variety of neuronal processes, many efforts are already made to generate gadolinium complexes that can act as a calcium-sensors in MRI.¹ We developed a series of the DO3A-based macrocyclic and bismacrocylic gadolinium chelates, bearing phosphonate groups as an additional coordination sites. These complexes are hypothesized to change the MRI contrast dynamically with Ca²⁺ concentration. Different lengths of the phosphonate side chains are exploited for fine-tuning the sensitivity of the agent to calcium ion concentration.

Methods. The macrocyclic ligands DO3A-**alkyl** aminobis(methylenphosphonates) (alkyl = propyl **1**; amyl **2**; hexyl **3**), and the bismacrocylic bisDO3A-ethyleneaminobis (alkyl)phosphonates (alkyl = methyl **4**; ethyl **5**) were prepared in multistep reactions from cyclen. Gadolinium complexes were obtained by treating the ligands with GdCl₃ at pH 7.0 -7.5 and 70°C for 24h. Longitudinal and transverse relaxivities r_1 and r_2 were measured on a vertical 7 T/60 cm MRI Biospec system (Bruker Biospin, Germany), at pH 7.4 and different concentrations of Ca²⁺.

Results. The sensitivity of the complexes **1-3** for the changes in Ca²⁺ concentrations increased with the chain length between the DO3A unit and the phosphonate functions (Fig. 1). Maximal observed change in relaxivity in presence of the physiological extracellular concentration of calcium (~1 mM) was up to 35% for compound **3**, and was selective to Ca²⁺ compared to other physiologically important metals such as magnesium or zinc. The significant changes in the relaxivity of the bismacrocylic compounds **4** and **5** were found in the presence of a higher concentration of Ca²⁺ ions.

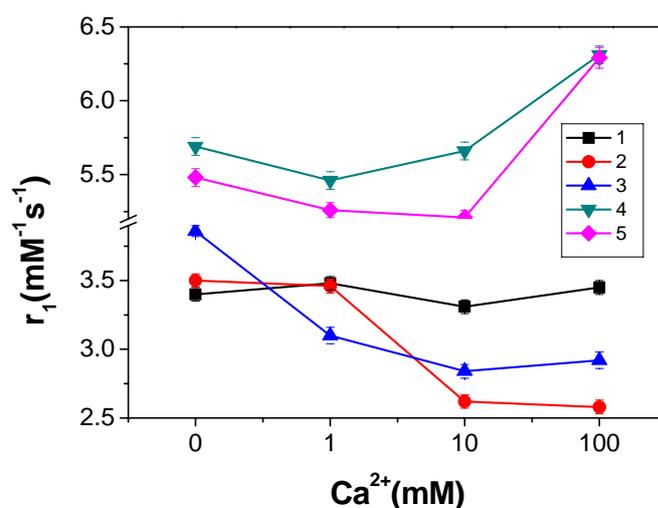


Figure 1. Dependence of the complex relaxivity to the Ca²⁺ ions concentration

1. Li et al. J.Am.Chem.Soc., **1999**,121,1413-1414

This work is supported by the Max-Planck Society, the Louis-Jeantet Foundation and the Hertie Foundation.