Oxygen uptake by aquatic sediments measured with a novel non-invasive eddy-correlation technique

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ABSTRACT: This paper presents a new non-invasive technique for measuring sediment O_2 uptake that, in its concept, differs fundamentally from other methods used to date. In almost all natural aquatic environments, the vertical transport of O_2 through the water column toward the sediment surface is facilitated by turbulent motion. The new technique relies on measuring 2 parameters simultaneously and at the same point in the water above the sediment: the fluctuating vertical velocity using an acoustic Doppler velocimeter and the fluctuating O_2 concentration using an O_2 microelectrode. From these 2 parameters, which typically are measured 10 to 50 cm above the sediment surface for a period of 10 to 20 min and at a frequency of 15 to 25 Hz, the vertical flux of O_2 toward the sediment surface is derived. Based on measurements performed under actual field conditions and comparisons with in situ flux-chamber measurements, we believe that this new technique is the optimal approach for determining O_2 uptake by sediments. The technique is superior to conventional methods as measurements are done under true in situ conditions, i.e. without any disturbance of the sediment and under the natural hydrodynamic conditions. Furthermore, this technique can be used for bio-irrigated or highly permeable sediments, such as sands, where traditional methods often fail. While this paper only focuses on O_2 uptake by sediments, the technique can also be applied to other solutes that can be measured at a sufficiently high temporal resolution.

KEY WORDS: Oxygen uptake · Sediment · Eddy-correlation measurements

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INTRODUCTION

Background. Consumption of O_2 in aquatic sediments is attributed to 3 types of processes: aerobic decomposition of organic matter, animal respiration and oxidation of reduced products of anaerobic decay, including NH₄⁺, Mn²⁺, Fe²⁺, H₂S, FeS, and FeS₂. Sediment O_2 uptake provides valuable information on these processes and is a frequently measured parameter in studies of aquatic sediments. At the same time, it is a parameter that is often difficult to measure accurately with the methods available to date.

Oxygen uptake by sediments is affected strongly by the transport processes responsible for the movement of O_2 from the overlying water down through the sediment. These transport processes include molecular diffusion, bioturbation (the diffusion-like transport of solutes and solids caused by movements of fauna), bioirrigation (the transport of solutes caused by pumping activity of tube-dwelling animals), and current- or wave-driven advection. Recent studies have shown that bioturbation is often a significant transport process, comparable in strength to molecular diffusion (Aller & Aller 1992, Forster et al. 1995, Berg et al. 2001). Likewise, bio-irrigation can be a dominant transport process in sediments densely populated by tube-dwelling animals (Aller 1983, Pelegri et al. 1994, Wang & Van Cappellen 1996). Other transport processes can also stimulate O_2 uptake by sediments. Currents over an uneven bottom or wave action at shallow sites can induce a significant advective transport, either through the interstitial pores in more permeable sediments such as sand, or through tubes and borrows created by fauna (Rutgers van der Loeff 1981, Savant et al. 1987, Thibodeaux & Boyle 1987, Huettel & Gust 1992, Shum 1992, Webster 1992, Webster & Taylor 1992). This advection can be the dominant transport process in sediments, and enhances O_2 uptake manyfold relative to uptake by molecular diffusion (Malan & McLachlan 1991, Forster et al. 1996, Lohse et al. 1996).

The 2 commonly used methods for determining O_2 uptake by sediments are measurements in recovered sediment cores kept in a controlled environment as close to the *in situ* conditions as possible (e.g. Mortimer 1941, Hayes & MacAulay 1959, Rasmussen & Jørgensen 1992), and measurements in *in situ* chambers which isolate a fraction of the sediment surface and bottom water from the surroundings (e.g. Odum 1957, Pamatmat & Fenton 1968, Jahnke & Christiansen 1989). In both cases, the sediment O_2 uptake is determined by measuring O_2 depletion in the overlying water over time. Typically, sediment cores have a diameter of 5 to 10 cm, while chambers cover a larger area of the sediment surface, for example 30×30 cm (Glud et al. 1998, 1999).

One problem with these methods is that they may affect active transport processes that influence O₂ uptake in the undisturbed sediment. While a good representation of in situ molecular diffusion and also in situ bioturbation caused by smaller animals (meiofauna) can be obtained in sediment cores, this method tends to underestimate the effects of bioturbation and bio-irrigation caused by larger animals (macrofauna). Several studies comparing O2 uptake measured in cores and in situ chambers for the same sediment have found O2 uptake to be several-fold lower in cores (Glud et al. 1998, 1999, 2003). The difference was explained in these comparisons as a result of a more realistic representation of irrigating macrofauna in the larger chambers. In addition, when measurements are made in both sediment cores and in situ chambers, the overlying water is kept fully mixed by stirring. This induced rotational flow differs from naturally occurring flow patterns over the sediment surface and represents another problem of these conventional methods. Furthermore, representative O2 uptake generally can not be measured in cores or in in situ chambers, where advective transport processes within a highly permeable sediment are created by current or wave action (Huettel & Webster 2001, Reimers et al. 2001). In addition to these shortcomings, measurements in sediment core and *in situ* chambers are difficult to perform when objects such as rocks, mussels, epifauna, and macroalgae are abundant either on the sediment surface or in the upper sediment layer. Under such circumstances, it is difficult not to disturb the sediment considerably when cores are collected or when *in situ* chambers are deployed.

This paper presents a new non-invasive technique for measuring O₂ uptake by aquatic sediments. In almost all natural environments, O₂ taken up by sediments is transported down through the water column by turbulent motion. The new technique relies on measuring turbulent fluctuations in the vertical velocity and the corresponding O₂ concentration simultaneously and at the same point above the sediment surface. If such measurements are done with an adequate temporal resolution to capture these fluctuations and for a period long enough to obtain a statistically sound representation of their variations, then the vertical flux of O_2 can be derived. If possible to perform from a technical point of view, such measurements obviously do not suffer from the same shortcomings as core and in situ chamber measurements because O2 uptake is determined under true in situ conditions, i.e. without any disturbance of the sediment and under the natural hydrodynamic conditions. This technique has been used for several decades to determine land-atmosphere and sea-atmosphere exchanges of CO₂, moisture, and energy in the atmospheric boundary layer (e.g. Wyngaard 1989), and is commonly referred to as flux measurement by eddy-correlation. This term is adopted here. Despite the clear advantages of the technique, it has only been used a few times in aquatic environments to determine fluxes of energy, for example between water and sea ice (Fukuchi et al. 1997, Shirasawa et al. 1997). The obvious reason is that, other than for temperature, it is difficult to construct a sensor that can measure scalar properties such as O₂ concentration in water at a well-defined point and with a sufficiently fast response to capture all turbulent fluctuations. However, as documented below, we have been able to make these measurements with an O2 microelectrode constructed to have a fast response. The fluctuating vertical velocity is measured using an acoustic Doppler velocimeter (ADV). The ADV gives the full 3D turbulent velocity field and is widely used in aquatic environments to study processes such as the transport of particles in the shear layer over sediment surfaces.

Theory. Fig. 1 shows an example of a vertical profile of the O_2 concentration in a marine muddy sediment and the overlying water measured *in situ* with a microelectrode. The O_2 is consumed in the upper 1.5 mm of the sediment, and the smoothness of the profile suggests that the important mechanisms for the vertical transport of O_2 in the sediment are molecular diffusion

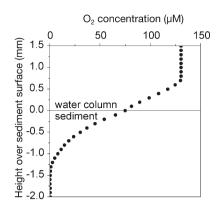


Fig. 1. Vertical profile of O₂ concentration measured *in situ* with a microelectrode in marine muddy sediment and overlying water (Aarhus Bay, Denmark)

and possibly bioturbation caused by meiofauna. In the water column, where the 2 possible transport mechanisms are molecular diffusion and advection, no concentration gradients are visible at 0.7 mm or higher above the sediment surface. This reflects a shift in the dominant transport process from molecular diffusion within the diffusive boundary layer to advection above the layer. Since there was no continuous advective flow directed toward the sediment at this site, the advective transport of O2 results exclusively from turbulent motions. The typical variation in the O₂ concentration and the vertical velocity is that the O_2 concentration is slightly higher when the vertical velocity is pointing down toward the sediment and lower when the velocity is pointing upward. Over time, this correlation gives rise to a net transport of O₂ toward the sediment. An example of this pattern is shown in Fig. 2, from some of our measurements with high temporal resolution of the vertical velocity and the associated O₂ concentration.

The mathematical expression for the vertical O_2 flux due to advection and molecular diffusion at any position in the water column at any point in time is

$$Flux = u_z C - D \frac{dC}{dz}$$
(1)

where u_z is the vertical velocity, *C* is the O₂ concentration, *D* is the molecular diffusivity of O₂ in water, and *z* is the vertical coordinate (e.g. Berner 1980, Boudreau 1997).

Turbulent advection is the dominant vertical transport in the water column above the diffusive boundary layer in almost all natural aquatic environments. For that reason, the diffusive term in Eq. (1) can be neglected. Furthermore, it is common practice when dealing with turbulent motions to separate both instantaneous values of u_z and C into 2 components equal to $\overline{u}_z + u'_z$ and $\overline{C} + C'$, where \overline{u}_z is the mean vertical velocity, u'_z is the vertical turbulent fluctuating velocity, \overline{C} is the mean concentration, and C' is the turbulent fluctuating concentration (e.g. Reynolds 1895, Stanišić 1985, Boudreau 1997). It should be noted that there are no simplifications hidden in these separations. The separations are now substituted into Eq. (1), which is then averaged over a period of time significantly longer than the time scale of the turbulent fluctuations, in which case averages of both u'_z and C' become equal to zero. It is finally assumed that that the mean vertical velocity is equal to zero ($\overline{u}_z = 0$). With these assumptions, Eq. (1) gives the following expression for the vertical O₂ flux averaged over time:

$$\overline{\text{Flux}} = \overline{u'_z C'} \tag{2}$$

where the bars symbolize the averaging. The derivation of Eq. (2) from Eq. (1), often referred to as the Reynolds' decomposition, is described in more detail in numerous publications (e.g. Reynolds 1895, Stanišić 1985, Boudreau 1997). Eq. (2) has been used for several decades to determine fluxes by eddy-correlation in the atmospheric boundary layer (e.g. Wyngaard 1989). In practical applications of Eq. (2), where a time series of u'_{z} (u'_{z1} , u'_{z2} , u'_{z3} , ... u'_{zN}) and C' (C'_{1} , C'_{2} , C'_{3} , ... C'_{N}) is extracted from measurements, the flux is simply calculated as the average of the sum u'_{z1} $C'_{1} + u'_{z2}$ $C'_{2} +$ u'_{z3} $C'_{3} + ... + u'_{zN}$ C'_{N} .

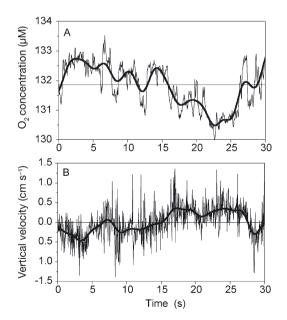


Fig. 2. Time series of (A) fluctuating O_2 concentration (thin line) and (B) associated vertical velocity (thin line). Positive velocity values indicate flow up and away from sediment surface. Data were measured 15 cm above a sediment surface at a frequency of 25 Hz. Mean values for time series are also shown (horizontal lines), as are the smoothed O_2 concentration and the smoothed vertical velocity (thick lines) (Aarhus Bay, Denmark)

When measuring u_z in the water column under timeinvariant conditions, i.e. where no changes in size or direction of the current occur, \overline{u}_z can be defined as the mean of all measured u_z . Since it is difficult to position the ADV so that the u_z is measured exactly perpendicular to the sediment surface (which in turn may not be completely even and well-defined), \overline{u}_z calculated from the measurements is likely to have some small value, and a correction must be made for Eq. (2) to be valid. The correction can be made by rotating the measured 3D velocity field (u_x, u_y, u_z) where u_x and u_y are the 2 horizontal velocity components) so \bar{u}_z equals zero. If \bar{u}_z is only slightly off zero, the correction can be made simply by subtracting \overline{u}_z from each individual u_z value. In the latter case, u'_z becomes equal to $u_z - \overline{u}_z$ and the average of all u'_z will consequently always equal zero, as assumed in the derivation of Eq. (2). If time-variant conditions are present, the correction must vary in time, and u'_{z} can, for example, be calculated from the running average of u_z . A challenging aspect of this more advanced correction, as discussed in detail below, is to determine how many adjacent data points to include in the running average. If too few points are included, some of the larger scale turbulent fluctuations are filtered out, and if too many points are included, non-turbulent motions may affect the flux calculation. The isolation of C' from the measured O_2 concentration (C), is done along the same lines.

Analysis of our measurements have shown that u'_z can often be calculated as $u_z - \overline{u}_{z_i}$ while C' in most situations technically should be calculated from the running average of C, since non-turbulent variations in C occur, presumably caused by a slow drift in the O_2 microelectrode calibration. However, the running average of C can be avoided if these variations happen on a time scale significantly larger that the scale of the turbulent fluctuations. In such situations C can be used directly in the flux calculation as

$$\overline{\text{Flux}} = \overline{u'_z C} \tag{3}$$

Eq. 3 represents the simplest way to determine sediment O_2 uptake from a measured time series of u_z and C.

If both u'_z and C' are extracted from the measured data, other advantages exist. Both Eqs. (2) and (3) represent an integration in the time domain. This integration can also be performed in the frequency domain after fast Fourier transformation of the time series of u'_z and C'. More specifically, the vertical O₂ flux can also be expressed as the integral over the frequency of the 1-sided cospectrum of u'_zC' (e.g. Stull 1988, Priestley 1992):

$$\overline{\text{Flux}} = \int_{0}^{\infty} \text{Co}_{u'_{z}c'}(f) df$$
(4)

where f is the frequency. This means that the cospectrum of $u'_z C'$ itself expresses the magnitude of the vertical O_2 flux at different frequencies. As shown below, this information can be used to evaluate if the O_2 microelectrode responds fast enough to capture all turbulent fluctuations facilitating the vertical transport of O_2 .

MATERIALS AND METHODS

Instruments. The ADV is a commercially available instrument manufactured by companies such as Nortek AS and SonTek/YSI. The ADV measures the 3D velocity in a small cylindrical volume, approximately 1.5 cm long and 0.6 cm in diameter, located on the ADV's center line, 10 cm from the base of the 3 sensors (Fig. 3). Velocities are determined with a frequency as high as 25 Hz. The ADV measurements are largely unaffected by water quality (O'Riordan et al. 1996). However, the speed of sound in water is correlated to temperature and salinity, and these parameters must be known in order to obtain the most accurate determination of the 3D velocity field.

The O_2 microelectrode is of the Clark-type (Revsbech 1989) and, using a thin silicone membrane and a compact tip, is constructed so as to have a short response time. The sensor has a 10 µm diameter tip and a 0.2 s response time for a 90% response to concentration changes. We constructed the electrode our-



Fig. 3. Acoustic Doppler velocimeter and O_2 microelectrode mounted on a rack. Arrow indicates position of acoustic Doppler velocimeter's measuring volume and tip of O_2 microelectrode

selves, but electrodes with similar specifications are produced and sold by companies such as Unisense A/S. The O_2 microelectrode was connected to a pico-ampere meter. To maximize signal-to-noise ratio, sensitivity and signal speed, the cable connections between sensor and pico-ampere meter were kept as short as possible. This was achieved by using a customized submersible pico-ampere meter in which the sensor was inserted directly into the amplifier housing (Fig. 3). Since seawater acts as a shield, this picoampere meter can resolve significantly smaller fluctuations *in situ* than under normal laboratory conditions.

Both the ADV and the O_2 microelectrode were mounted on a rack constructed for initial tests of the technique (Fig. 3). The tip of the microelectrode was located immediately next to the border of the ADV's measuring volume. The rack was designed so that the position of the measuring volume and the tip of the microelectrode could be adjusted 10 to 55 cm above the sediment surface. Prior to measurements, the rack was positioned manually on the sediment surface with the sensors facing into the current, and the ADV and pico-ampere meter were connected to the surface through independent cables. Data acquisition was controlled above the water surface by an IBM-compatible PC. The PC and supporting electronics were powered by batteries and were isolated galvanically from all other electronic devices.

Both the velocity and the O_2 concentration measurements contained some high frequency noise (Fig. 2). The noise from the ADV is purely random, can be assumed to follow a Gaussian distribution, and consequently converges to the value zero when averaged over multiple data points without introducing bias (SonTek/YSI 2001). The electronic circuitry and components behind the O_2 concentration measurements are all well-known and well-described (e.g. Tietze & Schenk 2003) and have noise characteristics similar to the ADV. This was confirmed by measurements during which the O_2 microelectrode and the ADV were positioned apart. These measurements led to the calculation of an O_2 uptake of zero.

Measurements. The technique was tested under field conditions on 3 occasions.

The first series of measurements was done in the River Wümme near Bremen, Germany, which is approximately 7 m wide, with a water depth of 1 m and a sandy bottom. This site allowed us to keep all equipment not mounted on the rack on land, and also to make easy adjustments of the orientation of the rack and the measuring height above the sediment surface. Additional measurements for comparison of the O_2 uptake in recovered sediment cores or *in situ* chambers were not made, since these 2 methods are likely to fail in permeable sands, as described above.

At the time of measurement, the river current was 13 cm s⁻¹. We sampled three 10 min time series for each of 4 measuring heights: 15, 20, 40 and 55 cm above the sediment surface. Each time series was sampled at a frequency of 25 Hz to ensure that all concentration changes that could be registered by the O_2 microelectrode were sampled.

The second series of measurements was done in Aarhus Bay, Denmark. The sediment was fine-grained marine mud and was located at 12 m water depth. At the time of measurement, the current was 2 cm s⁻¹ at 15 cm above the sediment surface. We sampled four 10 min time series 15 cm above the sediment surface at a frequency of 25 Hz. For comparison with the eddycorrelation measurements, 4 in situ chambers were deployed. These chambers had an inner diameter of 19 cm and were 32 cm in height. The lid on each chamber had a large opening that was closed by a stopper after deployment, a sampling port with syringe holder for water samples, and a second port for replacement of sampled water. The water inside the chambers was stirred during measurements by a flat rotating disk, 15 cm in diameter and 1 cm in height. Throughout the 3 h incubation, the disks were rotated at a speed of 20 rpm in a position approximately 7 cm above the sediment surface. Prior to measurements, the chambers were pushed approximately 20 cm into the sediment, enclosing a water volume of approximately 3.2 l. A more accurate estimate of this water volume was calculated from the dilution of 50 ml of a neutrally buoyant sodium bromide solution injected into each chamber. Oxygen concentrations in the chambers were monitored during incubation with O₂ optrodes (Presens) mounted in the lids. In addition, the O₂ concentrations in the chambers were determined by Winkler titration of water samples retrieved at the beginning and the end of the incubation. Oxygen decreased by approximately 20% in the chambers during the incubation. For additional comparison, 3 vertical profiles of O2 concentration were measured with microelectrodes in sediment cores collected at the site. The deployment of the rack holding the ADV and the O2 microelectrode, the deployment of the in situ chambers and the collection of sediment cores were done by divers.

The third and last series of measurements was carried out in Limfjorden Sound, Denmark. The sediment was a fine-grained marine mud and was located at 8 m water depth. At the time of measurement, directional changes in the current were observed and its strength varied between 1.5 and 4 cm s⁻¹ at 40 cm above the sediment surface. We sampled six 10 min time series 40 cm above the sediment surface at a frequency of 25 Hz. For comparison with the eddy-correlation measurements, 6 *in situ* chambers were deployed and operated as described for the Aarhus Bay site.

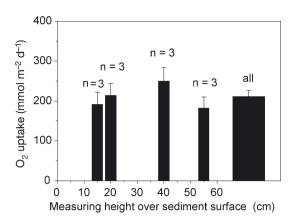


Fig. 4. Mean (+1 SE) O₂ uptake by sandy river sediment (River Wümme, Germany) determined by the eddy-correlation technique

RESULTS AND DISCUSSION

The results of the eddy-correlation measurements in the River Wümme are shown in Fig. 4. The sediment O_2 uptake was calculated from Eq. (3), with u'_z determined as $u_z - \overline{u}_z$. Some variation in uptake for the different measuring heights was expected, since the rates represent an increasing upstream area of the sediment surface as the measuring heights increased. However, the mean values of the O₂ uptake did not differ significantly (ANOVA, p > 0.18), and an overall mean of 210 \pm 16 mmol m⁻² d⁻¹ was calculated from the combined data. It is expected that some advective transport in the upper sediment layer was induced at a main current of 13 cm s^{-1} during the eddy-correlation measurements, which presumably explains the relatively high O_2 uptake. All potential shortcomings and errors in the eddy-correlation measurements, such as the response time of the O₂ microelectrode being too long to capture

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Profiles

Fig. 5. Mean (+1 SE) O_2 uptake by muddy marine sediment (Aarhus Bay, Denmark) determined by O_2 micro-profiles, *in situ* chambers and the new eddy-correlation technique

Chambers

New

technique

all important turbulent fluctuations, would lead to an underestimation of the O_2 uptake. In the light of this, the high uptake, combined with the relatively small error estimates, was viewed as a promising result for the new technique.

Also for the time series from Aarhus Bay, u'_z was determined as $u_z - u_{z_l}$ and the sediment O₂ uptake was calculated from Eq. (3). The calculated uptake is shown in Fig. 5, as is the result from the in situ chambers and the uptake determined from O₂ micro-profiles using the gradients in the diffusive boundary layer. Bio-irrigation is not included in the latter method, which explains the lower estimated O₂ uptake relative to the chamber measurements. The O₂ uptake determined in chambers and by the new technique differ significantly (ANOVA, p = 0.03), and the difference is probably due to an underrepresentation of faunal activity, especially the irrigating macro-fauna, in the chambers. Since the current was low when the measurements were made, and given the impermeable structure of the muddy sediment, it seems unlikely that advective transport, other than that induced by irrigating fauna, would influence the O_2 uptake in the sediment. The relative standard error estimate of 7% for the O₂ uptake determined by eddy-correlation was of the same magnitude as the 10 and 6% standard errors in the chamber measurements and the flux calculations from micro-profiles, respectively.

As a result of the time-variant current during the measurements in Limfjorden, u'_z was determined from the running average of u_z before the sediment O_2 uptake was calculated from Eq. (3). The number of adjacent data points used in the running average was determined in an analysis as detailed below. The calculated O_2 uptake is shown in Fig. 6, as is the result from the *in situ* chambers. The 2 uptake estimates do not differ significantly (ANOVA, p = 0.50), and have

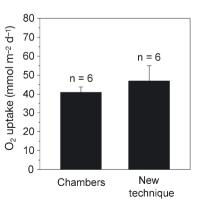


Fig. 6. Mean (+1 SE) O₂ uptake by muddy marine sediment (Limfjorden Sound, Denmark) determined by *in situ* chambers and the new eddy-correlation technique

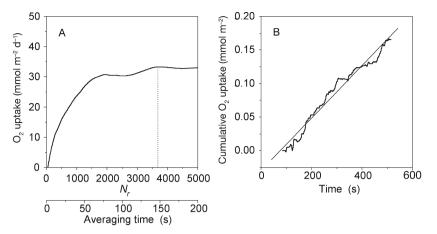


Fig. 7. (A) Determination of number of data points (N_r) used in calculation of running average O_2 concentration; dotted line represents chosen N_r value of 3700, corresponding to averaging over 148 s (lower x-axis). (B) Cumulative O_2 uptake over time; thin line represents linear fit

relative standard error estimates of 7% for the chamber measurements and 17% for the new technique. The latter standard error is considerably larger than for the other sites due to several factors. First, some natural variation in the uptake was expected with the observed changes in current, both in direction and strength. Second, the first and the last data points in the 10 min time series were used only to calculate the running average of u_z and were not used directly in the flux calculation. The uptake was for that reason calculated from values of u'_z and *C* that only covered a time interval of less than 8 min; this obviously increased the uncertainty of the calculation. A smaller standard error would likely have been found if time series longer than 10 min had been measured at this site.

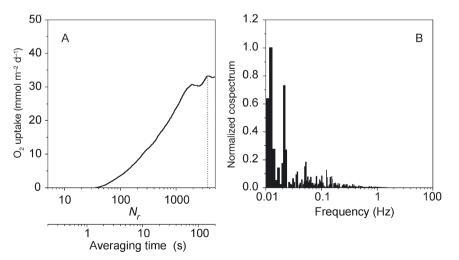


Fig. 8. (A) Data in Fig. 7A on logarithmic x-axis. (B) Normalized 1-sided cospectrum of $u'_z C'$ (see Eq. [1] showing frequencies of turbulent fluctuations responsible for vertical O₂ transport)

A further analysis of the eddy-correlation measurements requires that the fluctuating component of the O₂ concentration, C', is also known. Data from the Aarhus Bay site were used in the example presented below. Since the measured O₂ concentration generally contained a slow variation over time, presumably representing a drift in the O_2 microelectrode calibration, C' must be determined from the running average of C. The number of adjacent data points, N_n to be included in the running average was found by calculating the O_2 uptake repeatedly from Eq. (2) for increasing N_r and observing how the uptake changed. An example for one time series is shown in Fig. 7A. Obviously, if $N_r = 1$, all fluctuations in the time series are filtered out, which

means that C' equals zero and results in an O_2 uptake of zero. For increasing N_r , an increasing number of turbulent fluctuations are included in the calculation, those with the highest frequency first. The resulting increase in O_2 uptake levels off and reaches a near-constant value of 33 mmol m⁻² d⁻¹. It should be noted that the same O_2 uptake was calculated for this particular time series by using Eq. (3). The value of N_r chosen from these results and used in the further calculations was 3700, corresponding to averaging over 148 s (Fig. 7A). By knowing both C' and u'_z , the cumulative O_2 uptake for the time series can be calculated; this exhibited a relatively steady increase through time, as shown in Fig. 7B. The first and the last 1850 data points of a total of 15000 were used only to calculate the run-

ning average of *C*, and were not used directly in the flux calculation. This represents a disadvantage of the running average method or other similar averaging methods.

One of our initial concerns in evaluating the new technique was whether the O_2 microelectrode would respond fast enough to capture all turbulent fluctuations responsible for the vertical O_2 transport toward the sediment. If not, the result would obviously be an underestimation of the O_2 uptake. Although the high uptake calculated using the technique suggested that all fluctuations had been accounted for, we investigated this further. From Fig. 8A, which shows the same data as Fig. 7A but on a logarithmic x-axis, it is evident that no contributions to O_2 uptake are calculated for N_r smaller than 40. At the sampling frequency of 25 Hz used, this value of N_r corresponds to averaging over 1.6 s. Thus, it is reasonable to assume that turbulent fluctuations with frequencies higher than 1 Hz, if any, would be included in the calculation at N_r equal to 40. It should be emphasized that all measurements were done using an O₂ microelectrode with a 0.2 s response time for a 90% response to concentration changes. With this in mind, the results in Fig. 8A indicate that the turbulent transport was facilitated exclusively by fluctuations with frequencies lower than 1 Hz. This result is supported by a calculation of the 1-sided cospectrum of $u'_z C'$ (Fig. 8B), for the same time series. Although longer time series are required to obtain a smoother variation of the cospectrum, no contributions to the O₂ uptake are evident for frequencies higher than 1 Hz. Similar results were obtained in an equivalent analysis of the first series of measurements for the sandy river sediment, where current velocity was more than 6 times greater (data not shown).

The tests of the new technique presented here were conducted under actual field conditions and resulted in successful determination of O_2 uptake (Figs. 4, 5 & 6). An analysis of the turbulent fluctuations showed that the response of the O2 microelectrode was sufficiently fast to capture all turbulent eddies contributing to the vertical transport of O₂ toward the sediment (Figs. 7 & 8). Based on these results, we conclude that O_2 uptake by aquatic sediments can be determined by the new eddy-correlation technique, and we believe that the technique will become a standard approach for determining O₂ uptake by sediments in the future. The technique is superior to conventional methods, as measurements are done under true in situ conditions, i.e. without any disturbance of the sediment and under natural hydrodynamic conditions. In addition, the technique can be used for bio-irrigated or highly permeable sediments, such as sands, where traditional methods often fail. While this paper only focuses on O_2 uptake by sediments, the technique can also be applied to other solutes that can be measured at sufficiently high temporal resolution.

We are currently constructing a new device that can position the ADV and the O_2 microelectrode at a defined height above the sediment surface, turn these instruments in the horizontal plane, lock them into a position facing the current, and perform measurements, all without support from divers. All electronic equipment will be mounted on the device and we anticipate that sampling can be done under most weather conditions down to a water depth of 200 m. The new device will be used to collect data for various sediments and field conditions, including different current velocities and in the presence of density gradients which have the potential for dampening vertical turbulence. A future refinement will be to modify the device into a freefalling lander that can descend, land, collect data, and then ascend completely free from any connection to the surface for use at geater depths. We are also currently working on predicting the area of the sediment surface that contributes to the measured O_2 uptake. This is done through numerical-mathematical modeling of the 3D transport of O_2 in the layer between the sediment surface and the ADV and O_2 microelectrode.

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