Analysis of general rate model of linear chromatography considering finite rates of the adsorption and desorption steps

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Abstract

This article presents semi-analytical solutions and temporal moments of the general rate model of chromatography with a focus on evaluating the effect of finite rates for the adsorption and desorption steps, typically considered to be in equilibrium. The model equations are analytically solved in the Laplace domain and numerical Laplace inversion is applied to get back solutions in the actual time domain. The expression of first four temporal moments are derived from the analytical solutions in the Laplace domain. The derived analytical solutions and moments are helpful tools to predict dynamic behaviors inside the column and to evaluate the influence of model parameters on the elution profiles, in particular the effect of finite rates of the intrinsic adsorption and desorption steps. The correctness of analytical solutions are verified through the numerical solutions of a high resolution finite volume scheme. Several case studies are considered to quantify effects of the rate constants for adsorption and desorption, axial dispersion, film mass transfer resistance, intraparticle diffusion resistance, and inlet boundary conditions on the elution profiles.

Keywords: Chromatography, general rate model, dynamic sorption kinetics, analytical solutions, moment analysis.

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1. Introduction

Chromatography is known as one of the powerful separation and purification techniques used in petrochemical, fine chemical, pharmaceutical, biotechnical and food industries. For instance, this technique is applied to separate chiral molecules, enzymes, sugar and to purify proteins or to produce insulin. It has capability to rapidly produce milligrams to tons of purified products. In particular, this technology is highly effective within those process industries where because of physicochemical limitations traditional separation operations such as distillation, solvent extraction, crystallization, evaporation are not applicable. In the last few decades, this technique has achieved considerable industrial popularity, see e.g. Ruthven (1984); Guiochon and Lin (2003); Guiochon et al. (2006).

Chromatographic techniques are based on selective adsorption of mixture compounds on a solid phase (or liquid phase) with high surface area. In liquid-solid column chromatography, a mobile (liquid) phase, carrying mixture components is passed through a tabular column containing the stationary (solid) phase. During migration, each component of the mixture interact to varying degrees with the stationary phase. Thus, components of the mixture are continuously partitioned between solid (adsorbent) and mobile phases. Components having strong interaction with the stationary phase propagate slowly along the column compared to the weakly interacting components. A complete separation of the components can be achieved if the column length is long enough Ruthven (1984); Guiochon et al. (2006). Other applications of adsorption phenomena can be found in Lemlikchi et al. (2015); Kitous et al. (2016).

Mathematical modeling plays an important role in the design, understanding, and optimization of several chemical engineering processes. Modeling of the chromatographic columns are helpful to understand the transport mechanism and to approximate chemical-physical parameters from the analysis of experimental data. It offers a technique for predicting the dynamic behaviors of the solute in the columns without carrying out expensive and time-consuming experiments in the laboratories. The numerical tools are often more valuable than the conventional experimental methods in terms of providing more

profound insights and complete information that cannot be directly measured or observed, see Guiochon et al. (2006); Schmidt-Traub et al. (2012). A common practice in the mathematical modeling of chromatographic system is to consider a column packed with mono dispersed spherical particles and to write unsteady-state mass balances of both the mobile and the stationary phases and to impose some simplifying assumptions incompliance with the physics of the system. This leads to a system of partial differential equations (PDEs) coupled with some algebraic and/or differential equations.

In the literature, a number of dynamic models have been introduced to simulate the chromatographic process, see for example Ruthven (1984); Guiochon and Lin (2003); Guiochon et al. (2006). Each model considers a different level of complexity to describe the process. The most important to mention are the equilibrium dispersive model (EDM), the lumped kinetic model (LKM), and the general rate model (GRM). The general rate model (GRM) is regarded as the most comprehensive model for the simulation of fixed-bed adsorbers. In this model, the axial dispersion and all the mass-transfer resistances are taken into account, such as the external mass transfer of adsorbate from the bulk phase to the external surface of the adsorbent, the diffusion transport through the pores of the adsorbent, and the adsorption-desorption kinetics at the active sites. Herby, the latter aspect is studied less frequently. In most models applied to simulate chromatographic processes, permanently established adsorption equilibria are assumed. The GRM can be used to study the effects of different mass-transfer kinetics on the elution profiles. Either analytical or numerical solution techniques can be used to solve the model equations, depending on the degree of tractability of the governing system of equations.

The analytical solutions of the packed-beds for mono-dispersed solid spherical particles dates back to the pioneering works of Hougen and Marshal (1947), Thomas (1948, 1951), and Rosen et al. (1952). Afterwards, several other researchers presented analytical solutions of different models for fixed-bed absorber, see e.g. Rasmuson (1985); Carta (1988); Li et al. (2003, 2004); Guiochon et al. (2006); Shams (2014) and references therein.

The analytical solutions of different chromatographic models in the Laplace domain have been recently derived considering the assumption of a linear isotherm, see e.g. Felinger and Guiochon (2004); Javeed et al. (2013); Miyabe (2014); Qamar et al. (2014, 2015). When the Laplace transform is applied in terms of time variable, a function of time is transformed into a function of complex frequency. Now, the phenomenon is converted to the Laplace domain in which certain operations can be carried out, for example temporal moments can be found using the Laplace transformed function. After deriving the solution in the Laplace domain, it requires a back transformation to the actual time domain. However, the analytical back transformation is not possible in all situations, for example in the current situation of GRM. Thus, the numerical Laplace inversion is applied in such a scenario, see Rice et al. (1995).

The temporal moment analysis is a useful technique to extract information about the retention equilibrium and the mass transfer kinetics in the column. It provides accurate information about band broadening caused by several mass transfer kinetic processes. The technique has been comprehensively discussed in the literature, see instance Kubin (1965a,b); Kucera (1965); Schneider and Smith (1968); Suzuki (1973); Wolff et al. (1979, 1980); Ruthven (1984); Lenhoff (1987); Gengliang and Zhide (1996); Antos (2003); Guiochon et al. (2006), Miyabe et al. (2003, 2007, 2009)

This study extends our previous analysis of general rate model from quasi-stationary sorption model to dynamic linear sorption kinetic model, see Qamar et al. (2014). The previous model is a limiting case of the current model when the adsorption rate constant is very large. Analytical solutions are derived for the two sets of boundary conditions (BCs), i.e. the Dirichlet and Danckwers BCs. For further analysis, the first four temporal moments are derived from the Laplace transformed solutions. The analytical expressions of height equivalent to theoretical plate (HETP) are derived for both sets of BCs. The influences of adsorption and desorption rates, axial dispersion, film mass transfer resistance, intraparticle diffusion resistance, and inlet boundary conditions on the elution profiles are discussed and analyzed.

The current study also extends and generalizes the recent work of Miyabe (2014) who already analyzed GRM with finite adsorption and desorption rates. The major novelty of this article include: (a) the derivation of analytical solutions for both Dirichlet and

Danckwerts BCs considering pulse injections of finite widths, (b) the derivation of analytical expressions for the first four temporal moments and equations of HETP curve for the considered two sets of boundary conditions, (d) consideration of different case studies to quantify effects of the rate constants for adsorption and desorption, axial dispersion, film mass transfer resistance, intraparticle diffusion resistance, and inlet boundary conditions on the elution profiles, (c) Correction of Eq. (3) of Miyabe (2014) by incorporating the pore and surface diffusions in appropriate manner, see Eq. (2) of this manuscript.

2. The GRM with finite adsorption and desorption rates

This study considers an isothermal adsorption column packed with porous particles. Concentration pulses of an adsorbate are periodically injected to a flowing stream passing through the column. The adsorption column is subjected to axial dispersion, external mass-transfer resistance, intraparticle diffusion resistance, and adsorption-desorption kinetics.

Based on the aforementioned assumptions, the mass balance for the bulk phase is given as

$$\frac{\partial C_b}{\partial t} + u \frac{\partial C_b}{\partial z} = D_b \frac{\partial^2 C_b}{\partial z^2} - \frac{3Fk_{\text{ext}}}{R_p} \left(C_b - C_p|_{r=R_p} \right). \tag{1}$$

In the above equation, C_b and C_p are the concentrations of a solute in the bulk of the fluid and in particle pores, respectively. The phase ratio F is defined as $F = (1 - \epsilon_e)/\epsilon_e$, where ϵ is the external porosity. Moreover, u is the interstitial velocity, D_b represents the axial dispersion coefficient, k_{ext} is the external mass transfer coefficient, R_p is the radius of the adsorbent, and t and z denote time and axial coordinate of the column. In addition, rdenotes the radial coordinate

The mass balance equation for the solute in the stationary phase can be expressed assuming two mechanisms of intraparticle transport, namely pore and surface diffusions:

$$\epsilon_p \frac{\partial C_p}{\partial t} + (1 - \epsilon_p) \frac{\partial Q_p}{\partial t} = \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \left[\epsilon_p D_p \frac{\partial C_p}{\partial r} + (1 - \epsilon_p) D_s \frac{\partial Q_p}{\partial r} \right] \right), \tag{2}$$

where Q_p is the non-equilibrium concentration of the solute in the stationary phase, ϵ_p is the internal porosity, D_p is the pore diffusivity, and D_s is the surface diffusivity. In this study, linear adsorption kinetics of finite rate are assumed. The following mass balance for the amount of the solute adsorbed on the stationary phase is used

$$\frac{\partial Q_p}{\partial t} = K_d \left(aC_p - Q_p \right),\tag{3}$$

where the well-known Henry's constant a, used in linear equilibrium models, is defined as

$$a = \frac{K_a}{K_d}. (4)$$

In the above equations, K_a and K_d denote the adsorption and desorption rate constants, respectively. Eq. (3) is a simplified adsorption-desorption kinetic expression assuming that solute loading is sufficiently low at all times so that the number of available sites for adsorption does not change as the solute moves through the column. For rapid desorption (and adsorption) steps, i.e. $K_d \to \infty$, Eq. (3) gives $Q_p = aC_p$. In that case, the current model reduces to the linear GRM recently analyzed by Qamar et al. (2014).

In the context of this work, we will study in particular the impact of finite K_d -values on the course of elution profiles. Eqs. (1) and (2) are connected at $r = R_p$ via the following expression that quantifies the temporal change of the average loading of the particles:

$$\left[\epsilon_p D_p \frac{\partial C_p}{\partial r} + (1 - \epsilon_p) D_s \frac{\partial Q_p}{\partial r}\right]_{r=R_p} = k_{\text{ext}} (C_b - C_p|_{r=R_p}).$$
 (5)

To simply the notations and reduce the number of variables, the following dimensionless quantities are introduced:

$$c_{b} = \frac{C_{b}}{C_{\text{inj}}}, \ c_{p} = \frac{C_{p}}{C_{\text{inj}}}, \ q_{p} = \frac{Q_{p}}{aC_{\text{inj}}} \ \tau = \frac{ut}{L}, \ \rho = r/R_{p}, \quad x = \frac{z}{L}, \ Pe = \frac{Lu}{D_{b}}, \ \eta_{p} = \frac{D_{p}L}{uR_{p}^{2}},$$

$$\eta_{s} = \frac{D_{s}L}{uR_{p}^{2}}, \ Bi_{p} = \frac{k_{\text{ext}}R_{p}}{D_{p}}, \ Bi_{s} = \frac{k_{\text{ext}}R_{p}}{D_{s}}, \ \xi = 3F\frac{k_{\text{ext}}L}{R_{p}}u.$$
(6a)

Moreover, a dimensionless desorption rate is expressed as

$$k_{\rm d} = \frac{K_d L}{u}. (6b)$$

In Eq. (6a), C_{inj} denotes the non-zero injected bulk concentration, Pe is the Peclet number based on column length, Bi represents modified Biot number, and η_p and η_s describe

the ratio of space time and interaparticle diffusion times. Using the above dimensionless variables, the model Eqs. (1) to (3) can be rewritten as

$$\frac{\partial c_b}{\partial \tau} + \frac{\partial c_b}{\partial x} = \frac{1}{Pe} \frac{\partial^2 c_b}{\partial x^2} - \xi \left(c_b - c_p |_{\rho=1} \right), \tag{7}$$

$$\epsilon_p \frac{\partial c_p}{\partial \tau} + (1 - \epsilon_p) a \frac{\partial q_p}{\partial \tau} = \frac{1}{\rho^2} \frac{\partial}{\partial r} \left(\rho^2 \left[\epsilon_p \eta_p \frac{\partial c_p}{\partial \rho} + (1 - \epsilon_p) a \eta_s \frac{\partial q_p}{\partial r} \right] \right), \tag{8}$$

and

$$\frac{\partial q_p}{\partial \tau} = k_d(c_p - q_p). \tag{9}$$

The Eqs. (7)-(9) are also subjected to the initial and boundary conditions.

The initial conditions for an initially regenerated column are given as

$$c_b(0,x) = 0$$
, $c_p(0,x,\gamma) = 0$, $q_p(0,x) = 0$, $\forall x, \gamma \in (0,1)$. (10)

For Eq. (8), the following boundary conditions at $\rho = 0$ and $\rho = 1$ are assumed:

$$\frac{\partial c_p}{\partial \rho}\Big|_{\rho=0} = 0, \qquad \frac{\epsilon_p}{Bi_p} \frac{\partial c_p}{\partial \rho} + \frac{1 - \epsilon_p}{Bi_s} a \frac{\partial q_p}{\partial \rho}\Big|_{\rho=1} = c_b - c_p|_{\rho=1}.$$
(11)

Appropriate inlet and outlet BCs are required for Eqs. (7). The following two sets of BCs are considered.

Boundary conditions of type I: Dirichlet inlet BCs

As a first choice, the simpler Dirichlet boundary conditions could be applied at the column inlet

$$c_b|_{x=0} = \begin{cases} 1, & \text{if } 0 < \tau \le \tau_{\text{inj}}, \\ 0, & \tau > \tau_{\text{inj}}, \end{cases}$$

$$(12a)$$

where $\tau_{\rm inj}$ is the dimensionless time of injection. The following Neumann boundary condition is applied at the outlet of a column of hypothetically infinite length, $x = \infty$

$$\frac{\partial c_b(\tau, \infty)}{\partial x} = 0. \tag{12b}$$

For sufficiently small dispersion coefficient, for example $D_b \leq 10^{-5} \ m^2/s$, this Dirichlet inlet boundary condition is well applicable.

Boundary conditions of type II: Danckwerts (or Robin) type inlet BCs

Alternatively, the Danckwerts (or Robin) type boundary condition could be applied at the column inlet (e.g. Danckwerts (1953))

$$-\frac{1}{Pe}\frac{\partial c_b}{\partial x} + c_b \bigg|_{x=0} = \begin{cases} 1, & \text{if } 0 < \tau \le \tau_{\text{inj}}, \\ 0, & \tau > \tau_{\text{inj}}. \end{cases}$$
 (13a)

At the outlet of the column of finite length (i.e. at x = 1), the following Neumann outflow boundary condition is applied:

$$\frac{\partial c_b(1,\tau)}{\partial x} = 0. \tag{13b}$$

3. Analytical solutions

In this section, semi-analytical solutions of the linear GRM are presented for Dirichlet (Eq. (12a)) and Danckwerts (Eq. (13a)) inlet boundary conditions. The current model can conveniently be solved by means of Laplace transformation. The Laplace transformation is defined as

$$\bar{c}(s,x) = \int_{0}^{\infty} e^{-s\tau}(\tau, x) d\tau, \quad \tau \ge 0.$$

By applying the above Laplace definition on Eq. (7), we obtain

$$s\bar{c}_b = \frac{1}{Pe} \frac{\partial^2 \bar{c}_b}{\partial x^2} - \frac{\partial \bar{c}_b}{\partial x} - \xi(\bar{c}_b - \bar{c}_p|_{\rho=1}). \tag{14}$$

The Laplace transformation of Eqs. (8) is expressed as

$$\epsilon_p s \bar{c_p} + (1 - \epsilon_p) a s \bar{q_p} = \frac{1}{\rho^2} \frac{\partial}{\partial \rho} \left(\rho^2 \left[\epsilon_p \eta_p \frac{\partial \bar{c_p}}{\partial \rho} + (1 - \epsilon_p) a \eta_s \frac{\partial \bar{q_p}}{\partial r} \right] \right). \tag{15}$$

Moreover, the Laplace transformation of Eqs. (9) is given as

$$\bar{q} = \frac{k_d \bar{c_p}}{s + k_d} \,. \tag{16}$$

On using Eq. (16) in (15), we obtain

$$\frac{\partial^2 \bar{c_p}}{\partial \rho^2} + \frac{2}{\rho} \frac{\partial \bar{c_p}}{\partial \rho} = A(s)\bar{c_p}, \qquad (17)$$

where

$$A(s) = s \frac{\epsilon_p + (1 - \epsilon_p) \frac{ak_d}{s + k_d}}{\epsilon_p \eta_p + (1 - \epsilon_p) \eta_s \frac{ak_d}{s + k_d}}.$$
(18)

The general solution of (17) is given as

$$\bar{c}_p(s,x) = \frac{1}{\rho} [k_1 \cosh \sqrt{A(s)}\rho + k_2 \sinh \sqrt{A(s)}\rho]. \tag{19}$$

Using the Laplace transformation of the outlet boundary condition (BC) in Eq. (11), i.e. $\frac{\partial \bar{c}_p}{\partial \rho}|_{\rho=0}$, Eq. (19) gives $k_1=0$. Thus, we obtain

$$\bar{c}_p(s,x) = \frac{k_2}{\rho} \sinh \sqrt{A(s)}\rho. \tag{20}$$

Now, taking the Laplace transformation of the inlet BC in Eq. (11), we obtain

$$\frac{1}{B(s)} \left. \frac{\partial \bar{c_p}}{\partial \rho} \right|_{\rho=1} = \bar{c_p} - \left. \bar{c_p} \right|_{\rho=1}, \tag{21}$$

where

$$B(s) = \frac{Bi_p Bi_s}{Bi_s \epsilon_p + (1 - \epsilon_p) Bi_p \frac{ak_d}{s + k_d}}.$$
 (22)

On using Eq. (20) in Eq. (21), we obtain

$$k_2 = \frac{B(s)\bar{c}_b}{(B(s) - 1)\sinh(\sqrt{A(s)}) + \sqrt{A(s)}\cosh(\sqrt{A(s)})}.$$
 (23)

Thus, Eq. (20) at $\rho = 1$ gives

$$\bar{c}_p(s,x)|_{\rho=1} = \frac{B(s)}{B(s) - 1 + \sqrt{A(s)} \coth(\sqrt{A(s)})} \bar{c}_b = f(s)\bar{c}_b,$$
(24)

where

$$f(s) = \frac{B(s)}{B(s) - 1 + \sqrt{A(s)} \coth(\sqrt{A(s)})}.$$
 (25)

On using Eq. (24) in Eq. (14), we get

$$\frac{\partial^2 \bar{c}_b}{\partial x^2} - Pe \frac{\partial \bar{c}_b}{\partial x} - \phi(s) Pe \bar{c}_b = 0, \qquad (26)$$

where

$$\phi(s) = s + \xi(1 - f(s)). \tag{27}$$

The general solution of Eq. (26) is given as

$$\bar{c}_b(s,x) = A_1 e^{\lambda_1 x} + B_1 e^{\lambda_2 x},$$
 (28)

where

$$\lambda_{1,2}(s) = \frac{Pe}{2} \left(1 \mp \sqrt{1 + \frac{4\phi(s)}{Pe}} \right) .$$
 (29)

The integration constants A_1 and B_1 can be obtained by using BCs either in Eqs. (12a) and (12b) or in Eqs. (13a) and (13b).

Boundary conditions of type I: Dirichlet BCs

Here, we consider the boundary conditions given by Eqs. (12a) and (12b). Their Laplace transformations are given as

$$\bar{c}_b(s,0) = \frac{(1 - e^{-s\tau_{\rm inj}})}{s}, \quad \frac{d\bar{c}_b}{dr}(s,\infty) = 0.$$
 (30)

Using Eq. (30) in Eq. (28), the values of A_1 and B_1 come out to be

$$A_1 = \frac{(1 - e^{-s\tau_{\text{inj}}})}{s}, \qquad B_1 = 0.$$
 (31)

On plugging these values of A_1 and B_1 in Eq. (28), we get

$$\bar{c}_b(s,x) = \frac{(1 - e^{-s\tau_{\text{inj}}})}{s} e^{\lambda_1 x}, \qquad (32)$$

where, $\lambda_1(s)$ is given by Eq. (29) for the upper sign. When $\tau_{\rm inj} \to \infty$, the injection causes a complete breakthrough curve. For this case, the solution in Eq. (32) reduces to the following form

$$\bar{c}_b(s,x) = \frac{1}{s}e^{\lambda_1 x}. \tag{33}$$

Boundary conditions of type II: Danckwerts (or Robin) type BCs

The Laplace transformation of the inlet Danckwerts BC in Eq. (13a) gives

$$\bar{c}_b(s,0) = \frac{1}{s} \left(1 - e^{-s\tau_{\text{inj}}} \right) + \left. \frac{1}{Pe} \frac{d\bar{c}_b}{dx} \right|_{x=0}$$
 (34)

Similarly, the Laplace transformation of the outlet BC in Eq. (13b) gives

$$\left. \frac{d\bar{c}_b(s,x)}{dx} \right|_{x=1} = 0. \tag{35}$$

Using Eqs. (34) and (35) in Eq. (28), the values of A_1 and B_1 are obtained as

$$A_{1} = \frac{(1 - e^{-s\tau_{\text{inj}}})}{s} \frac{\lambda_{2}e^{\lambda_{2}}}{(1 - \frac{\lambda_{1}}{P_{e}})\lambda_{2}e^{\lambda_{2}} - (1 - \frac{\lambda_{2}}{P_{e}})\lambda_{1}e^{\lambda_{1}}},$$
(36)

$$B_{1} = -\frac{(1 - e^{-s\tau_{\text{inj}}})}{s} \frac{\lambda_{1}e^{\lambda_{1}}}{(1 - \frac{\lambda_{1}}{Pe})\lambda_{2}e^{\lambda_{2}} - (1 - \frac{\lambda_{2}}{Pe})\lambda_{1}e^{\lambda_{1}}}.$$
 (37)

At x = 1, the complete solution has the following simple form:

$$\bar{c}_b(s,x) = \frac{(1 - e^{-s\tau_{\text{inj}}})}{s} \frac{\lambda_2 e^{\lambda_1 + \lambda_2} - \lambda_1 e^{\lambda_1 + \lambda_2}}{(1 - \frac{\lambda_1}{P_e})\lambda_2 e^{\lambda_2} - (1 - \frac{\lambda_2}{P_e})\lambda_1 e^{\lambda_1}}.$$
 (38)

When $\tau_{\rm inj} \to \infty$, the injection again causes a breakthrough curve. In such a situation, Eq. (38) reduces to

$$\bar{c}_b(s,x) = \frac{1}{s} \frac{\lambda_2 e^{\lambda_1 + \lambda_2} - \lambda_1 e^{\lambda_1 + \lambda_2}}{(1 - \frac{\lambda_1}{P_e})\lambda_2 e^{\lambda_2} - (1 - \frac{\lambda_2}{P_e})\lambda_1 e^{\lambda_1}}.$$
 (39)

This completes the discussion of analytical solutions for the single component linear GRM in the Laplace domain.

There is no possibility to analytically transform back the Laplace domain solution in the actual time domain. The numerical Laplace inversion can be applied to obtain a discrete solution in time. In this technique, the integral of inverse Laplace transformation is approximated by Fourier series, see for example Rice et al. (1995). However, this solution is not helpful for the parameters estimation from experimental data. For that purpose, temporal moments are derived in the next section.

4. Moment analysis

Moment analysis is an effective technique for extracting important information about the retention and mass transfer processes in chromatographic columns, see e.g. Kucera (1965); Schneider and Smith (1968); Ruthven (1984); Guiochon et al. (2006); Miyabe et al. (2007, 2009). The Laplace domain solutions can be used to obtain moments. In this section, the temporal moments up to fourth order for the GRM are derived. The following moment generating property of the Laplace transform is exploited (e.g. Van der Laan (1958))

$$\mu_0 = \lim_{s \to 0} (\bar{c}(s, x = 1)), \quad \mu_n = (-1)^n \frac{1}{\mu_0} \lim_{s \to 0} \frac{d^n(\bar{c}(s, x = 1))}{ds^n}, \quad n = 1, 2, 3, \dots$$
 (40)

The second, third and fourth central moments can be obtained as

$$\mu_2' = \mu_2 - \mu_1^2, \quad \mu_3' = \mu_3 - 3\mu_1\mu_2 + 2\mu_1^3, \quad \mu_4' = \mu_4 - 4\mu_1\mu_3 + 6\mu_1^2\mu_2 - 3\mu_1^4.$$
 (41)

It is well known that the zeroth moment μ_0 gives the mass (or peak area), the first moment μ_1 corresponds to the retention time τ_R . The effects of longitudinal diffusion are not significant with respect to retention time or first moment. The second central moment μ'_2 , i.e. the variance of the elution profile, provides information about the rates of the mass transfer processes in the column. The third central moment μ'_3 quantifies the front asymmetries. Lastly, the fourth central moment μ'_4 measures the kurtosis. The complete derivations of moments are presented using the considered two types of boundary conditions.

Boundary conditions of type I: Dirichlet BCs

The final solution for Dirichlet BCs (c.f. Eq. (30)) is given by Eq. (33). The moments of this solution are given as follows.

Zeroth moment: The zeroth moment for rectangular profiles is given as

$$\mu_0 = \lim_{s \to 0} (\bar{c}_b(s, x = 1)) = \tau_{\text{inj}}.$$
 (42)

The zeroth moment for continuous breakthrough curves is simply $\mu_0 = 1$.

First moment: The first temporal moment for rectangular profiles is given as (c.f. Eq. (40))

$$\mu_1 = \frac{\tau_{\text{inj}}}{2} + 1 + a^* F \,, \tag{43}$$

where

$$a^* = \epsilon_p + (1 - \epsilon_p)a. \tag{44}$$

For continuous breakthrough curves, the first term on the right hand side of Eq. (43) in is zero, i.e.

$$\mu_1 = 1 + a^* F \,, \tag{45}$$

Second central moment: For a rectangular profile it is given as

$$\mu_2' = \frac{\tau_{\text{inj}}^2}{12} + \frac{2}{Pe} \left(1 + a^* F \right)^2 + F \left(\frac{2Fa^{*2}}{\xi} + \frac{2a^{*2}}{15\eta} \right) + 2 \frac{F(a^* - \epsilon_p)}{k_d}, \tag{46}$$

where

$$\eta = L \frac{\epsilon_p D_p + (1 - \epsilon_p) a D_s}{u R_p^2} \,. \tag{47}$$

For breakthrough curves the first term on the right hand side of Eq. (46) is zero, i.e.

$$\mu_2' = \frac{2}{Pe} \left(1 + a^* F \right)^2 + F \left(\frac{2Fa^{*2}}{\xi} + \frac{2a^{*2}}{15\eta} \right) + 2 \frac{F(a^* - \epsilon_p)}{k_d}. \tag{48}$$

The first and second central moments can be used to obtain a relation for the flowrate dependent Height Equivalent to Theoretical Plate (HETP) number. To derive the classical HETP-curve, let us consider for the sake of simplicity a Dirac injection, i.e. $\tau_{\rm inj} = 0$. Then, Eqs. (43), and (46) gives (e.g. Guiochon et al. (2006); Van Deemter et al. (1956))

$$\text{HETP}_{\text{Diric}}(u) = \frac{L\mu_2'}{\mu_1^2} = \frac{2D_b}{u} + \frac{2LF}{u(1+a^*F)^2} \left[\frac{Fa^{*2}}{\xi} + \frac{a^{*2}}{15\eta} + \frac{(a^* - \epsilon_p)}{k_d} \right] u. \tag{49}$$

In the above equation, the first term captures the band broadening by dispersion, the second term describes the contribution of external mass transfer, and the third term provides the contribution of diffusive migration of sample molecules inside the stationary phase, and the last term represents the contribution of desorption rate constant. The first term is frequently splitted using the following simplifying expression (e.g. Ruthven (1984); Guiochon et al. (2006))

$$D_b = \gamma_1 2R_p u + \gamma_2 D_{\text{mol}} \,. \tag{50}$$

Here, D_{mol} is the molecular diffusivity of the solute in the solvent, while γ_1 and γ_2 represent weight factors for the relative impact of the particle size/flow rate dependent and the molecular diffusion based contributions to band broadening. Thus, Eq. (49) can be rewritten as

$$HETP_{Diric}(u) = 4\gamma_1 R_p + \frac{2\gamma_2 D_{mol}}{u} + C_{core}u = A + \frac{B}{u} + Cu, \qquad (51)$$

where

$$A = 4\gamma_1 R_p, \quad B = 2\gamma_2 D_{\text{mol}}, \quad C = \frac{2LF}{u(1 + a^*F)^2} \left[\frac{Fa^{*2}}{\xi} + \frac{a^{*2}}{15\eta} + \frac{(a^* - \epsilon_p)}{k_d} \right]. \tag{52}$$

Third central moment: The third central moment is calculated as (c.f. Eq. (40))

$$\mu_{3}' = \frac{12}{Pe^{2}} (1 + a^{*}F)^{3} + \frac{6(1 + a^{*}F)F}{Pe} \left(\frac{2Fa^{*2}}{\xi} + \frac{2a^{*2}}{15\eta} \right) + Fa^{*3} \left(\frac{4}{105\eta^{2}} + \frac{4F}{5\xi\eta} + \frac{6F^{2}}{\xi^{2}} \right) + 12\frac{aF(1 - \epsilon_{p})}{k_{d}} \left[\frac{1 + a^{*}F}{Pe} + \frac{a^{*}F}{\xi} + \frac{a^{*}}{15\eta} + \frac{1}{2k_{d}} \right].$$
(53)

Moreover, μ'_3 is the same for rectangular and continuous breakthrough curves.

The second and third central moments can be used to calculate the skewness that measures the degree of asymmetry of elution profiles. It can be evaluated as

$$\delta_{\text{skew}} = \frac{\mu_3'}{(\mu_2')^{3/2}}. (54)$$

This relation is useful to quantify deviations from Gaussian peak shapes. As Eq. (54) indicates, chromatographic peaks corresponding to GRM have some degree of asymmetry. Thus, their front and rare parts have not the same shape.

Fourth central moment: The fourth temporal moment is obtained as (c.f. Eq. (40))

$$\mu_{4} = \frac{12(1+a^{*}F)^{4}}{Pe^{3}}(Pe+10) + \frac{24(1+a^{*}F)^{2}F}{Pe^{2}}\left(\frac{Fa^{*2}}{\xi} + \frac{a^{*2}}{15\eta}\right)(6+Pe)$$

$$+ \frac{48F^{3}a^{*3}}{\xi^{2}Pe} + \frac{48F^{4}a^{*4}}{\xi^{2}Pe} + \frac{48F^{3}a^{*4}}{5\xi\eta Pe} + \frac{32F^{2}a^{*3}}{5\xi\eta Pe} + \frac{32Fa^{*3}}{105\eta^{2}Pe} + \frac{72F^{2}a^{*4}}{175\eta^{2}Pe}$$

$$+ \frac{12F^{4}a^{*4}}{\xi^{2}} + \frac{24F^{4}a^{*4}}{\xi^{3}} + \frac{24F^{3}a^{*4}}{5\eta\xi^{2}} + \frac{72F^{2}a^{*4}}{175\xi\eta^{2}} + \frac{8F^{3}a^{*4}}{5\xi\eta} + \frac{4aF^{2}a^{*4}}{75\eta^{2}}$$

$$-24F(\epsilon_{p}-1) + 12(F(\epsilon_{p}-1))^{2} + \frac{8F((\epsilon_{p}-1))^{2}}{Bi\eta} + \frac{8F((\epsilon_{p}-1))^{2}}{5\eta}$$

$$-\frac{8aF^{2}a^{*2}(\epsilon_{p}-1)}{8i\eta k_{d}} - \frac{16aFa^{*}(\epsilon_{p}-1)}{5\eta k_{d}} - \frac{8aF^{2}a^{*2}(\epsilon_{p}-1)}{5\eta k_{d}} - \frac{16aFa^{*2}(\epsilon_{p}-1)}{5Bi\eta^{2}k_{d}}$$

$$-\frac{8aF^{2}a^{*2}(\epsilon_{p}-1)}{Bi\eta k_{d}} - \frac{8aFa^{*2}(\epsilon_{p}-1)}{k_{d}Bi^{2}\eta^{2}} - \frac{16aFa^{*2}(\epsilon_{p}-1)}{35\eta k_{d}}$$

$$+\frac{\tau_{\rm inj}^{4}}{80} + \tau_{\rm inj}^{2}\frac{(1+a^{*}F)^{2}}{Pe} + \tau_{\rm inj}^{2}F\left[\frac{Fa^{*2}}{\xi} + \frac{a^{*2}}{15\eta}\right], \tag{55}$$

where

$$Bi = \frac{Bi_p Bi_s}{Bi_s \epsilon_p + (1 - \epsilon_p) a Bi_p}.$$
 (56)

For breakthrough curves all terms in μ'_4 containing $\tau_{\rm inj}$ are zero. The fourth central moment can be utilized to calculate kurtosis which measures the profiles peakedness or flatness relative to a normal distribution. In general, the kurtosis is a descriptor of the shape of a probability distribution. It is instructive to use an adjusted version of Pearson's kurtosis, the excess kurtosis, (see DeCarlo (1997)). The excess kurtosis compares the shape of a given distribution to that of the normal distribution. Distributions with negative or positive excess kurtosis are called platykurtic distributions or leptokurtic distributions, respectively. The following definition quantifies the excess kurtosis

$$\gamma = \frac{\mu_4'}{(\mu_2')^2} - 3. \tag{57}$$

A high kurtosis distribution has a sharper peak and a broader tails than the normal distribution, while a low kurtosis distribution has a more rounded peak and thinner tails. Distributions with zero excess kurtosis, as the normal distribution, are called mesokurtic.

The above moments were presented in terms of the dimensionless time coordinate τ . However, they can be easily expressed in term of the actual time coordinate $t := L\tau/u$ as follows:

$$\mu_0(t) = \frac{L}{u}\mu_0(\tau), \quad \mu_i(t) = \left(\frac{L}{u}\right)^i \mu_i(\tau), \quad i = 1, 2, 3.$$
 (58)

Boundary conditions of type II: Danckwerts (or Robin) type BCs

In this case, the BCs given by Eqs. (34) and (35) are considered. The final solution is given by Eq. (38). The moments of this solution are given below.

Zeroth moment: The zeroth moment for rectangular profiles is given as

$$\mu_0 = \tau_{\rm inj} \,. \tag{59}$$

First moment: The first temporal moment using Eq. (40) is calculated as

$$\mu_1 = \frac{\tau_{\text{inj}}}{2} + 1 + a^* F \tag{60}$$

Second central moment: It is given as

$$\mu_2' = \frac{\tau_{\text{inj}}^2}{12} + \frac{2(1+a^*F)^2}{Pe} \left[1 + \frac{1}{Pe} (e^{-Pe} - 1) \right] + F \left[\frac{2Fa^{*2}}{\xi} + \frac{2a^{*2}}{15\eta} \right] + \frac{2F(a^* - \epsilon_p)}{k_d}, \quad (61)$$

In this case the expression in Eqs. (60) and (61) can be used to calculate HETP-curve for the Danckwerts BCs. Thus, we finally obtain

$$HETP_{Danck}(u) = \frac{L\mu_2'}{\mu_1^2} = 4\gamma_1 R_p + \frac{2\gamma_2 D_{mol}}{u} + Cu = \tilde{A} + \frac{\tilde{B}}{u} + Cu, \qquad (62)$$

where C is given by Eq. (52) and \tilde{A} and \tilde{B} are given as

$$\tilde{A} = 4\gamma_1 R_p \left[1 + \frac{1}{Pe} (e^{-Pe} - 1) \right], \quad \tilde{B} = 2\gamma_2 D_{\text{mol}} \left[1 + \frac{1}{Pe} (e^{-Pe} - 1) \right].$$
 (63)

Third central moment: It is expressed as

$$\begin{split} \mu_{3}' = & \frac{12}{Pe^{2}} \left(1 + a^{*}F \right)^{3} \left[\left(1 + \frac{2}{Pe} \right) e^{-Pe} + \left(1 - \frac{2}{Pe} \right) \right] \\ & + \frac{6(1 + a^{*}F)F}{Pe} \left(\frac{2Fa^{*2}}{\xi} + \frac{2a^{*2}}{15\eta} \right) \left[\frac{1}{Pe} (e^{-Pe} - 1) + 1 \right] + Fa^{*3} \left(\frac{4}{105\eta^{2}} + \frac{4F}{5\xi\eta} + \frac{6F^{2}}{\xi^{2}} \right) \\ & + 12 \frac{aF(1 - \epsilon_{p})}{k_{d}} \left(\frac{1 + a^{*}F}{Pe} \left[\frac{1}{Pe} (e^{-Pe} - 1) + 1 \right] + \frac{a^{*}F}{\xi} + \frac{a^{*}}{15\eta} + \frac{1}{2k_{d}} \right). \end{split}$$
(64)

The expression of μ'_4 was very large for Danckwarts BCs. Therefore, only plots of that moment are presented.

5. Numerical test problems

In this section, at first the effect of K_d , i.e. the desorption rate, is analyzed. Then, the effects of Pe, η (c.f. Eq. (47)) and Bi (c.f. Eq. (56)), characterizing the axial dispersion, intraparticle diffusion resistance and film mass transfer resistance, on the concentration profiles and moments are investigated. Moreover, effects of external mass transfer coefficient k_{ext} , as well as pore and surface diffusivities $(D_p \text{ and } D_s)$ on elution profiles are analyzed. A second-order accurate finite volume scheme (FVS) is chosen to numerically approximate the model equations for verifying the correctness of derived analytical results, see Javeed et al. (2011). All parameters used in the test problems are given in Table 1. Figures 1a shows a comparison of the concentration profiles for different values of the K_d . As K_d increases from 0.1 to $10^4 sec^{-1}$, the elution profiles become narrow, tall and symmetric. Thus, the column has better efficiency for large value of K_d and is more useful for purification and separation of multi-component mixtures. Beyond $K_d = 10^4$, the remaining dispersion is due to the other kinetic effects being active in the GRM and the adsorption equilibrium is permanently fulfilled. Since, there are typically no precise data regarding the system specific rate of adsorption and desorption available, a more general evaluation of the border value for K_d and an independent validation are not easy and out of the scope of our paper. Here, we essentially illustrate just the possibility to quantify the consequences of the effect. Figures 1b sows that analytical and numerical solutions of the finite volume scheme (FVS) are in good agreement with each other for all values of K_d . This is an indication of the accuracy of proposed numerical algorithm and the derived analytical solutions.

The effects of model parameters Pe, Bi, η , and the considered two sets of BCs on the elution curves are shown in Figure 2. It is evident that if axial dispersion or film mass transfer resistance is important, the peak becomes wider and time corresponding to the peak maximum is slightly lower. The effect of K_d is same as in Figure 1. For small values of B_i and η , the diffusion rate is very slow and fluid retention time is large. The breakthrough time becomes shorter due to the limitation of the intraparticle diffusion resistance. For low

efficient columns, i.e. small Pe numbers (low flowrates, short columns, large back-mixing effects), the selection of boundary conditions influences the results and the more realistic Danckwerts BC and the corresponding solutions should be applied (c.f. Figure 2d). If Pe is very large Eq. (63) converges into Eq. (52).

The elution profiles for different values of external mass transfer coefficient are shown in Figure 3a. Here, $K_d = 10 \, sec^{-1}$ is fixed. It is interesting that the band broadening decreases significantly when $k_{\rm ext}$ increases from 10^{-6} cm/sec to $10^{-4} \, cm/sec$, but when $k_{\rm ext}$ increases to $10^{-1} \, cm/sec$, the decrease of band broadening is not remarkable and for $10^{-4} \, cm/sec$ and $10^{-1} \, cm/sec$ the calculated profiles can hardly be distinguished. The result suggests us that the effect of the external mass transfer resistance can be neglected when the value of $k_{\rm ext}$ is more than $10^{-1} \, cm/sec$. The effect of $D_p = D_s$ on the eluted band profile is shown in Figure 3a for $K_d = 10 \, sec^{-1}$. The figure indicates that the band broadening is sensitive to the values of $D_p = D_s$. It can be seen that on increasing $D_p = D_s$, the band broadening decreases.

Figure 4 displays the first four dimensional temporal moments (c.f. Eq. (58)) which endorse all observations of Figure 1. Eqs. (43) and (44) indicate that retention time of the solute molecules is not correlated with K_d . Thus, as required, the retention time (i.e. μ_1) was not effected on changing the value of K_d . However, the variance μ'_2 , the third central moment μ'_3 , and the fourth central moment μ'_4 were reduced in different ways on increasing the value of K_d . Obviously, the results are very similar for $K_d = 10^3$ and 10^4 . Thus, for values of K_d larger then 10^4 the adsorption equilibrium is more or less established and the reduced GRM can be used.

Figure 5 displays the skewness (c.f. Eq. (54)) and excess kurtosis (c.f. Eq. (57)) as functions of velocity for different values of K_d . It can be observed that the concentration profile is more asymmetrical for small values of velocity u and K_d (see also Figures 1). Moreover, the excess kurtosis or flatness of the profile increases on increasing K_d . Once again, it is visible that for $K_d = 10^4$ the equilibrium is reached.

Figure 6a shows the plots of plate heights HETP (c.f. Eqs. (51) and (52)) as functions of the velocity u. The illustrating calculations were done for the Danckwerts BC and

different values of K_d . It can be seen that HETP-values become smaller with an increase in K_d . Figure 6b illustrates differences in the HETP of considered two types of boundary conditions. HETP is smaller for the Dankweets BCs due to the joint effects of the different inlet and out boundary conditions (Eqs. (12a) and (12b) vs Eqs. (13a) and (13b)).

6. Conclusion

Semi-analytical solutions and temporal moments of the GRM were derived considering dynamic linear sorption kinetics and two different sets of boundary conditions. The model equations were analytically solved in the Laplace domain and numerical Laplace inversion was applied to get back solutions in the actual time domain. Analytical expression of the first four temporal moments were derived from solutions in the Laplace domain. The derived solutions and moments were used to analyze the effect of finite adsorption-desorption rates, axial dispersion, film mass transfer resistance and intraparticle diffusion resistance on the elution curves. In the parametric calculations carried out, using the solutions derived, emphasis was put on providing insight in the effect of finite desorption rate. The results showed that a small value of K_d produces spreading in the elution profiles and the profiles become more asymmetric, causing reduction in the efficiency of the column. The analytical results were validated against the numerical results of second order finite volume scheme. Good agreements between analytically and numerically determined results verified the correctness of analytical solutions and accuracy of the suggested numerical scheme.

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Table 1: Parameters of the problems.

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Parameters	values
Column length	$L = 15 \ cm$
Stationary phase particle radius	$R_p = 5 \times 10^{-6} \ cm$
External porosity	$\epsilon = 0.4$
Geometrical constant	$\gamma_1 = 0.5$
Geometrical constant	$\gamma_2 = 1.0$
Internal porosity	$\epsilon_p = 0.333$
Interstitial velocity	$u = 0.4 \ cm/sec$
Axial dispersion coefficient	$D_b = 10^{-3} \ cm^2/sec$
Molecular diffusivity	$D_{\rm mol} = 10^{-5} \ cm^2/sec$
Pore diffusivity	$D_p = 10^{-6} \ cm^2/sec$
Surface diffusivity	$D_s = 10^{-6} \ cm^2/sec$
Henry's constant	a = 2.5
Desorption rate constant	$K_d = 0 \text{ to } 10^4 \ sec^{-1}$
External mass transfer coefficient	$k_{\rm ext} = 1.0 \ cm/sec$
Initial concentration	c(0,z) = 0 g/l
Initial concentration	$c_p(0,z) = 0 g/l$
Concentration at inlet	$c_{\rm inj} = 1.0 g/l$
Injection time	$t_{\rm inj} = 20~sec$

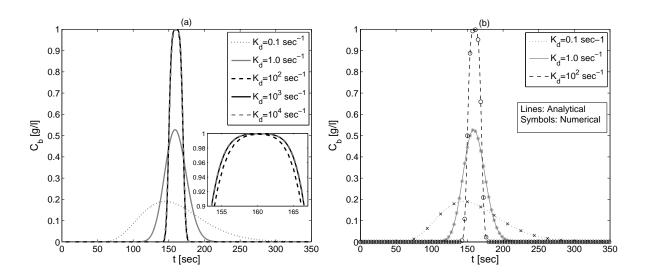


Figure 1: Effect of K_d on the concentration profiles. All parameters are given in Table 1.

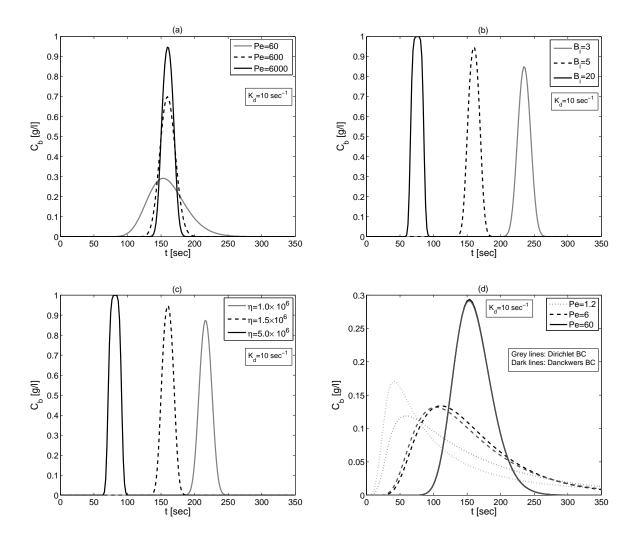


Figure 2: Effect of Pe, B_i and η , and BCs on the concentration profiles for fixed $K_d = 10 \, sec^{-1}$. Other parameters are given in Table 1.

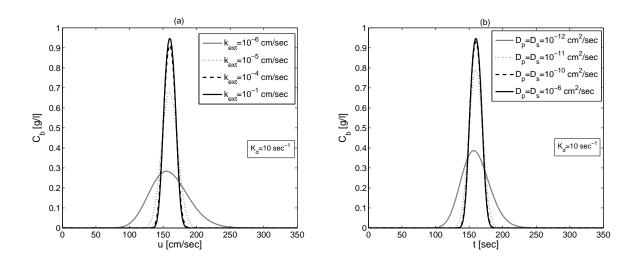


Figure 3: Effect of k_{ext} and $D_p = D_s$ on the concentration profiles for fixed $K_d = 10 \, sec^{-1}$. Other parameters are given in Table 1.

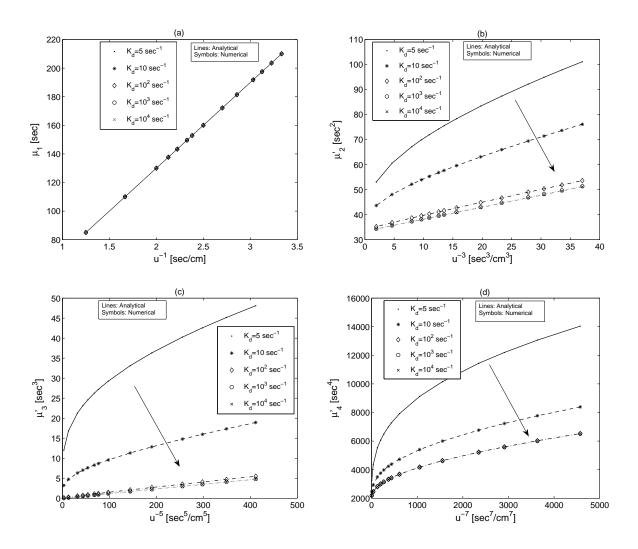


Figure 4: Effect of K_d on temporal moments. Other parameters are given in Table 1.

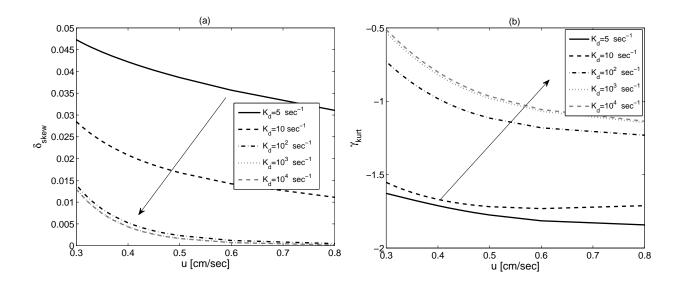


Figure 5: Effect of K_d on the skewness and kurtosis. Other parameters are given in Table 1.

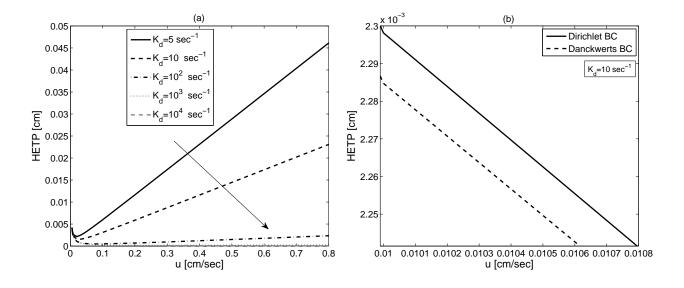


Figure 6: (a) Effect of K_d on the HETP curve, (b) effect of BCs on the HETP curve. Other parameters are given in Table 1.