

Supporting Information

Study of UV-photoexcitation and Ultrafast Dynamics of HCFC-132b (CF₂ClCH₂Cl)

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Figure S1. Graphical representation of the molecular orbitals composing the complete active space (CAS(12,12).

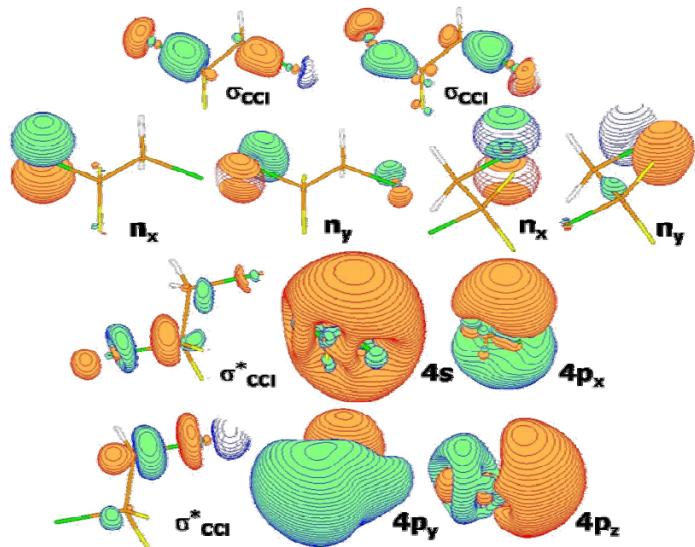
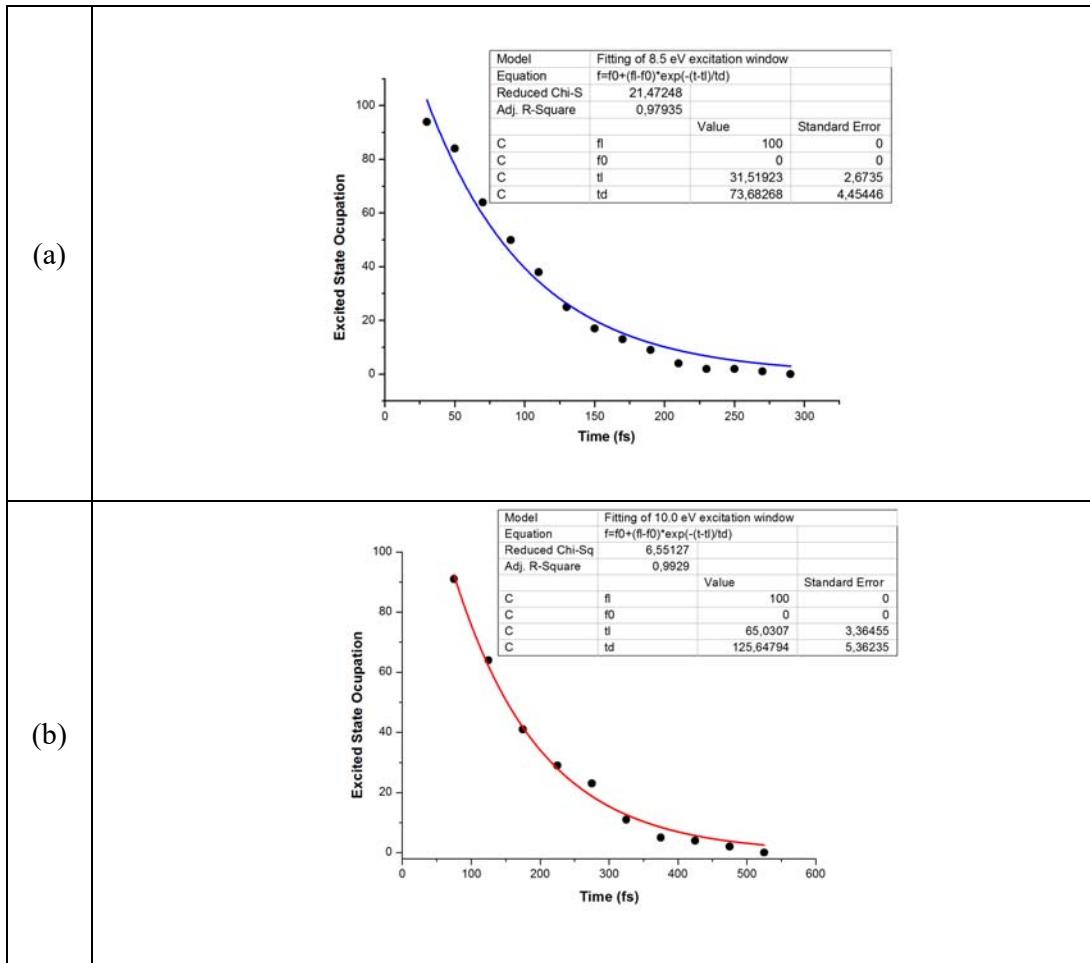


Figure S2. Fitting of the excited-state occupation as a function of time for 8.5 eV (a) and 10 eV (b) excitation windows.



S3. Cartesian Coordinates

Ground state of HCFC-132b. The geometry is in standard XYZ format in Angstrom.

C	0.032504	0.486085	0.000000
C	-0.519714	-0.921613	0.000000
Cl	-1.323952	1.630556	0.000000
Cl	0.790714	-2.111749	0.000000
F	0.790714	0.717639	1.082645
F	0.790714	0.717639	-1.082645
H	-1.122280	-1.062027	-0.891737
H	-1.122280	-1.062027	0.891737

S4. Vertical-excitation energies (eV), oscillator strengths, and configuration weights obtained for $\text{CF}_2\text{ClCH}_2\text{Cl}$ molecule at the TD-CAM-B3LYP and TD-M062X levels with the aug-cc-pVDZ (C, F, H)/d-aug-cc-pVDZ (Cl) basis set.

States	TD-CAM-B3LYP			TD-M062X		
	ΔE	Weights	$f \times 10^2$	ΔE	Weights	$f \times 10^2$
Gs	0.00	-	-	0.00	-	-
$n_{y\text{Cl}(3)}-\sigma^*_{\text{CCl}(3)}$	7.25	0.57	745.90	7.13	0.54	0.14
$n_{x\text{Cl}(3)}-\sigma^*_{\text{CCl}(3)}$	7.35	0.58	753.70	7.24	0.55	1.51
$n_{y\text{Cl}(3)}-4p_z$	7.67	0.56	752.90	7.58	0.47	0.89
$n_{y\text{Cl}(4)}-4p_z$	7.68	0.55	759.00	7.60	0.49	0.01
$n_{x\text{Cl}(4)}-\sigma^*_{\text{CCl}(3)}$	8.04	0.64	819.80	8.02	0.65	1.18
$n_{y\text{Cl}(4)}-\sigma^*_{\text{CCl}(3)}$	8.18	0.64	820.80	8.14	0.65	1.53
$n_{y\text{Cl}(4)}-\sigma^*_{\text{CCl}(4)}$	8.85	0.54	906.90	8.82	0.57	7.87
$n_{x\text{Cl}(3)}-4p_z$	8.86	0.54	919.50	8.85	0.59	1.58
$n_{y\text{Cl}(3)}-4s$	8.91	0.55	921.00	8.92	0.64	1.01
$n_{x\text{Cl}(3)}-4s$	8.93	0.50	938.00	8.98	0.60	0.19
$n_{x\text{Cl}(4)}-\sigma^*_{\text{CCl}(4)}$	9.00	0.63	941.40	9.00	0.50	4.07
$n_{y\text{Cl}(3)}-4p_z$	9.10	0.66	968.70	9.00	0.52	7.77
$n_{x\text{Cl}(3)}-4p_y$	9.55	0.54	979.30	9.39	0.51	0.13
$n_{y\text{Cl}(4)}-4p_x$	9.58	0.38	990.80	9.46	0.34	0.33
$n_{x\text{Cl}(4)}-4p_x$	9.65	0.43	991.50	9.51	0.39	2.39
$n_{y\text{Cl}(3)}-4p_x$	9.69	0.50	998.10	9.56	0.45	7.06
$n_{y\text{Cl}(4)}-4s$	9.71	0.50	1003.50	9.75	0.62	0.00
$n_{x\text{Cl}(4)}-4p_y$	9.81	0.32	1008.60	9.77	0.43	17.70
$n_{y\text{Cl}(3)}-\sigma^*_{\text{CCl}(4)}$	9.82	0.46	1026.80	9.78	0.41	0.11
$n_{y\text{Cl}(4)}-4p_y$	9.88	0.64	1035.40	9.85	0.42	0.72
$n_{x\text{Cl}(4)}-4s$	9.94	0.57	1043.70	9.90	0.42	0.68
$n_{y\text{Cl}(3)}-4p_y$	10.03	0.55	1044.30	9.91	0.44	1.49
$n_{x\text{Cl}(3)}-\sigma^*_{\text{CCl}(4)}$	10.04	0.46	1058.50	10.00	0.37	5.19
$n_{x\text{Cl}(3)}-4p_x$	10.11	0.60	1064.00	10.02	0.49	0.26

Figure S5. Comparison between the potential energy curves along the C-Cl bonds, obtained at the CASPT2 and TD- ω B97XD levels.

