

Nanostructuring of 100 thousand tons

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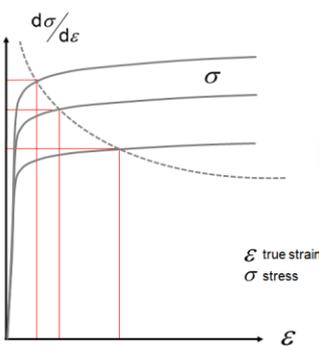
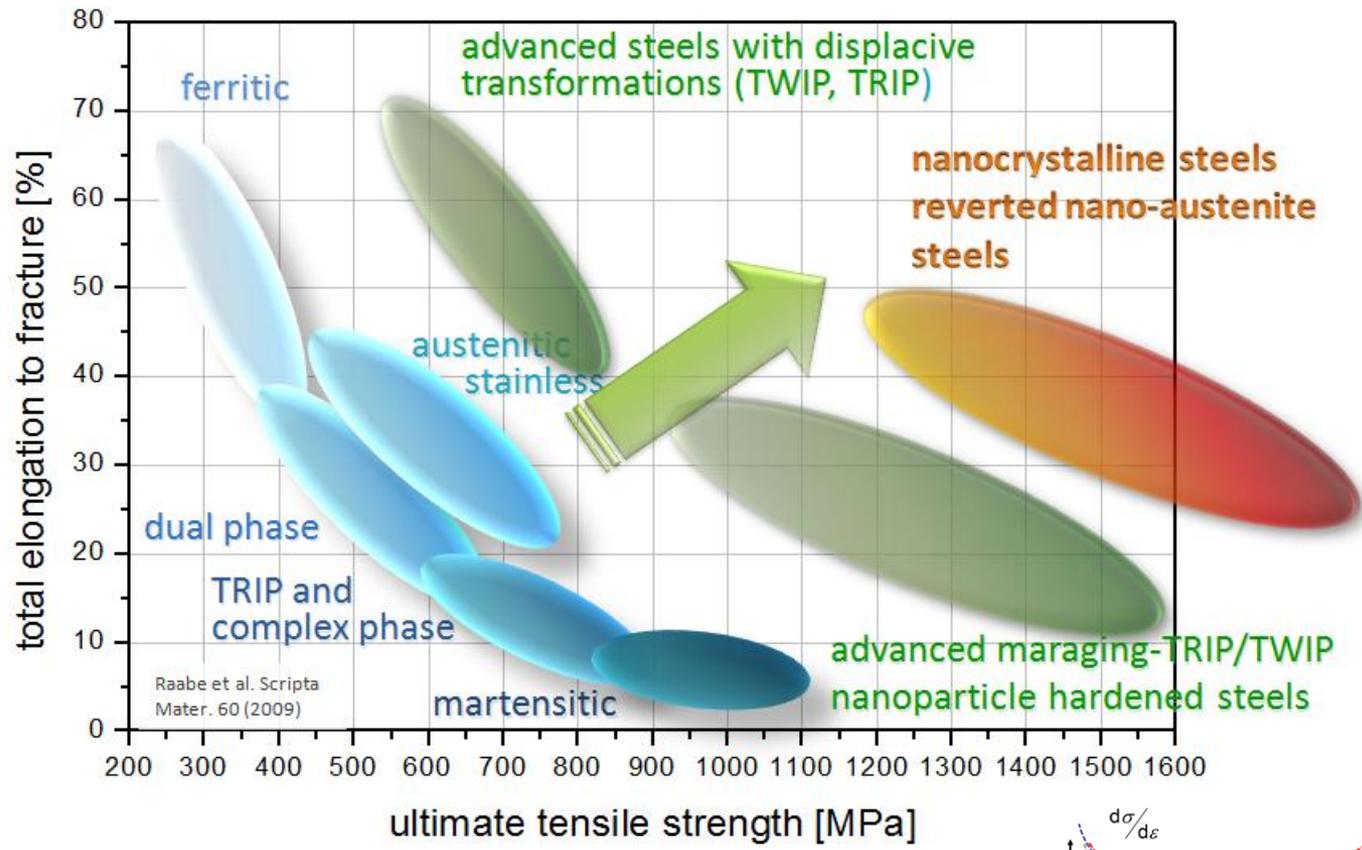
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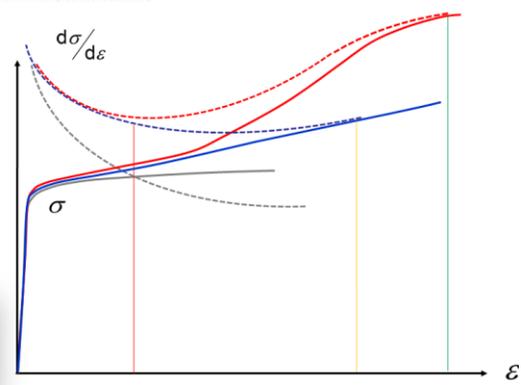
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SHORTENED VERSION



Inverse strength-ductility relation

Design strain hardening only where needed



Self-organized nanostructuring via lattice defect decoration and selective phase transformation: nano-particles and nano-TRIP

Motivation for nanostructuring alloys via self-organization:

Properties of alloys change profoundly at nanoscale (mech. properties)

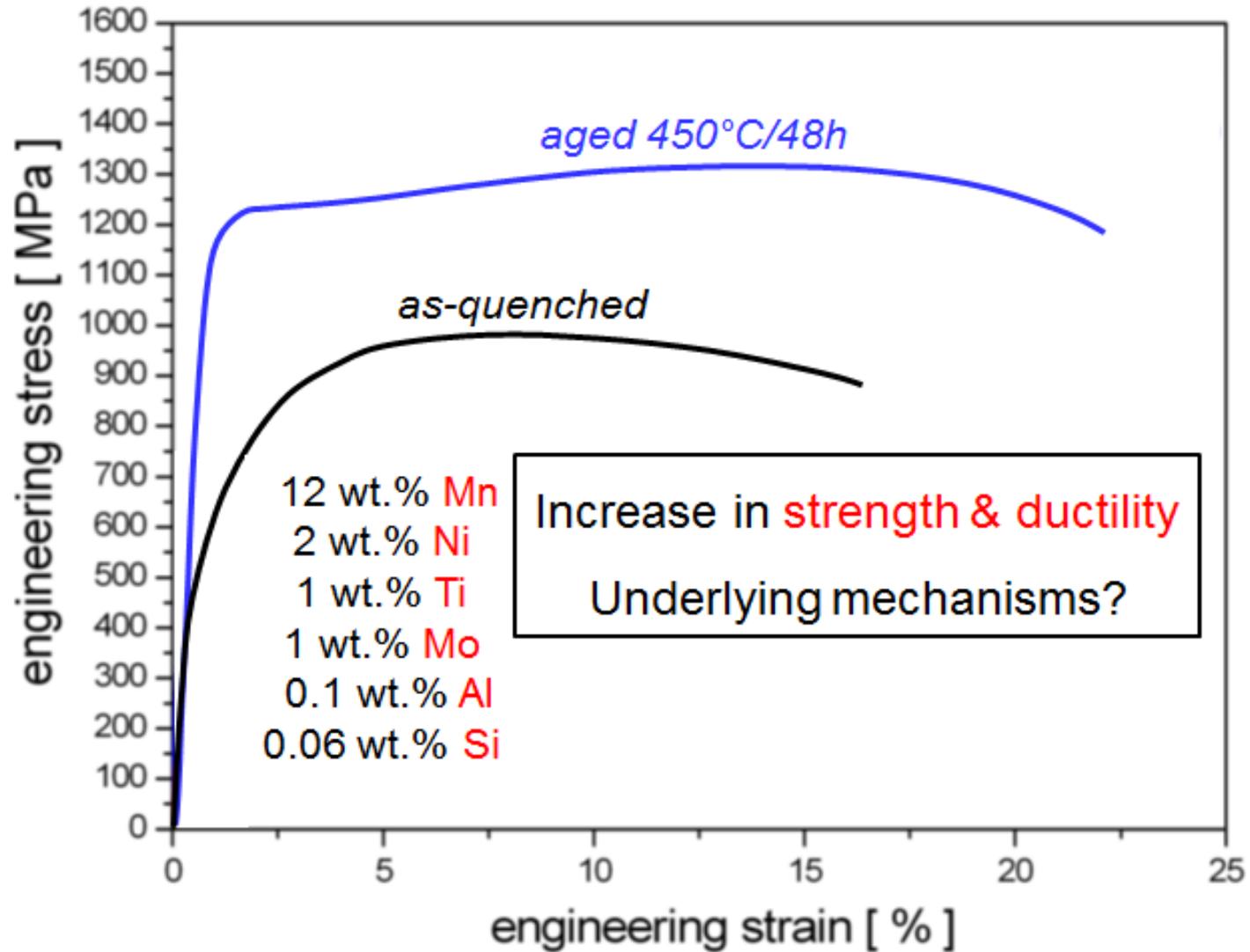
Self organization essential for nanostructuring large quantities under real-world process conditions:

e.g.

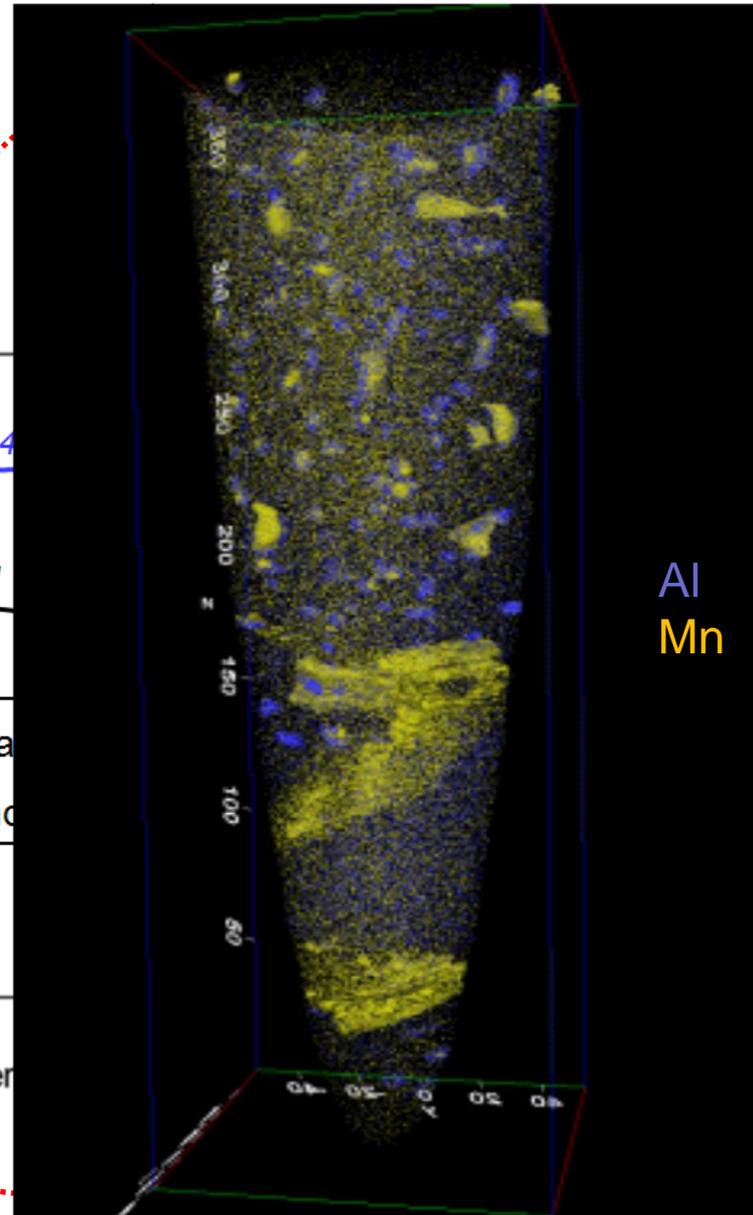
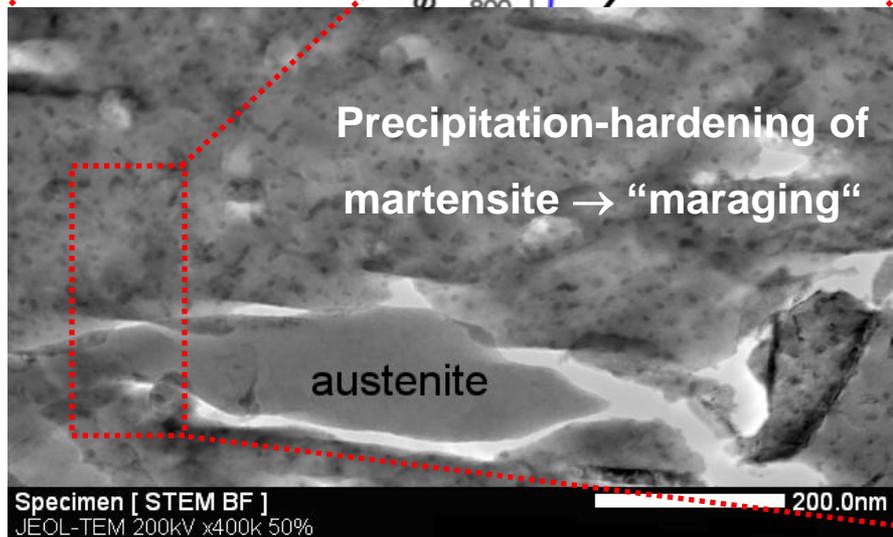
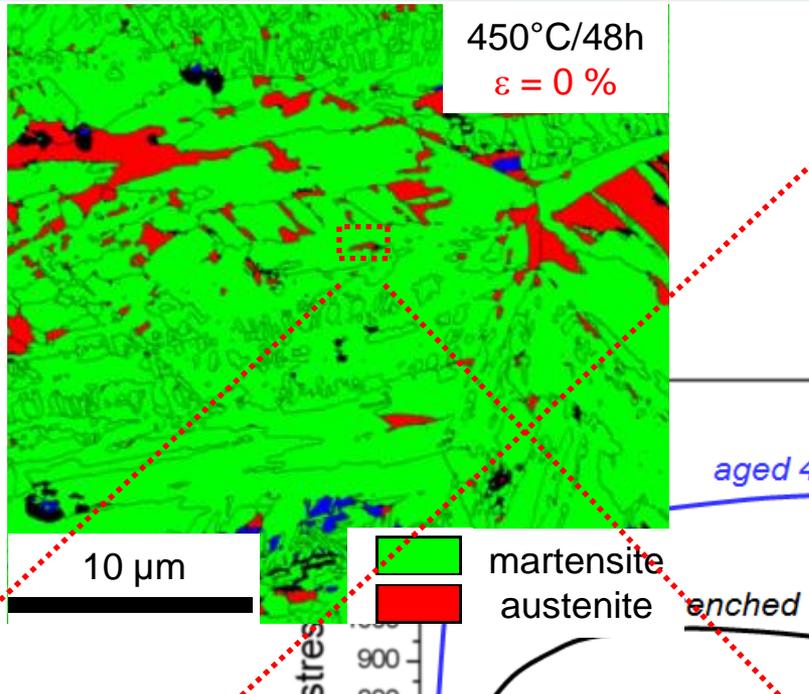
Fe: 1.4 Billion t/a

Al: 24 Million t/a

Ti: 150 Thousand t/a



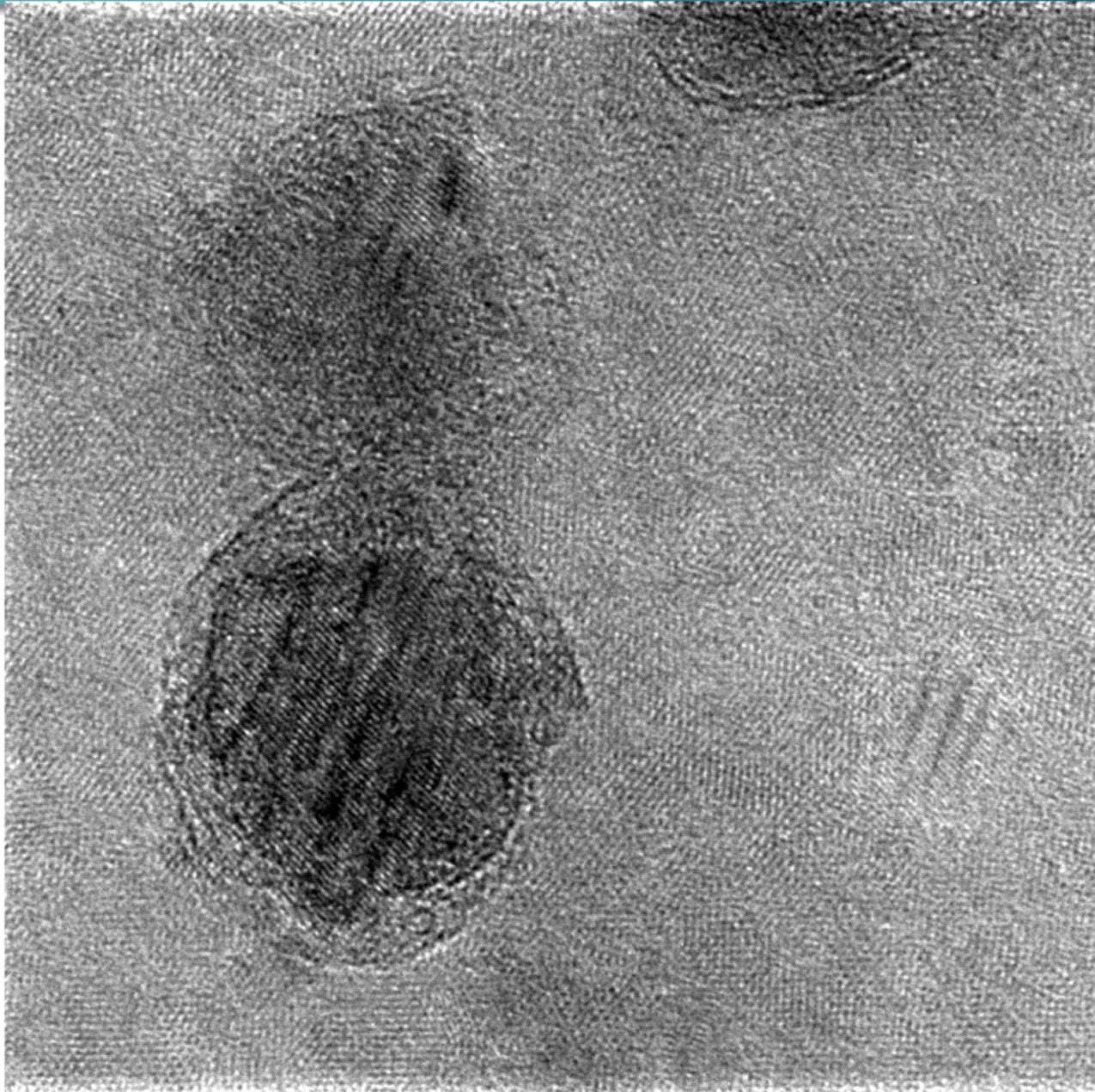
Hierarchy of maraging TRIP / TWIP steels



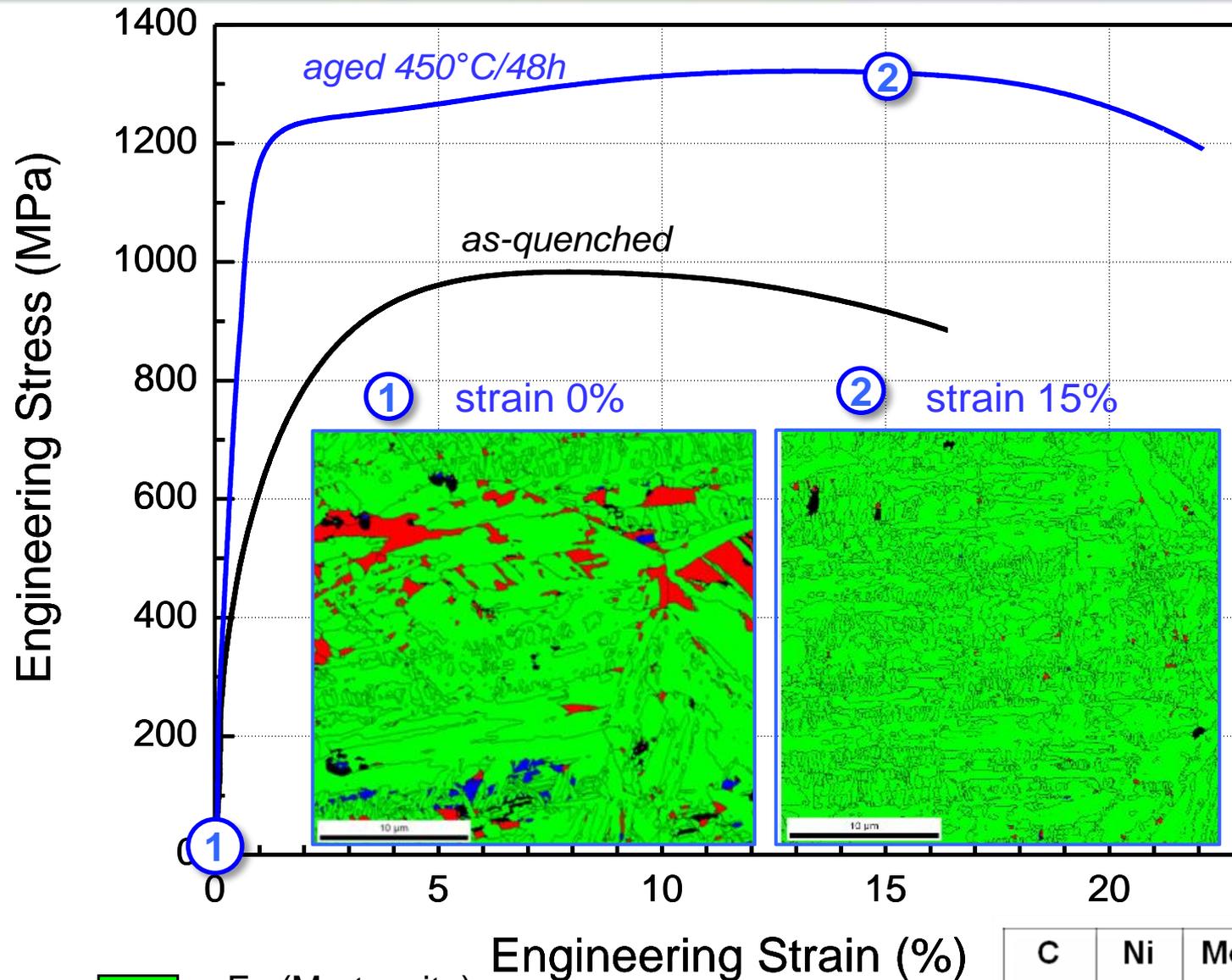
martensite

austenite

martensite

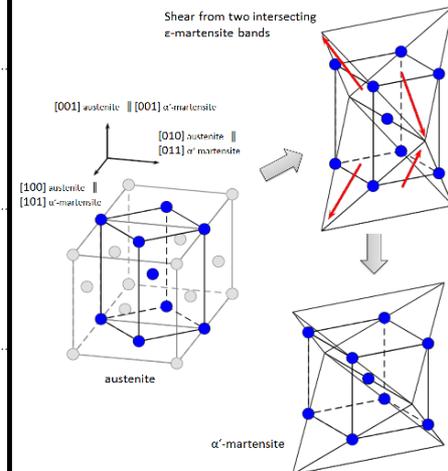


Effect of aging on ductility



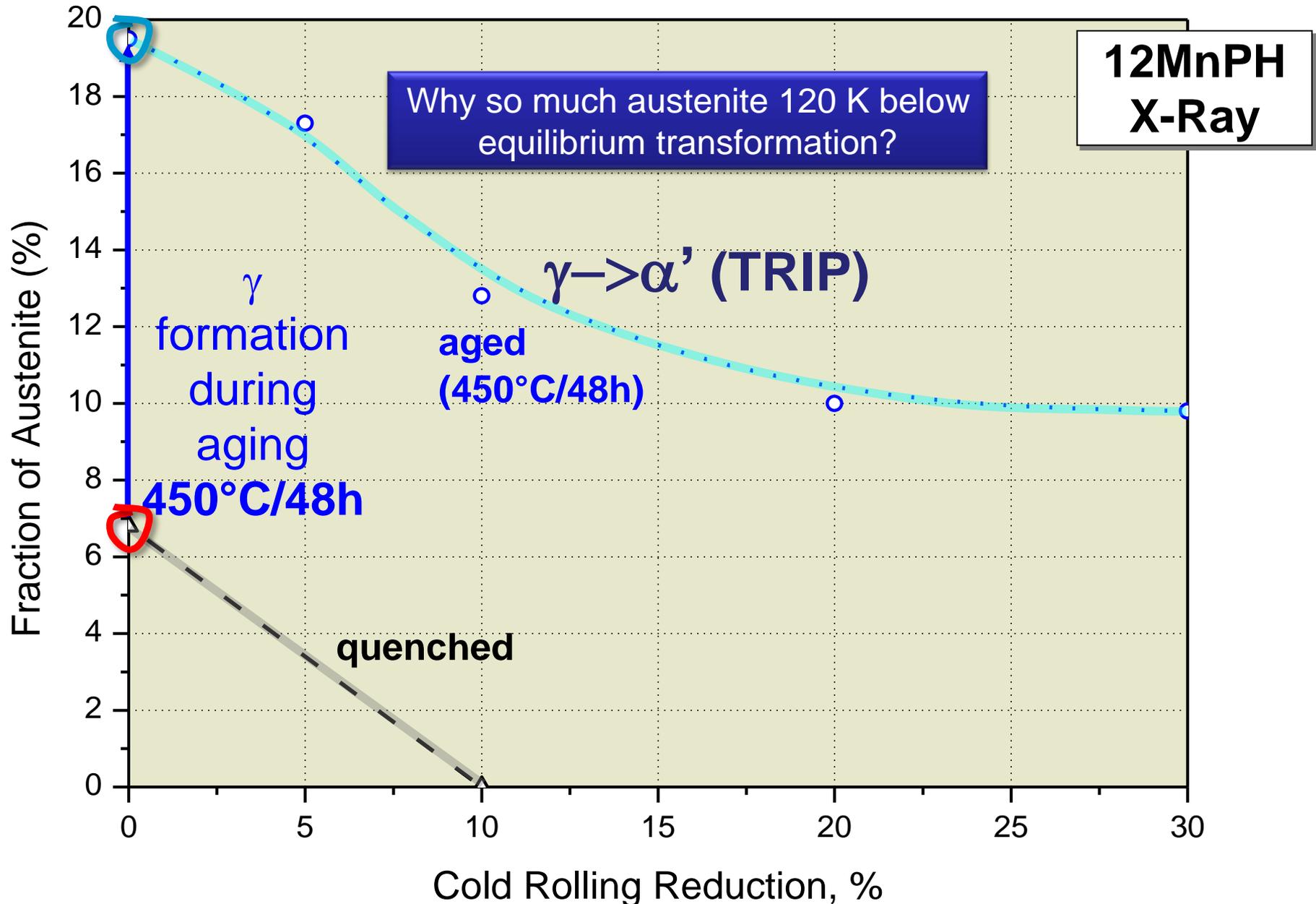
Precipitation hardening

increase of austenite fraction during aging



α -Fe (Martensite)
 γ -Fe (Austenite), vol. fraction 15-20%

C	Ni	Mo	Ti	Al	Mn	Fe
0.01	2.0	1.0	1.0	0.15	12	bal.



Aging-induced austenite reversion

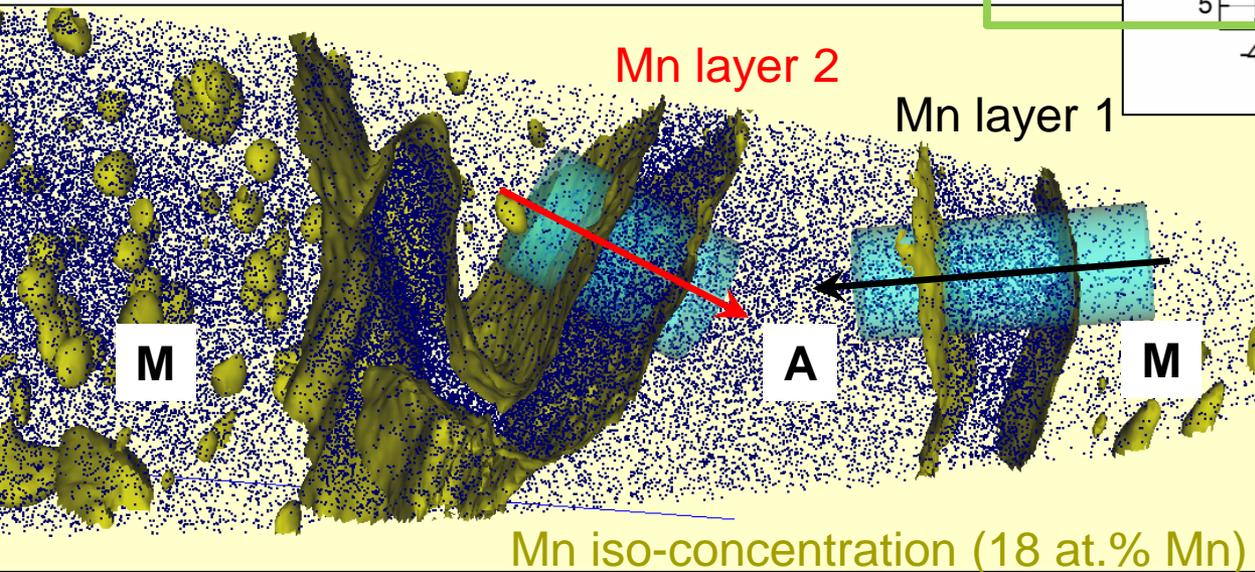
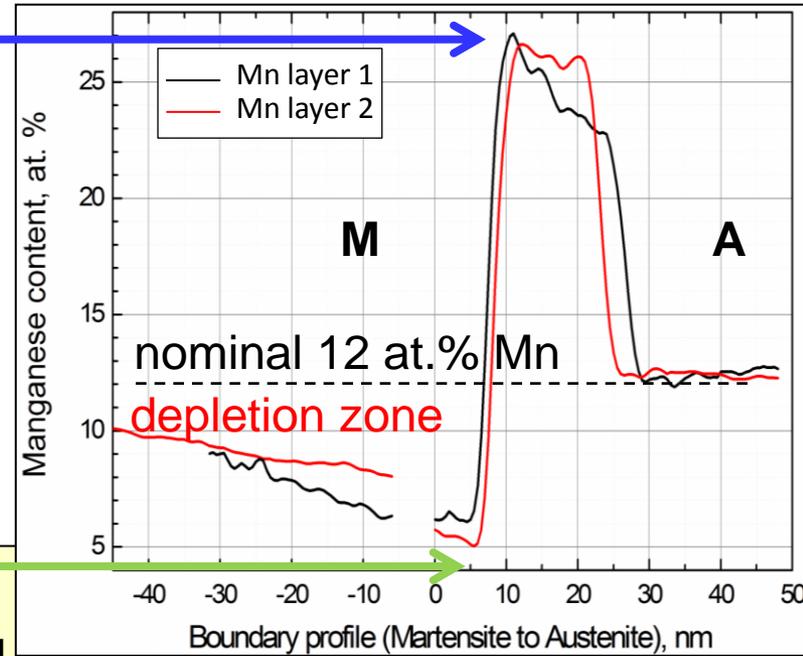


Thermo-Calc \Rightarrow

equilibrium Mn-conc.:

27 at. % Mn in austenite (A)

3 at. % Mn in ferrite (martensite) (M)



precipitates in α'

$$x_{Diff} \cong 2\sqrt{Dt} \cong 30nm$$

no precipitates in
austenite

$$x_{Diff} \cong 2nm$$



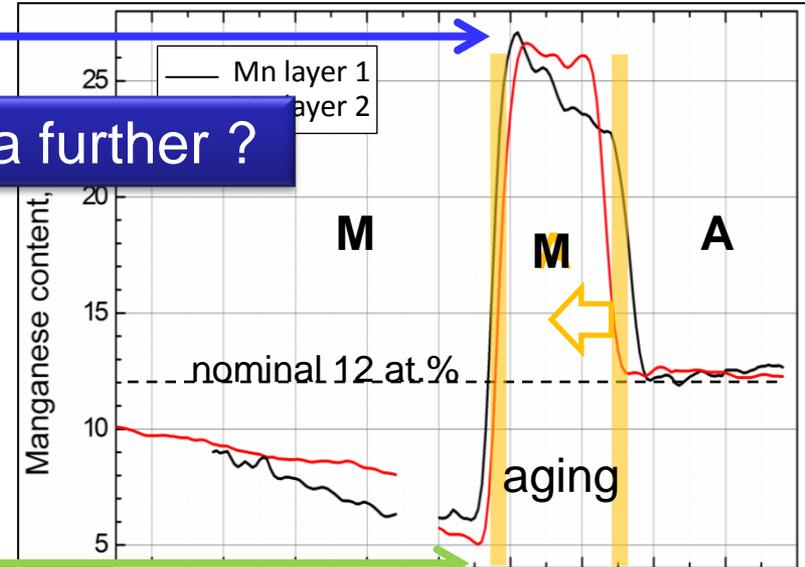
Thermo-Calc

Can I push this idea further ?

equilibrium Mn-conc.:

27 at. % Mn in austenite (A)

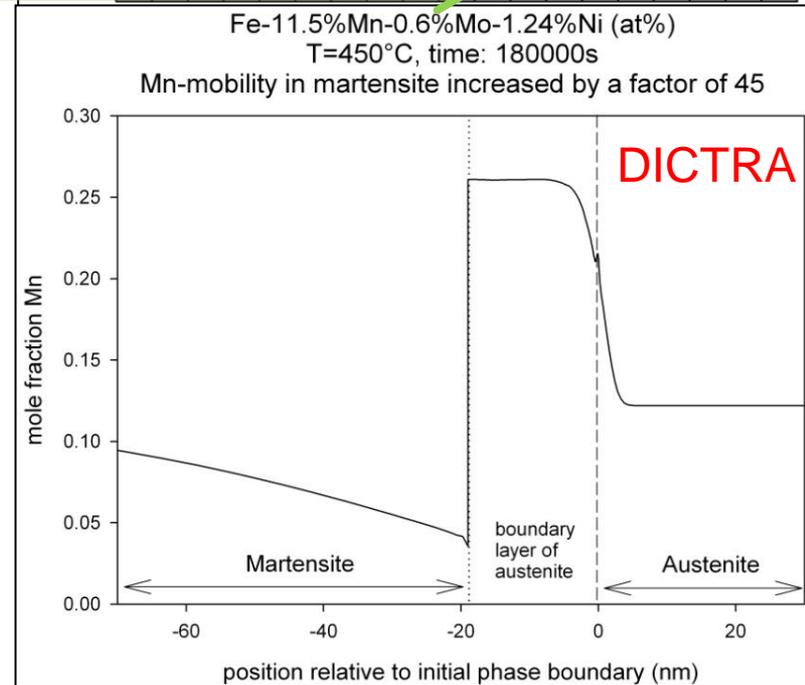
3 at. % Mn in ferrite (martensite) (M)

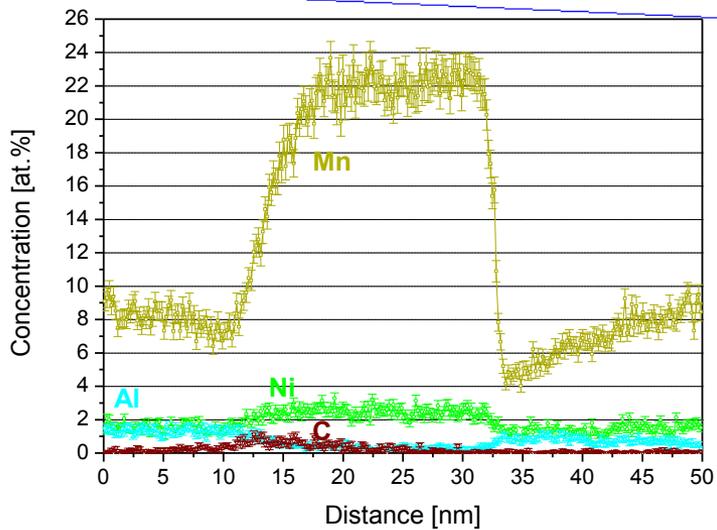
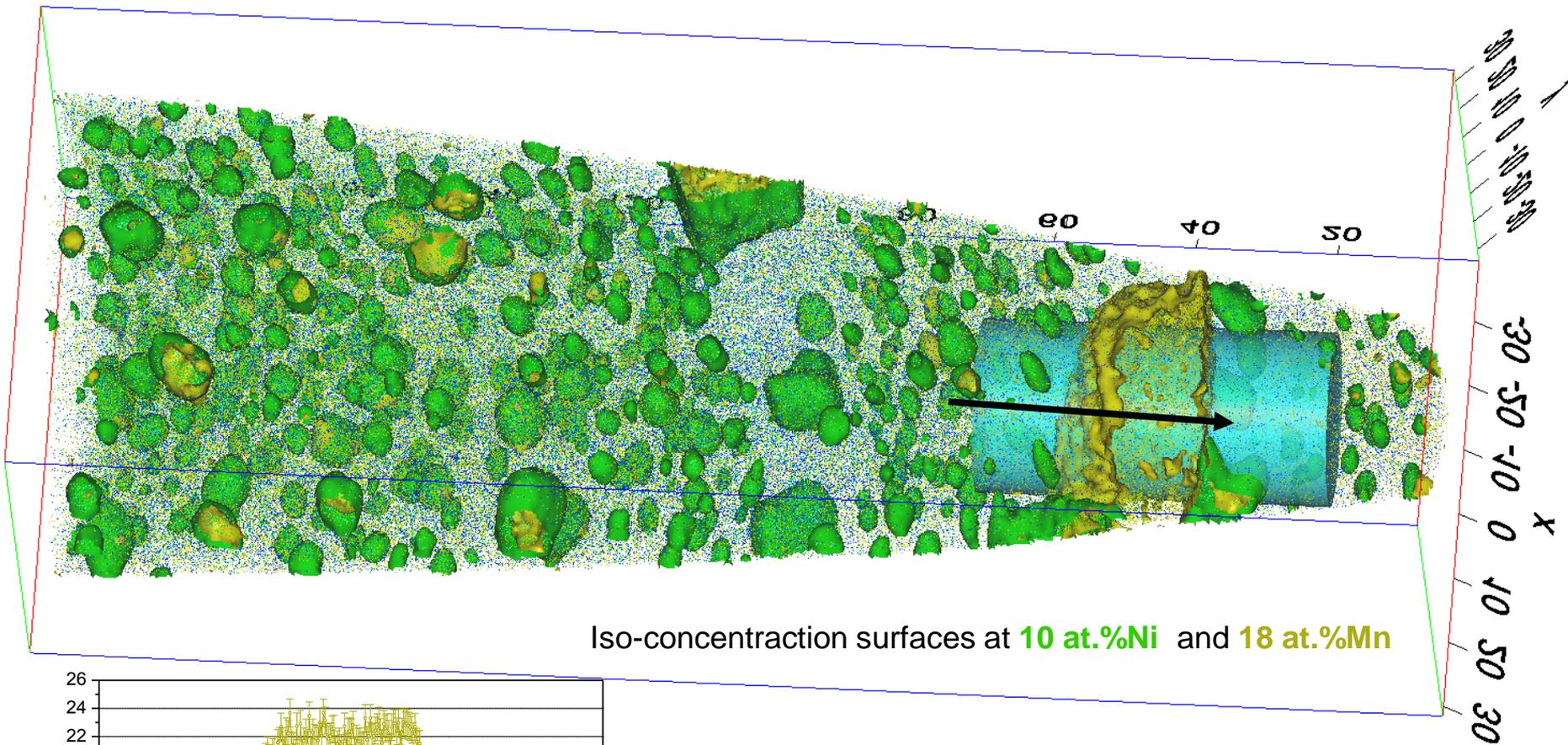


Excellent agreement between experiment & simulation !



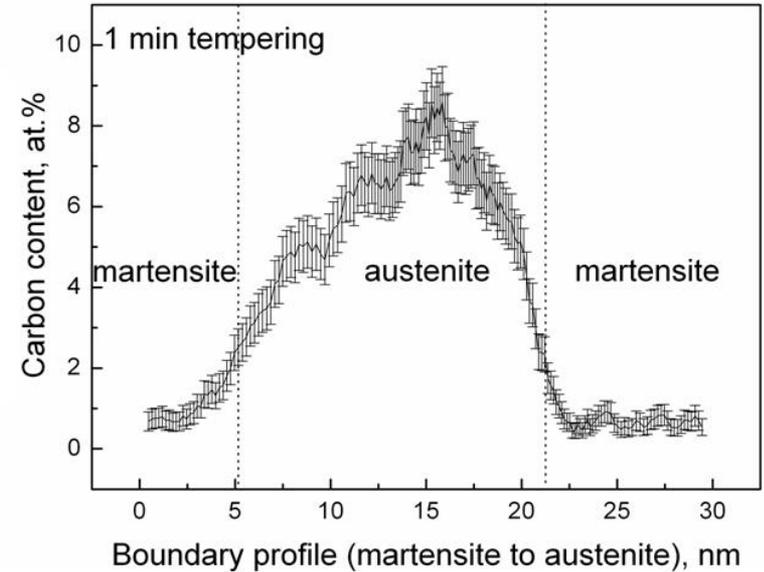
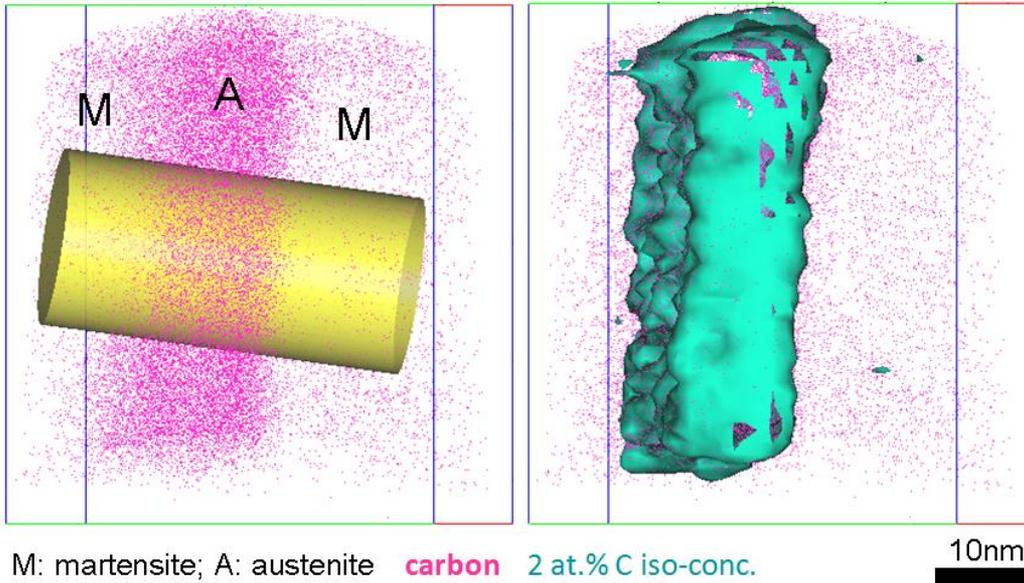
Kinetic freezing and associated austenite reversion !





Equilibrium segregation on g.b. and associated austenite reversion !

at 5.45 at.% C, austenite forms at 400°C

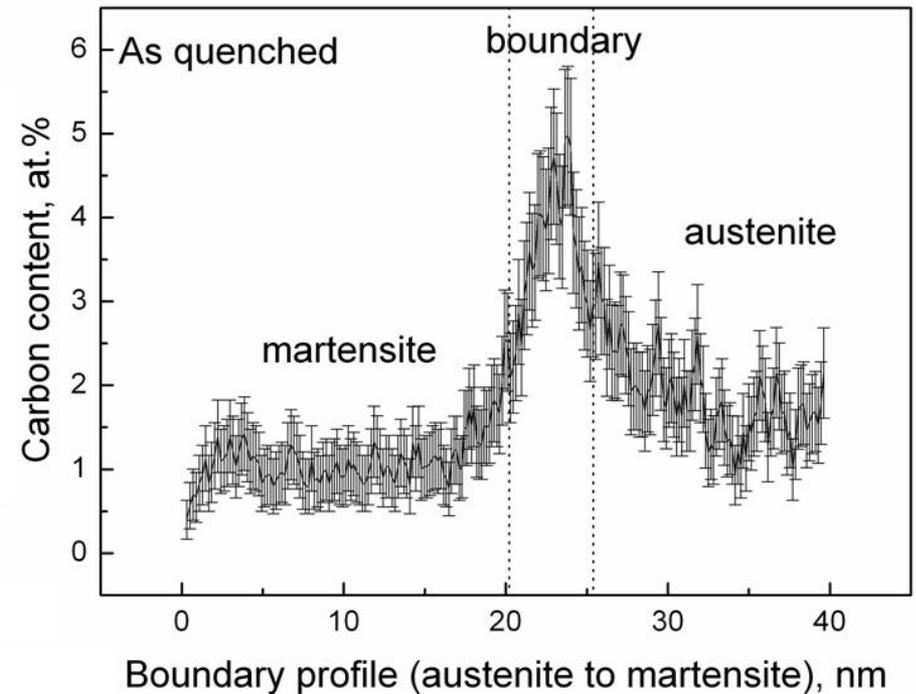
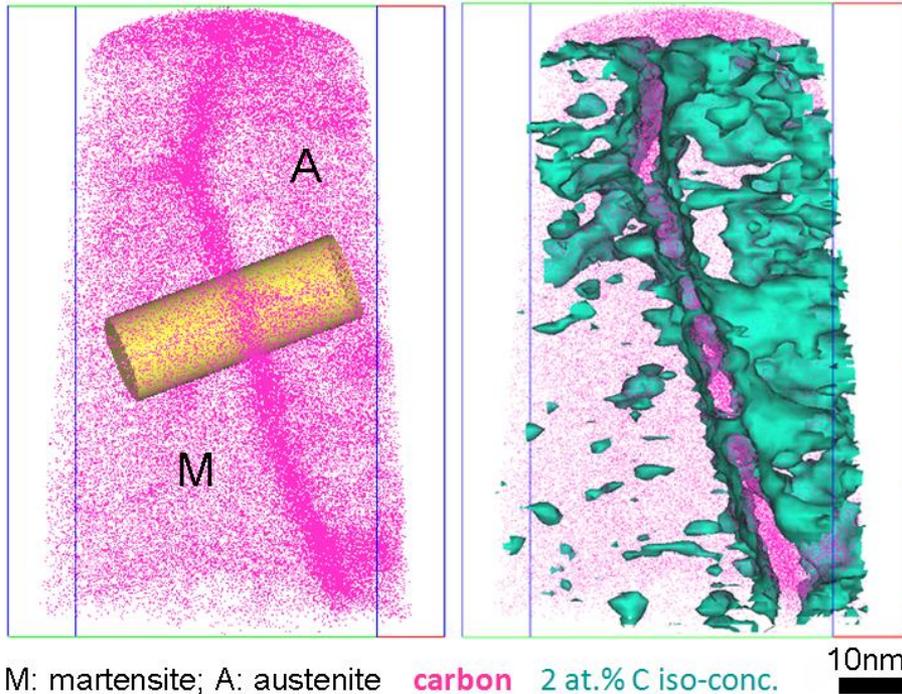


C has 'Λ' shape in austenite layer: inheritance from austenite, Gibbs adsorption isotherm;

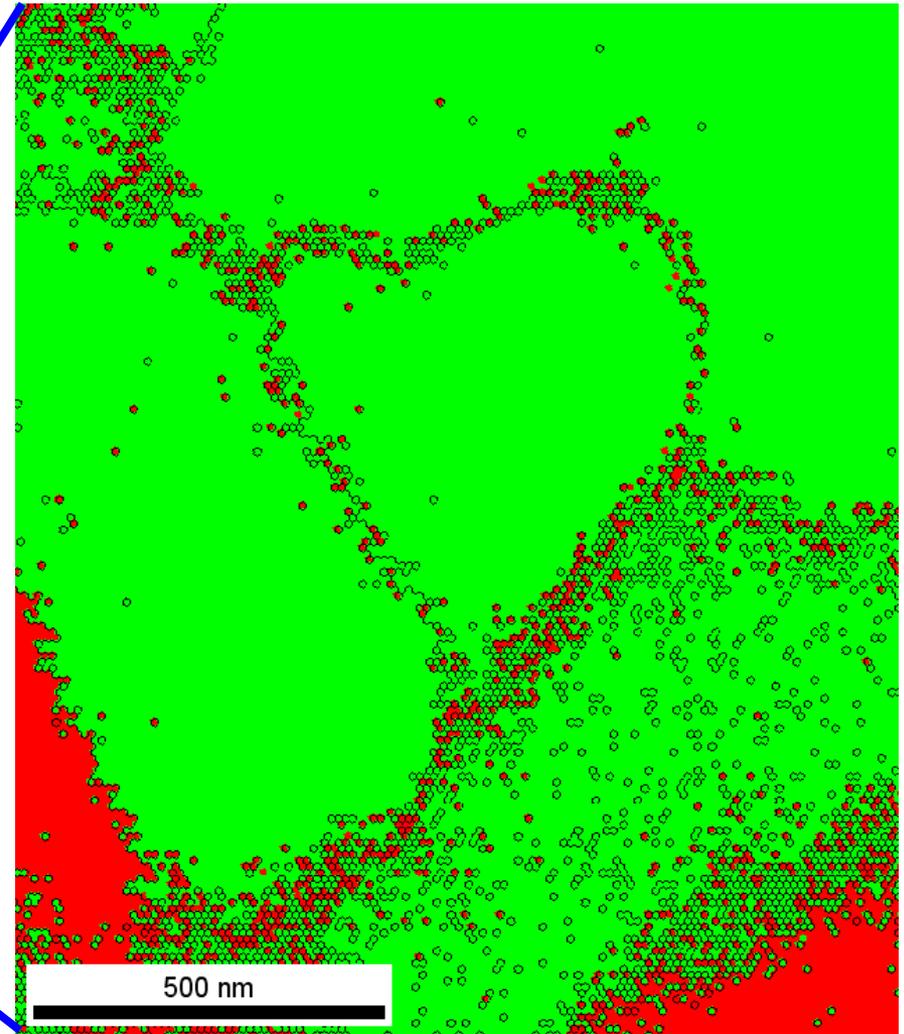
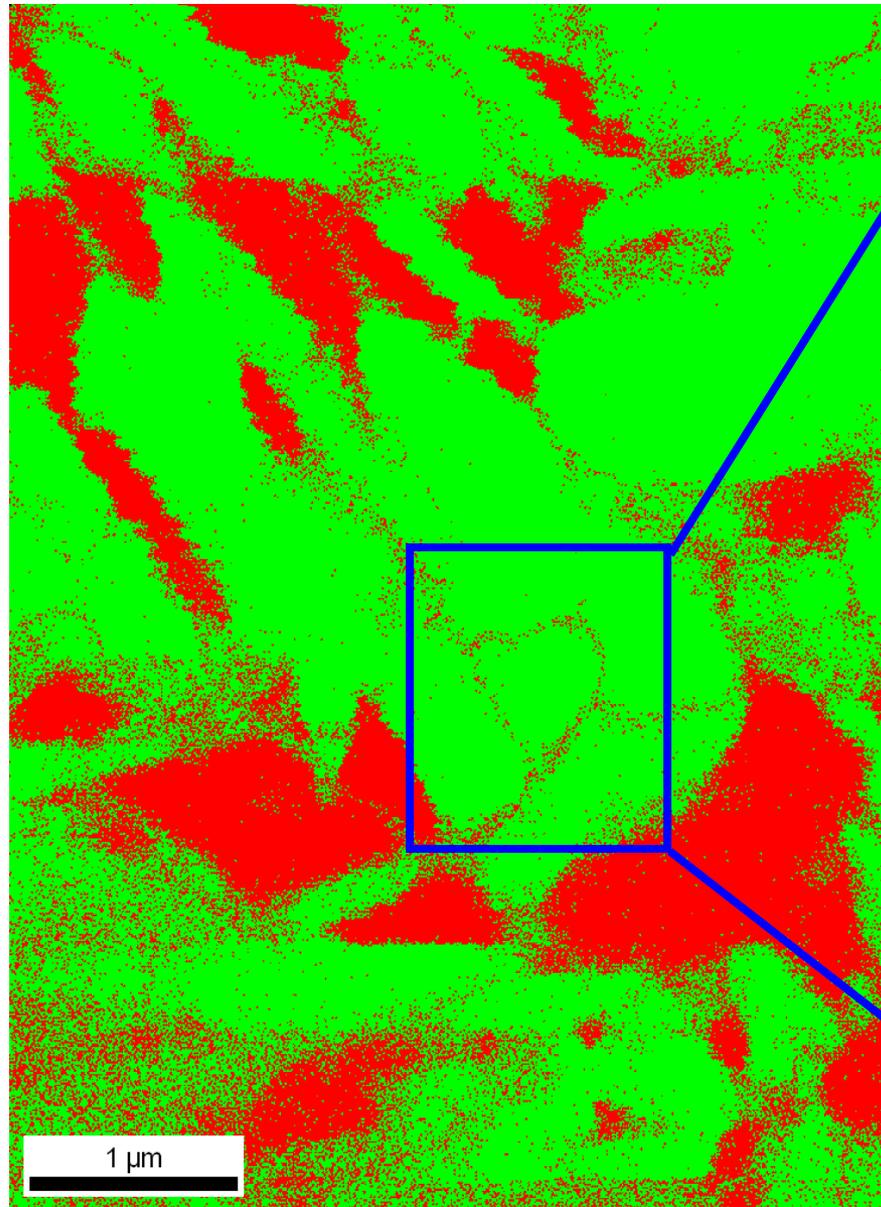
C on martensite grain boundaries

C has 'V' shape in austenite layer: austenite reversion through partitioning and kinetic freezing

at 5.45 at.% C, austenite forms at 400°C



C has 'Λ' shape in austenite layer: Gibbs adsorption isotherm; C on martensite grain boundaries
 C has 'V' shape in austenite layer: austenite reversion through partitioning and kinetic freezing



Phase

	Iron - Gamma
	Iron - Alpha