

Kinetics of the Interactions between Yeast Elongation Factors 1A and 1B α , Guanine Nucleotides, and Aminoacyl-tRNA*

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The interactions of elongation factor 1A (eEF1A) from *Saccharomyces cerevisiae* with elongation factor 1B α (eEF1B α), guanine nucleotides, and aminoacyl-tRNA were studied kinetically by fluorescence stopped-flow. eEF1A has similar affinities for GDP and GTP, 0.4 and 1.1 μM , respectively. Dissociation of nucleotides from eEF1A in the absence of the guanine nucleotide exchange factor is slow (about 0.1 s^{-1}) and is accelerated by eEF1B α by 320-fold and 250-fold for GDP and GTP, respectively. The rate constant of eEF1B α binding to eEF1A (10^7 – 10^8 $\text{M}^{-1} \text{s}^{-1}$) is independent of guanine nucleotides. At the concentrations of nucleotides and factors prevailing in the cell, the overall exchange rate is expected to be in the range of 6 s^{-1} , which is compatible with the rate of protein synthesis in the cell. eEF1A·GTP binds Phe-tRNA^{Phe} with a K_d of 3 nM, whereas eEF1A·GDP shows no significant binding, indicating that eEF1A has similar tRNA binding properties as its prokaryotic homolog, EF-Tu.

The eukaryotic translation elongation factor 1 (eEF1)³ comprises eEF1A and eEF1B (1). eEF1A, a 50-kDa protein homologous to prokaryotic elongation factor Tu (EF-Tu), delivers aminoacyl-tRNAs (aa-tRNAs) to the A site of the ribosome. Similarly to EF-Tu, eEF1A is a member of the GTPase superfamily and can bind GTP and GDP. The dissociation of GDP from eEF1A is accelerated by a guanine nucleotide exchange factor (GEF), eEF1B, which is composed of two subunits, eEF1B α and eEF1B γ , in yeast, or three subunits, eEF1B α , eEF1B γ , and eEF1B β , in mammals. eEF1B α contains the catalytic domain necessary for nucleotide exchange and is thus the functional equivalent to the bacterial GEF of EF-Tu, EF-Ts.

Although eEF1B α and EF-Ts have the same function, the two proteins do not exhibit any significant sequence homology. While the isolated structure of the C terminus of eEF1B α was initially reported to be similar to one domain of EF-Ts (2), the two GEFs bind their G-proteins in fundamentally different ways. eEF1B α interacts with domains 1 and 2 of eEF1A (3), disrupting the switch 2 region of eEF1A, which forms part of the binding pocket for Mg^{2+} and the γ -phosphate of GTP, and inserting the highly conserved Lys²⁰⁵ of eEF1B α into the Mg^{2+} and GDP/GTP binding sites of eEF1A. This prevents the binding of the β - and γ -phosphates to the P loop (4). The structures of the sugar- and base-binding pockets of eEF1A are mostly unperturbed by eEF1B α , which is not the case in the EF-Tu·EF-Ts complex. EF-Ts, on the other hand, binds EF-Tu via domains I and III (5, 6). Because eEF1B α instead interacts with domains 1 and 2 of eEF1A, the latter of which is the binding site of aa-tRNA (3), binding of eEF1B α and aa-tRNA to eEF1A may either be mutually exclusive or either contribute to forming a binding pocket for aa-tRNA.

eEF1A from many different organisms was reported to have similar affinities for GTP and GDP, for example 0.7 μM and 1 μM for GTP and GDP, respectively, as measured for eEF1A from *S. cerevisiae* (7). Thus, the formation of active eEF1A·GTP is thermodynamically favored by the higher intracellular concentration of GTP over GDP. In addition, the high concentration of aa-tRNA in the cell is expected to further shift the equilibrium toward the GTP-bound state due to the formation of EF-Tu·GTP·aa-tRNA, without affecting the kinetics of nucleotide exchange. However, the rate of spontaneous GDP release from eEF1A, about 0.17 s^{-1} (8) appears to be too slow to maintain the rate of protein synthesis, about 10 s^{-1} in yeast cells (9, 10), which explains the necessity for eEF1B α . On the other hand, the rate of spontaneous GDP release from eEF1A is almost 100-fold faster than that from EF-Tu, 0.002 s^{-1} (11, 12), which suggests why overexpression of eEF1A allows protein synthesis in the absence of eEF1B α *in vivo* (13), *i.e.* when at any time sufficient amounts of eEF1A·GTP are available to bind aa-tRNA, despite the slow nucleotide exchange.

By analogy to the prokaryotic EF-Tu·EF-Ts (11) and in agreement with crystallographic studies on eEF1A·eEF1B α complexes (4), the exchange reaction is initiated by the binding of eEF1B α to eEF1A·GDP to form the eEF1A·GDP·eEF1B α complex, which dissociates into GDP and eEF1A·eEF1B α . Next, GTP binds to the eEF1A·eEF1B α complex to form the intermediary eEF1A·GTP·eEF1B α complex, which dissociates into

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³ The abbreviations used are: eEF, eukaryotic elongation factor; EF-Tu, elongation factor Tu; EF-Ts, elongation factor Ts; GEF, guanine nucleotide exchange factor; aa-tRNA, aminoacyl-tRNA; mant-GDP/GTP, (2'/3'-O-(N-methyl-anthraniloyl)-GDP/GTP); Prf, proflavin.

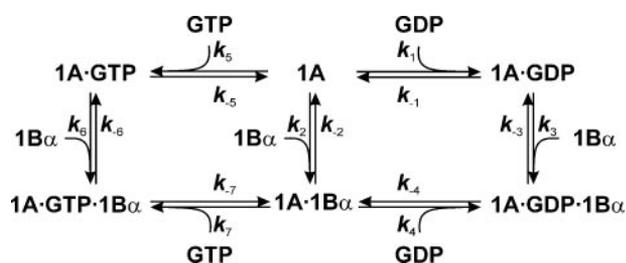


FIGURE 1. **Kinetic scheme of eEF1A interactions with guanine nucleotides and eEF1B α .** eEF1A can bind GDP (rate constant k_1), GTP (k_5), or eEF1B α (k_2) to form the respective binary complexes, which dissociate with the rate constants, k_{-1} , k_{-5} , and k_{-2} , respectively. The binary complexes formed with either GDP or GTP bind eEF1B α (k_3 and k_6) resulting in ternary complexes consisting of the two elongation factors and the respective nucleotide. The ternary complexes can dissociate by releasing either the nucleotide with the rate constants k_{-4} (GDP) or k_{-7} (GTP), or eEF1B α with the rate constants k_{-3} or k_{-6} . Finally, the eEF1A·eEF1B α complex can bind guanine nucleotides, rate constants k_4 (GDP) or k_7 (GTP), or dissociate, rate constant k_{-2} . Note that k_1 to k_7 are second-order association rate constants ($\text{M}^{-1} \text{s}^{-1}$), while k_{-1} to k_{-7} are first-order dissociation rate constants (s^{-1}).

eEF1B α and eEF1A·GTP, thereby completing the exchange reaction (Fig. 1). Nucleotide exchange in eEF1A in the presence of eEF1B α was suggested to be the rate-controlling step of eukaryotic translation (14). Recent rapid kinetics measurements indicated that the rate constant of eEF1B-catalyzed GDP dissociation from eEF1A is quite high, 100–200 s^{-1} at physiological Mg^{2+} concentrations (8); hence GDP release by itself does not seem to be rate-limiting for protein elongation. However, as the rate constants of all other steps are unknown, it cannot be excluded that some other step of nucleotide exchange, e.g. formation of the eEF1A·GDP·eEF1B complex or binding of GTP to the eEF1A·eEF1B complex, is rate-limiting.

Like prokaryotic EF-Tu, yeast eEF1A is able to bind aa-tRNA in a GTP-dependent manner and promote its binding to the mRNA-programmed 80S ribosome (15). While the ternary complex from bacteria is very well-characterized from a biochemical (11, 16–18) as well as a structural (19, 20) point of view, the available information about the corresponding eukaryotic eEF1A·GTP·aa-tRNA complex is more sparse and divergent. Notably, the existence of non-canonical mammalian eEF1A·GDP complexes with deacylated tRNA was suggested (21). The equilibrium dissociation constant, K_d , of the latter complex was estimated to 20 nM (21), a value which is comparable to that of canonical EF-Tu·GTP·aa-tRNA, and 1000 times lower than the K_d of the EF-Tu·GDP·aa-tRNA complex (16). Here we report the rate and equilibrium affinity constants of interactions among eEF1A, eEF1B α , GDP, or GTP, and aa-tRNA, as determined by stopped-flow kinetics. Nucleotide binding/dissociation was studied using fluorescent derivatives of GDP/GTP, mant-GDP/GTP, which were shown to closely mimic unmodified guanine nucleotides in their interactions with eEF1A (8). The binding of eEF1A to eEF1B α was monitored by fluorescence changes of intrinsic Trp residues. Binding of aa-tRNA to eEF1A was monitored using a fluorescence reporter group in Phe-tRNA^{Phe}(Prf16/17). The combination of these observables allowed us to solve the kinetic mechanism of nucleotide exchange in eEF1A and of aa-tRNA binding to the factor, and to estimate the effective rate of the reactions at the concentrations of components prevailing in the cell.

EXPERIMENTAL PROCEDURES

Biochemical Methods—eEF1A was purified as described (22). eEF1B α was expressed and purified by nickel chelation affinity chromatography followed by further purification on a Source-Q anion exchange column (GE Healthcare) equilibrated by 20 mM Tris-Cl, pH 7.6 and 0.5 mM dithiothreitol. eEF1B α was eluted using a linear gradient from 120 mM to 600 mM KCl. Protein concentrations were determined both colorimetrically (Bradford Assay, Bio-Rad) and by absorbance measurements at 205, 210, and 280 nm, using an extinction coefficient (280 nm) of 44,920 $\text{M}^{-1} \text{cm}^{-1}$ for eEF1A and 20,970 $\text{M}^{-1} \text{cm}^{-1}$ for eEF1B α (23–25). eEF1A preparations were free of GTP or GDP as determined by HPLC analysis (26). tRNA^{Phe}(Prf16/17) was prepared as described (27, 28). Aminoacylation was carried out with tRNA^{Phe}(Prf16/17) (4 μM), purified yeast phenylalanyl-tRNA synthetase (3% v/v), [¹⁴C]phenylalanine (30 μM), ATP (3 mM) in buffer A (50 mM Tris-HCl pH 7.5, 70 mM NH_4Cl , 30 mM KCl, 10 mM MgCl_2) for 15 min at 37 °C. The extent of aminoacylation was determined by trichloroacetic acid precipitation and filtration through GF/C filters.

Rapid Kinetic Measurements—The interactions of eEF1A with guanine nucleotides and eEF1B α were studied essentially as previously described for EF-Tu (11, 12). Fluorescence stopped-flow measurements were performed on a SX-18MV spectrometer (Applied Photophysics) in buffer A (50 mM Tris-HCl, pH 7.5, 70 mM NH_4Cl , 30 mM KCl, 10 mM MgCl_2) at 20 °C, if not stated otherwise. The interaction between eEF1A and eEF1B α was monitored by changes in tryptophan fluorescence (11, 29). eEF1A contains eight tryptophan residues of which six are located in the G domain of eEF1A. Tryptophan fluorescence was excited at 280 nm and measured after passing KV335 filters (Schott). To prepare the complexes of eEF1A with the fluorescent nucleotides mant-GTP or mant-GDP, the protein was preincubated with a 5-fold excess of the respective nucleotide; purification of complexes from unbound nucleotides was not possible because of dissociation of unstable eEF1A·nucleotide complexes during purification. The fluorescence of mant-GDP/GTP bound to eEF1A was excited via fluorescence resonance energy transfer (FRET) from tryptophan (excitation wavelength, 280 nm) and measured after passing KV408 filters (Schott). Proflavin fluorescence was excited at 470 nm and measured after passing KV500 filter (Schott).

Stopped-flow experiments were performed by rapidly mixing equal volumes (60 μl each) of the reactants and monitoring the time course of fluorescence change. Time courses depicted in the figures were obtained by averaging 5–10 individual transients. Data were evaluated by fitting to a single exponential function with a characteristic time constant (k_{app}), amplitude (A), and another variable for the final signal (F_∞) according to the equation $F = F_\infty + A \cdot \exp(-k_{app} \cdot t)$ where F is the fluorescence at time t . Where necessary, two exponential terms were used with two characteristic time constants (k_{app1} , k_{app2}), amplitudes (A , B), and another variable for the final signal (F_∞) according to the equation $F = F_\infty + A \cdot \exp(-k_{app1} \cdot t) + B \cdot \exp(-k_{app2} \cdot t)$. Calculations were performed using TableCurve software (Jandel Scientific) or Prism (Graphpad Soft-

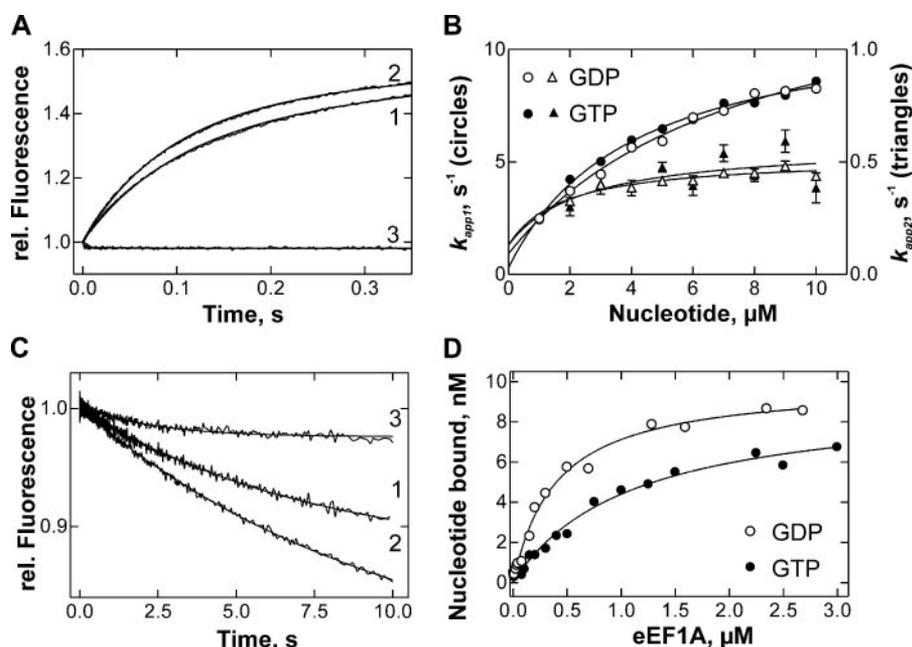


FIGURE 2. Binding of mant-GDP/mant-GTP to eEF1A. A, FRET change upon mixing nucleotide-free eEF1A (0.2 μM) with mant-GDP (3 μM) (1), mant-GTP (3 μM) (2) or buffer (3). B, concentration dependence of k_{app1} (circles) and k_{app2} (triangles) of mant-GDP (\circ , \triangle) and mant-GTP (\bullet , \blacktriangle). Values of k_{app1} and k_{app2} were determined by exponential fitting of time courses as in Fig. 2A. Goodness of hyperbolic fits of k_{app} values ($R^2 > 0.99$ for k_{app1} with GTP or GDP, 0.86 for k_{app2} with GDP and 0.66 for k_{app2} with GTP) was significantly better than that of linear fits (not shown; $R^2 = 0.92$ for k_{app1} with GTP or GDP, 0.65 for k_{app2} with GDP and 0.30 for k_{app2} with GTP); C, dissociation of eEF1A-mant-GDP (1) or eEF1A-mant-GTP (2) complexes (0.1 μM) in the presence of GDP or GTP (25 μM), respectively, or in the absence of excess unlabeled nucleotide (3). D, equilibrium titration of mant-GDP (\circ) or mant-GTP (\bullet) (0.01 μM) with eEF1A.

ware). Standard deviations were calculated using the same software.

Equilibrium Titrations—To determine the equilibrium dissociation constant (K_d) of the eEF1A·eEF1B α complex, titrations were carried out by adding increasing amounts of eEF1B α to a fixed amount of eEF1A (0.1 μM). The increase of the tryptophan fluorescence of eEF1A upon complex formation was measured in a PTI fluorimeter (excitation at 280 nm, emission at 333 nm). As eEF1B α contains three tryptophan residues, a control titration was carried out in the absence of eEF1A and the resulting fluorescence signal subtracted from the signal obtained in the presence of eEF1A. The measured fluorescence was corrected for dilution. Fluorescence titrations were evaluated as described in detail in (30). The affinity of mant-GTP/mant-GDP to eEF1A was measured at constant nucleotide concentration (0.01 μM) and increasing concentrations of eEF1A. Fluorescence emission at 448 nm was measured upon excitation at 355 nm. To determine the K_d values of the eEF1A·GTP·Phe-tRNA^{Phe} complex, a fixed amount of Phe-tRNA^{Phe}(Prf16/17) (0.01 μM) aminoacylated *in situ* (see above) was mixed with increasing amounts of eEF1A in the presence of GTP (2 mM). Fluorescence emission at 510 nm was measured upon excitation at 470 nm, and the data were treated as described above.

RESULTS

Interactions of eEF1A with GDP and GTP—Interactions of GDP and GTP with eEF1A were studied as described before (11, 12), using FRET from tryptophan residues in eEF1A to the mant group of mant-GDP or mant-GTP. The intrinsic GTPase

activity of eEF1A was very low, $<0.14 \text{ h}^{-1}$; hence no appreciable conversion of GTP to GDP occurred during the experiments. Upon binding of labeled nucleotide, a 60% increase of mant fluorescence was observed. To determine association rate constants, a fixed concentration of nucleotide-free eEF1A was mixed with varying concentrations of fluorescent guanine nucleotides. The time curves obtained were described best by two-exponential fitting, resulting in apparent rate constants k_{app1} and k_{app2} . Both k_{app1} and k_{app2} values increased with nucleotide concentration (Fig. 2B). In both cases, the concentration dependence deviated from the linear behavior expected for a second-order binding reaction. Rather, the k_{app} values saturated at high nucleotide concentration, suggesting that the observed fluorescence changes reported steps following the bimolecular binding step. The observed concentration dependence would be consistent with a reaction scheme $A + B \rightleftharpoons C \rightleftharpoons D \rightleftharpoons E$,

where the bimolecular reaction $A + B \rightleftharpoons C$ is too fast to be measured or does not result in an appreciable fluorescence change, while the first-order reactions $C \rightleftharpoons D$ and $D \rightleftharpoons E$ yield k_{app1} and k_{app2} , respectively. However, the observed concentration dependence of k_{app} values would be equally consistent with eEF1A being heterogeneous. In such a case, a fraction of the protein would bind nucleotides faster and yield k_{app1} , while another fraction would be less active and give the lower k_{app2} ; for both fractions, the k_{app} would reflect a monomolecular transition in the reaction scheme $A + B \rightleftharpoons C \rightleftharpoons D$. Because the two reaction mechanisms cannot be distinguished, and information about the second-order step is not available, the complete set of rate constants could not be calculated. Furthermore, attempts to fit the concentration dependences depicted in Fig. 2B using the values of k_{-1} and k_{-5} , as well as the equilibrium dissociation constants determined below yielded satisfactory fits for both mechanisms. The values of the rate constants varied significantly depending on the assumed mechanism, and, as the mechanisms cannot be distinguished, are not reported here. Nevertheless, two qualitative statements can be made: (i) eEF1A binds to GDP and GTP in a very similar way, and (ii) structural rearrangements, rather than bimolecular binding steps, result in FRET changes.

Nucleotide dissociation rate constants were determined upon mixing eEF1A·mant-GDP or eEF1A·mant-GTP with an excess unlabeled nucleotide. The release of the labeled nucleotide from the elongation factor resulted in a fluorescence decrease, and the time courses were single-exponential. Given the complicated nucleotide binding mechanism, the observa-

tion of a single dissociation step suggests that this step is rate-limiting in nucleotide release; hence we assigned the respective values to the effective rate constants of nucleotide dissociation, k_{-1} and k_{-5} . From single-exponential fitting of the time courses of Fig. 2C, the following rate constants were obtained: $k_{-1} = 0.13 \pm 0.01 \text{ s}^{-1}$ (GDP), $k_{-5} = 0.10 \pm 0.01 \text{ s}^{-1}$ (GTP).

Nucleotide binding affinities of eEF1A were determined by titrating mant-GTP or mant-GDP with increasing concentrations of eEF1A ("Experimental Procedures") (Fig. 2D). From the hyperbolic fits, the values $K_d = 0.4 \pm 0.1 \mu\text{M}$ (GDP) and $K_d = 1.1 \pm 0.2 \mu\text{M}$ (GTP) were obtained. Because the detailed kinetic mechanism of nucleotide binding could not be determined, we made estimations for the effective constants of nucleotide association with eEF1A, assuming a single binding step. The resulting association rate constants were $3.3 \cdot 10^5 \text{ M}^{-1} \text{ s}^{-1}$ (GDP) and $0.9 \cdot 10^5 \text{ M}^{-1} \text{ s}^{-1}$ (GTP), which were assigned to the rate constants k_1 and k_5 , respectively. Note that these rate constants, and probably also k_{-1} and k_{-5} , reflect effective rate constants that characterize the overall binding reaction where the individual equilibrium steps are grouped into one.

Binding of eEF1 β to eEF1A in the Absence of Nucleotides—The association of eEF1 β with eEF1A was monitored by the increase in intrinsic tryptophan fluorescence (Fig. 3A). Because there are eight tryptophan residues in eEF1A and three in eEF1 β , and probably only part of them exhibits a fluorescence change upon complex formation, the amplitude of the signal change was small, but reproducible, particularly when up to 10 individual transients were averaged. To calculate the association rate constant k_2 , time courses were measured at a fixed concentration of eEF1A and varying eEF1 β concentrations. Apparent rate constants were determined by exponential fitting and plotted against the eEF1 β concentration (Fig. 3B); from the slope of the linear plot $k_2 = (12 \pm 2) \cdot 10^6 \text{ M}^{-1} \text{ s}^{-1}$ was determined, from the Y-axis intercept $k_{-2} = 1.0 \pm 0.8 \text{ s}^{-1}$. To obtain a more precise estimation for k_{-2} , we determined the equilibrium dissociation constant of eEF1A binding to eEF1 β , and calculated k_{-2} from k_2 and K_d . To determine K_d , a fixed amount of nucleotide-free eEF1A was titrated with eEF1 β and tryptophan fluorescence was measured at equilibrium (Fig. 3C). Hyperbolic fitting resulted in a value of $K_d = 0.16 \pm 0.02 \mu\text{M}$. Using these values, $k_{-2} = 1.9 \pm 0.4 \text{ s}^{-1}$ was obtained, in agreement with the value estimated from kinetic experiments.

Interactions of eEF1A with eEF1 β in the Presence of GDP—Dissociation of the eEF1A·mant-GDP complex after binding of eEF1 β was monitored by the decrease of FRET from tryptophan to mant-GDP. An excess of unlabeled GDP was included with eEF1 β to prevent rebinding of mant-GDP. Time courses of dissociation at non-saturating eEF1 β concentrations showed two exponential phases, a faster, which reflected the dissociation of mant-GDP from the eEF1A·mant-GDP·eEF1 β complex, and a slower, caused by the spontaneous dissociation of the eEF1A·mant-GDP complex (Fig. 4A). The apparent rate constant of mant-GDP release from the eEF1A·mant-GDP·eEF1 β complex exhibited a hyperbolic dependence on the concentration of eEF1 β (Fig. 4B). At low concentrations of eEF1 β , the apparent rate constant of the dissociation of eEF1A·mant-GDP·eEF1 β increased linearly with the concentration eEF1 β (Fig. 4B, inset), indicating the

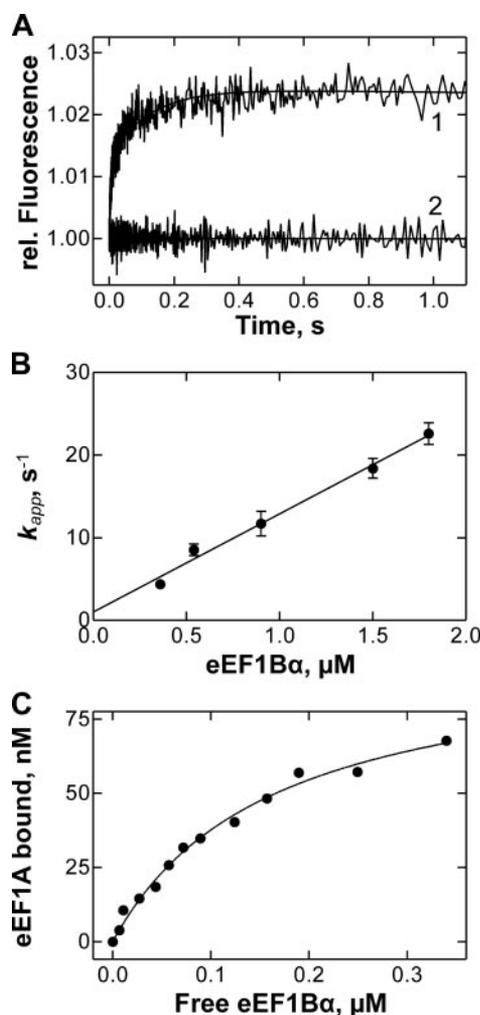


FIGURE 3. eEF1A·eEF1 β interactions. A, tryptophan fluorescence change upon mixing nucleotide-free eEF1A ($0.25 \mu\text{M}$) with eEF1 β ($0.9 \mu\text{M}$) (1) or buffer (2). The apparent rate constant of the interaction, k_{app} , was determined by exponential fitting. B, concentration dependence of k_{app} . From the slope of the plot, the value of the bimolecular association rate constant (k_2) was calculated. C, equilibrium titration of eEF1A ($0.1 \mu\text{M}$) with eEF1 β . The relative change of the intrinsic tryptophan fluorescence (ΔF) was corrected for dilution and linear increase of free eEF1 β fluorescence.

concentration range where the binding of eEF1 β to eEF1A·mant-GDP is rate-limiting. As rebinding of mant-GDP is negligible, the initial slope (modified from equation on p. 124 of Ref. 31) is equal to $k_3/(1 + k_{-3}/k_{-4}) = 20 \pm 2 \mu\text{M}^{-1} \text{ s}^{-1}$, and k_3 can be calculated when k_{-3} and k_{-4} are known. At higher concentration of eEF1 β , the dissociation of GDP from the ternary complex becomes rate-limiting, and the apparent rate constant measured at saturation with eEF1 β yields $k_{-4} = 42 \pm 1 \text{ s}^{-1}$ (Fig. 4B).

To determine k_{-3} , a constant concentration of eEF1A·eEF1 β complex was titrated with increasing concentrations of GDP and the dissociation of the binary complex detected by the decrease of the intrinsic tryptophan fluorescence (Fig. 4C). Apparent rate constants were estimated by single exponential fitting. The concentration dependence of the apparent rate constants was hyperbolic (Fig. 4D). At saturation, the dissociation of eEF1 β is rate-limiting, thus k_{-3} can be determined directly, with $k_{-3} = 117 \pm 8 \text{ s}^{-1}$. The initial slope was equal to $k_4/(1 + k_{-4}/k_{-3}) = 3.3 \pm 0.5 \mu\text{M}^{-1} \text{ s}^{-1}$, as the binding of GDP

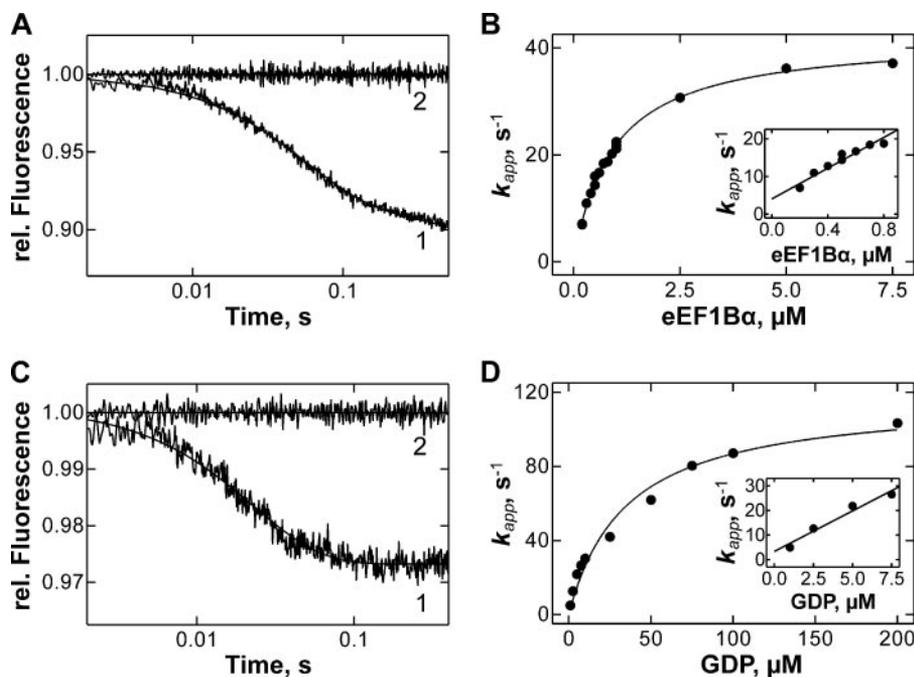


FIGURE 4. **Interactions between eEF1A, eEF1B α , and GDP.** A, dissociation of eEF1A·mant-GDP complex (0.1 μM) upon addition of EF1B α (0.7 μM) and GDP (25 μM) (1) or buffer (2) monitored by FRET. B, concentration dependence of k_{app} of mant-GDP dissociation from the eEF1A·mant-GDP complex in the presence of increasing eEF1B α concentration and excess GDP (25 μM). *Inset*, initial slope of the curve. C, dissociation of eEF1A·eEF1B α (0.5 μM) upon addition of GDP (25 μM) (1) or buffer (2), monitored by tryptophan fluorescence. D, dependence of k_{app} of eEF1A·eEF1B α dissociation on GDP concentration. *Inset*, initial slope of the curve.

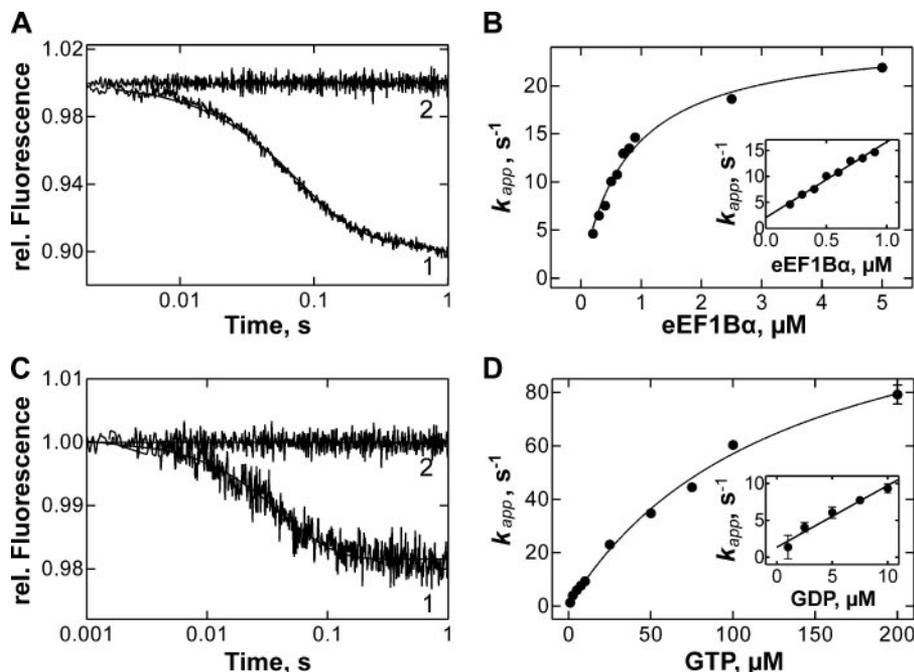


FIGURE 5. **Interactions between eEF1A, eEF1B α , and GTP.** A, dissociation of eEF1A·mant-GTP complex (0.1 μM) upon addition of EF1B α (0.7 μM) and GTP (25 μM) (1) or buffer (2) measured by FRET. B, eEF1B α concentration dependence of k_{app} of mant-GTP dissociation from the eEF1A·mant-GTP complex in the presence of eEF1B α as indicated and excess GTP (25 μM). *Inset*, initial slope of the curve. C, dissociation of eEF1A·eEF1B α (0.5 μM) upon addition of GTP (25 μM) (1) or buffer (2), monitored by tryptophan fluorescence. D, dependence of k_{app} of eEF1A·eEF1B α dissociation on GTP concentration. *Inset*, initial slope of the curve.

to the binary complex was rate-limiting at low GDP concentrations (Fig. 4D, *inset*). With the known values for k_{-3} and k_{-4} and the initial slopes for both hyperbolic curves, the

values for k_3 and k_4 can be calculated, $k_3 = 7.5 \cdot 10^7 \text{ M}^{-1} \text{ s}^{-1}$ and $k_4 = 4.5 \cdot 10^6 \text{ M}^{-1} \text{ s}^{-1}$.

Interactions of eEF1A with eEF1B α in the Presence of GTP—The interaction of eEF1A with eEF1B α in the presence of GTP was studied essentially in the same way as for GDP, except that the GTP solutions were preincubated with phosphoenol pyruvate and pyruvate kinase to convert any GDP present into GTP. Similarly to eEF1A·mant-GDP, the addition of eEF1B α to eEF1A·mant-GTP resulted in a biphasic FRET decrease, with the first phase reflecting the dissociation of mant-GTP from the eEF1A·mantGTP·eEF1B α complex, and the second reflecting the spontaneous eEF1A·mant-GTP dissociation (Fig. 5A). The dependence of the apparent rate constant of mant-GTP release on eEF1B α concentration was hyperbolic (Fig. 5B). The rate constant of mant-GTP dissociation from the eEF1A·mantGTP·eEF1B α determined from k_{app} at saturation was $k_{-7} = 25 \pm 1 \text{ s}^{-1}$; the initial part of the plot yielded $k_6/(1 + k_{-6}/k_{-7}) = 15 \pm 1 \text{ } \mu\text{M}^{-1} \text{ s}^{-1}$ (Fig. 5B, *inset*).

The dissociation of the eEF1A·eEF1B α complex upon addition of GTP (Fig. 5C) was monitored by tryptophan fluorescence at varying GTP concentrations (Fig. 5D). The apparent rate constant of eEF1B α release from eEF1A·mantGTP·eEF1B α , as calculated by hyperbolic fitting of Fig. 5D, was $k_{-6} = 130 \pm 9 \text{ s}^{-1}$, and the slope of the initial part of the plot was $k_7/(1 + k_{-7}/k_{-6}) = 0.8 \pm 0.1 \text{ } \mu\text{M}^{-1} \text{ s}^{-1}$. From these values, the rate constants $k_6 = 9.0 \cdot 10^7 \text{ M}^{-1} \text{ s}^{-1}$ and $k_7 = 1.0 \cdot 10^6 \text{ M}^{-1} \text{ s}^{-1}$ were calculated.

Comparison of the GDP and GTP Cycles—To test the consistency of the kinetic model, we compared the thermodynamic cycles of GDP and GTP exchange. As the binding of eEF1A to eEF1B α is a step that is common to both GDP and GTP cycles, the same K_2 value should arise regardless of how the value was calculated, as $K_2 = K_5 \cdot K_6 / K_7$ or $K_2 = K_1 \cdot K_3 / K_4$, and should be close to the measured K_2 value (Fig. 3). Using the values in Fig. 7, identical K_2 values, 0.07 μM , were calculated from the GDP and GTP cycles, which is within the

Functions of Yeast eEF1A

statistical significance identical to the value calculated from the rate constants, $0.08 \mu\text{M}$, and is similar to the measured K_d ($0.16 \mu\text{M}$).

aa-tRNA Binding—Association of aa-tRNA with eEF1A was measured utilizing the fluorescence change of yeast Phe-tRNA^{Phe} containing a fluorescent dye, proflavin, at positions 16 or 17 in the D loop. Upon addition of eEF1A·GTP to Phe-tRNA^{Phe}(Prf17/16), a decrease in fluorescence was observed (Fig. 6A). The apparent rate constants of association were determined by single-exponential fitting. The concentration dependence of k_{app} was not linear (Fig. 6B), suggesting that the fluorescence change reflected a rearrangement step following an initial rapid-equilibrium binding step, which was not accompanied by a fluorescence change. From hyperbolic fitting of the concentration dependence of k_{app} , a K_{d1} value for the first step of $0.9 \pm 0.3 \mu\text{M}$ was determined, and the forward rate constant of the second step was $5.7 \pm 0.3 \text{ s}^{-1}$; the value for the backward rate constant of the second step was very close to 0 (Y-axis intercept) and could not be determined with precision. The amplitudes of fluorescence changes did not change with eEF1A concentration, suggesting that the affinity of binding was $<0.2 \mu\text{M}$ (the first concentration point measured). The lower limit for the bimolecular rate constant of aa-tRNA binding to eEF1A was estimated from the initial slope of the concentration dependence, about $2 \cdot 10^6 \text{ M}^{-1} \text{ s}^{-1}$.

To determine the overall affinity of Phe-tRNA^{Phe} to eEF1A·GTP, equilibrium fluorescence titrations were performed with constant amounts of Phe-tRNA^{Phe}(Prf16/17) and varying eEF1A concentration (Fig. 6C). The K_d value for the eEF1A·GTP·Phe-tRNA^{Phe} complex was $3 \pm 1 \text{ nM}$. Taking into account that the overall affinity of the two-step binding equilibrium is the product of the K_d values of each step, *i.e.* $K_d = K_{d1} \cdot k_{-2}/k_2$, the k_{-2} value can be calculated using the values of K_{d1} and k_2 determined above, $k_{-2} = 0.02 \pm 0.01 \text{ s}^{-1}$.

It has been reported that eukaryotic eEF1A·GDP, unlike bacterial EF-Tu·GDP, can promote the binding of aa-tRNA to the mRNA-programmed ribosome, though to a smaller extent than with GTP (32). However, given the similarity of the affinities of GTP and GDP for eEF1A, a small GTP contamination in GDP solutions would yield a certain amount of eEF1A·GTP which could bind aa-tRNA. In fact, when binding experiments were carried out with GDP solutions that did not contain any detectable GTP (33), no fluorescence change of Phe-tRNA^{Phe}(Prf16/17) was observed upon addition of eEF1A·GDP up to $1 \mu\text{M}$. This suggests that the affinity of aa-tRNA for eEF1A in the GDP form is very low, and at least 300 times lower than that for eEF1A·GTP.

Previously, mammalian eEF1A was reported to bind deacylated tRNA^{Phe} in the presence of GDP. To test whether the yeast factor is also able to bind deacylated tRNA as suggested (21), we performed stopped-flow and fluorescence titration experiments as described above for Phe-tRNA^{Phe} with deacylated tRNA^{Phe}(Prf16/17) and eEF1A·GTP or eEF1A·GDP. No fluorescence change was observed, even at high concentrations of factors. Furthermore, if deacylated tRNA binds to eEF1A, it should be able to compete with Phe-tRNA^{Phe} for the binding to the factor. However, even a large excess of deacylated tRNA^{Phe} did not chase Phe-tRNA^{Phe}(Prf16/17) from eEF1A, regardless

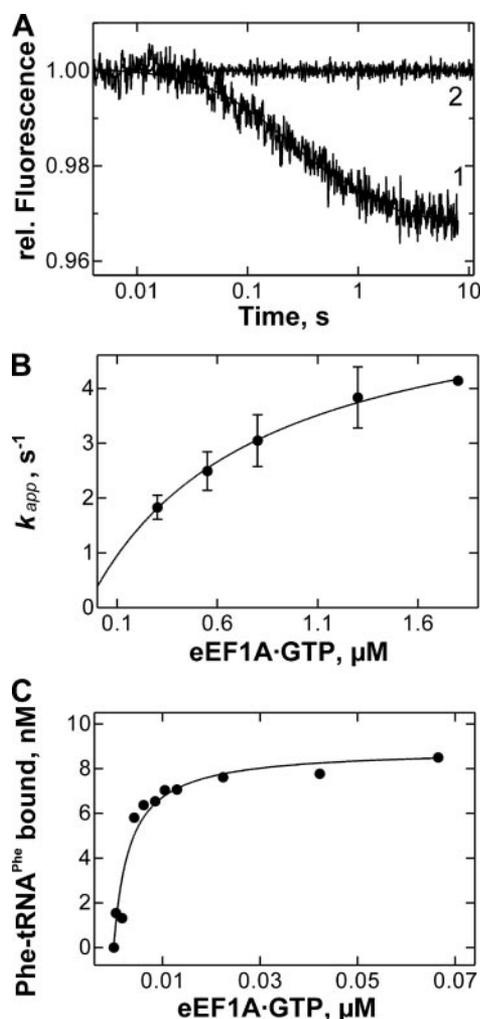


FIGURE 6. Binding of eEF1A to Phe-tRNA^{Phe}. A, time courses of Phe-tRNA^{Phe}(Prf16/17) ($0.2 \mu\text{M}$) binding to eEF1A·GTP ($1 \mu\text{M}$); (2), control in the absence of eEF1A. B, concentration dependence of k_{app} of the reaction between Phe-tRNA^{Phe}(Prf16/17) and eEF1A·GTP. C, equilibrium titrations of Phe-tRNA^{Phe}(Prf16/17) ($0.01 \mu\text{M}$) with eEF1A·GTP.

of whether GTP or GDP was present. Thus, we have to conclude that deacylated tRNA most likely does not bind to yeast eEF1A to any significant extent.

DISCUSSION

Nucleotide Binding to eEF1A—eEF1A from *S. cerevisiae* binds GDP and GTP with similar affinities, 0.4 and $1.1 \mu\text{M}$, respectively, in agreement with earlier reports (K_d GDP = $1 \mu\text{M}$ and K_d GTP = $0.7 \mu\text{M}$ (7); $K_d = 0.18 \mu\text{M}$ for mant-GDP (8); both at somewhat different reaction conditions). The nucleotide binding properties of eEF1A differ from those of other GTPases, most notably of its prokaryotic counterpart EF-Tu (11, 12), eukaryotic initiation factor eIF2 (34), translation termination factors RF3 (35) and eRF3 (36), as well as from those of most Ras-like GTPases and the G_α subunits of heterotrimeric G-proteins, which bind GDP about 10–100 times more tightly than GTP. The intrinsic rate of GDP dissociation from EF-Tu, eIF2, RF3, or Ras-like GTPases is very slow, limiting GTP binding (11, 37, 38). However, in a number of GTPases, including translation factors IF2 (39) and its eukaryotic ortholog eIF5B (40), EF-G (33), SelB (41), eRF3 (36) or the GTPases of the signal

recognition particle pathway (42, 43), the GDP-to-GTP exchange is rapid and spontaneous. Structures of EF-G and SelB crystallized in the nucleotide-free forms or with GDP or GTP bound show similar overall domain arrangement, regardless of the nucleotide (44–49), in contrast to many other GTPases that undergo a large conformational change when switching from the GTP-bound to GDP-bound form (50). Particularly striking differences are found between the three elongation factors that deliver aa-tRNA to the ribosome: while EF-Tu has a strong preference for GDP and practically does not release GDP in the absence of EF-Ts, which acts as a GEF, eEF1A has similar affinities for GTP and GDP, and SelB has a preference for GTP and does not require a GEF.

The binding of nucleotides to eEF1A entails at least two steps. The complex binding mechanism can be understood by assuming that in the absence of a nucleotide the active site of eEF1A has an “open” structure. The first step of binding probably represents a diffusion-controlled association of the nucleotide with the binding pocket. The resulting weak complex rearranges to a “closed” form of the binding pocket, which stabilizes nucleotide binding and thus increases the nucleotide affinity, and leads to a change in mant fluorescence. A similar biphasic nucleotide binding was observed with eRF3 (36), Ffh (43), and Ras (51). Spontaneous dissociation of GDP from eEF1A takes place at a rate of 0.13 s^{-1} , in excellent agreement with the results of a recent kinetic study (0.17 s^{-1}) (8). The rates of GTP and GDP dissociation from eEF1A are similar, which is the major difference to the nucleotide release from the prokaryotic homolog, EF-Tu, which, in the absence of EF-Ts, releases GDP more slowly (0.002 s^{-1}) than GTP (0.03 s^{-1}) (11, 12). This suggests that the nucleotide binding pocket in the GTP- and GDP-bound form of eEF1A, even in its “closed” form, is more open than that of EF-Tu and that the structural differences between the GTP- and GDP-bound forms of eEF1A is probably less dramatic than in the case of EF-Tu (52–54).

Nucleotide Exchange in eEF1A—In the presence of eEF1B α , the eEF1A·GDP complex dissociates rapidly, at a maximum rate of 42 s^{-1} (at 10 mM Mg^{2+} and $20 \text{ }^\circ\text{C}$). This value is again in very good agreement with the published value of 75 s^{-1} , measured at the same Mg^{2+} concentration and $37 \text{ }^\circ\text{C}$ (8). In the presence of eEF1B α , the release of GDP and GTP is accelerated 320-fold and 250-fold, respectively, relative to the spontaneous release (Fig. 7). This is a rather modest effect, compared with the 60,000-fold acceleration of GDP dissociation from EF-Tu by EF-Ts. However, the rate of the nucleotide release from the ternary complex eEF1A·GDP·eEF1B α is only 3-fold lower than from EF-Tu·GDP·EF-Ts (125 s^{-1}) (11). The main difference is a much faster (65-fold) spontaneous dissociation of GDP from eEF1A compared with EF-Tu, which is also one of the main quantitative differences in the kinetic mechanisms of nucleotide exchange between eEF1A and EF-Tu. Another difference is the 30-fold lower stability of the eEF1A·eEF1B α complex in the absence of a nucleotide compared with EF-Tu·EF-Ts, which most likely reflects the fact that the contacts in the eEF1A·eEF1B α complex are quite different from those in EF-Tu·EF-Ts (3).

Another interesting implication of the data is that the bimolecular rate constant of eEF1B α binding to eEF1A is largely

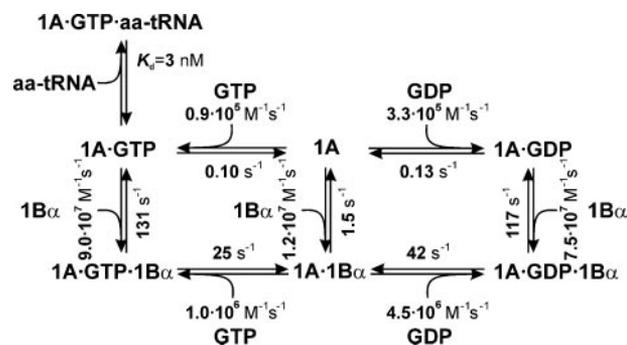


FIGURE 7. Kinetic mechanism of nucleotide exchange and aa-tRNA binding to eEF1A from *S. cerevisiae*.

independent of the nucleotide binding state of eEF1A. Furthermore, the rates of eEF1B α binding to eEF1A were very similar to those of the formation of the EF-Ts·EF-Tu complex with or without nucleotides, which is remarkable, given that the GEFs are entirely unrelated in sequence. The values of the association rate constants k_3 and k_6 suggest a diffusion-controlled reaction, assuming the encounter frequency of $7 \cdot 10^9$ and a steric factor of about 0.01 (31). This would explain why the association rates are insensitive to the structure of the contact surface and are similar to those determined for other GTPase-GEF complexes such as Ran-RCC1 (55). Specific interactions in the respective protein-protein complexes are expected to affect the following rearrangements steps which were however not observed in the present study. The crystal structures indicate that binding of the eEF1B α fragment to eEF1A, eEF1A·GDPNP, or eEF1A·GDP produced only minor changes in the relative orientations of the domains in all complexes (4). The changes in the two proteins were limited to new conformations of a few side chains surrounding the nucleotide binding site. Such small rearrangements may be very rapid and will probably not give rise to discernible kinetic steps.

Assuming that the concentrations of eEF1A and tRNAs in yeast cells are comparable, about $100 \mu\text{M}$ (56), the ratio of eEF1A to eEF1B α is 4:1 to 3:1 (7), and GTP is present in a 10–30-fold excess over GDP, the effective rate of nucleotide exchange can be calculated for these concentrations *in vivo*. Even though a fraction of eEF1A is likely to be bound to actin (57) and thus may be inaccessible for eEF1B α binding, the rate of eEF1B α binding to eEF1A·GDP, calculated from the values of k_3 , k_{-3} , and k_{-4} (observed rate = $k_3[\text{eEF1B}\alpha]/(1 + k_{-3}/k_{-4})$), is very high, because of the high concentration of eEF1B α , and clearly not rate-limiting for the reaction. The effective dissociation rate of GDP from the ternary complex is in the range of $8\text{--}18 \text{ s}^{-1}$ depending on the assumed GTP/GDP ratio (rate = $k_{-4}/[1 + (k_4[\text{GDP}])/(k_7[\text{GTP}])]$). The following step of GTP binding is very fast ($>1000 \text{ s}^{-1}$), because of the high intracellular concentration of GTP. The dissociation of eEF1B α from the eEF1A·GTP·eEF1B α complex takes place at about 13 s^{-1} (rate = $k_{-6}/[1 + (k_6[\text{eEF1B}\alpha])/k_{\text{aa-tRNA-binding}}[\text{aa-tRNA}]]$), which appears to be the second partially rate-limiting step of nucleotide exchange, in addition to GDP dissociation. In total, the overall rate of nucleotide exchange is expected to be in the range of about 6 s^{-1} . The rate of protein synthesis in yeast cells is about 2.3 s^{-1} at doubling times characteristic for the temper-

ature, 20 °C, used in the present experiments (58, 59), suggesting that nucleotide release is not limiting at conditions of normal growth. However, because the effective rate of eEF1 α binding to eEF-1A·GDP linearly depends on eEF1 α concentration, the binding step may become at least partially rate-limiting upon eEF1 α depletion, thus decreasing the overall rate of nucleotide exchange and potentially affecting the rate of protein synthesis in the cell.

Structural and kinetic analyses of the nucleotide exchange in EF-Tu suggested that several interactions with EF-Ts contribute to nucleotide exchange. EF-Ts induces a movement of helix D of the G domain of EF-Tu that shifts residues that are involved in the stabilization of the ribose and the guanine base away from the nucleotide-binding site, thereby relaxing the interactions of those residues with the ribose and/or guanine base (5, 6). A residue in the conserved TDFV sequence motif of EF-Ts, Phe⁸¹, intrudes between switch II region and His¹¹⁸ of EF-Tu and disrupts the binding of the β -phosphate of GDP and shifts the position of helix B of EF-Tu and alters the position of the residues coordinating the Mg²⁺ ion in the EF-Tu·GDP complex. Somewhat disappointingly, the disruption of any of the putative key interactions in the EF-Tu·EF-Ts complex resulted in only small to moderate changes in the efficiency of nucleotide exchange (60–64). It is possible that in addition to the contacts indicated by crystal structures, EF-Ts binding induces many small rearrangements of EF-Tu that contribute synergistically to efficient exchange of guanine nucleotides. Recognition of eEF1A by eEF1 α is very different from that of EF-Tu by EF-Ts (3, 4). The residues of eEF1A that interact with the base and sugar moieties of GTP are undisturbed. The structure of the Mg²⁺ binding site is altered by a reorganization of the switch 2 region and the insertion of Lys²⁰⁵ of eEF1 α into the binding site. As a result, the interactions with the β - and γ -phosphates of GDP or GTP are disrupted, which may accelerate nucleotide release. The peptide flip in the P loop takes place in both eEF1A·eEF1 α and EF-Tu·EF-Ts, and this makes GDP binding unfavorable (3, 6). Lys²⁰⁵ of eEF1 α appears to be important for the mechanism of nucleotide exchange, and the K205A mutation is lethal due to impaired GEF function (4, 8). However, mutagenesis and kinetic studies showed that the mutation reduced the rate of GDP release from eEF1A by a factor of 13 only (at 1 mM Mg²⁺) (8). Likewise, removal of Mg²⁺ increased the rate of eEF1 α -induced GDP dissociation no more than 6-fold (8). This suggests that, similarly to EF-Tu·EF-Ts, any contact in the eEF1A·eEF1 α complex alone is expected to contribute moderately to the destabilization of nucleotide binding, but together they act synergistically to bring about the overall acceleration of nucleotide exchange.

The role of the nucleotide exchange co-factor, eEF1 β , is not clear. eEF1 β by itself has no measurable exchange activity, but has a small positive effect on nucleotide exchange in the complex with eEF1 α , *i.e.* it increases the rate of GDP dissociation from yeast eEF1A in the presence of eEF1 α by a factor of two (7), or from *Artemia* eEF1A by a factor of 1.6 (14). However, eEF-1 γ contains a hydrophobic tail and appears to have an affinity toward membrane and cytoskeletal elements (65) and RNA (66), and could thereby contribute to the anchoring of mRNAs and translation components in the vicinity of cytoskeleton- or membrane-

bound ribosomes (67). In addition, loss of the two yeast genes encoding eEF1 β results in constitutive resistance to oxidative stress (68). This may indicate that the activity of the eEF1B complex responds to stress, which remains to be determined.

eEF1A Interaction with aa-tRNA—Yeast eEF1A·GTP binds aa-tRNA very tightly, with nanomolar affinity. This suggests that the local conformation of eEF1A at the aa-tRNA binding pocket is sufficiently different for aa-tRNA to select between the GTP- and the GDP form, despite the similar affinities of eEF1A binding to GTP and GDP. Although gross rearrangements in the eEF1A·eEF1 α complex were not observed upon binding of different nucleotides (4), conformational changes that may occur in the absence of eEF1 α cannot be excluded. In this respect, yeast eEF1A seems to be quite similar to its prokaryotic homolog EF-Tu.

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