

**MAX-PLANCK-INSTITUT FÜR PLASMAPHYSIK**  
**GARCHING BEI MÜNCHEN**

SCATTERING OF RARE GAS IONS FROM  
METAL SURFACES

STREUUNG VON EDELGASIONEN AN  
METALLOBERFLÄCHEN

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ABSTRACT

The scattering of rare gas ions from polycrystalline and single crystal metal targets is studied in the energy range of 2 to 16 keV. This report deals with three aspects of ion scattering: 1. Noble gas ion scattering as an analytical tool for surface analysis, 2. The inelastic energy losses in the scattering processes, and 3. The influence of the crystal structure on the scattering. The last part contains some model calculations to explain the experimental results.

Die Streuung von Edelgasionen an poly- und einkristallinen Metall Targets wurde im Energiebereich von 2 bis 16 keV untersucht. Dieser Bericht handelt von drei Aspekten der Ionenstreuung: 1. Die Edelgasstreuung als Mittel zur Oberflächenanalyse, 2. Die inelastischen Energieverluste bei der Streuung und 3. Der Einfluß der Kristallstruktur auf die Streuung. Der letzte Teil enthält einige Modellrechnungen, die zum Verständnis der Experimente angestellt wurden.

## 1. INTRODUCTION

The scattering of rare gas ions from metal surfaces has been the subject of investigations in various laboratories for a number of years. The comprehensive literature is discussed in detail by Mashkova and Molchanov in a recently published review article /1/. The present paper therefore only makes reference to a few special studies.

Our investigations with rare gas ions were conducted parallel to measurements of the backscattering of hydrogen ions /2, 3/ and were likewise done with the BOMBARDON apparatus, which has already been described in detail /4/. In this device targets can be bombarded with ion beams of 2 - 20 keV. The target is placed in a UHV chamber, where it is mounted on a goniometer head, on which it can be rotated about three mutually perpendicular axes. The backscattered ions are analyzed with an electrostatic energy spectrometer which can be swept about the target and are detected with a multiplier.

In this report three aspects of rare gas ion scattering from metal surfaces are discussed on the basis of experimental results and model calculations:

- Scattering as a method of analysis
- Inelastic energy losses due to scattering
- Influence of crystal structure on scattering.

## 2. SCATTERING OF RARE GAS IONS AS A METHOD OF ANALYSIS

In many cases rare gas ions are scattered from the individual surface atoms of the solid. The energy loss due to the scattering and the scattering angle are then related by the purely kinematic formula:

$$\begin{aligned} E_1/E_0 &= f(A, \vartheta) \\ f(A, \vartheta) &= \frac{1}{(1+A)^2} \left\{ \cos \vartheta \pm [A^2 - \sin^2 \vartheta]^{1/2} \right\}^2 \end{aligned} \quad (1)$$

where the notations are as follows:

- $E_0$  - energy of ions before collision
- $E_1$  - energy of ions after collision
- $\vartheta$  - scattering angle in laboratory system
- $M_1$  - mass of colliding particle
- $M_2$  - mass of target atoms
- $A = \frac{M_2}{M_1}$

By determining the energy of the particles after collision for a fixed scattering angle it is possible to obtain the mass  $M_2$  of the scattering atom. This affords a method of surface analysis. This technique was first used by D.P. Smith /5/ and later elaborated by Ball et al. /6/, Goff /7/, and Heiland and Taglauer /8/.

Given the energy resolution  $E/\Delta E$  of the spectrometer used, it is possible to calculate the attainable mass resolution:

$$\frac{M_2}{\Delta M_2} = g(A, \vartheta) \frac{E}{\Delta E} \quad (2)$$

$$g(A, \vartheta) = \frac{2A}{A+1} \cdot \frac{A + \sin^2 \vartheta - \cos \vartheta \sqrt{A^2 - \sin^2 \vartheta}}{A^2 - \sin^2 \vartheta + \cos \vartheta \sqrt{A^2 - \sin^2 \vartheta}} \quad \text{for } A > 1$$

Figure 1 shows the dependence of the function  $g$  on the mass ratio  $A$  for two scattering angles, and Fig. 2 the dependence on the scattering angle  $\vartheta$ . The curves show that the mass resolution is most favourable for small  $A$  and large  $\vartheta$ . The mass resolution is limited, however, by the smallest detectable scattering intensity.

The scattering cross-section, and hence the intensity, decreases as the scattering angle becomes larger. This is shown in Fig. 3, where the scattering cross-section is plotted as a function of the scattering angle for 15.1 and 4.8 keV  $\text{Ne} \rightarrow \text{Ni}$  according to the tables of Robinson /9/. The cross-section does become larger with increasing  $A$  for a certain ion mass  $M_1$ , but at the same time the mass resolution undergoes a corresponding decrease (Fig. 2).

When an electrostatic spectrometer is used, the recorded scattering intensity decreases with decreasing energy because the energy interval is  $\Delta E \propto E$ . The detection probability is therefore low for small  $A$  and large  $\vartheta$  since the energy of the backscattered ions is small.

It should also be noted that it is only for  $A > 1$  that all masses can be detected at all scattering angles. For  $A < 1$  scattering only occurs at angles for which  $\sin^2 \vartheta < A^2$ .

An experimental example is given in Fig. 4, which shows the energy spectra of  $\text{Ne}^+$  ions backscattered from a Ni target. The increase in resolution and decrease in intensity as the scattering angle becomes larger are obvious. At  $\vartheta = 120^\circ$  and  $\vartheta = 150^\circ$  the two Ni isotopes with masses 58 and 60 can be separated without trouble. In this case the value for  $A$  and  $E/\Delta E$  were  $A = 2.9$  and  $E/\Delta E = 140$ . The small peaks occurring at half the energy of the large maxima can be ascribed to  $\text{Ne}^{++}$  ions. In the spectra the  $\text{Ni}^{62}$  isotope (3.7% in the natural isotope mixture) can just be recognized. The detection limit in this case is thus in the region of a few per cent.

Another example showing the detection of oxygen on a vanadium surface is given in Fig. 5. The upper spectrum was recorded after the target had been heated for a few minutes to about  $750^\circ\text{C}$ . This causes oxygen to diffuse to the surface, and this can be detected as a maximum in the spectrum. After a certain bombardment dose the oxygen is sputtered and the oxygen maximum vanishes. At the same time a small maximum (denoted by " $\text{V}^+$ ") appears which can be ascribed to vanadium "sputtered" in a binary collision. Besides presenting ion scattering as a method of analysis, Fig. 5 also demonstrates a certain disadvantage, namely that the surface can be changed relatively fast by the measurement if the bombardment dose required for analysis is not kept as low as possible. In this case the oxygen maximum vanishes after a bombardment dose of approximately  $1.5 \cdot 10^{-2} \text{ Cb/cm}^2$ , while about  $2 \cdot 10^{-3} \text{ Cb/cm}^2$  was needed to record the spectrum.

The backscattering spectra here have a somewhat different shape to those at lower energies ( $< 1200 \text{ eV}$ ) as was investigated by Heiland and Taglauer /8/, /18/. Particularly at small scattering angles there is an appreciable

scattering intensity at energies below the binary collision maximum (see also Figs. 6, 11 and 17). This is due to ions which are scattered deeper inside the solid. At low energies the backscattering maxima are approximately symmetric because only the ions backscattered from the surface have a chance of not being neutralized. Heiland et al. [18] find the maximum of the target atoms "sputtered" in binary collisions to be at  $E_0 < 1200$  eV only when bombardment is with  $Ar^+$  ions. They, too, observed an appreciable change of the surface during the measurements.

### 3. INELASTIC ENERGY LOSSES DUE TO SCATTERING

The energies of the ions elastically backscattered from the surface atoms in binary collisions are given by formula (1). It is experimentally observed, however, that the corresponding maxima are shifted towards lower energies. In Figs. 4, 5 and 6 the energies calculated with formula (1) are denoted by arrows. (In Figs. 5 and 6 at  $\vartheta = 30^\circ$  the maximum is broadened by multiple scattering). The shift of the maxima is distinctly larger than the experimental errors. It was possible to determine the energy exactly to within  $\pm 25$  eV and to set the scattering angle exactly to within  $\pm 0.3^\circ$ . The observed shifts would require deviations of the scattering angles from the nominal value of several degrees. It must therefore be assumed that the inelastic energy losses due to scattering cannot be neglected. Figure 7 shows the inelastic energy loss due to scattering of  $Ne^+$  ions from Ni as a function of the scattering angle. In Fig. 8 and Fig. 9 the inelastic energy losses are depicted as functions of the primary energy  $E_0$  and the energy  $E_1$  after the collision respectively.

In our energy range the estimate by Firsov [10] for the electronic energy loss should be valid. This would make the inelastic energy loss

$$\Delta E_{th} = \frac{(Z_1 + Z_2)^{5/3} \cdot 4.3 \cdot 10^{-8} u}{[1 + 3.1 (Z_1 + Z_2)^{1/3} \cdot 10^7 R_0]^5} \quad (3)$$

where  $Z_1$  and  $Z_2$  are the atomic numbers of the collision partners,  $u$  the velocity of the incoming particle (if the other particle is at rest), and  $R_0$  the distance of closest approach, which is a function of the primary energy  $E_0$  and the scattering angle  $\vartheta$ . The  $R_0$  values can be interpolated from the tables of Robinson (9) and were used to calculate the energy losses according to equation (3). A comparison between the experimental results and the Firsov theory is shown in Fig. 10, where the ratio of the experimental to the theoretical values are shown as function of  $E_1/E_0$ . As one expects the agreement is satisfactory only for small elastic energy losses i. e. when  $E_1$  is not very different from  $E_0$ . The theoretical values calculated with  $u = v_0$  are too large in most cases but too small if calculated with  $u = v_1$  (velocity of the scattered ion).

As the scattering angles decrease there is more probability not only of purely binary collisions but also of the multiple scattering processes described in the following section. These produce the "shoulders" on the high-energy side of the maxima in Fig. 6. At smaller scattering angles the shoulders cause the binary collision maximum to be shifted to higher energies. The measured energy losses for smaller scattering angles are therefore probably still too small.

#### 4. INFLUENCE OF CRYSTAL STRUCTURE ON SCATTERING

The maxima in Figs. 4, 5 and 6 ascribed to binary collisions with surface atoms according to formula (1) show more or less pronounced shoulders on the high-energy sides. These can be explained in terms of several successive binary collisions (multiple collisions) [11, 12, 13, 14]. This effect is best investigated by means of scattering from single crystal surfaces. Figure 11 shows the backscattering spectra of 10 keV  $\text{Ne}^+$  ions impinging at an angle of  $\psi = 20^\circ$  at a  $\langle 110 \rangle$  chain in the (100) surface of a vanadium single crystal. At some scattering angles two distinct maxima can be observed. The presence of two maxima at a fixed scattering angle can be explained in terms of multiple scattering from single chains [15, 16, 17, 18, 19]. Model calculations were made by computer to clarify the effects of multiple scattering. The results are discussed below.

#### 4.1 COMPUTER CALCULATIONS

The scattering of particles from an atomic chain is simulated according to the following scheme. An ion with energy  $E_0$  impinges at an angle  $\psi$  on a rigid one-dimensional chain of atoms, collides in succession with the individual atoms and leaves the chain again with a total scattering angle  $\vartheta$  and an energy  $E_1$ . The first impact parameter  $p_1$  is chosen such that the corresponding angle of deflection is smaller than  $0.01^\circ$ . The collision parameter for the next collision with the next chain atom at a distance  $d$  is found from Fig. 12 (according to Parilis /11/) to be

$$p_2 = p_1 - d \sin(\psi - \vartheta_1) \quad (4)$$

and the collision parameters for the subsequent collisions are similarly determined. For every collision the corresponding scattering angle is determined from the collision parameter and the corresponding energy by means of Lindhard's magic formula /20/ and Molière's interaction potential /21/<sup>+</sup>). The elastic energy loss is taken into account by subtracting after every collision the transmitted energy corresponding to the scattering angle from the energy of the scattered ion. The ion keeps colliding with chain atoms until the angle of deflection is smaller than  $0.01^\circ$ . An ion beam is simulated by using this method to calculate a series of ion paths with successively larger first collision parameter until the first ion path (shifted by one atomic spacing) is again obtained, i. e. when  $p_2$  becomes equal to the  $p_1$  of the first ion path.

An example is discussed on the basis of Figs. 13, 14 and 15.  $\text{Ne}^{20}$  ions with 8 keV impinge at an angle of  $\psi = 15^\circ$  on a  $\langle 110 \rangle$  chain of a  $\text{Ni}^{58}$  crystal and are scattered from this chain. In the lower part of Fig. 13 the possible energies of the scattered ions are plotted with a dashed line as a function of the total scattering angle. In this case, only a scattering angle range of  $26.7^\circ$  to  $85.8^\circ$  is possible. For a fixed scattering angle in this range two energies

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<sup>+</sup>) The potential is of the form

$$V(r) = \frac{Z_1 Z_2}{r} e^2 \phi(r/a) \quad \text{with} \quad \phi(r/a) = 0.1 e^{-6 r/a} + 0.55 e^{-12 r/a} + 0.35 e^{-0.3 r/a}$$

and  $a = 0.8853 a_0 (Z_1^{2/3} + Z_2^{2/3})^{-1/2}$ .



are possible. All possible energies lie on a closed loop located above the (solid) binary collision curve according to formula (1).

The generation of a loop is a shadowing effect. When an ion impinges on a chain at a sufficiently acute angle, small collision parameters and hence large scattering angles become impossible (see Fig. 10). For the same reason there can be no arbitrarily small total angles of deflection because the ions departing at small angles to the chain are then scattered by the following ions at larger angles. The extent of the loop, i.e. the possible scattering angle range, depends on the angle of incidence  $\psi$ , the energy  $E_0$  of the incident ion, the atomic spacing  $d$  in the chain, and the atomic numbers  $Z_1$  (i.e. interaction potential) of the ions and atoms. As  $\psi$ ,  $d$  or  $E_0$  becomes smaller or  $Z_1$  becomes larger, the shadowing effect increases and the loop becomes smaller. If the angle of incidence  $\psi$  is further reduced, the loop contracts to a point so that the angles of incidence and emergence become equal. The ions are then reflected from the chain; the transverse energy is conserved. If  $\psi$  is increased, an angle of incidence is generated for which the loop is no longer closed at large angles. Further increases of  $\psi$  cause the scattering intensity in the high-energy branch to decrease fast and the low-energy branch approaches the binary collision curve. Multiple scattering is then only of minor importance.

The two energies belonging to one  $\vartheta$  can be roughly estimated by combining the individual angles of deflection. The bottom branch of the loop (Fig. 13 bottom, dashed) is characterized by the fact that the scattering from the chain is mainly governed by one large angle of deflection. The top branch (dot-dashed) is characterized by the fact that two collisions make an essential contribution to the total deflection (see Fig. 14). This becomes more obvious at lower energies or higher atomic numbers.

The two types of symmetric scattering, for which the angles of incidence and emergence are equal, i.e.  $\vartheta = 2\psi$ , A (with even number of single collisions) and B (with odd number of single collisions) do not lead to special points on the loop in Fig. 13. At point C the loop comes closest to the binary collision curve. This point is characterized by the fact that, when the total

scattering angle is split into the largest individual scattering angle  $\vartheta_{\max}$  and the residual angle  $\Delta\vartheta$ , i. e.  $\vartheta = \vartheta_{\max} + \Delta\vartheta$ ,  $\Delta\vartheta$  has a minimum. The contribution of the main scattering angle to the total scattering angle  $\vartheta_{\max}/\vartheta$  is plotted in the upper part of Fig. 13 for the high-energy (dot-dashed) and low-energy (dashed) branches of the loop. In the high-energy branch  $\vartheta_{\max}$  is always smaller than in the low-energy branch since it follows from formula (1) that

$$f(\vartheta_1 + \vartheta_2) \leq f(\vartheta_1) \cdot f(\vartheta_2) \quad .$$

The difference between the scattering processes leading to the high and low-energy branches of the loop are slight. This is seen in, for instance, the statistics of the individual scattering processes occurring (Fig. 14). In the low-energy branch the angle range  $5^\circ$  to  $20^\circ$  does not occur. On the other hand, the individual scattering angles above  $20^\circ$  and below  $5^\circ$  are much more frequent than in the high-energy branch. In Fig. 15 the size of the individual scattering angles  $\vartheta_i$  is plotted as a function of the number of collisions preceding. The strongest deflection in both branches occurs at the fourth single collision (with the definition chosen for the first collision).

The intensity distribution along the loop is represented in Fig. 16. It was determined from the statistics of the calculated ion paths. The intensity in the low-energy branch is somewhat higher than in the high-energy branch. It is highest at the ends of the loop. When making a comparison with the experiment, however, allowance should be made for the fact that only the scattered ions can be detected. The neutralization probability can, however, be different for different branches of the loop.

#### 4.2 COMPARISON OF THE EXPERIMENTAL RESULTS WITH COMPUTER CALCULATIONS

A few typical experimental backscattering spectra of 10 keV  $\text{Ne}^+$  ions from a vanadium  $\langle 110 \rangle$  chain are shown in Fig. 11. In many cases the two maxima for the doubly charged backscattered  $\text{Ne}^{++}$  ions are better separated than for  $\text{Ne}^+$ . An example is given in Fig. 17. In Figs. 18, 19 and 20 the energy values of the maxima as derived from the spectra were plotted versus

the scattering angle set. The solid curves represent the values calculated according to the model described in Section 4.1.

For the scattering angles at which two energies are predicted by the model two energies are also found experimentally. The existence of loops is thus confirmed. The spacing between the two maxima observed agrees very well with the theoretical width of the loops. (The narrow loop for  $\psi = 10^\circ$  in Fig. 19 was not resolved experimentally). The starting point of the loops, i. e. the smallest possible scattering angle  $\vartheta_{\min}$  agrees well with the calculated value within the margins of error. Heiland et al. /18/, on the other hand, observed in a  $\text{Ne}^+$  scattering experiment with a Ni target using energies below 1 keV a shift of the whole loop towards smaller scattering angles. The experimentally observed loops are appreciably shifted towards lower energies compared with those calculated. This can be explained in terms of inelastic energy losses, as in Sektion 3, which were not allowed for in the computer calculation. Furthermore, maxima in the backscattering spectra are also found outside the theoretically possible scattering angle range. These are then close to the theoretical binary collision curve. Their intensity, however, is much lower than would be expected from the increasing cross-section at small angles.

In the experiment the ions incident on the crystal surface are scattered not only from the isolated chains, as was assumed for the calculations. The majority of the ions rather impinge between the chains. Van Veen et al. /22/ have shown, however, that in a two-dimensional calculation loops quite similar to those due to scattering from isolated chains are obtained, there being an additional slight scattering intensity outside the loops. This contributes to the background in the backscattering spectra. Furthermore, no ideal chains are obtained in the experiment. If defects in the chains are assumed in the calculations, the calculations show that the scattering can occur for other combinations of energy and angle. This is discussed in the next section.

### 4.3 INFLUENCE OF LATTICE DEFECTS ON CHAIN SCATTERING

The following lattice defects were taken into account for the computer calculations: a single atom is missing in a long chain ("vacancy") a second superposed chain starts or terminates ("upward step" and "downward step"). The result of such a calculation for bombardment of vanadium with  $\text{Ne}^+$  ions along a  $\langle 110 \rangle$  chain is shown in Fig. 21. The loop due to scattering from the ideal chain is reduced to a point (denoted by "chain" in Fig. 21) owing to the small bombardment angle  $\psi = 8^\circ$ . To the left of this point scattering intensity is obtained by downward steps, and to the right a new loop by upward steps, the low-energy branch practically coinciding with the binary collision curve. Within these loops scattering intensity is obtained on a new, narrow loop by defects in the chains (denoted by "vacancy").

This thus explains why scattering intensity was found in Figs. 18, 19 and 20 where it would not be expected after scattering from ideal chains. It is difficult to decide, however, what kind of chain defects are the main cause of the prohibited scattering intensity.

The energies of the target atoms sputtered in head-on-collisions (denoted by " $\text{V}^+$  sputtered") are well below the binary collision curve. They are therefore easy to detect experimentally with an energy analyzer alone. This is no longer true when the incident ions have higher masses than the target atoms. The calculation of  $\text{Kr}^+$  scattering from copper ( $M_1 > M_2$ ) is represented in Fig. 22. Here the gas ions scattered from ideal and imperfect chains intersect the curve of the "sputtered" target atoms several times. Measured spectra are therefore hard to interpret if energy analysis is not accompanied by mass analysis. On the other hand, the curves are far apart as regards energy, and so they are easier to resolve experimentally. The scattering of  $\text{Kr}^+$  with simultaneous energy and mass analyses was investigated in detail by Begeman et al /23/. They do not, however, find any target ions for  $\vartheta = 30^\circ$  and various  $\psi$ , as would be expected from Fig. 22. This may have various reasons. In Fig. 22 (and Fig. 21) there is no information on the intensity with which certain combinations of energy and scattering angle occur. Furthermore, the probability of the target ions being emitted in the charged state becomes less than that of the bombardment ions.

Figure 23 shows a few experimental backscattering spectra of 8.22 keV  $\text{Kr}^+$  from the  $\langle 110 \rangle$  chains of vanadium. The energy of the target atom scattered in the binary collision is denoted by " $\text{Kr}^+ \rightarrow \text{V}$ ", that of the target atoms "sputtered" in the binary collision by " $\text{V}^+$ ". This figure shows how the shape of the backscattering spectra and their intensity already changes with relatively small changes of the angle of incidence  $\psi$  for a fixed scattering angle  $\theta = 30^\circ$ . To interpret these spectra, even more extensive computer calculations are required. It will then perhaps be possible, however, to decide from the backscattering spectra how strongly disturbed the surface of the investigated crystals is.

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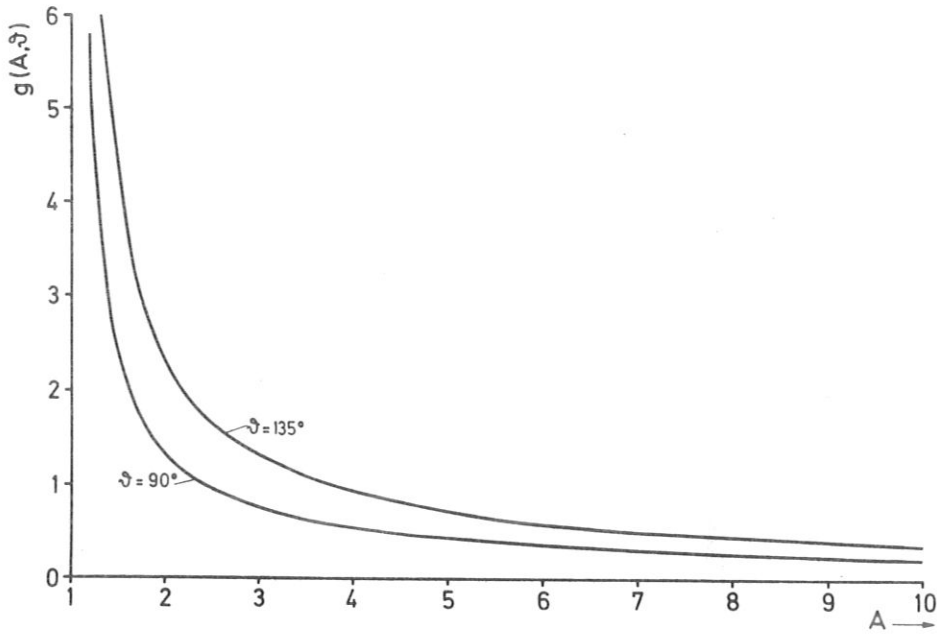


Fig. 1: The function  $g(A, \vartheta)$  governing the mass resolution, versus  $A = \frac{M_2}{M_1}$  for the scattering angles  $\vartheta = 90^\circ$  and  $\vartheta = 135^\circ$  (acc. to formula (2) on p.2).

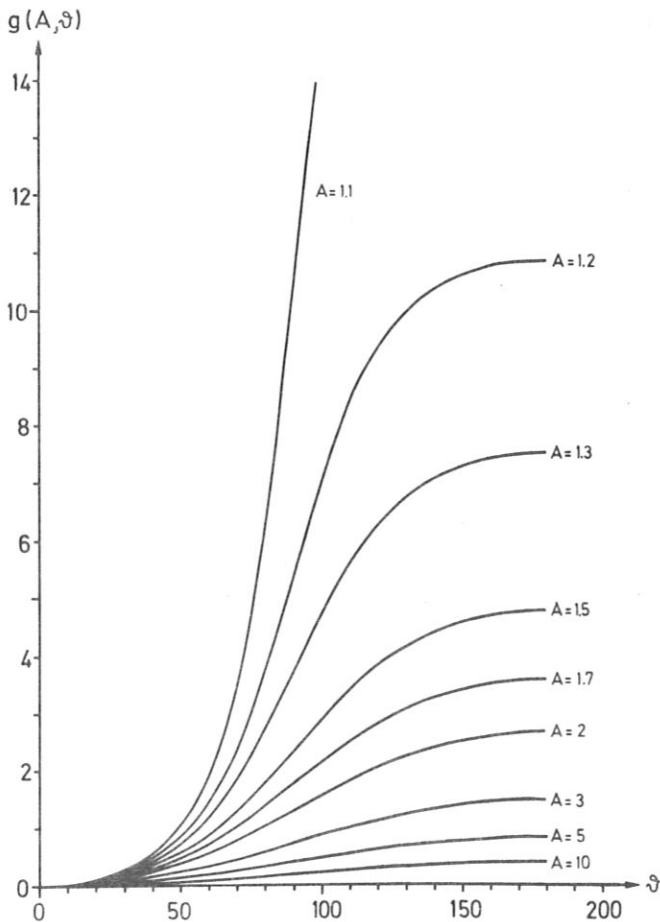


Fig. 2:

The function  $g(A, \vartheta)$  governing the mass resolution, versus the scattering angle  $\vartheta$  (acc. to formula (2) on p. 2).

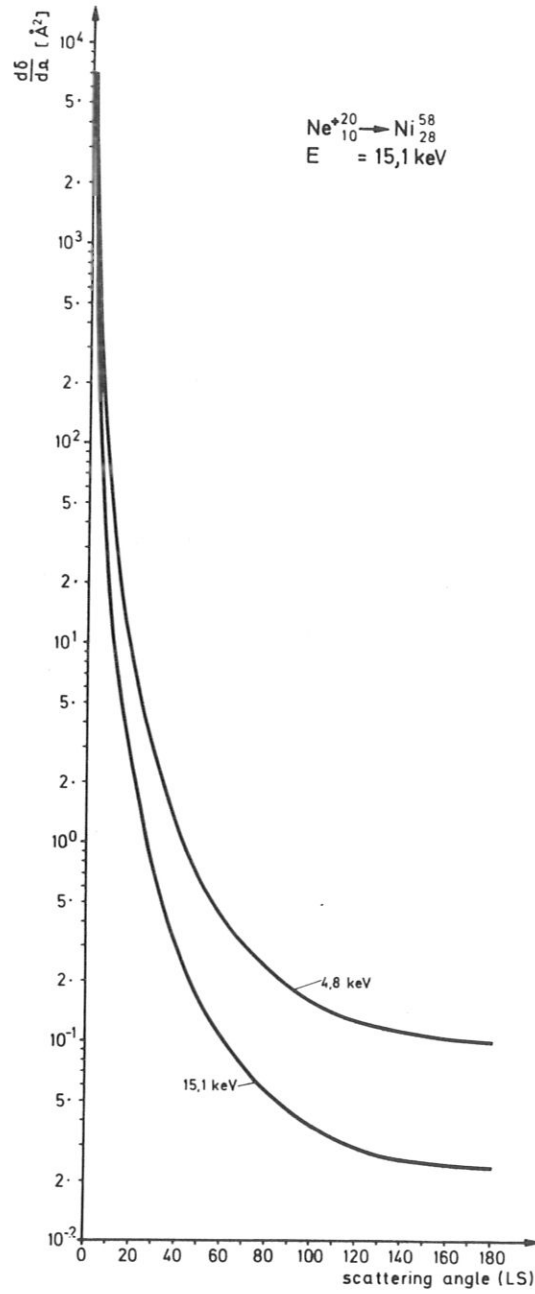


Fig. 3: Differential scattering cross-section versus the scattering angle in the laboratory system (acc. to M.T.Robinson (9) ).



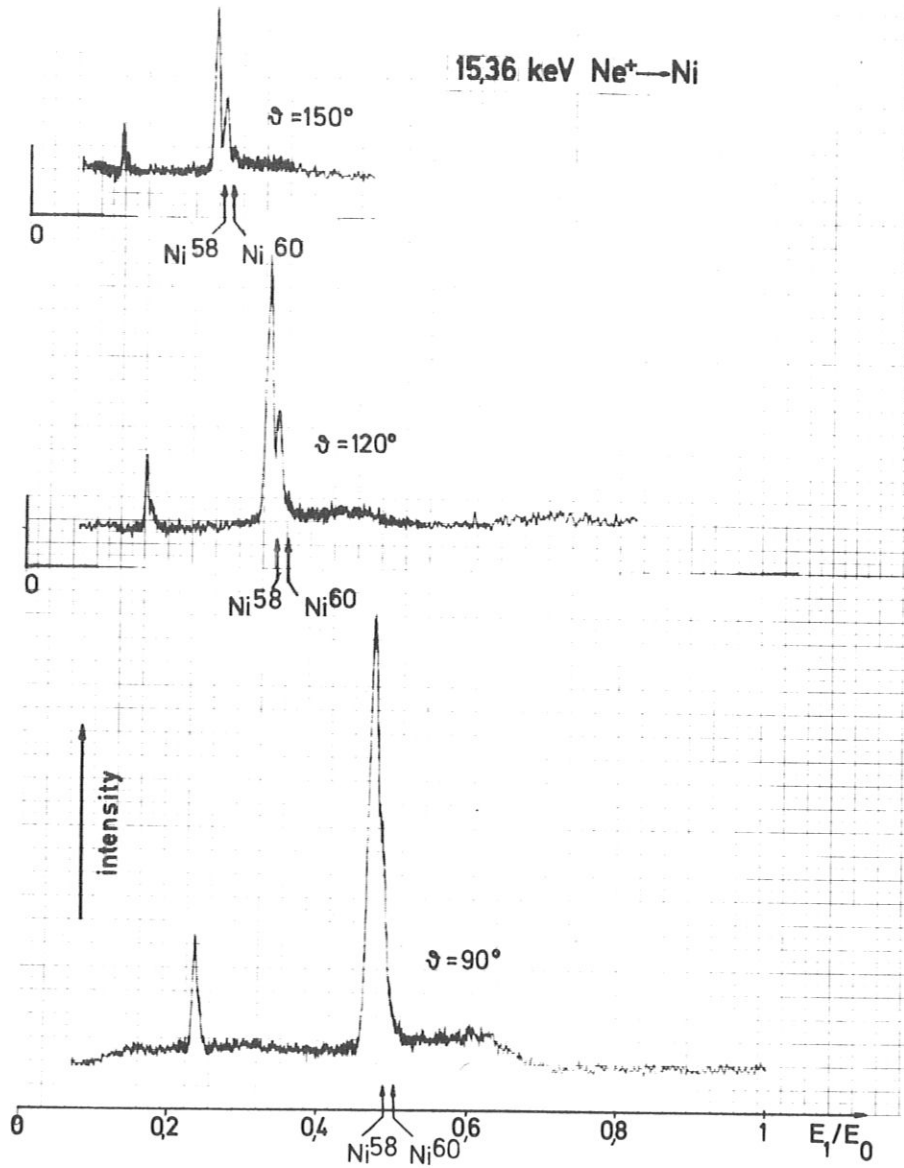


Fig. 4: Backscattering spectra of 15.36 keV  $\text{Ne}^+$  ions from Ni for the scattering angles  $\vartheta = 150^\circ$ ,  $120^\circ$  and  $90^\circ$ .

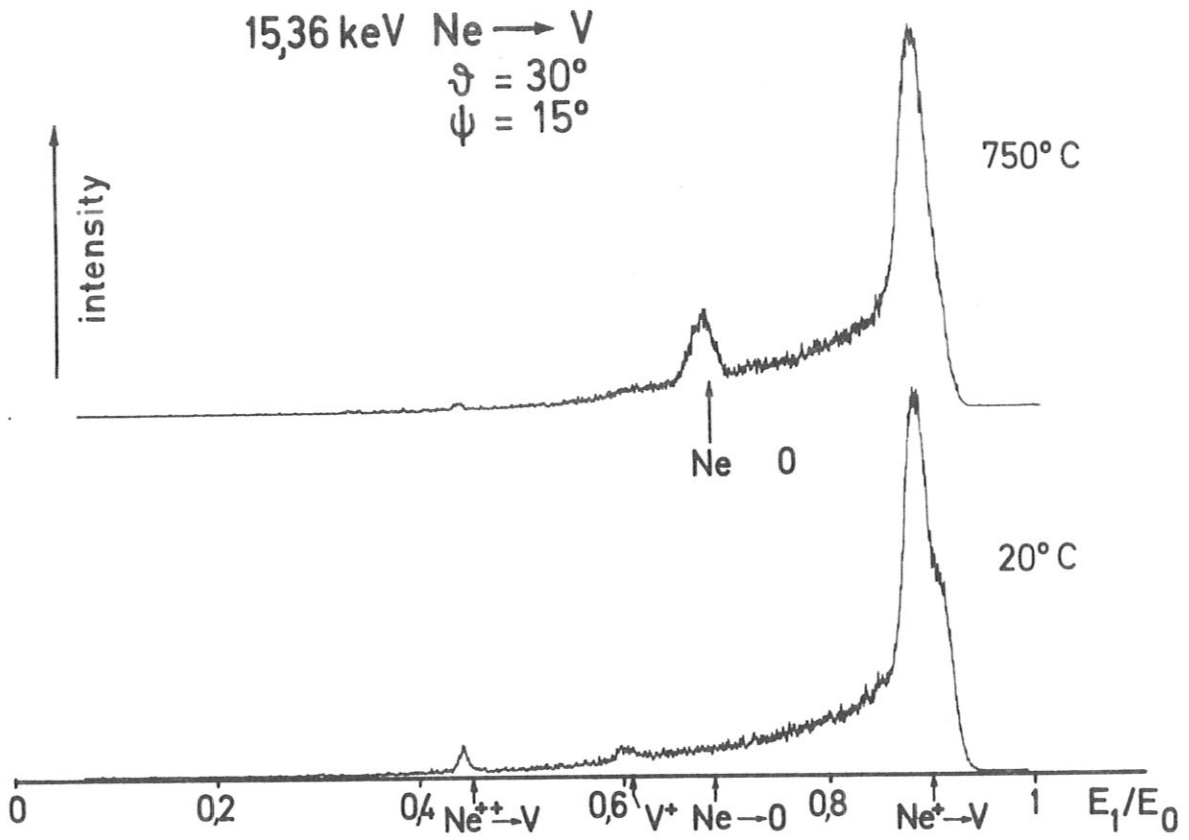


Fig. 5: Backscattering spectra of 15.36 keV  $\text{Ne}^{+}$  ions from vanadium.

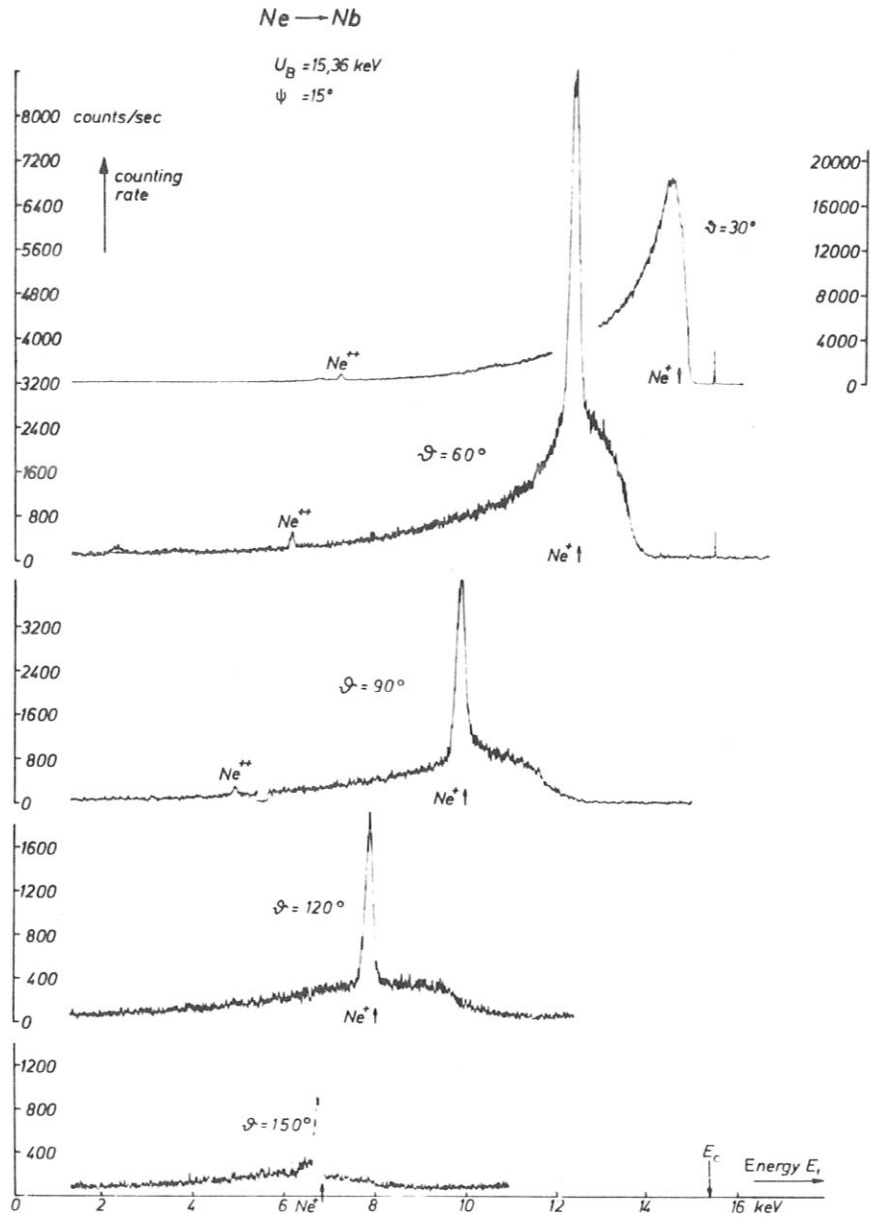


Fig. 6: Backscattering spectra of 15.36 keV  $Ne^+$  ions from polycrystalline Nb for various scattering angles  $\vartheta$ .

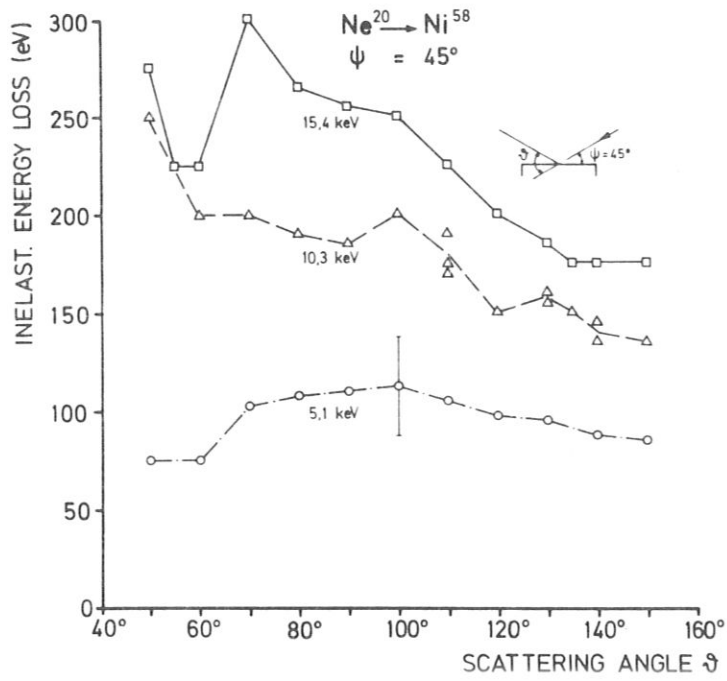


Fig. 7: Inelastic energy losses of Ne<sup>+</sup> ions scattered from Ni, as a function of the scattering angle  $\vartheta$ .

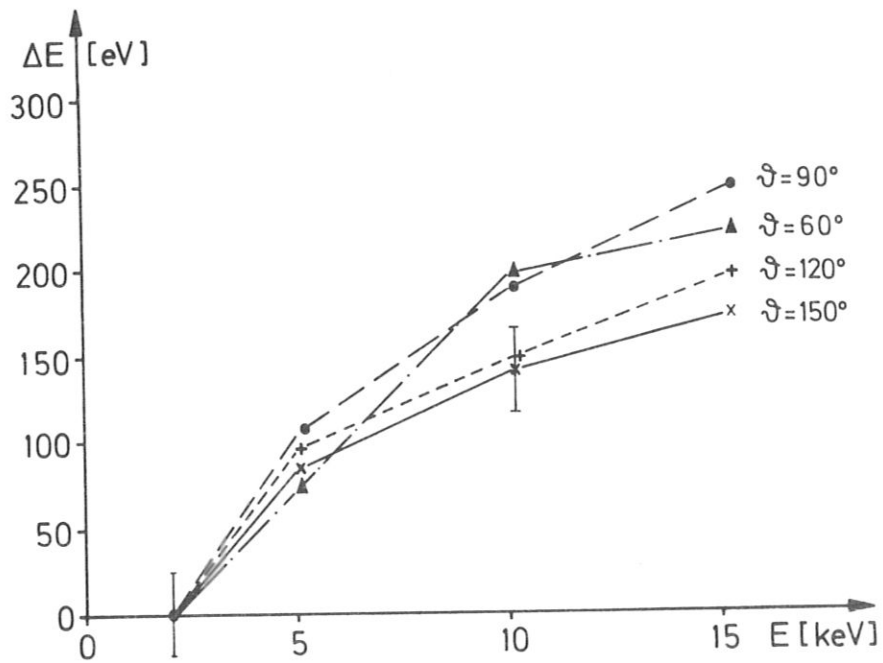


Fig. 8: Inelastic energy losses of Ne<sup>+</sup> ions scattered from Ni, as a function of the primary energy  $E_0$ .

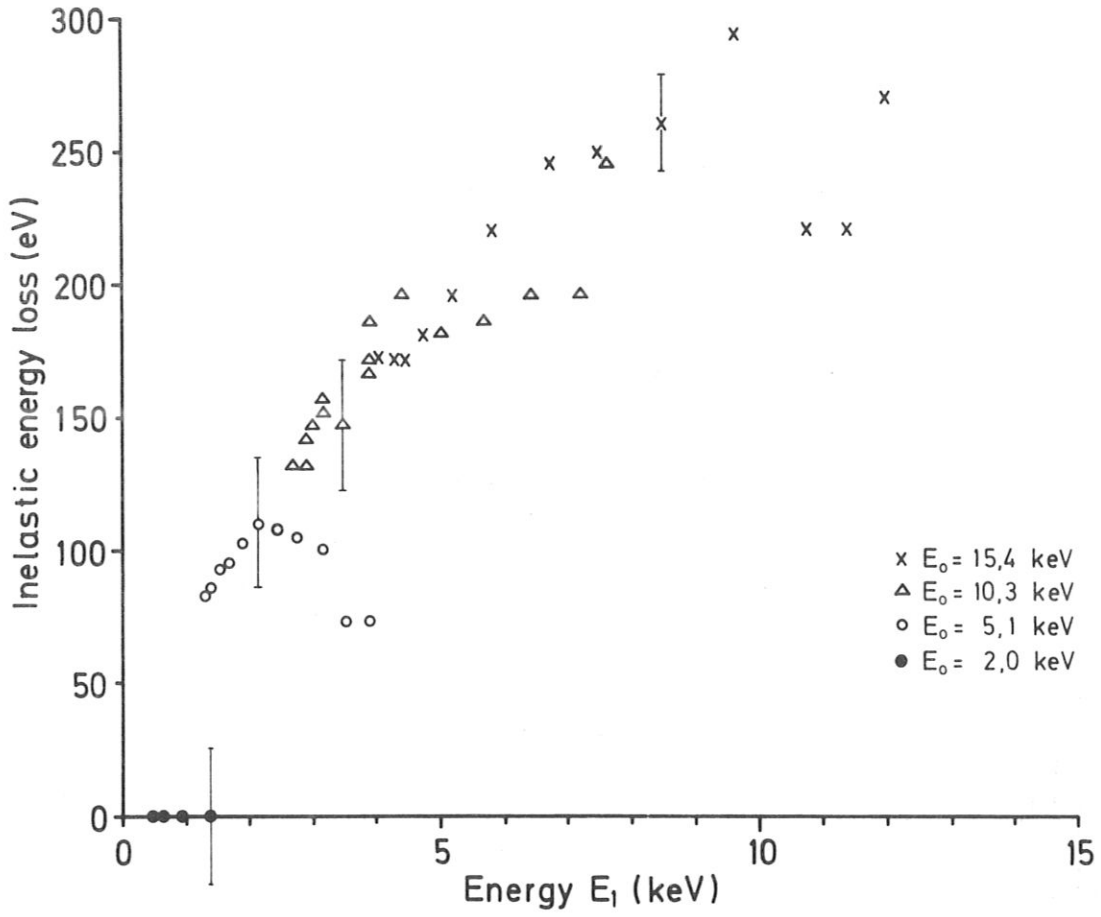


Fig. 9: Inelastic energy losses of Ne<sup>+</sup> ions scattered from Ni, as a function of their energy E<sub>1</sub> after scattering.

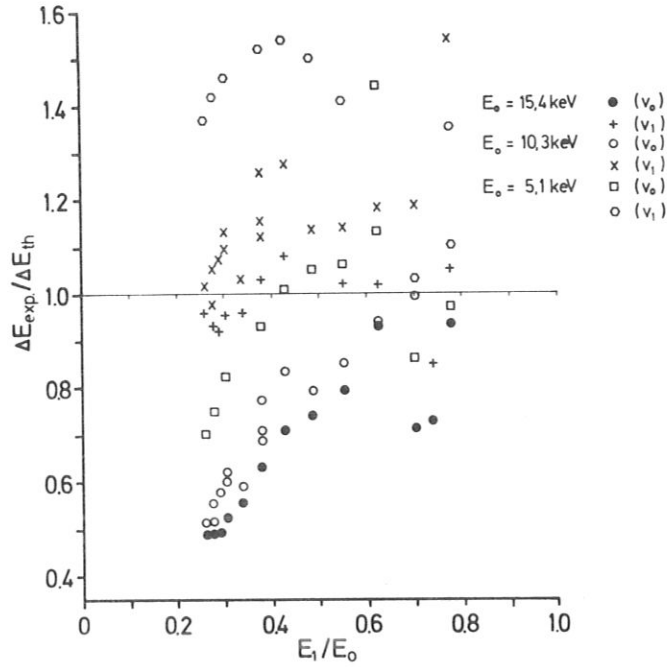


Fig. 10: Ratio of experimentally observed inelastic energy losses and theoretical values according to formula (3) page 4 as a function of the energy after elastic collisions.

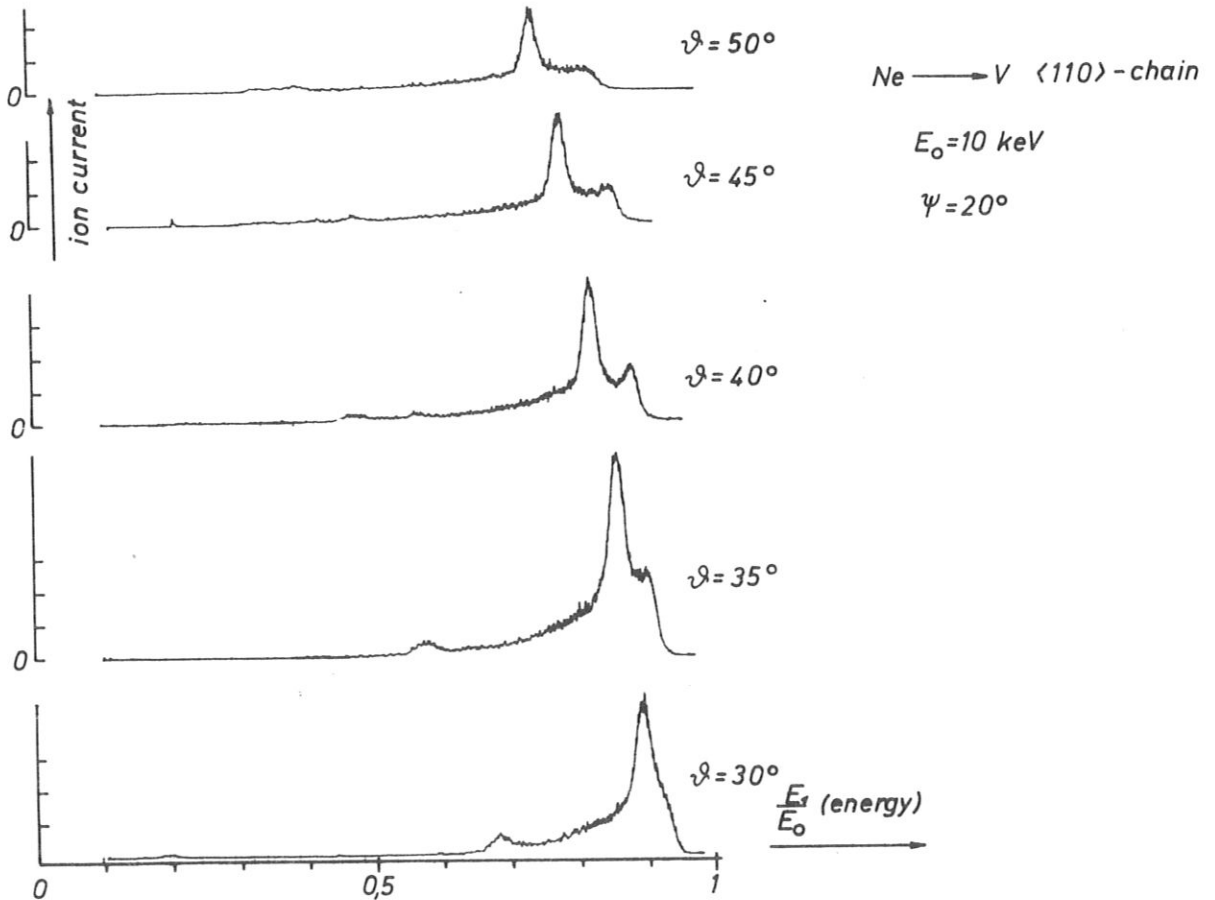


Fig. 11: Energy distributions of 10 keV  $\text{Ne}^+$  ions scattered from a  $\langle 110 \rangle$  chain of vanadium.

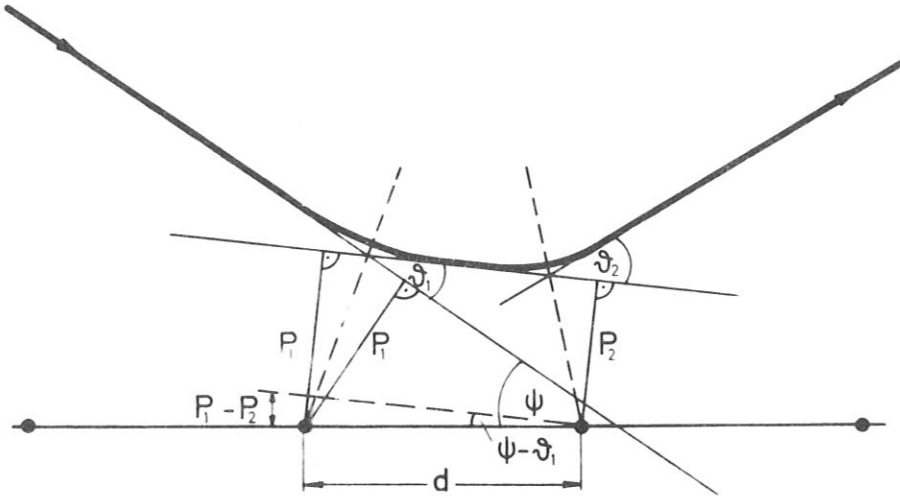


Fig. 12: Calculation of the impact parameters  $p$  in scattering from an atomic chain (acc. to Parilis /11/ ).

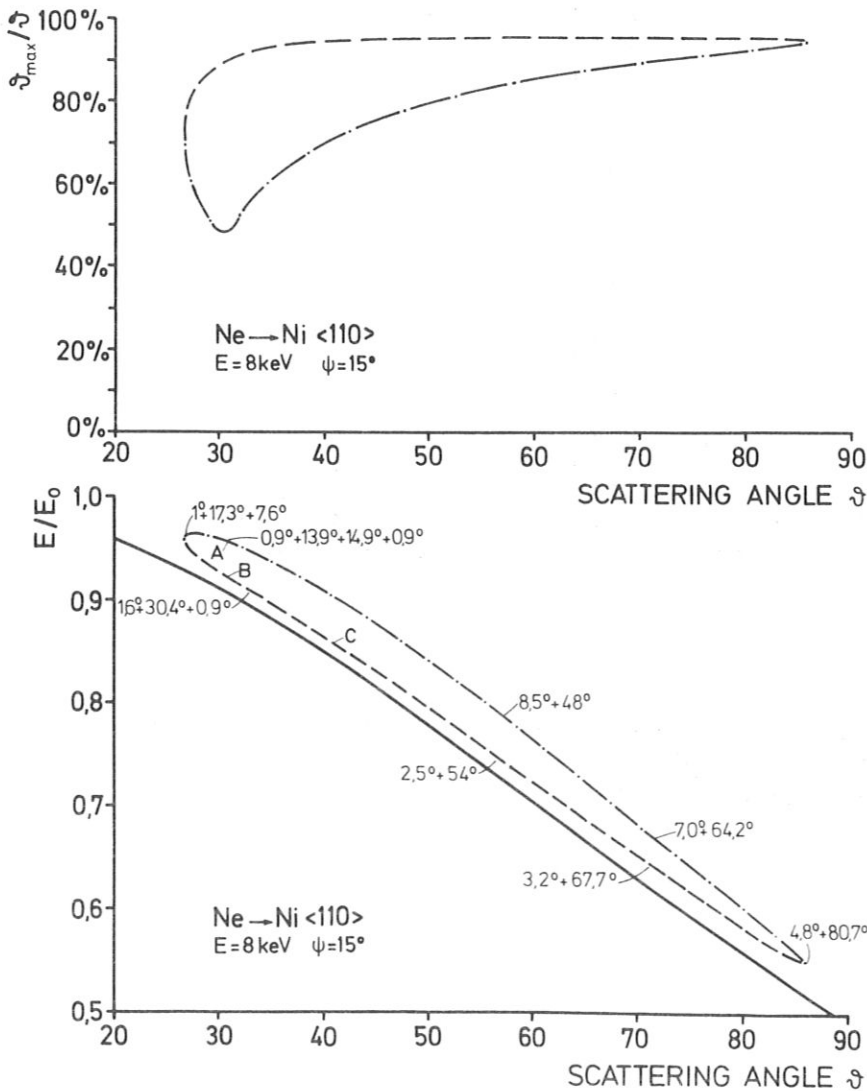


Fig. 13:

Bottom:

Possible energies of  $Ne^+$  ions scattered from an ideal  $\langle 110 \rangle$  chain of Ni (dashed, dot-dashed curves). The solid line represents the binary collision curve acc. to formula (1) (for points A, B, C see text).

Top:

Contribution of the largest individual scattering angle  $\theta_{max}$  to the total scattering angle  $\theta$  for the loop below.

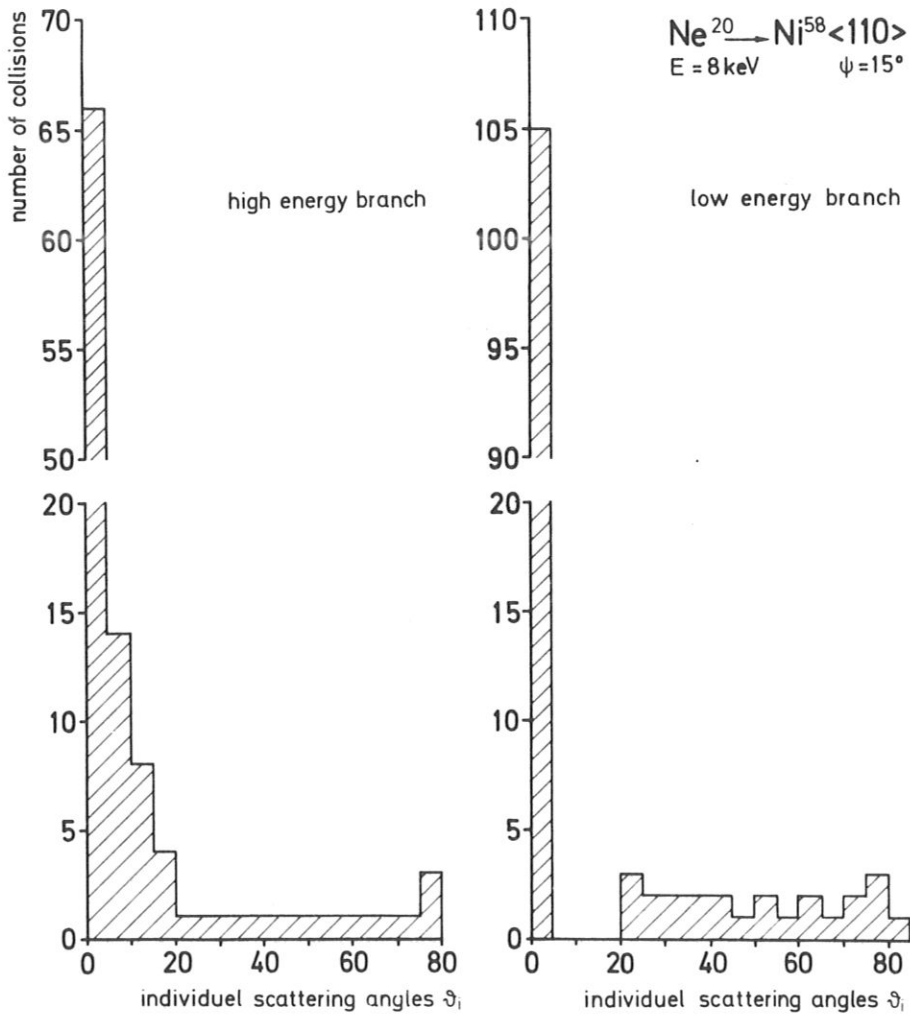


Fig. 14: Statistics of the single collision processes for the multiple scattering of  $\text{Ne}^+$  from a  $\langle 110 \rangle$  chain of Ni.



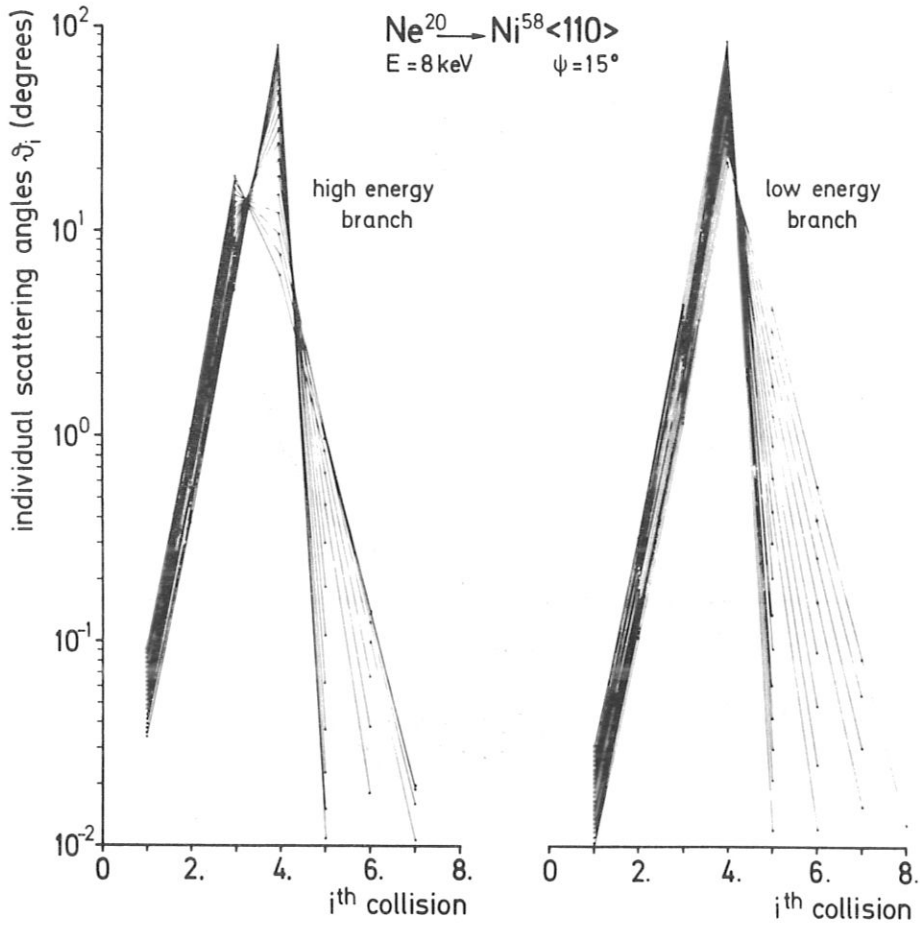


Fig. 15: Size of the individual scattering angles as a function of collision number in multiple scattering of Ne<sup>+</sup> from a <110> chain of Ni.

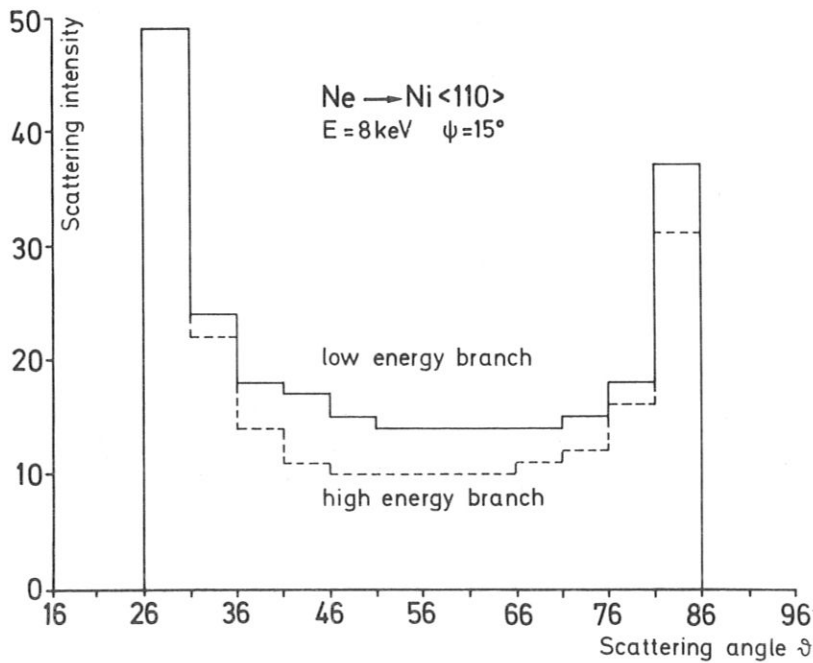


Fig. 16: Intensity distribution along the loop.

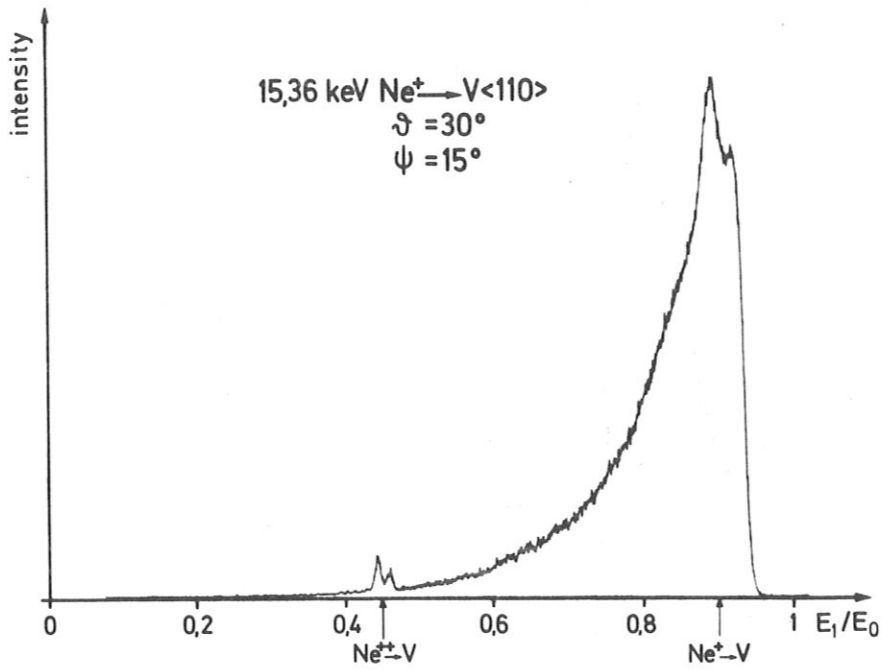


Fig. 17: Backscattering spectrum of  $\text{Ne}^+$  from a  $\langle 110 \rangle$  chain of V.

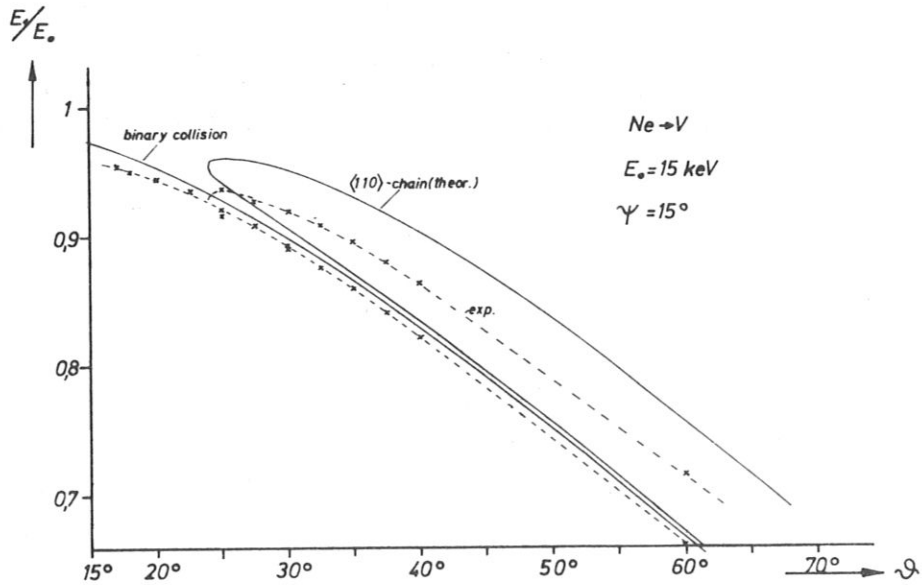


Fig. 18: Comparison of the theoretical and experimental energies of  $\text{Ne}^+$  ions with a primary energy of 15 keV scattered from a  $\langle 110 \rangle$  chain of V.

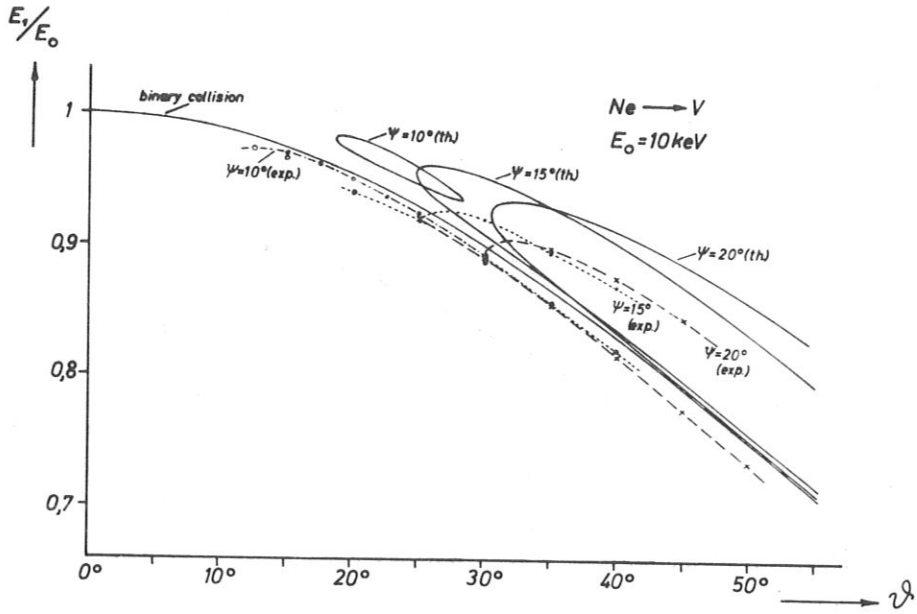


Fig. 19: like Fig. 18, but  $\text{Ne}^+$  with primary energy of 10 keV.

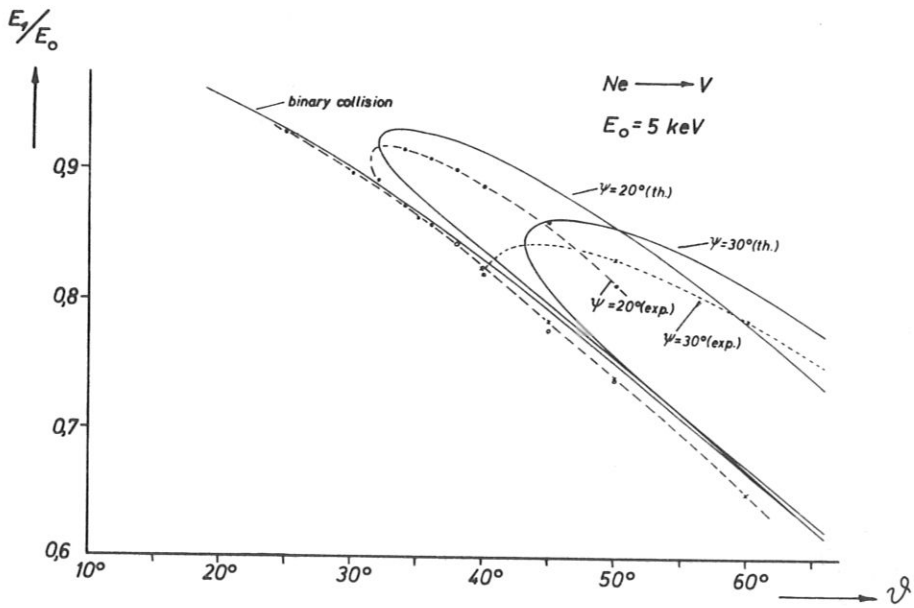


Fig. 20: like 18 and 19, but  $\text{Ne}^+$  with a primary energy of 5 keV.

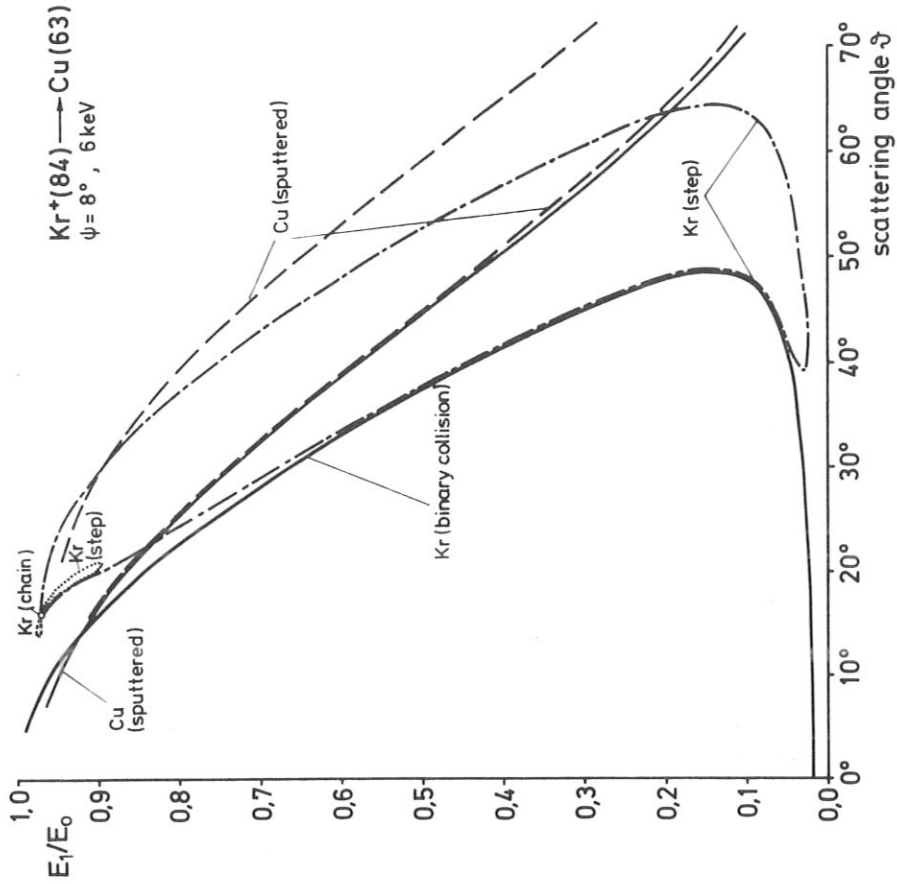


Fig. 22: Theoretical energy values of 6 keV  $Kr^+$  ions scattered from imperfect chains on copper.

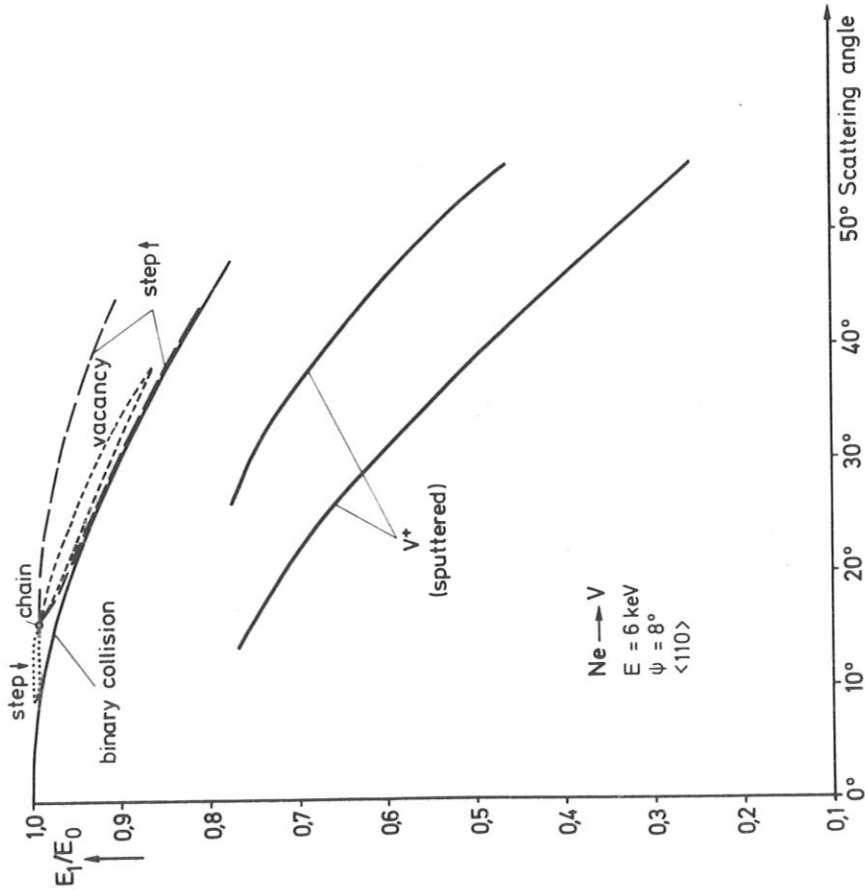


Fig. 21: Theoretical energy values of 6 keV Ne ions scattered from imperfect  $\langle 110 \rangle$  chains of vanadium.

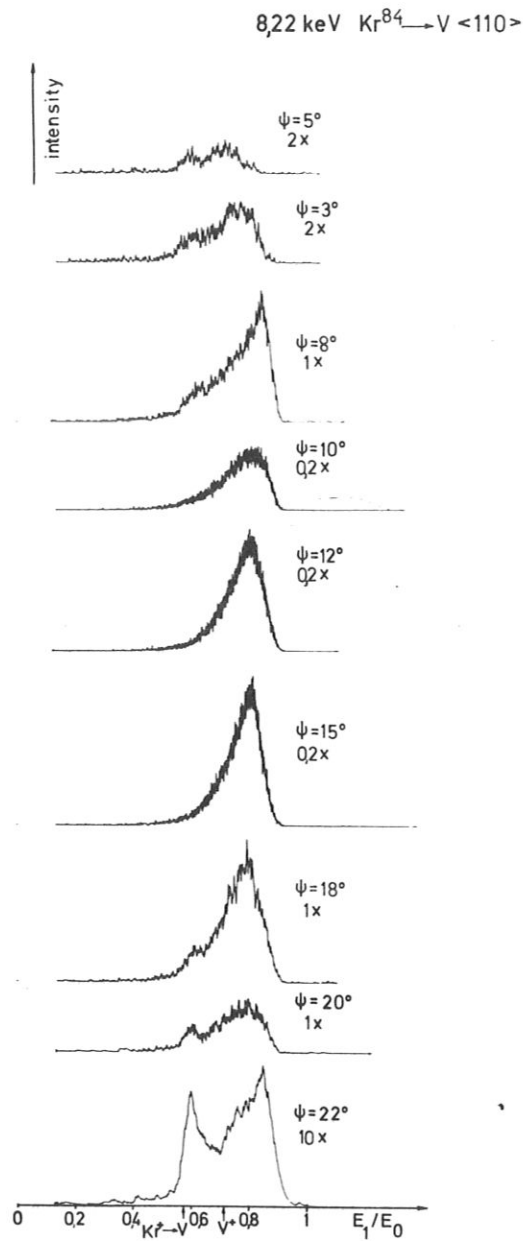


Fig. 23: Backscattering spectra of 8.22 keV  $\text{Kr}^{84}$  ions from  $\langle 110 \rangle$  chains of vanadium for a scattering angle  $\theta = 30^\circ$ .

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