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Supporting Information

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Homogeneous Catalytic Hydrogenation of Amides to Amines

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■ SECTION S-1: SUPPORTING FIGURES REFERRED TO IN THE MAIN TEXT

Figures S1-S3 given below should be viewed in conjunction with the main text as. For a more detailed explanation of what each figures represents, please refer back to the main text.

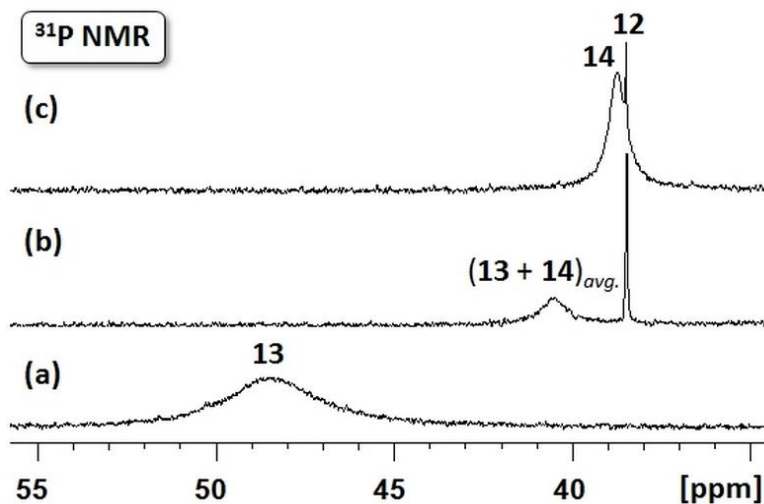


Figure S1. $^{31}\text{P}\{^1\text{H}\}$ NMR spectra for (a) **13** [obtained in the presence of a large excess (> 3 eq) of MSA]; (b) **12** together **13** and **14** (observed as an average); (c) the same sample as in (b) the mixture after equilibration overnight at R.T to give mainly **14**.

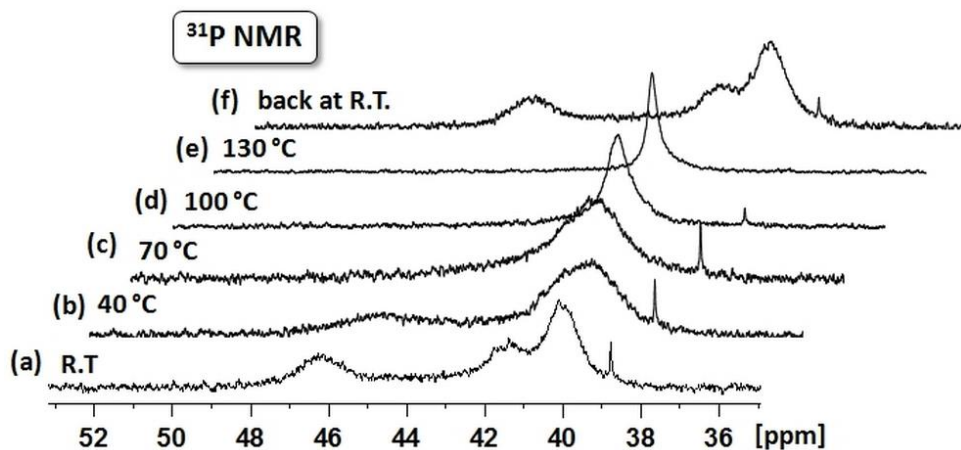


Figure S2. NMR spectra recorded during a VT HP-NMR study on the in solution behaviour of **9** at (a) R.T., (b) 40 °C, (c) 70 °C, (d) 100 °C, (e) 130 °C and (f) back at R.T. Conditions: *N*-phenylacetamide (0.75 mmol), thf- d_8 (3 ml), **9** (10 mol %), MSA (15 mol %) and H_2 (10 bar).

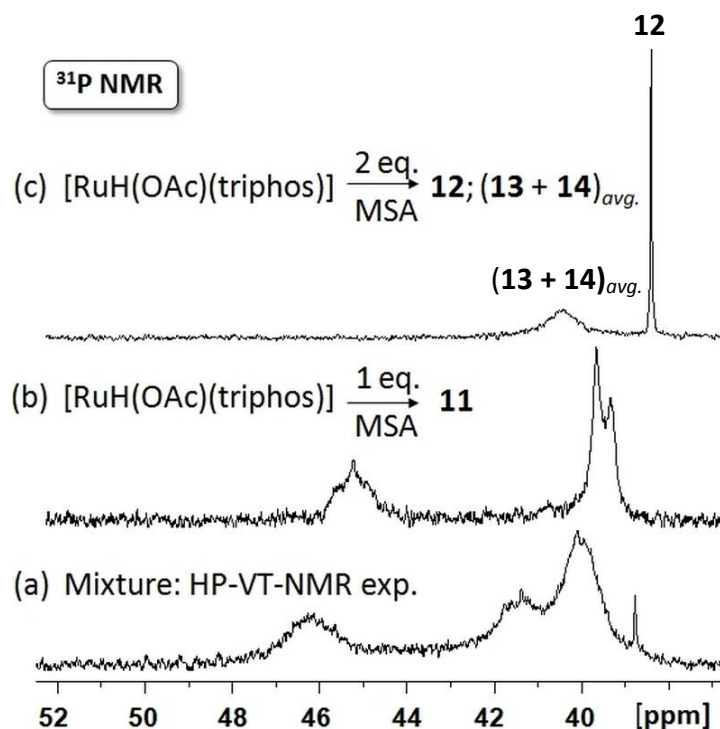


Figure S3. A comparison between the $^{31}\text{P}\{^1\text{H}\}$ NMR spectra of (a) a mixture of *N*-phenylacetamide, **9** and MSA in thf-d_8 under H_2 (10 bar) at R.T (b) **11** and (c) a mixture of **12**, **13** and **14**.

■ SECTION S-2: CRYSTALLOGRAPHIC DISCUSSION

The crystal and molecular structures of complexes **11**, **12** and two water coordinated analogues of **14** and **15**, $[\text{Ru}(\text{CH}_3\text{SO}_3\text{-}\kappa^1\text{O})_2(\text{H}_2\text{O})(\text{triphos})]$ (**19**) and $[\text{Ru}(\text{OAc-}\kappa^1\text{O})_2(\text{H}_2\text{O})(\text{triphos})]$ (**20**), were determined by single crystal X-ray-diffraction and are depicted in Figures S4–S7 (see main text for the compound numbering scheme). In all of the determined structures, the geometry about the Ru(II) centers are best described as distorted octahedral with the triphos ligands adopting the expected *facial* configuration. Complex **11** crystallises together with two molecules of dichloromethane as yellow prisms in the monoclinic spacegroup $P2_1/c$ (Figure S4). For **11** the Ru(1)–O(1), Ru(1)–O(2) and Ru–P bond lengths are comparable to those reported for the related complex, $[\text{RuCl}(\text{OAc})(\text{triphos})]$ [(Ru–O) $_{\text{avg}}$, 2.224 Å and (Ru–P) $_{\text{avg}}$ 2.278 Å].¹ Similarly the Ru(1)–O(3) distance of 2.203(4) Å compares well that between the oxygen coordinated dimethylsulfoxide (DMSO) ligands and the metal centre in, *fac*- $[\text{Ru}(\text{triphos})(\text{DMSO-}\kappa^1\text{O})_2(\text{H}_2\text{O})]\text{CF}_3\text{SO}_3$ [2.215(4)Å and 2.190(4) Å].² The small chelate angle of the acetato ligand [59.80(15)°] forces a distorted octahedral geometry, resulting in widening of the O(1)–Ru(1)–P(2) [105.73(11)°] and O(2)–Ru(1)–P(3) [109.63(11)°] angles. Furthermore, in order to minimise steric congestion the methanesulfonato ligand is bent away from the triphos phenyl rings to give an O(3)–Ru(1)–P(1) angle [171.97(10)°] which deviates significantly from the ideal linearity.

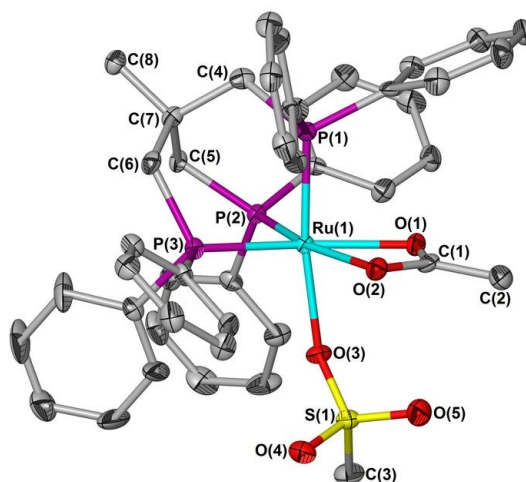


Figure S4 Molecular structure of **11** with thermal ellipsoids set at 50 % probability. Hydrogen atoms and solvent molecules are omitted for clarity. Selected bond lengths (Å) and angles(°): Ru(1)–O(1) 2.173(4), Ru(1)–O(2) 2.210(4), Ru(1)–O(3) 2.203(4), Ru(1)–P(1) 2.2519(15), Ru(1)–P(2) 2.2731(15), Ru(1)–P(3) 2.2551(13), O(1)–Ru(1)–O(3) 80.83(14), O(1)–Ru(1)–O(2) 59.80(15), O(3)–Ru(1)–O(2) 84.23(14), O(1)–Ru(1)–P(1) 91.50(11), O(3)–Ru(1)–P(1) 171.97(10), O(2)–Ru(1)–P(1) 89.85(11), O(1)–Ru(1)–P(3) 169.42(11), O(3)–Ru(1)–P(3) 99.20(11), O(2)–Ru(1)–P(3) 109.63(11), P(1)–Ru(1)–P(3) 87.89(5), O(1)–Ru(1)–P(2) 105.73(11), O(3)–Ru(1)–P(2) 95.00(11), O(2)–Ru(1)–P(2) 165.48(11), P(1)–Ru(1)–P(2) 89.37(6), P(3)–Ru(1)–P(2) 84.83(5).

The methanesulfonato bridged dimer, **12**, crystallises from dichloromethane, together with three dichloromethane molecules, as yellow prisms in the triclinic space group $P\bar{1}$. In the molecular structure of **12**, the coordination sphere around each Ru(II) centre adopts a slightly distorted octahedral geometry. When viewed along the Ru···Ru vector, the two triphos ligands approach an eclipsed conformation while the bridging sulfonato ligands are staggered with respect to the P-atoms (Figure S5). The average Ru–O bonding distance (2.187) of the bridging ligands is comparable to that of the monodentate methanesulfonato ligand to Ru in **11** [Ru(1)–O(3) 2.203(4)]. Similarly, no noteworthy differences exist between the Ru–P bond lengths of compound **11** and **12**.

During an attempt to separate **14** from a mixture of **12**, **13** and **14** by recrystallisation, the water coordinated analogue [Ru(CH₃SO₃-κ¹O)₂(H₂O)(triphos)] (**19**) crystallised from solution as yellow prisms in the monoclinic space group $P2_1/n$. Although the asymmetric unit contains two unique molecules of **19** co-crystallised with five solvent molecules, only one of them will be discussed here as no significant difference between their corresponding bond lengths and angles (Figure S6) exist. The methanesulfonato ligands are bent away from the triphos ligand towards the coordinated water molecule not only to minimise steric interactions but also to facilitate the formation of two six membered rings through hydrogen bonding, which aids in stabilising the structure. As for complex **11**, the Ru(1)–O(1) [2.209(5) Å], Ru(1)–O(3) [2.207(5) Å] and the average Ru–P (2.273 Å) bonding distances sit within the range of those reported for *fac*-[Ru(triphos)(DMSO-κ¹O)₂(H₂O)]CF₃SO₃ in the literature.²

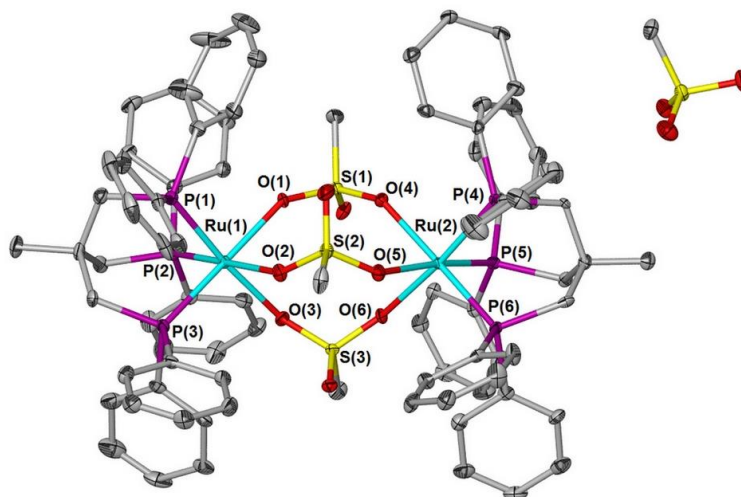


Figure S5 Molecular structure of **12** with thermal ellipsoids set at 50 % probability. Hydrogen atoms and solvent molecules are omitted for clarity. Selected bond lengths (Å) and angles(°): Ru(1)–O(1) 2.167(6), Ru(1)–O(2) 2.195(6), Ru(1)–O(3) 2.190(6), Ru(2)–O(4) 2.181(7), Ru(2)–O(5) 2.193(6), Ru(2)–O(6) 2.193(5), Ru(1)–P(1) 2.251(3), Ru(1)–P(2) 2.283(3), Ru(1)–P(3) 2.263(3), Ru(2)–P(4) 2.279(2), Ru(2)–P(5) 2.252(3), Ru(2)–P(6) 2.280(3), O(1)–Ru(1)–O(2) 83.5(3), O(1)–Ru(1)–O(3) 84.0(3), O(3)–Ru(1)–O(2) 81.9(3), O(1)–Ru(1)–P(1) 90.71(17), O(3)–Ru(1)–P(1) 174.48(17), O(2)–Ru(1)–P(1) 95.88(17), O(1)–Ru(1)–P(3) 178.94(15), O(3)–Ru(1)–P(3) 96.62(18), O(2)–Ru(1)–P(3) 95.79(19), P(1)–Ru(1)–P(3) 88.62(9), O(1)–Ru(1)–P(2) 94.10(18), O(3)–Ru(1)–P(2) 93.59(17), O(2)–Ru(1)–P(2) 175.05(16), P(1)–Ru(1)–P(2) 88.44(9), P(3)–Ru(1)–P(2) 86.70(9), O(1)–S(1)–O(4) 112.5(4), O(2)–S(2)–O(5) 112.0(4), O(3)–S(3)–O(6) 113.4(4).

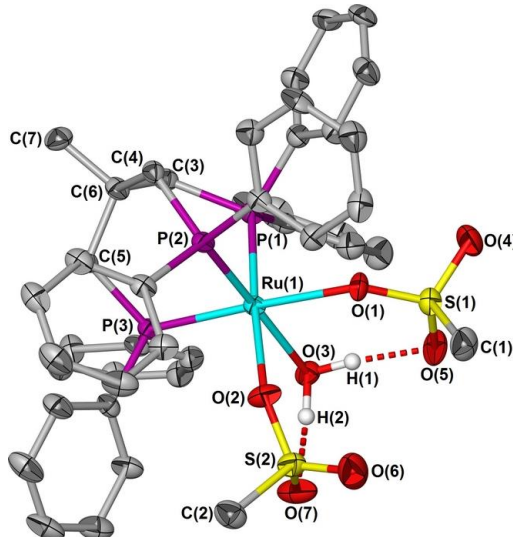


Figure S6 Molecular structure of **19** with thermal ellipsoids set at 50% probability. Included solvent molecules and hydrogen atoms (with the exception of those associated with water) are omitted for clarity. Selected bond lengths (Å) and angles(°): Ru(1)–O(3) 2.207(5), Ru(1)–O(1) 2.209(5), Ru(1)–O(2) 2.223(5), Ru(1)–P(3) 2.2615(19), Ru(1)–P(1) 2.2668(19), Ru(1)–P(2) 2.2914(18), S(1)–O(1) 1.472(5), S(1)–O(1) 1.472(5), P(1)–C(3) 1.841(8), P(2)–C(4) 1.837(7), P(3)–C(5) 1.833(7), O(3)–Ru(1)–O(1) 82.43(19), O(1)–Ru(1)–O(2) 81.8(2), O(3)–Ru(1)–O(2) 81.7(2), O(3)–Ru(1)–P(3) 95.71(15), O(1)–Ru(1)–P(3) 177.77(15), O(2)–Ru(1)–P(3) 96.77(17), O(3)–Ru(1)–P(1) 96.77(15), O(1)–Ru(1)–P(1) 94.92(15), O(2)–Ru(1)–P(1) 176.50(17), P(3)–Ru(1)–P(1) 86.50(7), O(3)–Ru(1)–P(2) 174.13(15), O(1)–Ru(1)–P(2) 94.93(14), O(2)–Ru(1)–P(2) 92.76(15), P(3)–Ru(1)–P(2) 86.81(7), P(1)–Ru(1)–P(2) 88.66(7).

Compound **20**, crystallised from a dichloromethane solution of **15** in the monoclinic space group $P2_1/n$. The structure of **20** is analogous to that of **11** in that the two acetato ligands are again bent towards the coordinated water molecule to allow for stabilisation through hydrogen bonding (Figure S7). Not surprisingly, the Ru(1)–O(1) and Ru(1)–O(3) bond lengths [2.1614(15) Å and 2.1510(15) Å] are slightly shorter than those of the bidentate acetato ligand in **11** [2.167(4) Å and 2.209(4) Å], where slight bond elongation is brought about by ring strain. Similar to complexes **11**, **12** and **19**, the constraints of the tridentate triphos ligand together with steric congestion around the Ru(II) centre causes all angles to deviate significantly from ideality with the largest divergence observed for the angle O(3)–Ru(1)–O(1) [80.86(6)°].

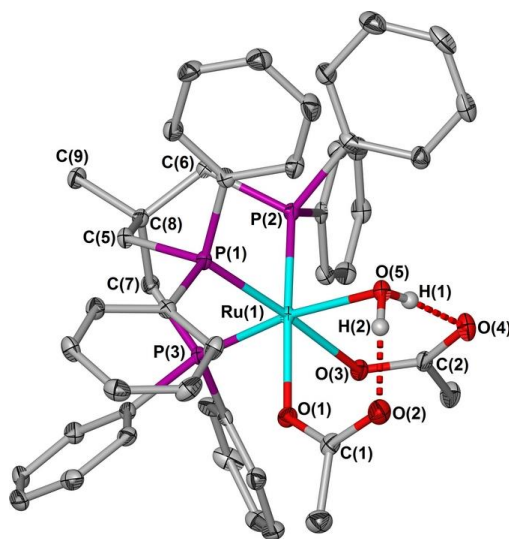


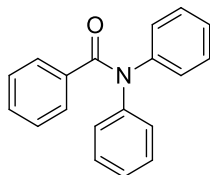
Figure S7 Molecular structure of **20** with thermal ellipsoids set at 50% probability. Hydrogen atoms (with the exception of those associated with water) are omitted for clarity. Selected bond lengths (Å) and angles(°): Ru(1)–O(1) 2.1614(15), Ru(1)–O(3) 2.1510(15), Ru(1)–O(5) 2.1988(15), Ru(1)–P(1) 2.2844(6), Ru(1)–P(2) 2.2776(7), Ru(1)–P(3) 2.2647(6), O(3)–Ru(1)–O(1) 80.86(6), O(3)–Ru(1)–O(5) 85.00(6), O(1)–Ru(1)–O(5) 86.63(6), O(3)–Ru(1)–P(3) 87.82(4), O(1)–Ru(1)–P(3) 87.58(4), O(5)–Ru(1)–P(3) 171.42(4), O(3)–Ru(1)–P(2) 95.07(4), O(1)–Ru(1)–P(2) 175.80(4), O(5)–Ru(1)–P(2) 94.11(5), P(3)–Ru(1)–P(2) 91.21(2), O(3)–Ru(1)–P(1) 175.35(4), O(1)–Ru(1)–P(1) 98.90(4), O(5)–Ru(1)–P(1) 99.62(5), P(3)–Ru(1)–P(1) 87.53(2), P(2)–Ru(1)–P(1) 85.07(2).

■ SECTION S-3: ANALYTICAL DATA FOR AMIDE SUBSTRATES PREPARED IN HOUSE

General procedure for the synthesis of amides

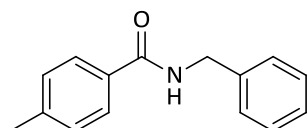
A solution of the corresponding amine (22 mmol) and triethylamine (22 mmol) in dichloromethane (50 ml) was cooled to 0 °C with an ice bath. The appropriate acid chloride (22 mmol) was added dropwise to the mixture over a period of 10 min at 0 °C. The solution was then allowed to warm to room temperature and stirred overnight. The mixture was then washed with a 2 M solution of HCl (3 × 50 ml) and a saturate NaHCO₃ solution (3 × 50 ml). The organic phase was subsequently dried over anhydrous MgSO₃ and the solvent removed *in vacuo* to give the crude product. Amides were further purified by recrystallisation, typically from hot toluene.³

N,N-diphenylbenzamide⁴



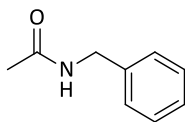
¹H NMR (CDCl₃, 400 MHz): δ = 7.38 (2H, m, CH), 7.22 (5H, m, CH), 7.07-7.15 (8H, m, CH); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ = 170.6 (C=O), 143.9 (C-N), 136.1 (C-C=O), 130.1 (CH), 129.2 (CH), 129.1 (CH), 127.9 (CH), 127.5 (CH), 126.3 (CH); GCMS: 273, 180, 167, 105, 77, 51; Elemental analysis: Found: C, 83.27; N, 5.33; H, 5.30. Calc. for C₁₉H₁₅NO: C, 83.49; N, 5.12; H, 5.53%.

N-benzyl-4-methylbenzamide³



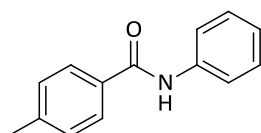
¹H NMR (CDCl₃, 500 MHz): δ = 7.71 (2H, d, ³J_{HH} = 8.1 Hz, CH), 7.37 (4H, d, J_{HH} = 4.4 Hz, CH), 7.32 (1H, m, CH), 7.24 (2H, d, ³J_{HH} = 7.9 Hz, CH), 6.47 (1H, s, NH), 4.65 (2H, d, J_{HH} = 5.6 Hz, CH₂), 2.42 (3H, s, CH₃); ¹³C{¹H} NMR (CDCl₃, 75 MHz): δ = 167.4 (C=O), 142.0 (CMe), 138.4 (CCH₂), 131.5 (CC=O), 129.3 (CH), 128.8 (CH), 127.9 (CH), 127.6 (CH), 127.0 (CH), 44.1 (CH₂), 21.5 (CH₃); GCMS: 225, 119, 106, 91; Elemental analysis: Found: C, 79.65; N, 6.33; H, 6.62. Calc. for C₁₅H₁₅NO: C, 79.97; N, 6.22; H, 6.71%.

***N*-benzylacetamide³**



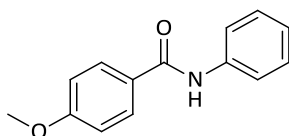
¹H NMR (CDCl₃, 400 MHz): δ = 7.33 (2H, m, CH), 7.27 (3H, m, CH), 5.88 (1H, s, NH), 4.42 (2H, d, ²*J*_{HH} = 5.7 Hz, CH₂), 2.01 (3H, s, CH₃); ¹H NMR ((CD₃)₂SO, 400 MHz): δ = 8.34 (1H, s, NH), 7.30 (2H, m, CH), 7.22 (3H, m, CH), 4.24 (2H, d, ²*J*_{HH} = 6.0 Hz, CH₂), 1.86 (3H, s, CH₃); ¹³C{¹H} NMR ((CD₃)₂SO, 100 MHz): δ = 169.1 (C=O), 139.5 (CCH₂), 128.2 (CH), 127.2 (CH), 126.7 (CH), 42.0 (CH₂), 22.5 (CH₃); Elemental analysis: Found: C, 72.88; N, 9.63; H 7.06,. Calc. for C₉H₁₁NO: C, 72.46; N, 9.39; H, 7.43%.

4-Methylbenzanilide³



¹H NMR (CDCl₃, 300 MHz): δ = 7.81 (1H, s, NH), 7.77 (2H, d, ³*J*_{HH} = 8.2 Hz, CH), 7.64 (2H, dd, ³*J*_{HH} = 8.5, ⁴*J*_{HH} = 1.0 Hz, CH), 7.37 (2H, t, *J*_{HH} = 8.0 Hz, CH), 7.28 (2H, d, ³*J*_{HH} = 7.9 Hz), 7.14 (1H, tt, ³*J*_{HH} = 7.4, ⁴*J*_{HH} = 1.1 Hz, CH), 2.43 (3H, s, CH₃); ¹H NMR ((CD₃)₂SO, 400 MHz): δ = 10.14 (1H, s, NH), 7.86 (2H, d, ³*J*_{HH} = 8.2 Hz, CH), 7.76 (2H, d, ³*J*_{HH} = 7.6 Hz, CH), 7.33 (4H, m, CH), 7.08 (1H, tt, ³*J*_{HH} = 7.4 Hz, ⁴*J*_{HH} = 1.1 Hz, CH), 2.37 (3H, s, CH₃); ¹³C{¹H} NMR ((CD₃)₂SO, 100 MHz): δ = 165.3 (C=O), 141.5 (CMe), 139.2 (C-NH), 132.0 (C-C=O), 128.8 (CH), 128.5 (CH), 127.6 (CH), 123.5 (CH), 120.3 (CH), 21.0 (CH₃).

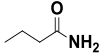
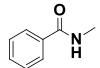
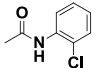
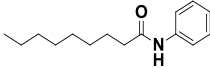
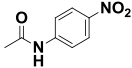
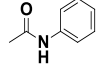
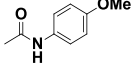
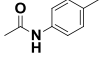
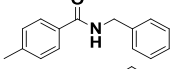
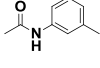
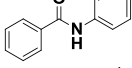
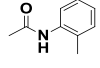
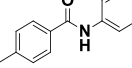
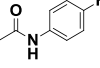
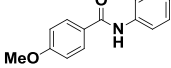
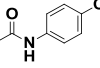
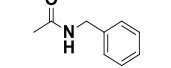
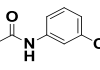
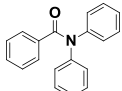
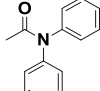
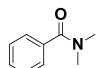
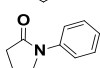
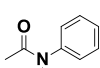

4-Methoxybenzanilide³



¹H NMR ((CD₃)₂CO, 300 MHz): δ = 10.09 (1H, s, NH), 7.97 (2H, dt, ³*J*_{HH} = 9.8 Hz, ⁴*J*_{HH} = 2.5 Hz, CH), 7.78 (2H, d, ³*J*_{HH} = 7.5 Hz, CH), 7.34 (2H, t, ³*J*_{HH} = 7.9 Hz, CH), 7.08 (3H, m, CH), 3.84 (3H, s, CH₃); ¹³C{¹H} NMR ((CD₃)₂CO, 100 MHz): δ = 220.8 (C=O), 161.8 (C-OMe), 139.3 (C-N), 129.5 (CH), 128.5 (CH), 126.9 (C-C=O), 123.4 (CH), 120.3 (CH), 113.5 (CH), 55.4 (CH₃); Elemental analysis: Found: C, 74.24; N, 6.52; H, 6.05. Calc. for C₉H₁₁NO: C, 73.99; N, 6.16; H, 5.77%.

■ SECTION S-4: GC-FID SPECTRA FOR THE POST RUN PRODUCT MIXTURES OBTAINED WITH SELECTED SUBSTRATES (LISTED IN TABLE S1).

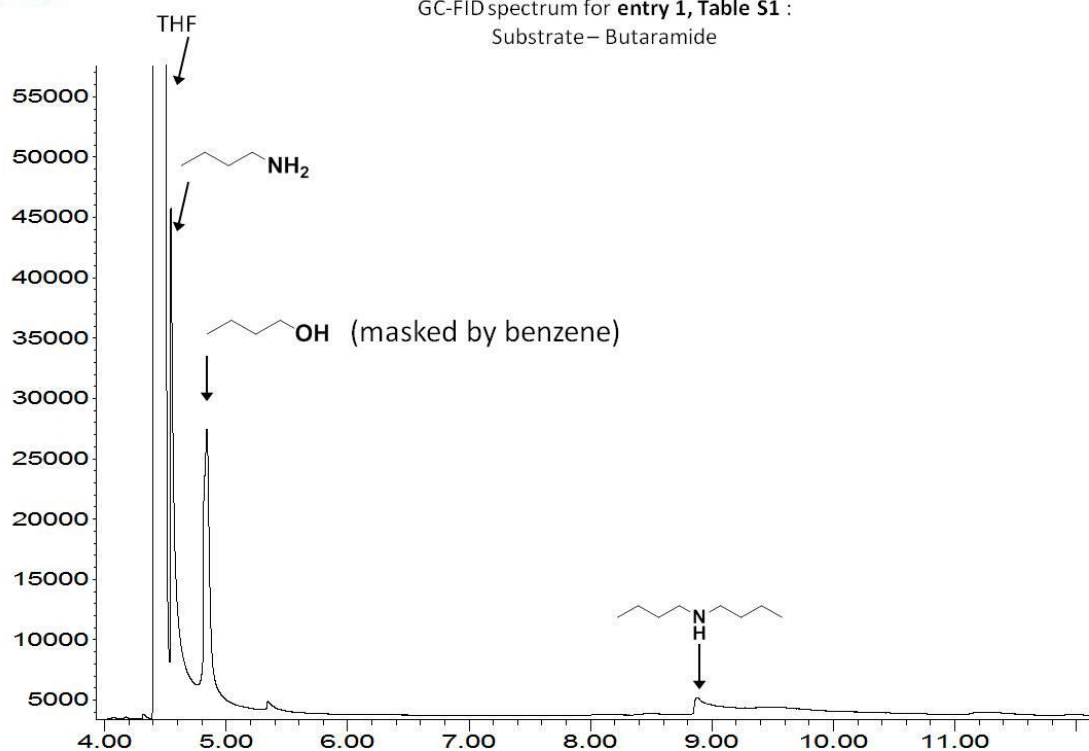
Table S1. Substrate scope for [Ru(acac)₃] / triphos catalysed amide hydrogenations.

Entry	Substrate	MSA (mol %)	Temp (° C)	P _{H₂} (bar)	Conv. (%)	Sel. (%)	Entry	Substrate	MSA (mol %)	Temp (° C)	P _{H₂} (bar)	Conv. (%)	Sel. (%)
<i>Primary amide</i> ^[b]													
1		1.5	200	10	100	61							
<i>Secondary amides</i>													
2		1.5	200	10	82 ^[c]	<5 ^[c]	11		1.5	200	10	45	0
3		1.5	200	10	92 ^[d]	92 ^[d]	12		1.5	200	10	100	0
4		1.5	200	10	100 ^[e]	92 ^[e]	13		1.0	220	10	97	78
5		1.0	220	10	98	78	14		1.5	200	10	15 ^[f]	<5 ^[f]
6		1.5	200	10	100	79	15		1.0	220	10	100	92
7		1.5	200	10	100	90	16		1.5	200	10	100	94
8		1.0	220	10	99	77	17		1.5	200	10	92 ^[c]	61 ^[c]
9		1.5	200	10	64	28	18		1.5	220	10	100	<5
10		1.5	200	10	75	75							
<i>Tertiary amides</i>													
19		1.5	220	10	92	73	22		1.0	220	40	83	42
20		1.5	220	10	33 ^[c]	7 ^[c]	23		1.5	200	10	19	100
21		1.5	200	10	19	63	24		1.5	200	10	0	0

[a] Conditions: Substrate (5 mmol), [Ru(acac)₃] (1 mol %), triphos (2 mol %), thf (10 ml), 16 h. Conv. and Sel. calculated using NMR integration. [b] Reaction performed in the presence of NH₃(aq) (10 ml). [c] Based on uncalibrated GC-FID integration. [d] Conv. 86%, Sel. 78% in Lit.⁵ [e] The Sel. for different reactions with *N*-phenylacetamide under identical conditions varies between 86-92 %. In the presence of added mercury, 100% Conv. with 84 % Sel. was obtained. [f] Small amounts of mixed tertiary amines formed.

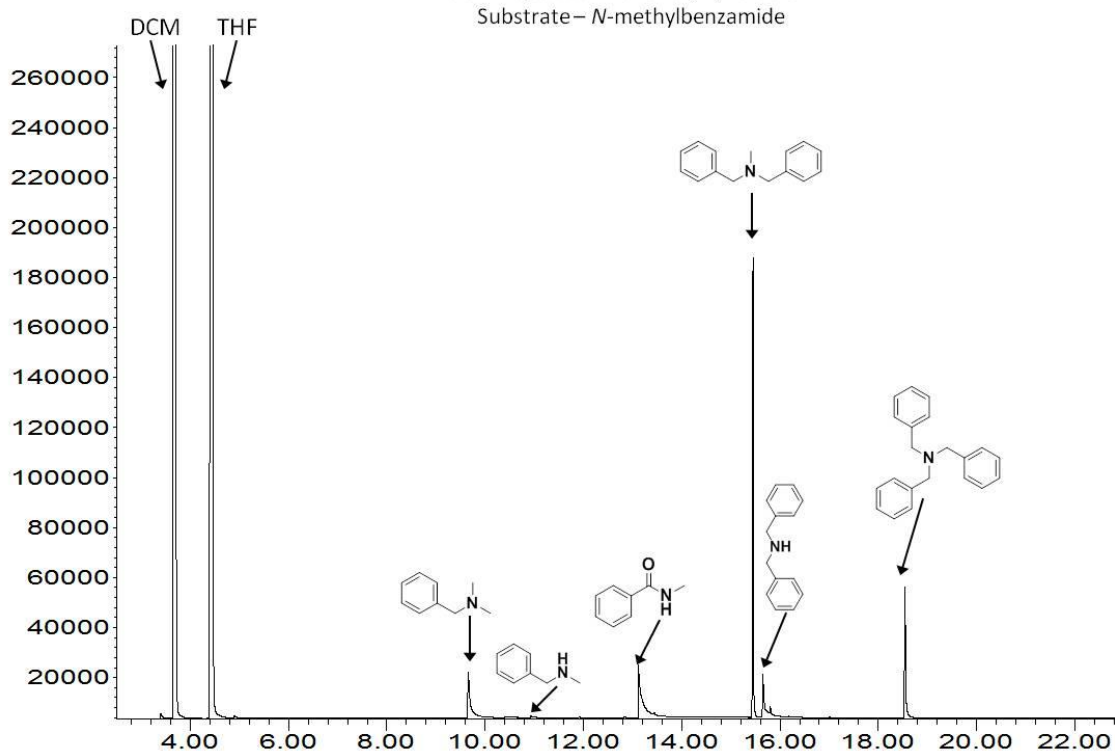
Response_

GC-FID spectrum for **entry 1, Table S1** :
Substrate – Butaramide



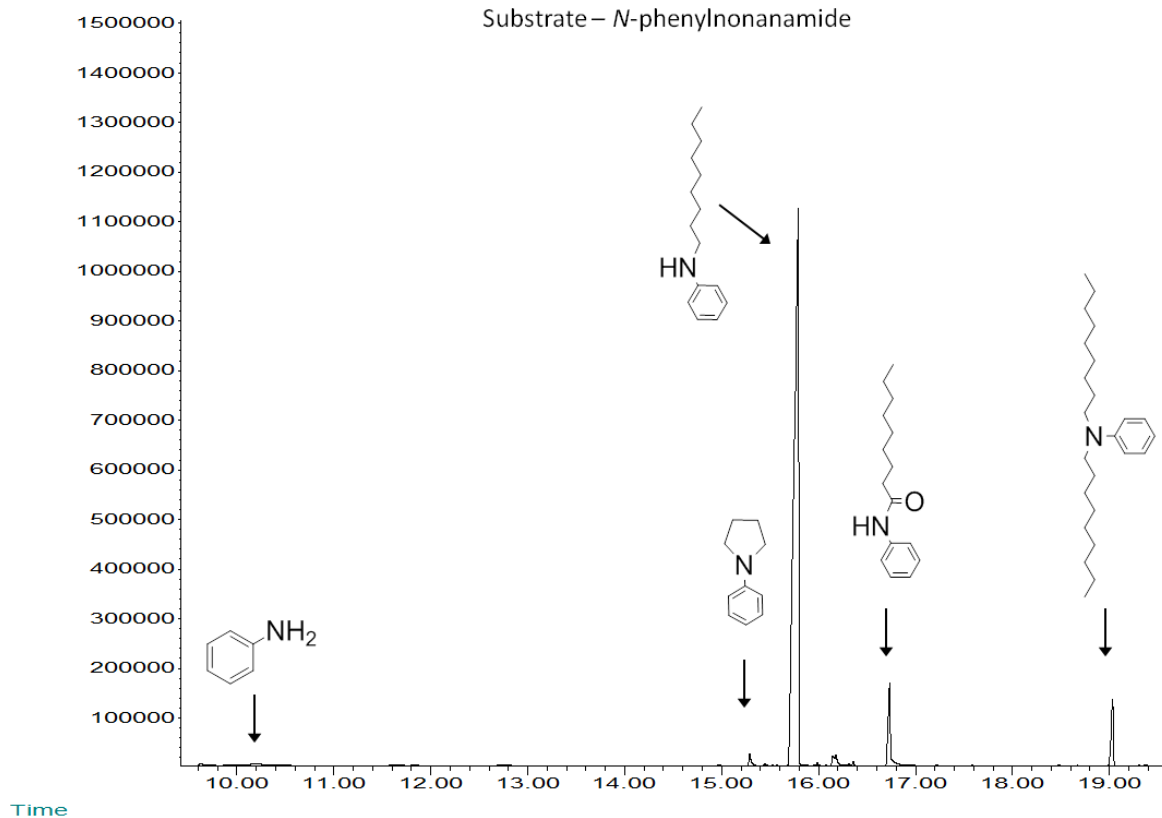
Response_

GC-FID spectrum for **entry 2, Table S1**:
Substrate – *N*-methylbenzamide



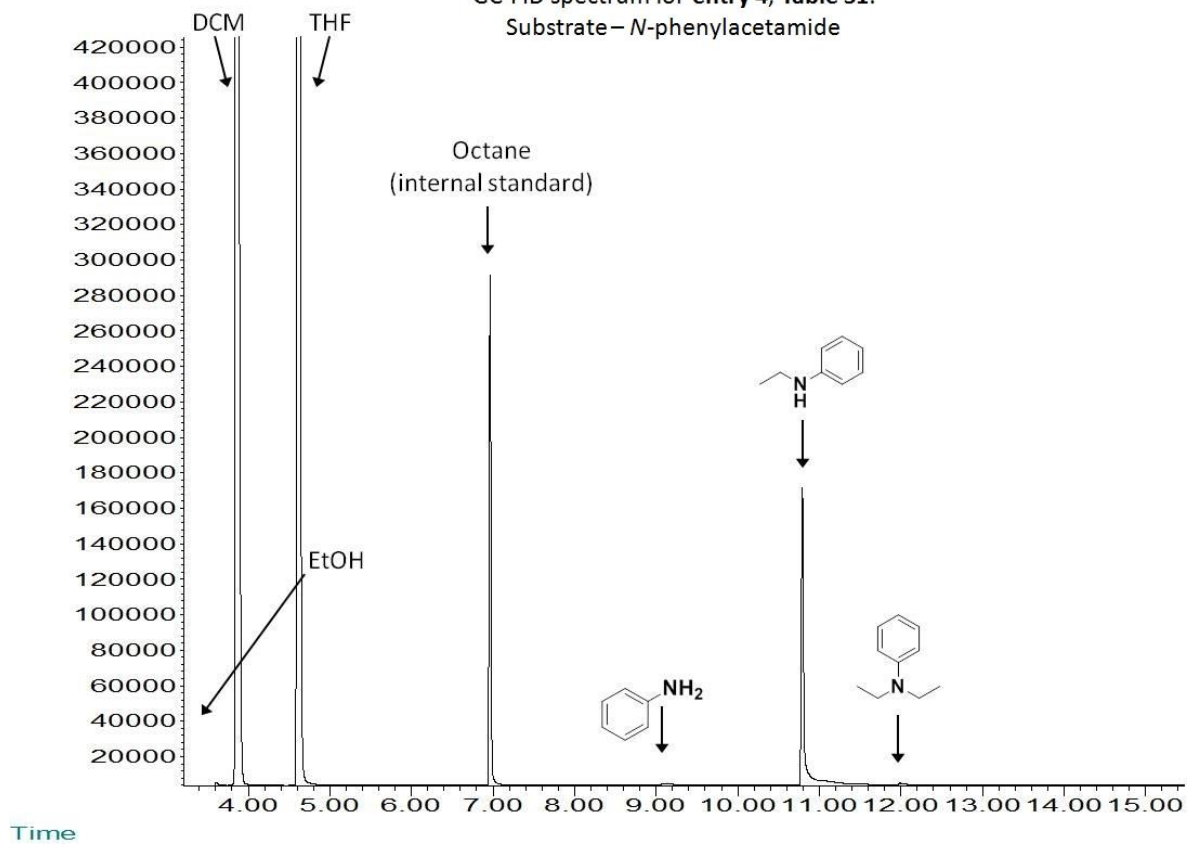
Response_

GC-FID spectrum for **entry 3, Table S1:**
Substrate – *N*-phenylnonanamide



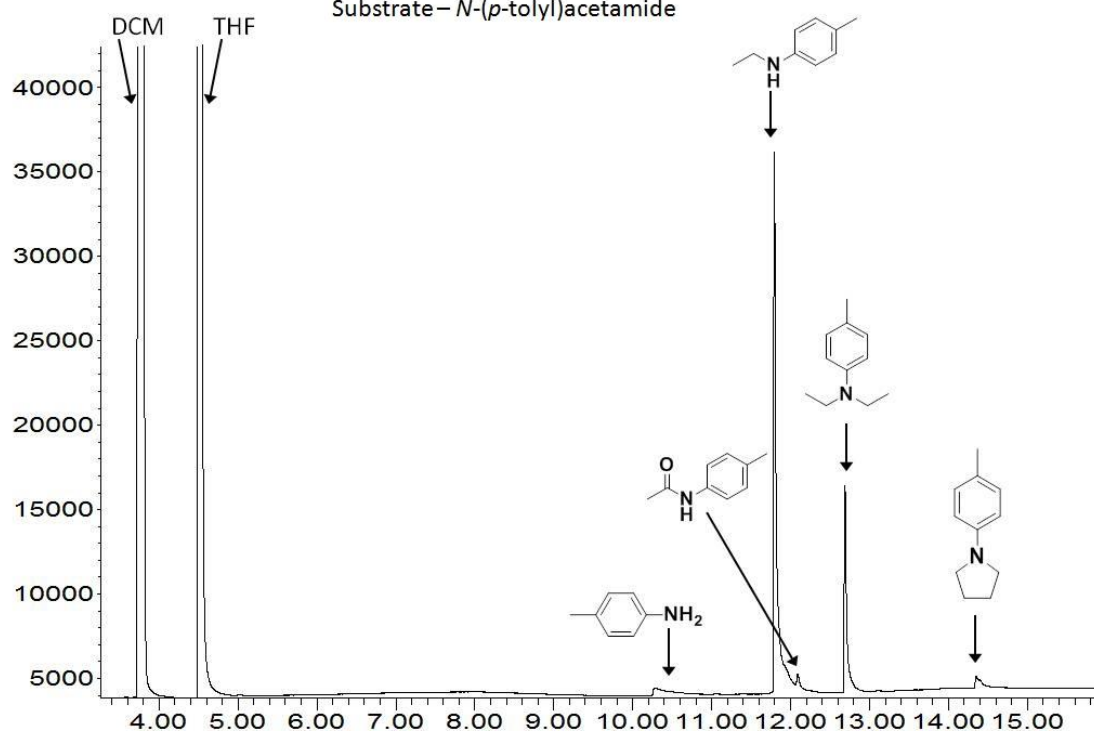
Response_

GC-FID spectrum for **entry 4, Table S1:**
Substrate – *N*-phenylacetamide



Response_

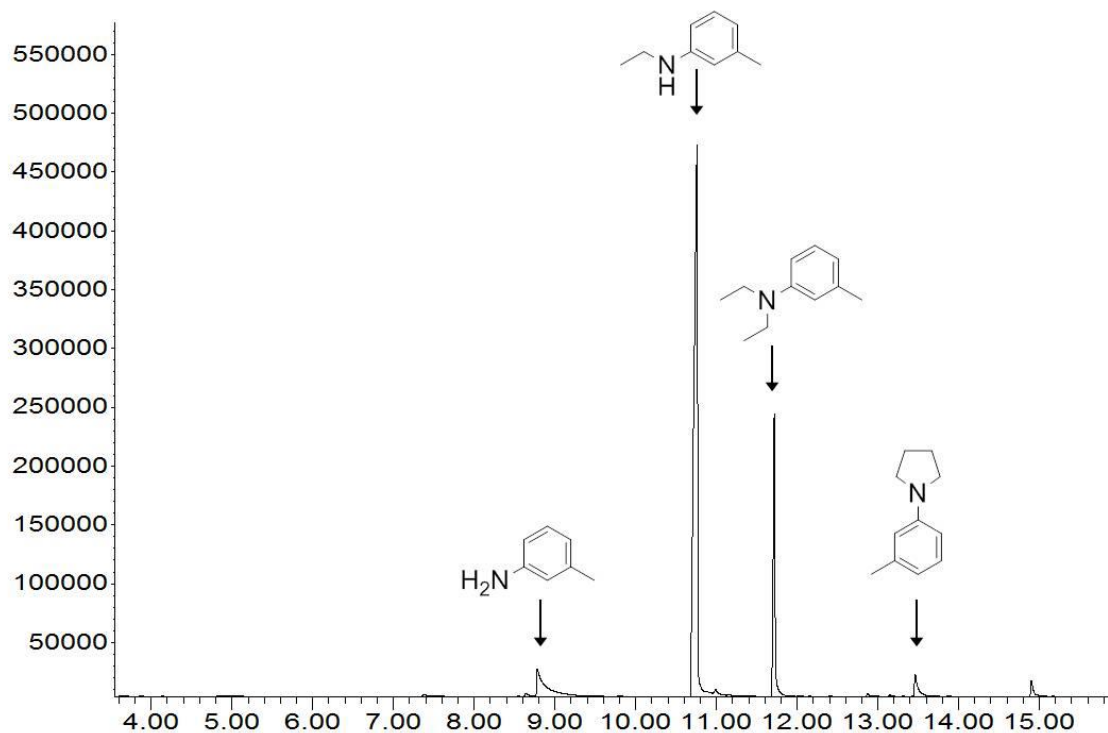
GC-FID spectrum for **entry 5, Table S1**:
Substrate – *N*-(*p*-tolyl)acetamide



Time

Response_

GC-FID spectrum for **entry 6, Table S1**:
Substrate – *N*-(*m*-tolyl)acetamide

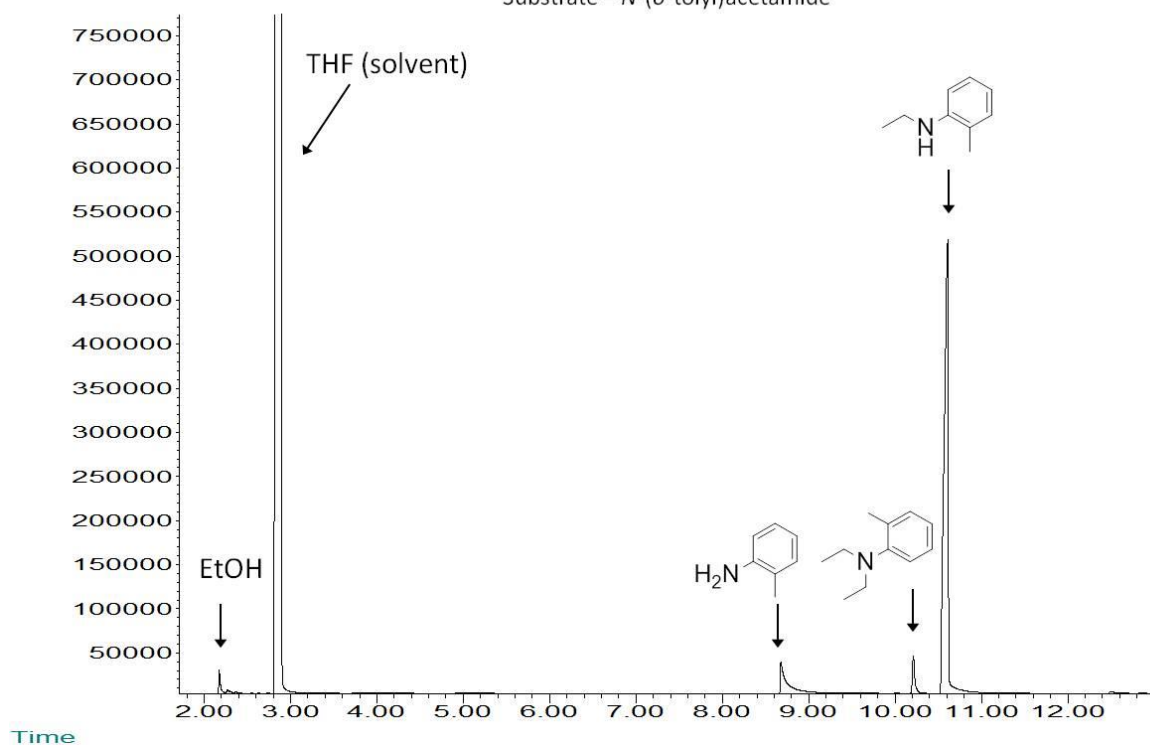


Time

Response_

GC-FID spectrum for **entry 7, Table S1**:

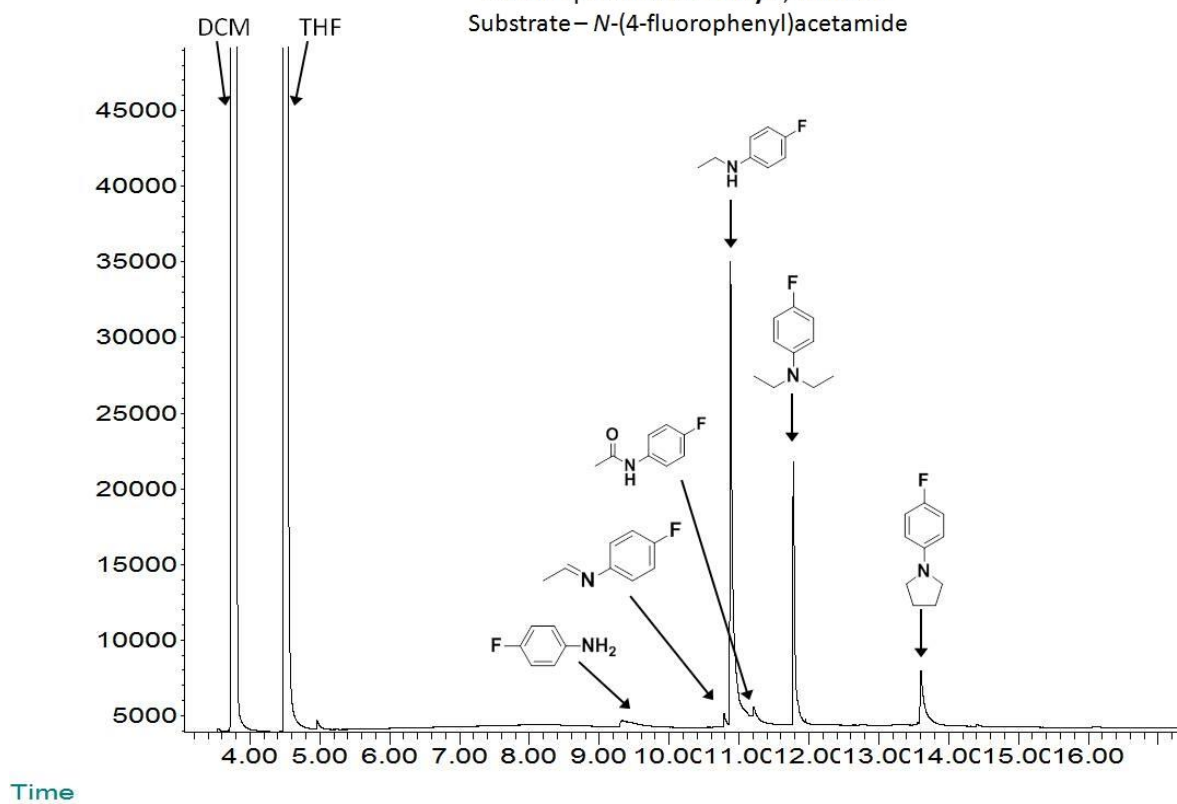
Substrate – *N*-(*o*-tolyl)acetamide



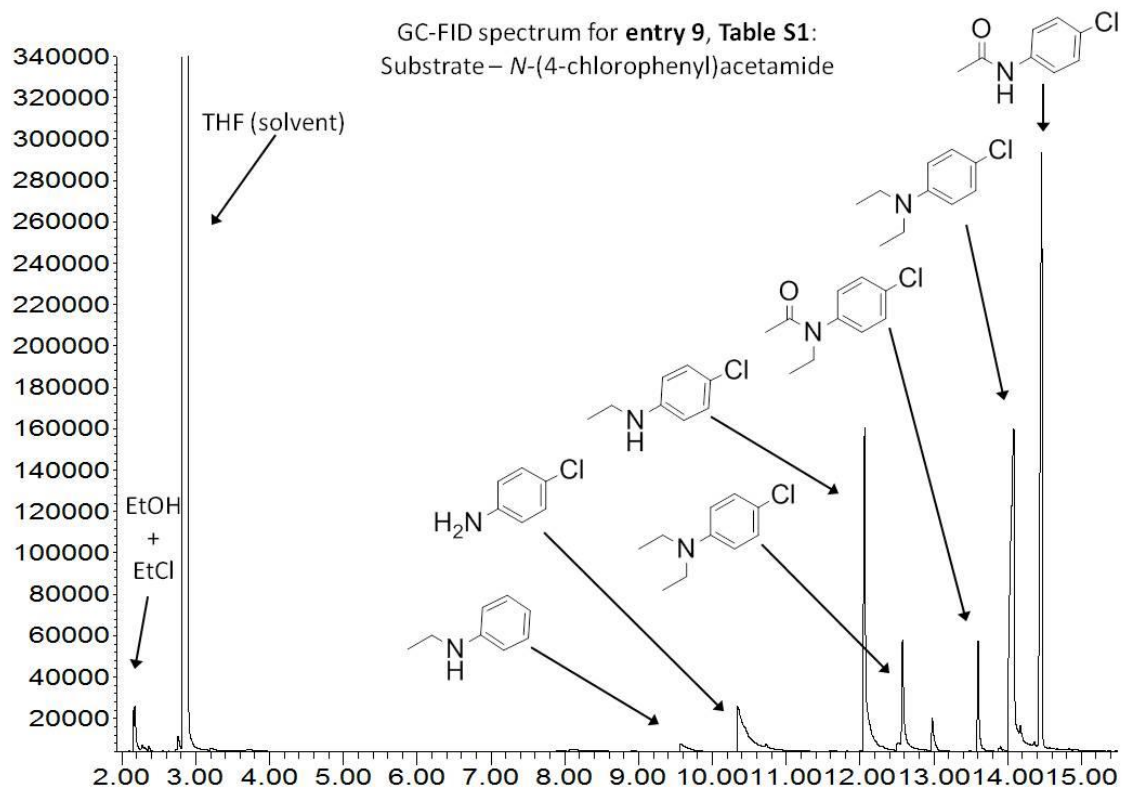
Response_

GC-FID spectrum for **entry 8, Table S1**:

Substrate – *N*-(4-fluorophenyl)acetamide

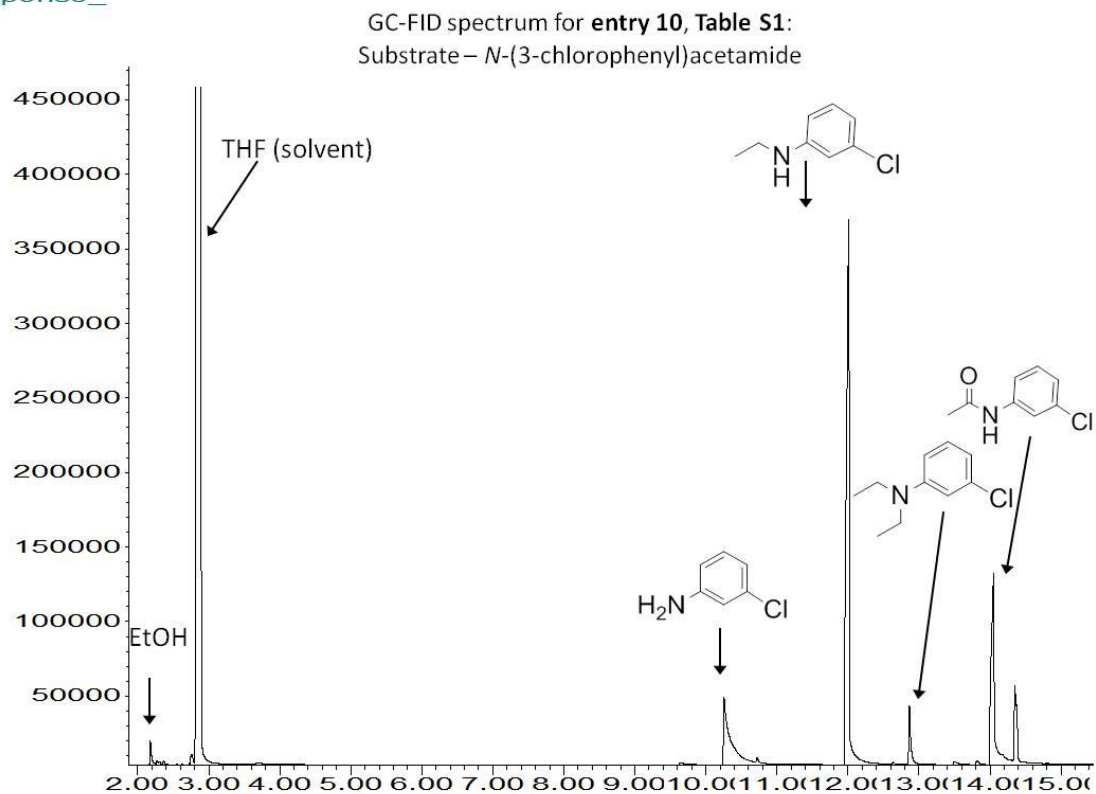


Response_



Time

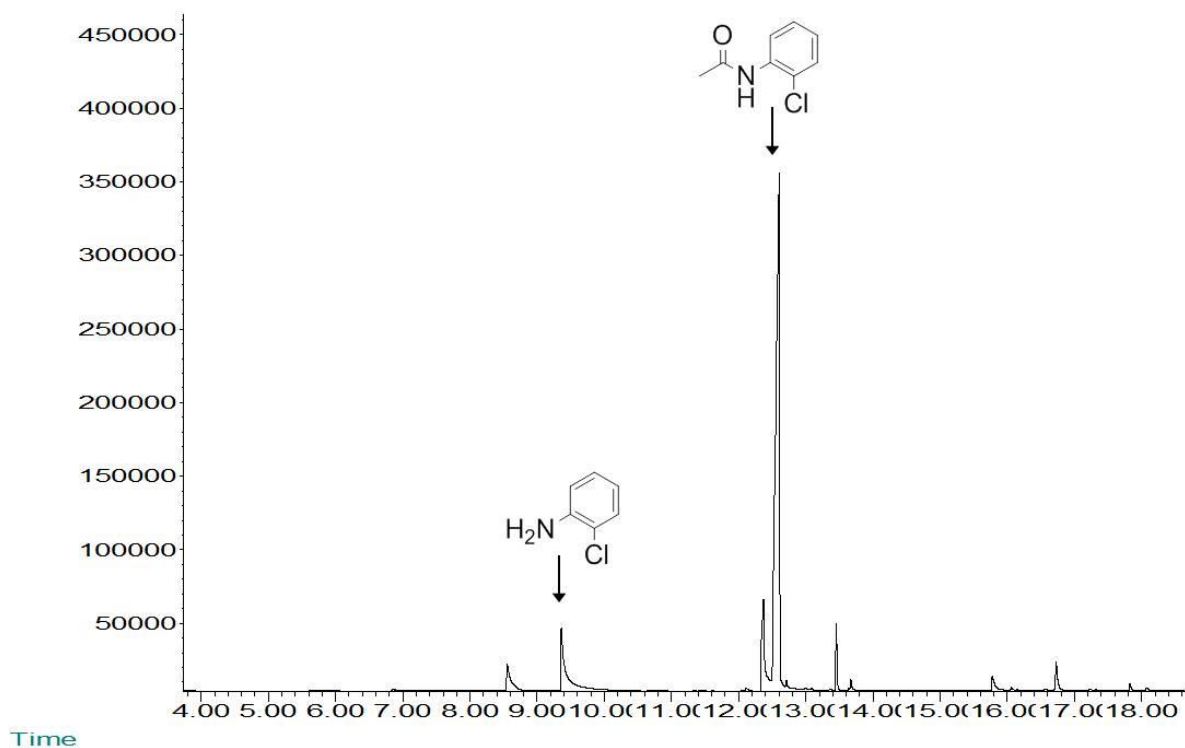
Response_



Time

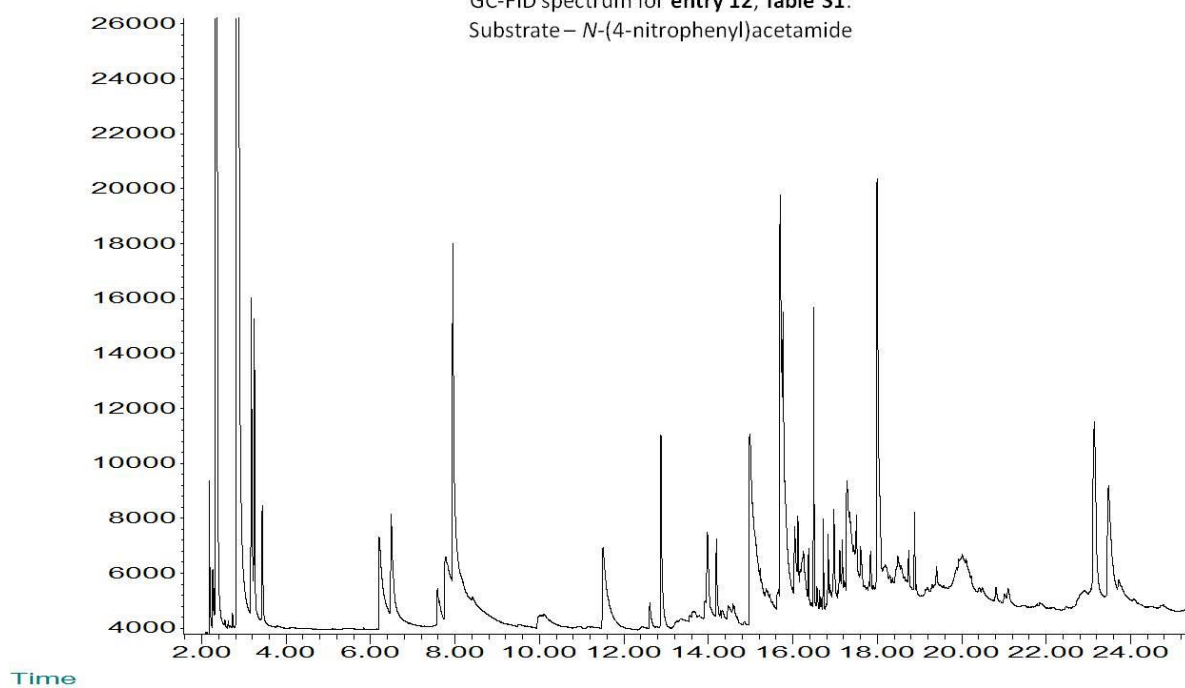
Response_

GC-FID spectrum for **entry 11, Table S1**:
Substrate – *N*-(2-chlorophenyl)acetamide



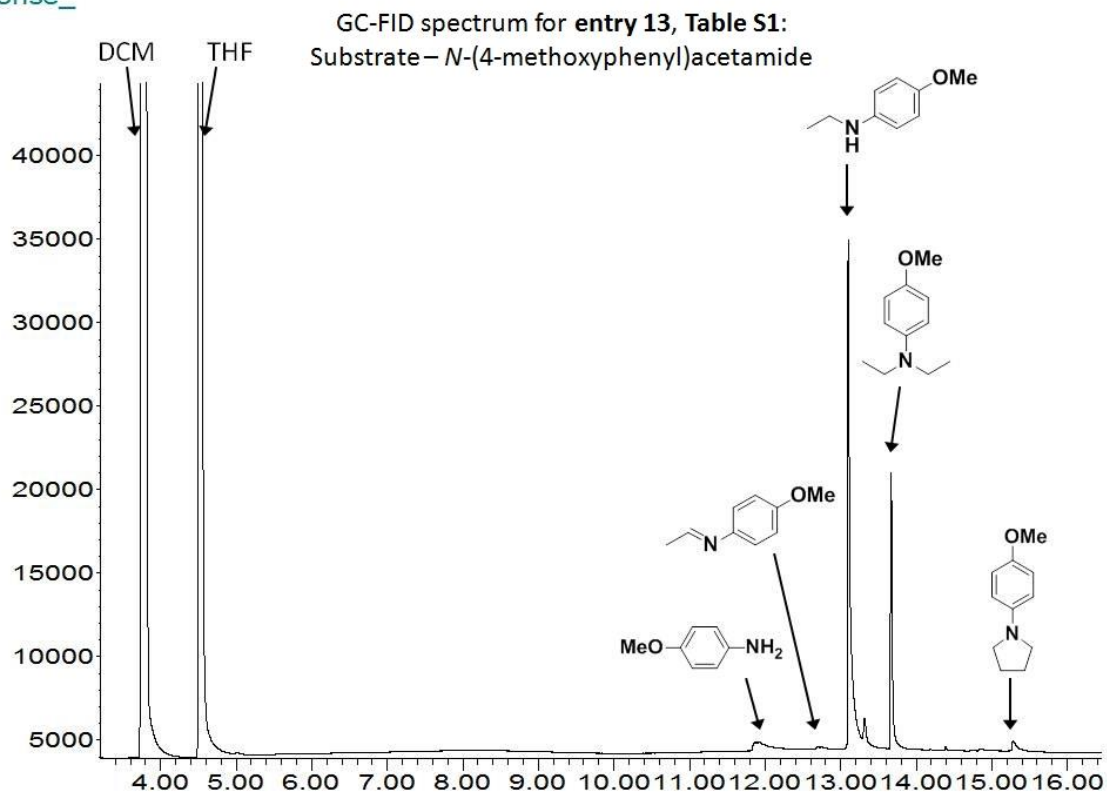
Response_

GC-FID spectrum for **entry 12, Table S1**:
Substrate – *N*-(4-nitrophenyl)acetamide

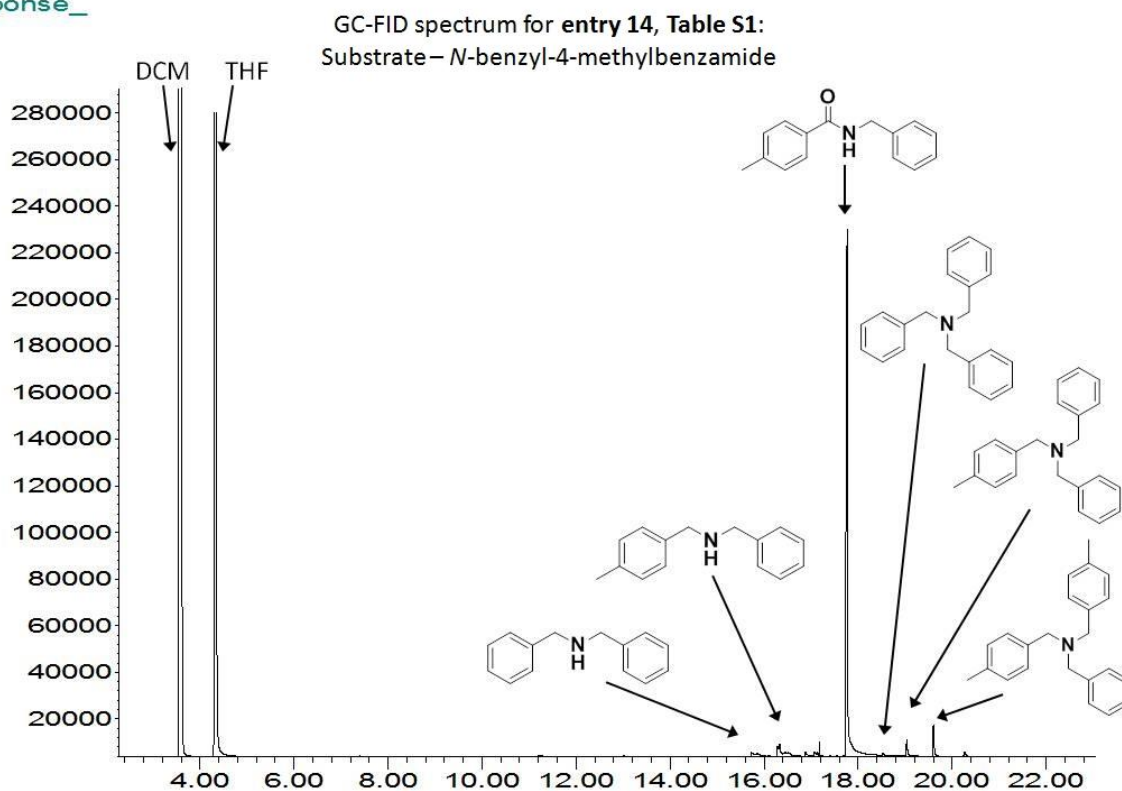


Time

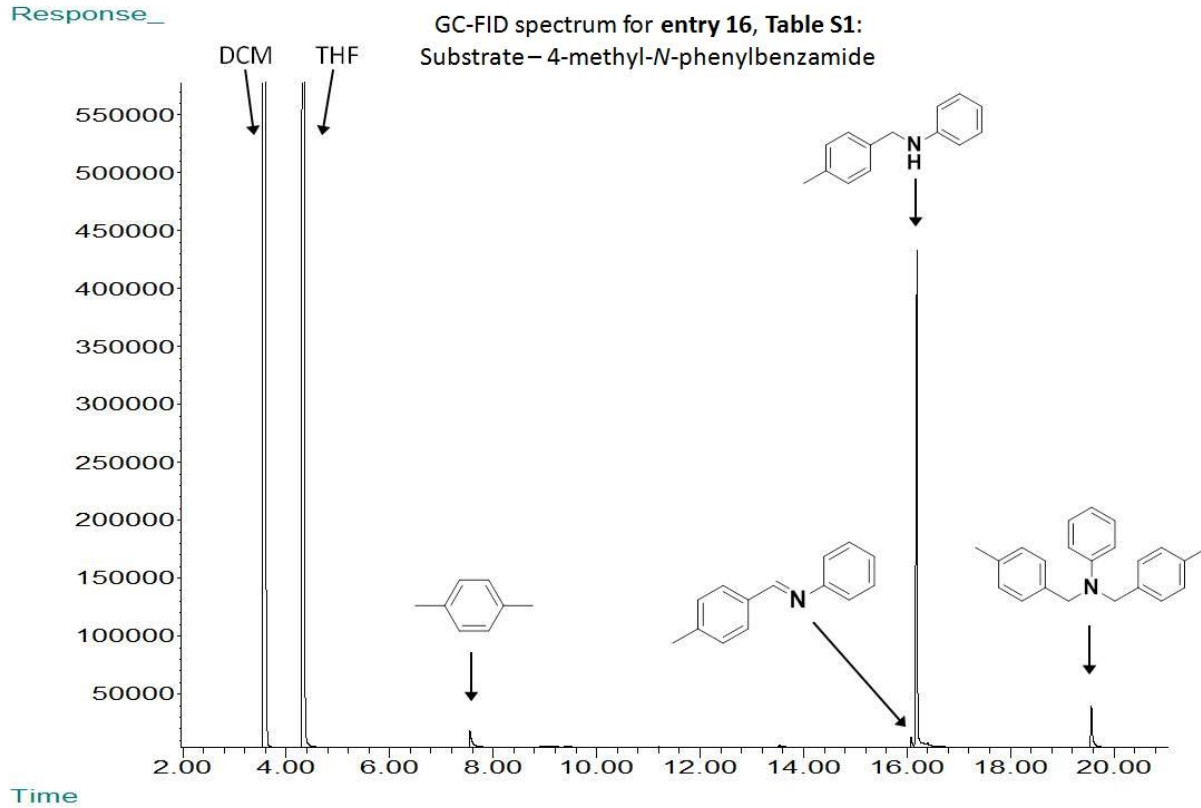
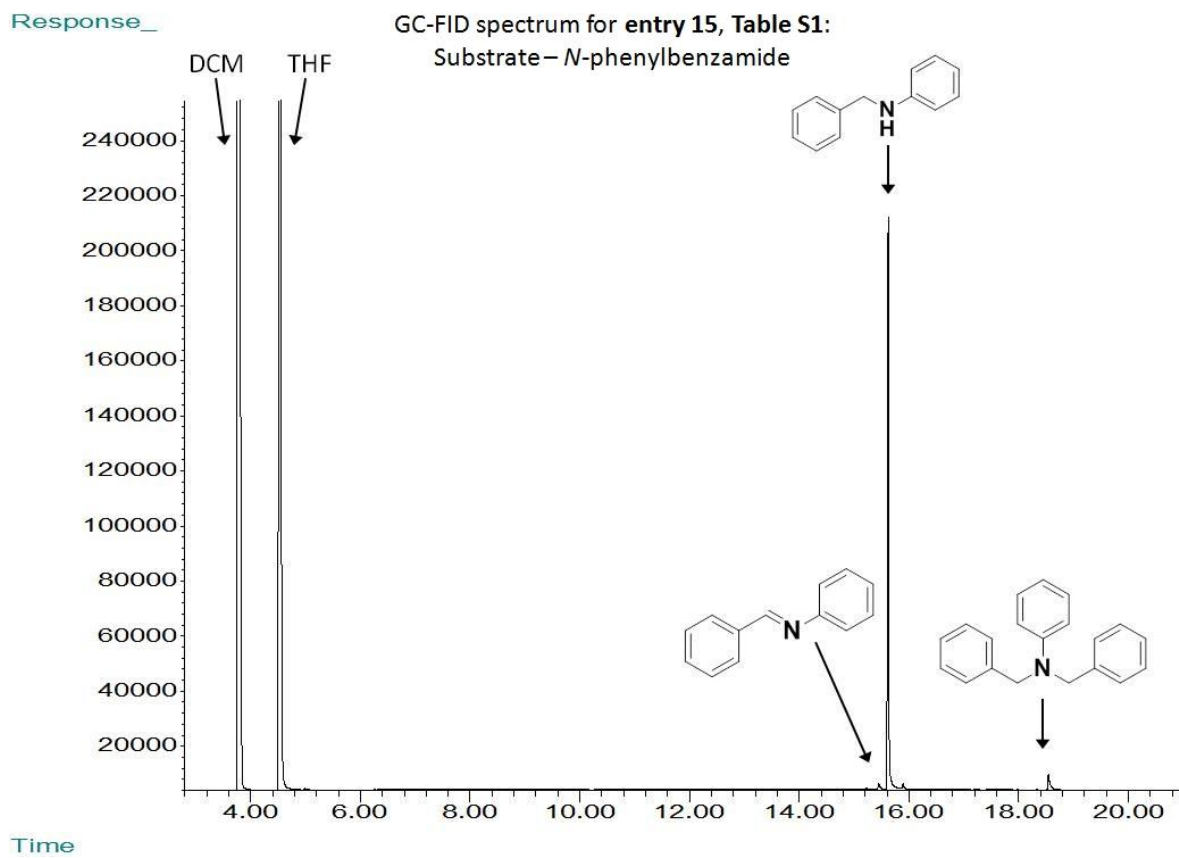
Response_

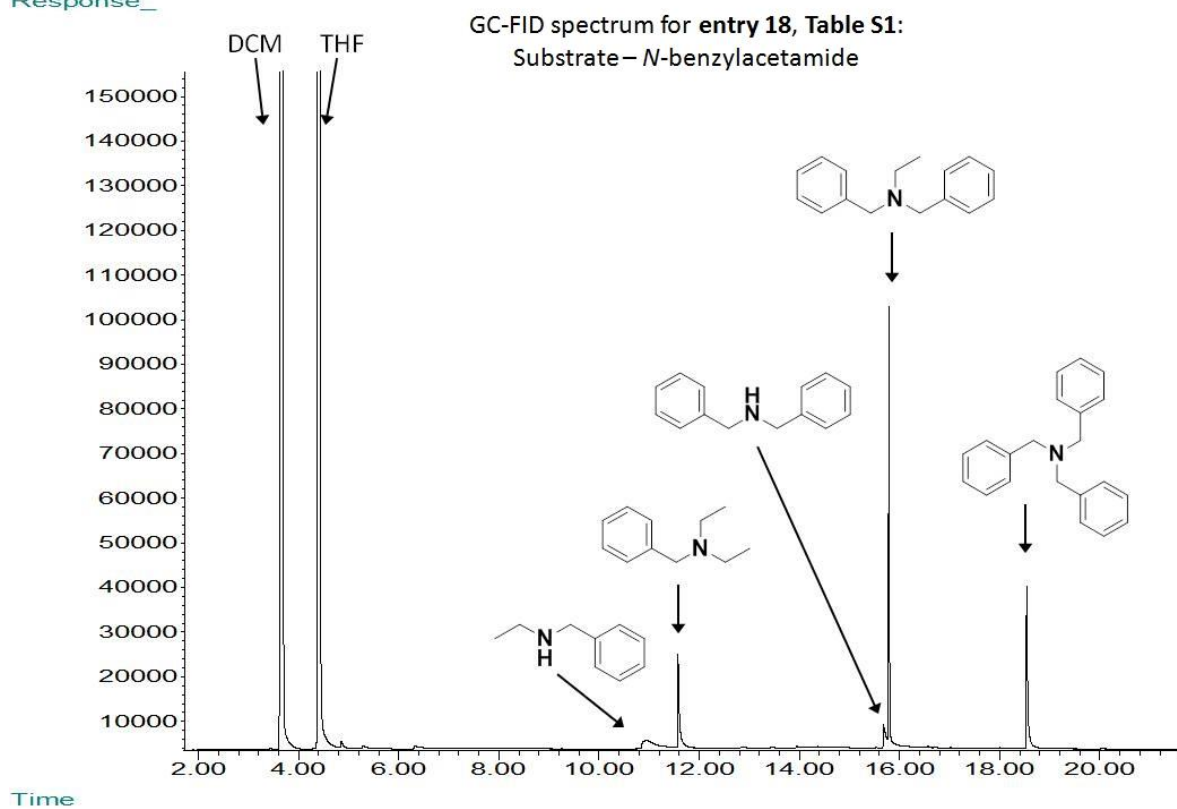
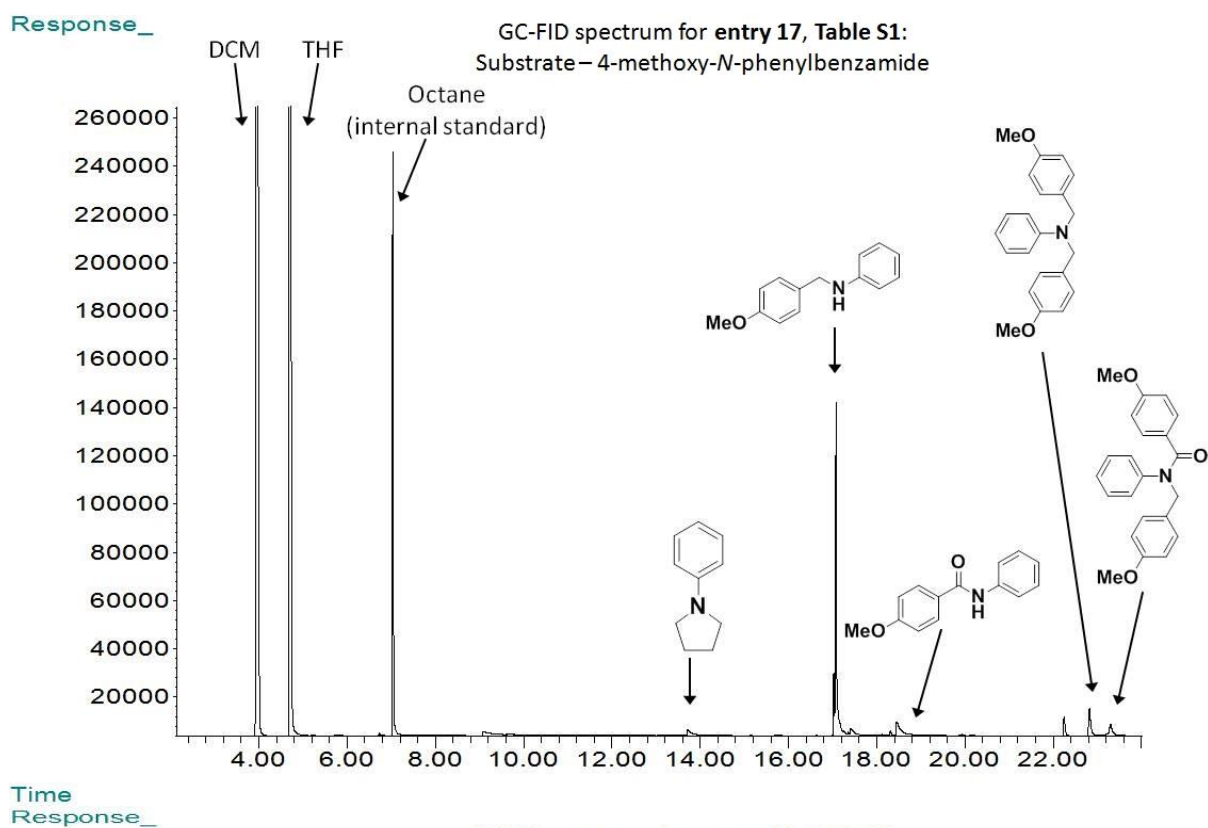


Time
Response_

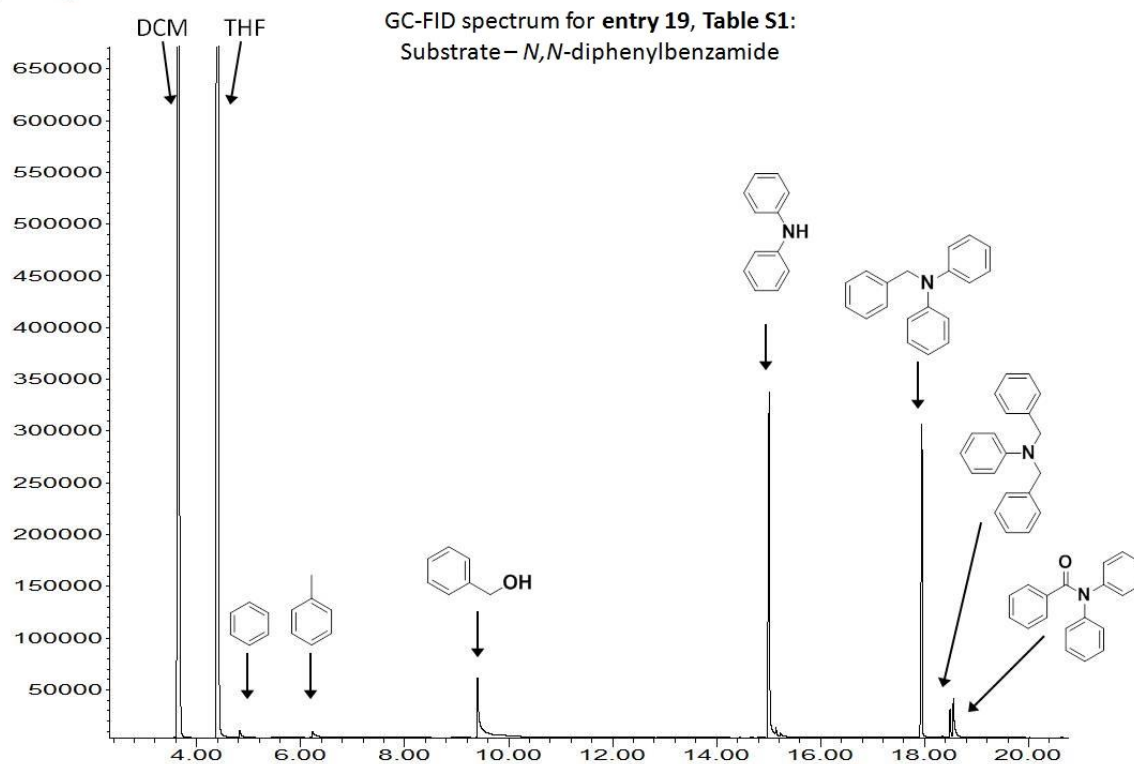


Time

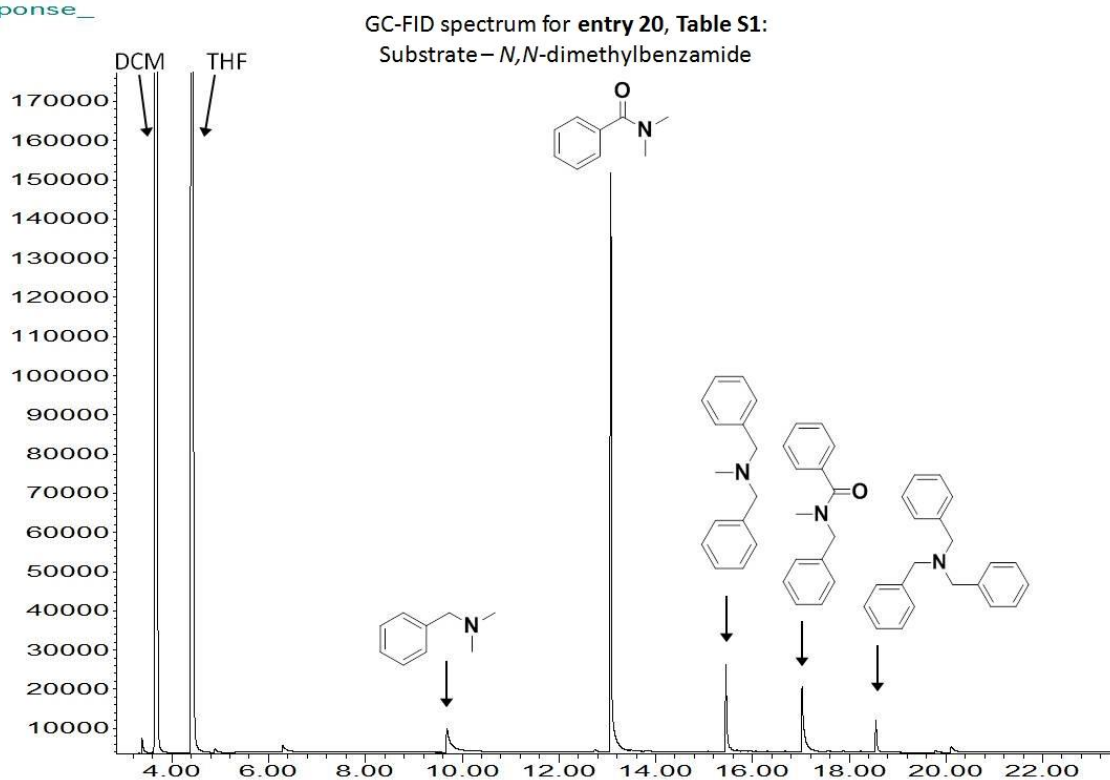




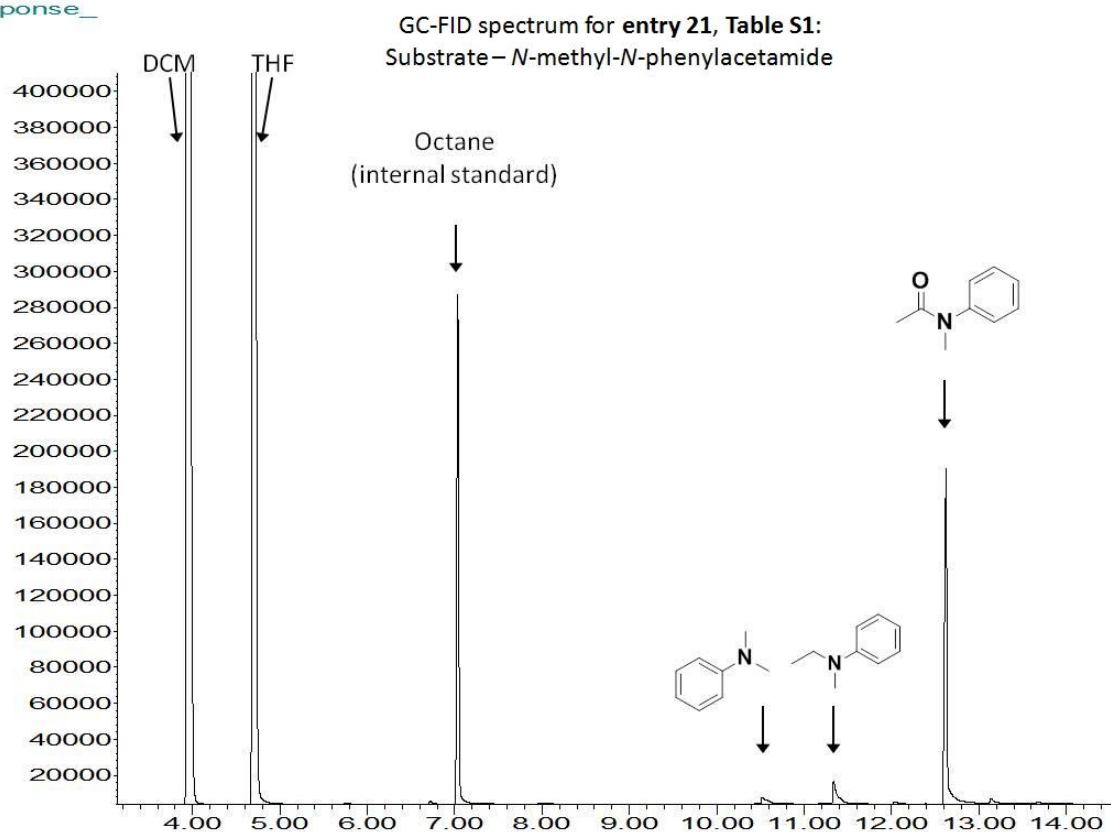
Response_



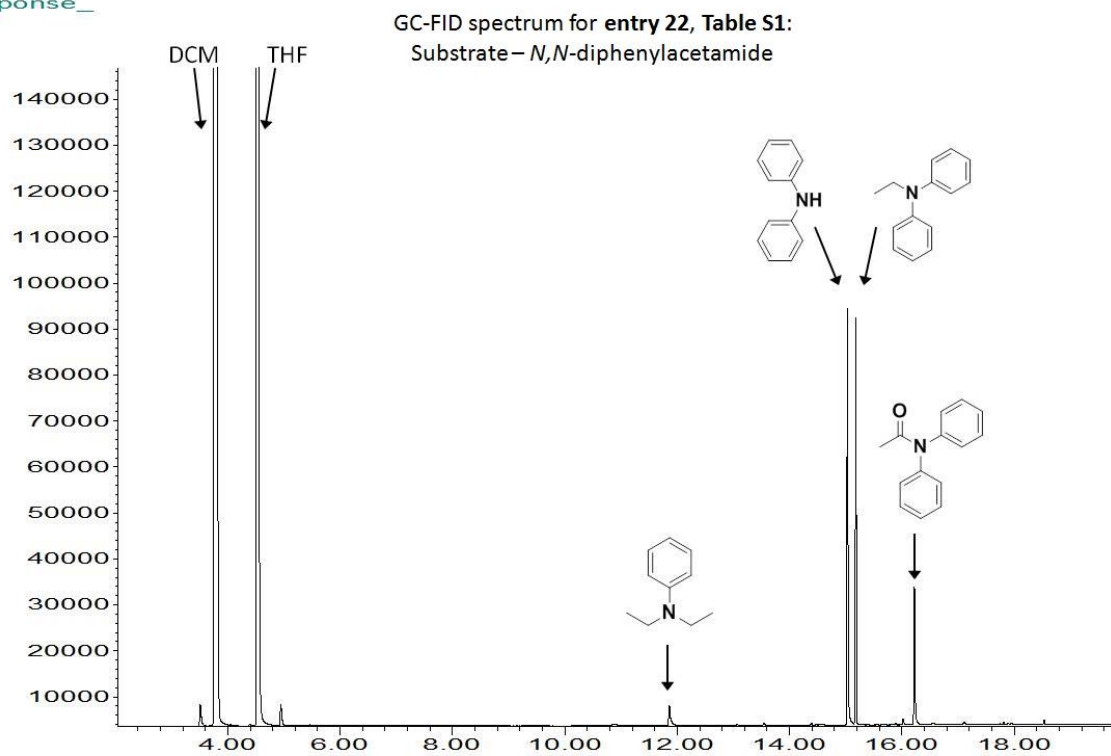
Time
Response_



Response_



Time
Response_

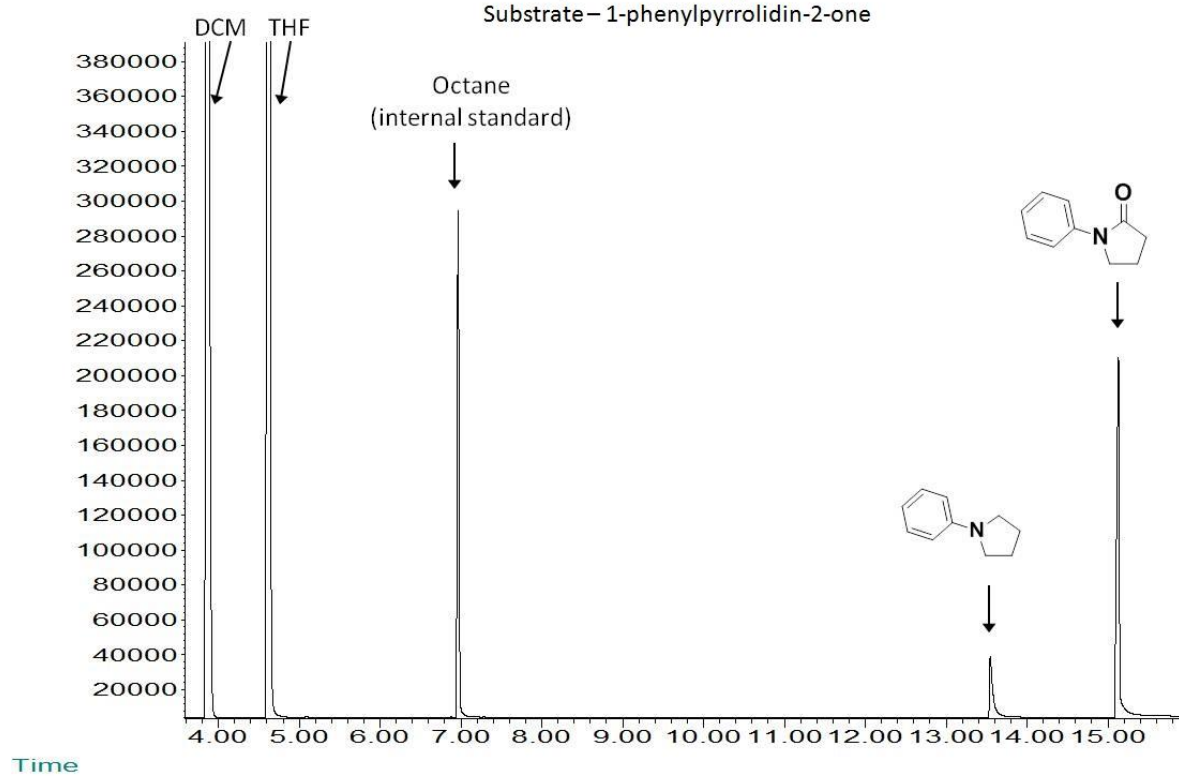


Time

Response_

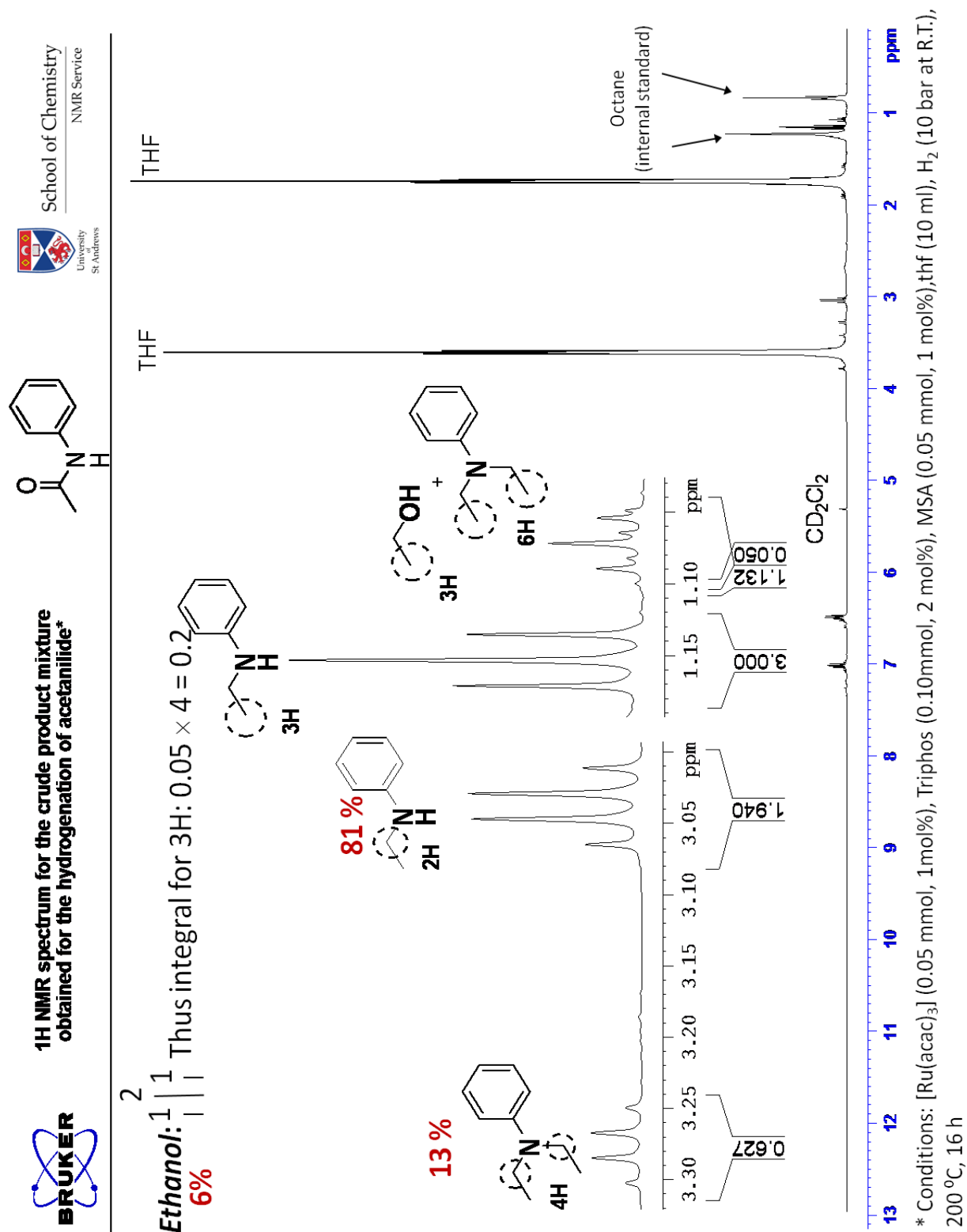
GC-FID spectrum for entry 23, Table S1:

Substrate– 1-phenylpyrrolidin-2-one



■ SECTION S-5: EXAMPLE OF QUANTIFICATION BASED ON ^1H NMR INTEGRATION

[example substrate – *N*-phenylacetamide (trivial name - acetanilide)]



■ SECTION S-6: CRYSTALLOGRAPHIC DATA TABLES

Table S2. Crystallographic data for compounds 11 and 12.

Parameter	Compound 11	Compound 12
Chemical formula	C ₄₅ H ₄₅ O ₄ P ₃ Ru	C ₈₆ H ₉₀ O ₁₂ P ₆ Ru ₂ S ₄
Formula weight (g/mol)	843.83	1831.87
Temp. (K)	93(2)	93(2)
Wavelength (Å)	0.71075	0.71075
Crystal system	Monoclinic	Triclinic
Crystal dimensions (mm ³)	0.20 × 0.20 × 0.20	0.03 × 0.03 × 0.03
Crystal shape and colour	Prism, yellow	Prism, yellow
Space group	<i>P</i> 2 ₁ / <i>c</i> (No. 14)	<i>P</i> $\bar{1}$ (No. 2)
<i>a</i> (Å)	10.057(2)	12.5217(5)
<i>b</i> (Å)	15.075(3)	12.6970(5)
<i>c</i> (Å)	29.442(6)	31.500(3)
α (°)	90.0000	78.307(5)
β (°)	96.663(5)	81.575(6)
γ (°)	90.0000	68.302(5)
Unit cell volume (Å ³)	4433.7(15)	4542.6(5)
No. of formula units per unit cell, <i>Z</i>	4	2
<i>d</i> _{calcd} (g/cm ³)	1.573	1.525
Radiation type	MoK α	MoK α
Absorption coefficient, μ (mm ⁻¹)	0.797	0.766
<i>F</i> (000)	2152	2140
θ -range for data collection (°)	2.04 to 25.35	3.06 to 25.41
Index range	-11 ≤ <i>h</i> ≤ 12, -18 ≤ <i>k</i> ≤ 14, -33 ≤ <i>l</i> ≤ 35	-10 ≤ <i>h</i> ≤ 10, -15 ≤ <i>k</i> ≤ 15, -37 ≤ <i>l</i> ≤ 37
No. of reflections measured	26754	46062
No. of unique reflections	8051	16487
Refinement parameters / restraints	562 / 0	1078 / 18
<i>R</i> _{int}	0.0943	0.0678
Final <i>R</i> ₁ values (<i>I</i> > 2σ(<i>I</i>))	0.0729	0.0864
Final <i>wR</i> (<i>F</i> ²) values (<i>I</i> > 2σ(<i>I</i>))	0.2099	0.2241
Final <i>R</i> ₁ values (all data)	0.0914	0.1126
Final <i>wR</i> (<i>F</i> ²) values (all data)	0.2393	0.2427
Goodness of fit on <i>F</i> ²	1.078	1.037
Largest diff. peak and hole (e.Å ⁻³)	1.500 and -2.250	7.640 and -2.560
Weighing scheme †	<i>a</i> = 0.1406 <i>b</i> = 8.3130	<i>a</i> = 0.1063 <i>b</i> = 70.0913

* The full Crystallographic Information Framework (CIF) files are also available as electronic supporting information.

† $wR2 = \{\sum[w(F_o^2 - F_c^2)^2] / \sum[w(F_o^2)^2]\}^{1/2}$; $w = 1/[\sigma^2(F_o^2) + (aP)^2 + bP + d + e \sin \theta]$; $P = [f(\text{Max}(0 \text{ or } F_o^2))] + (1-f) F_c^2$

Table S3. Crystallographic data for compounds 19 and 20.

Parameter	Compound 19	Compound 20
Chemical formula	C ₄₃ H ₄₇ O ₇ P ₃ Ru	C ₄₅ H ₄₇ O ₅ P ₃ Ru
Formula weight (g/mol)	869.82	861.84
Temp. (K)	93(2)	93(2)
Wavelength (Å)	0.71075	0.71075
Crystal system	Monoclinic	Monoclinic
Crystal dimensions (mm ³)	0.12 × 0.10 × 0.03	0.03 × 0.03 × 0.03
Crystal shape and colour	Prism, yellow	Prism, yellow
Space group	<i>P</i> 2 ₁ / <i>n</i> (No. 14)	<i>P</i> 2 ₁ / <i>n</i> (No. 14)
<i>a</i> (Å)	12.702(2)	11.969(2)
<i>b</i> (Å)	31.997(6)	20.106(4)
<i>c</i> (Å)	24.329(5)	16.576(3)
α (°)	90.00	90.00
β (°)	100.330(2)	102.415(3)
γ (°)	90.00	90.00
Unit cell volume (Å ³)	9727(3)	3895.6(12)
No. of formula units per unit cell, <i>Z</i>	8	4
<i>d</i> _{calcd} (g/cm ³)	1.612	1.469
Radiation type	MoK α	MoK α
Absorption coefficient, μ (mm ⁻¹)	0.887	0.573
<i>F</i> (000)	4824	1784
θ -range for data collection (°)	1.70 to 25.42	1.62 to 25.35
Index range	-15 ≤ <i>h</i> ≤ 15, -38 ≤ <i>k</i> ≤ 38, -29 ≤ <i>l</i> ≤ 29	-14 ≤ <i>h</i> ≤ 14, -23 ≤ <i>k</i> ≤ 24, -19 ≤ <i>l</i> ≤ 19
No. of reflections measured	97095	38423
No. of unique reflections	17854	7119
Refinement parameters / restraints	1159 / 4	498 / 0
<i>R</i> _{int}	0.0621	0.0346
Final <i>R</i> ₁ values (<i>I</i> > 2 σ (<i>I</i>))	0.0951	0.0299
Final <i>wR</i> (<i>F</i> ²) values (<i>I</i> > 2 σ (<i>I</i>))	0.2385	0.0622
Final <i>R</i> ₁ values (all data)	0.0989	0.0315
Final <i>wR</i> (<i>F</i> ²) values (all data)	0.2413	0.0630
Goodness of fit on <i>F</i> ²	1.150	1.078
Largest diff. peak and hole (e.Å ⁻³)	1.511 and -2.440	0.330 and -0.464
Weighting scheme †	<i>a</i> = 0.0973 <i>b</i> = 93.0135	<i>a</i> = 0.0193 <i>b</i> = 4.8610

* The full Crystallographic Information Framework (CIF) files are also available as electronic supporting information.

† $wR2 = \{\sum[w(F_o^2 - F_c^2)^2] / \sum[w(F_o^2)^2]\}^{1/2}$; $w = 1/[\sigma^2(F_o^2) + (aP)^2 + bP + d + e \sin \theta]$; $P = [f(\text{Max}(0 \text{ or } F_o^2))] + (1-f) F_c^2$

■ SECTION S-7: REFERENCES

- (1) Chaplin, A. B.; Dyson, P. J. *Inorg. Chem.* **2008**, 47, 381.
- (2) Iengo, E.; Zangrando, E.; Baiutti, E.; Munini, F.; Alessio, E. *Eur. J. Inorg. Chem.* **2005**, 1019.
- (3) Núñez Magro, A. A. *PhD Thesis* **2007**, 244.
- (4) Wei, P.; Bi, X.; Wu, Z.; Xu, Z. *Org. Lett.* **2005**, 7, 3199.
- (5) Stein, M.; Breit, B. *Angew. Chem. Int. Ed.* **2012**, DOI: 10.1002/anie.201207803.