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Supporting Information

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Keywords: carbon nitride thin film, directed growth, charge transfer, photocatalysis

Table S1. Carbon/nitrogen atomic ratio of Ph-CN thin films depending on the S mass mixed with the precursor.

S added (mg)	0	100	400	600	1200
C/N (atom)	0.85	0.86	0.90	0.94	0.95

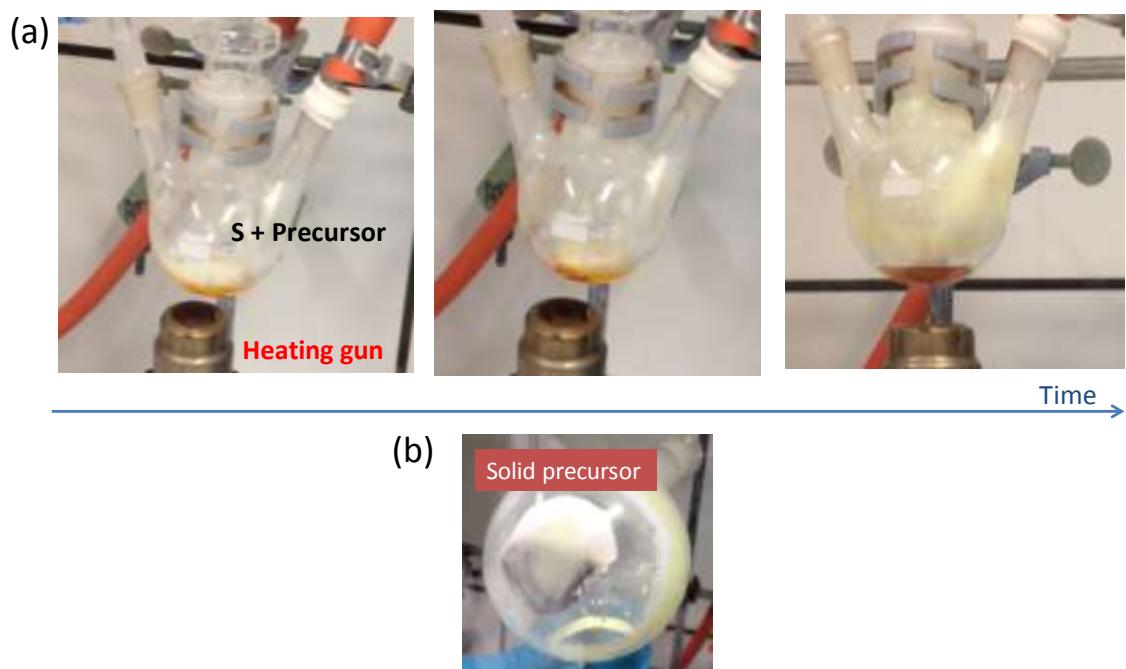


Figure S1. a) In-situ observation of the thermal behaviors of the mixture of S and the precursor, b) phase separation of the product by heating mixed S and cyanuric acid-melamine supramolecular complex that does not melt at elevated temperature.

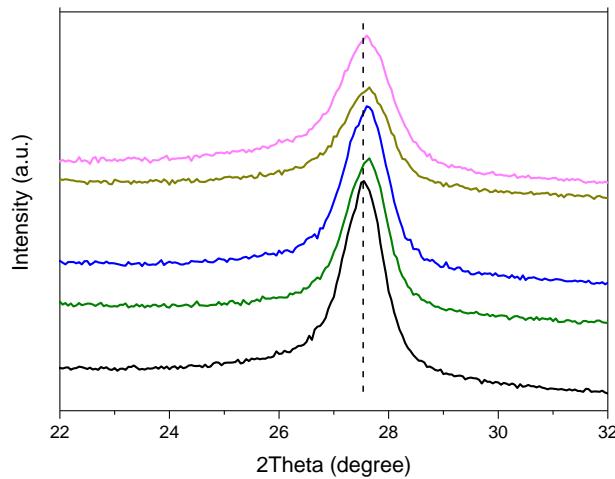


Figure S2. Zoomed in XRD pattern of the Ph-CN substrate, sulfur amount increasing from bottom to top.

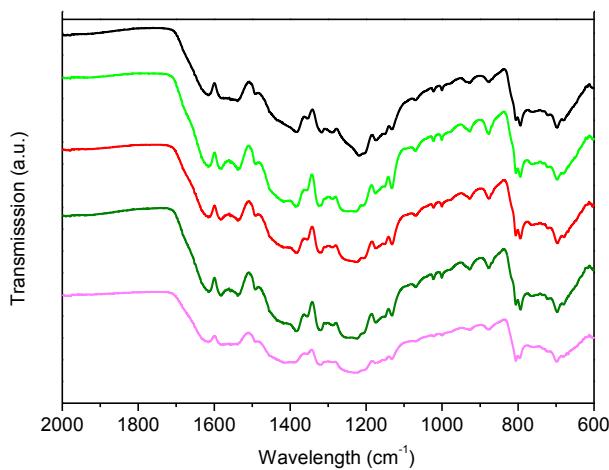


Figure S3. FTIR spectra of Ph-CN substrate with various S added, from top to bottom: 0, 100, 400, 600 and 1200 mg.

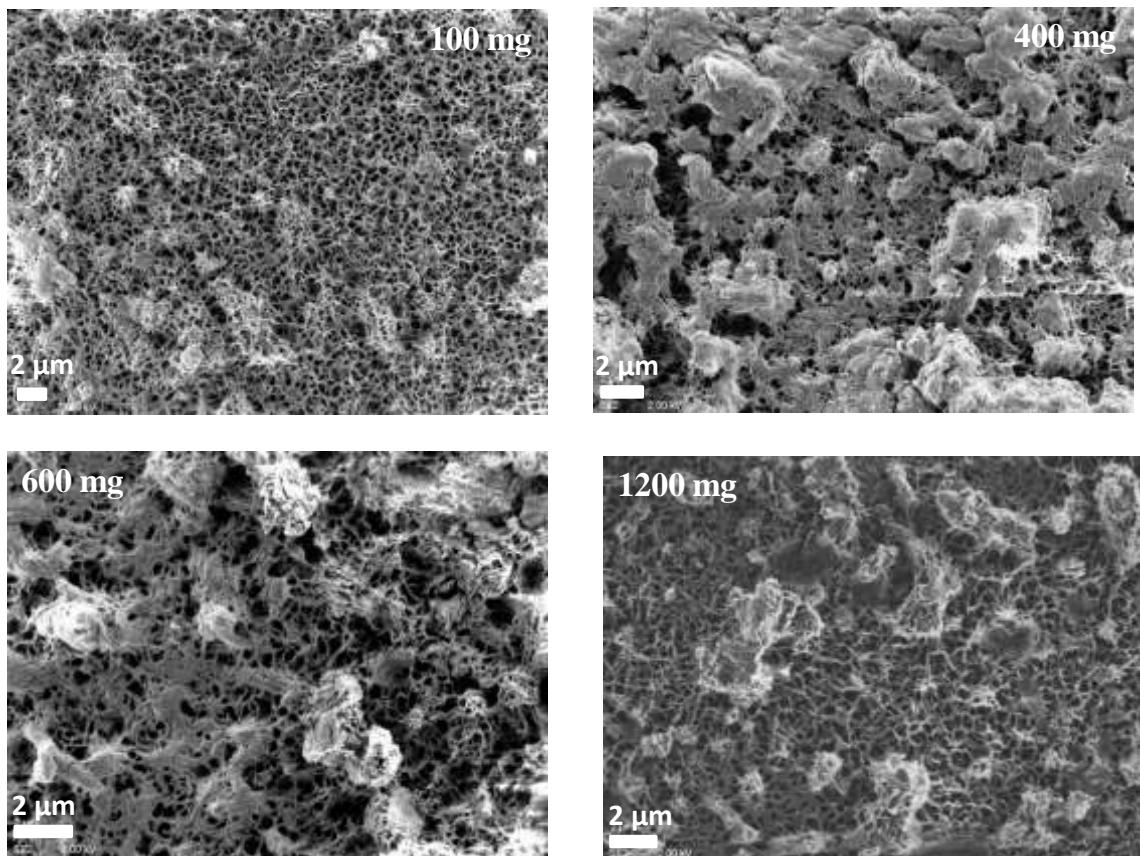


Figure S4. Top-view SEM images of Ph-CN thin films with different amount of S added.

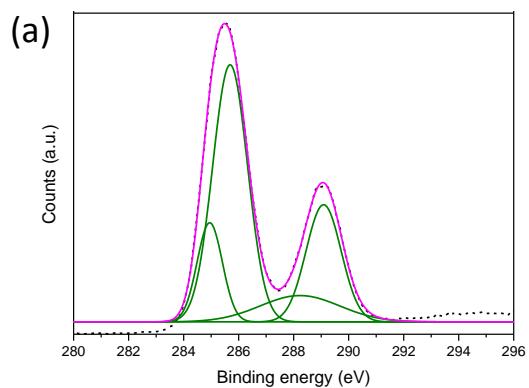


Figure S5. Fitted C 1s XPS spectrum of Ph-CN substrate.

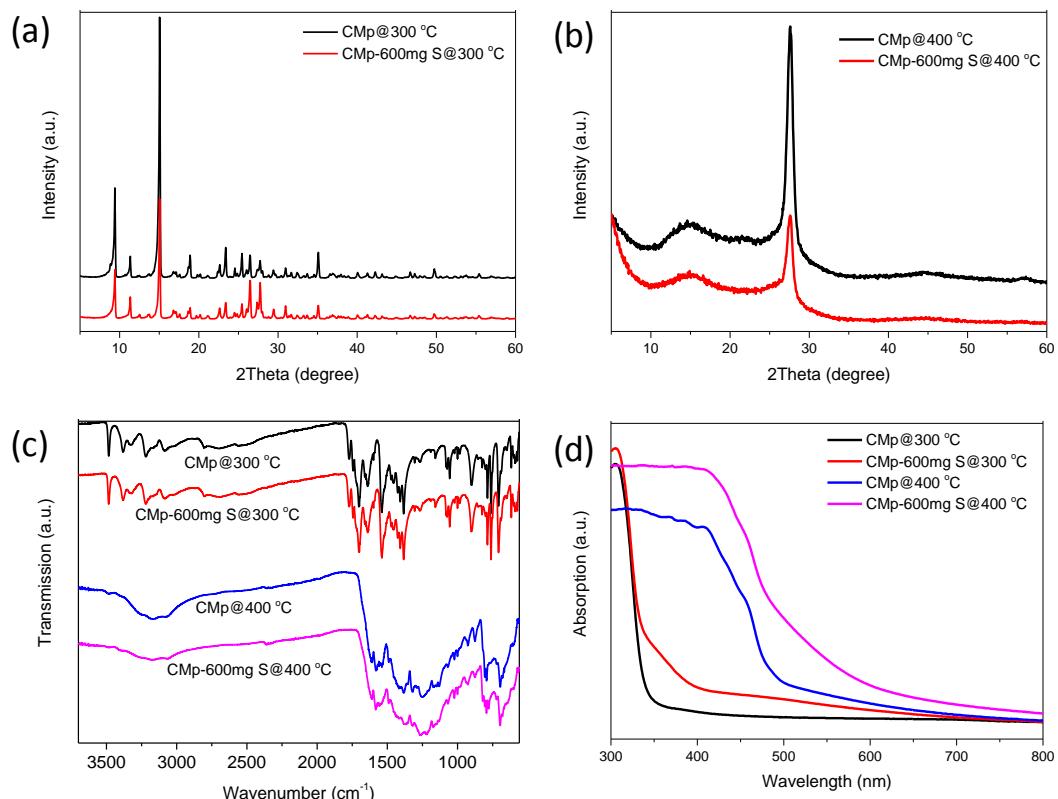


Figure S6. a,b) XRD patterns,c) FTIR spectra and d) UV-vis spectra of the intermediates obtained at 300 and 400 °C without and with 600 mg sulfur.

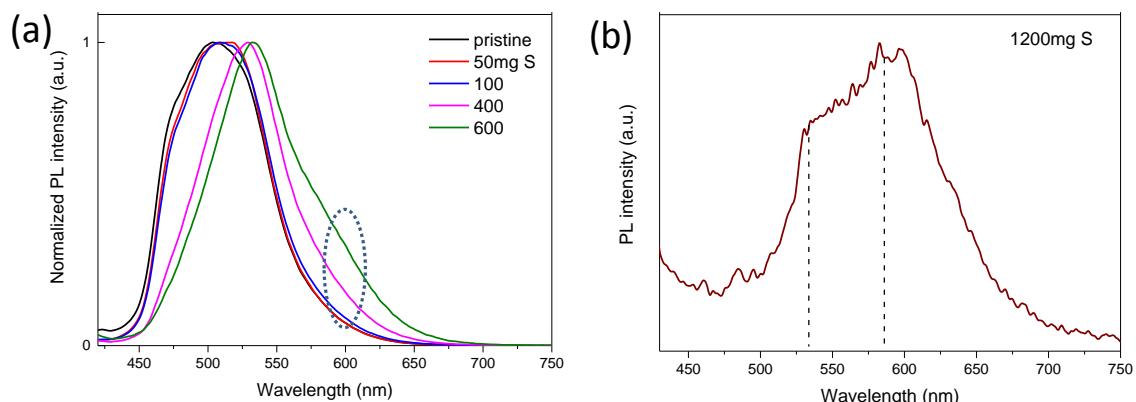


Figure S7. a) Normalized PL spectra of Ph-CN substrates with different S amount, b) PL spectrum of Ph-CNS₁₂₀₀.