

Supporting Information

Gold Difluorocarbenoid Complexes: Spectroscopic and Chemical Profiling

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Figure S1. Structure of complex 10b in the solid state

X-ray Crystal Structure Analysis of Complex 11b: (C₁₉ H₃₃ Au F₃ P), $M_r = 546.39 \text{ g} \cdot \text{mol}^{-1}$, colourless plate, crystal size 0.134 x 0.054 x 0.021 mm³, monoclinic, space group $P2_1$, a = 10.4678(13) Å, b = 9.1318(11) Å, c = 10.7661(13) Å, $\beta = 104.961(2)^\circ$, V = 994.2(2) Å³, T = 100(2) K, Z = 2, $D_{calc} = 1.825 \text{ g} \cdot \text{cm}^3$, $\lambda = 0.71073$ Å, $\mu(Mo-K_{\alpha}) = 7.504 \text{ mm}^{-1}$, Gaussian absorption correction (T_{min} = 0.58, T_{max} = 0.90), Bruker AXS Enraf-Nonius Kappa Mach3 IµS Apex-II diffractometer, 2.968 < θ < 34.337°, 35450 measured reflections, 8294 independent reflections, 7471 reflections with $I > 2\sigma(I)$, $R_{int} = 0.0445$, Absolute structure parameter = 0.001(4).

The structure was solved by direct methods and refined by full-matrix least-squares against F^2 to $R_1 = 0.027$ [$I > 2\sigma(I)$], $wR_2 = 0.048$, 217 parameters. Several low-angle reflections were shadowed by the beamstop and were omitted before the final refinement cycles.

The H atoms were refined using a rotational group riding model, S = 1.043, residual electron density 1.5 /-1.4 e Å⁻³. CCDC 1899834.



Figure S2. Structure of complex **10c** in the solid state; the gold atom is disordered over three positions (only one is shown for clarity)

X-ray Crystal Structure Analysis of Complex 10c: $(C_{34} H_{49} Au F_3 P)$, $M_r = 742.67 \text{ g} \cdot \text{mol}^{-1}$, colourless prism, crystal size 0.16 x 0.11 x 0.07 mm³, monoclinic, space group $P2_1/n$, a = 9.2347(5) Å, b = 14.5805(14) Å, c = 23.829(2) Å, $\beta = 97.258(7)^\circ$, V = 3182.8(5) Å³, T = 100(2) K, Z = 4, $D_{calc} = 1.550 \text{ g} \cdot \text{cm}^3$, $\lambda = 0.71073$ Å, $\mu(Mo-K_{\alpha}) = 4.711 \text{ mm}^{-1}$, Gaussian absorption correction ($T_{min} = 0.49$, $T_{max} = 0.73$), Bruker AXS Enraf-Nonius KappaCCD diffractometer, 2.924 < θ < 33.035°, 63545 measured reflections, 12033 independent reflections, 10475 reflections with $I > 2\sigma(I)$, $R_{int} = 0.0319$.

The structure was solved by direct methods and refined by full-matrix least-squares against F^2 to $R_I = 0.034 \ [I > 2\sigma(I)], \ wR_2 = 0.080, 366$ parameters. Several low-angle reflections were shadowed by the beamstop and were omitted before the final refinement cycles.

The H atoms were refined using a rotational group riding model, S = 1.101, residual electron density 1.7 /-1.8 e Å⁻³. CCDC 1899837.



Figure S3. Structure of complex **10d** in the solid state; disorder of one of the six *tert*-butyl groups is not shown for clarity (only one is shown for clarity)

X-ray Crystal Structure Analysis of Complex 10d: (C₄₃ H₆₃ Au F₃ O₃ P), $M_r = 912.87$ g · mol⁻¹, colourless prism, hexagonal, space group $P6_3/m$, a = 15.0613(2) Å, b = 15.0613(2) Å, c = 10.8267(2) Å, V = 2126.92(6) Å³, T = 80.15 K, Z = 2, $D_{calc} = 1.425$ g · cm³, $\lambda = 0.61990$ Å, μ (synchrotron) = 3.545 mm⁻¹, Semi-empirical absorption correction (T_{min} = 1.00, T_{max} = 0.73), PETRA III synchrotron, 2.1322 < θ < 32.0330°, 53844 measured reflections, 3535 independent reflections, 3284 reflections with $I > 2\sigma(I)$, $R_{int} = 0.1204$ Absolute structure parameter = 0.089(9).

The structure was solved by direct methods and refined by full-matrix least-squares against F^2 to $R_1 = 0.048 [I > 2\sigma(I)]$, $wR_2 = 0.119$, 165 parameters.

The H atoms were refined using a rotational group riding model, S = 1.131, residual electron density 3.5 /-0.5 e Å⁻³. CCDC 1899836.



Figure S4. Structure of complex 11 in the solid state

X-ray Crystal Structure Analysis of Complex 11: $(C_{45} H_{26} Au_2 Cl_2 F_{24} P_2)$, $M_r = 1549.43 \text{ g} \cdot \text{mol}^{-1}$, colourless plate, crystal size 0.091 x 0.069 x 0.027 mm³, monoclinic, space group *I*2, *a* = 15.640(3) Å, b = 8.6193(19) Å, c = 19.495(4) Å, $\beta = 110.974(18)^\circ$, V = 2453.9(9) Å³, T = 100(2) K, Z = 2, $D_{calc} = 2.097 \text{ g} \cdot \text{cm}^3$, $\lambda = 0.71073$ Å, $\mu(Mo-K_{\alpha}) = 6.271 \text{ mm}^{-1}$, Gaussian absorption correction ($T_{min} = 0.62$, $T_{max} = 0.86$), Bruker AXS Enraf-Nonius Kappa Mach3 IµS Apex-II diffractometer, $2.790 < \theta < 30.505^\circ$, 30928 measured reflections, 7407 independent reflections, 7406 reflections with $I > 2\sigma(I)$, $R_{int} = 0.0250$, Absolute structure parameter = 0.010(2).

The structure was solved by direct methods and refined by full-matrix least-squares against F^2 to $R_1 = 0.033$ [$I > 2\sigma(I)$], $wR_2 = 0.045$, 343 parameters. Several low-angle reflections were shadowed by the beamstop and were omitted before the final refinement cycles. The H atoms were refined using a rotational group riding model, S = 1.047, residual electron density 1.5 /-0.8 e Å⁻³. **CCDC 1899835**.

The structure of complex **11** in the solid state (Figure S4) suggests that the pairing of the discrete $[L_2Au]^+$ and $[Au(CF_3)_2]^-$ ions is reinforced by an aurophilic interaction (Au1....Au2 2.88 Å); this contact is close to the shorter end of Au...Au interactions, which typically range from 2.75 to 3.4 Å and are stabilizing to about 5-10 kcal·mol⁻¹.¹ The axes of the cationic unit and the bis(trifluoromethyl)aurate entity are almost orthogonal to each other likely for steric reasons (C1–Au1–Au2–P2 102.9°). For a discussion of related pairs of ionic gold complexes, see ref.²

General. All reactions were carried out under an Ar atmosphere in flame-dried glassware unless stated otherwise. Solvents were purified by distillation over the indicated drying agents and were transferred under Ar: Et_2O (Mg/anthracene), CH_2Cl_2 (CaH₂), pentane (Na/K).

NMR: Spectra were recorded on a Bruker Avance III HD 400 MHz or Avance III 500 MHz spectrometer in the indicated solvents; chemical shifts (δ) are given in ppm relative to TMS, coupling constants (*J*) in Hz. Signal assignments were established using NOESY, HSQC and HMBC and other 2D experiments; numbering schemes as shown in the Inserts.

IR: Perkin-Elmer Spectrum One spectrometer, wavenumbers (\tilde{v}) in cm⁻¹. MS: EI: Finnigan MAT 8400 (70 eV), ESI: Thermo Scientific LTQ-FT or Thermo Scientific Exactive, GC-EI: Thermo Scientific Trace GC Ultra with a Thermo Scientific ISQ spectrometer; accurate mass determinations: Finnigan MAT 95, Thermo Scientific LTQ FT, or Thermo Scientific Exactive. Flash chromatography: Merck Geduran® Si 60 (40–63 µm).

Unless stated otherwise, all commercially available reagents (Aldrich, TCI-Europe, Strem, ABCR) were used as received.

General procedure for the Preparation of Complexes 10a-d. Complexes 10a-d were prepared according to a modified literature procedure.³ To a solution of LAuCl (1 equiv) in CH_2Cl_2 (10 mL) and MeCN (10 mL) was added AgF (1.05 equiv) and TMSCF₃ (2 equiv). The resulting suspension was stirred in the dark for 12 h before it was filtered. The filtrate was evaporated to dryness and the residue was purified by flash chromatography to give the title compounds as colorless solid each.

The analytical and spectroscopic data of complexes 10a (98%) and 10b (93%) were in excellent agreement with those reported in the literature.³

Complex 10c. XPhosAuCl (0.24 mmol, 172 mg), AgF (0.26 mmol, 33 mg) and TMSCF₃



(0.68 mmol, 100 µL) were used; yield: 168 mg (93%). ¹H NMR (500 MHz, 298 K, CD₂Cl₂): δ 0.94 (d, J = 6.7 Hz, 6H, 19), 1.22 (m, 1H, 4(ax)), 1.25 (m, 1H, 5(ax)), 1.29 (d, J = 6.9 Hz, 6H, 21), 1.30 (m, 1H, 3(ax)), 1.31 (d, J = 6.9 Hz, 6H, 18), 1.37 (t3d J = 12.5 Hz, 12.3 Hz, 3.5 Hz, 5.8 Hz, 1H, 6(ax)), 1.51 (t3d J = 12.5 Hz, 12.3 Hz, 3.5 Hz, 6.0 Hz, 1H, 2(ax)), 1.69 (m, 1H, 4(eq)), 1.78 (m, 1H,

5(eq)), 1.82 (m, 1H, 6(eq)), 1.86 (m, 1H, 3(eq)), 2.08 (dm, J = 12.5 Hz, 1H, 2(eq)), 2.19 (ttd, J = 12.3 Hz, 3.1 Hz, 10.3 Hz, 1H, 1), 2.28 (qq, J = 6.9 Hz, 6.7 Hz, 2H, 17), 2.92 (sept, J = 6.9 Hz, 1H, 20), 7.11 (s, 2H, 15), 7.20 (dm, J = 4.4 Hz, 1H, 11), 7.49 (m, 1H, 9), 7.50 (m, 1H, 10), 7.63 (dm, J = 7.5 Hz, 1H, 8). ¹⁹F{¹H} NMR (470 MHz, CD₂Cl₂): δ –29.13 (d, J = 40.3 Hz). ³¹P{¹H} NMR (202 MHz, CD₂Cl₂): δ 37.81 (q, J = 40.3 Hz). ¹³C{¹H} NMR (125 MHz, CD₂Cl₂): δ 22.70 (C18), 23.21 (C21), 25.36 (C19), 25.77 (d, J = 1.7 Hz, C4), 26.77 (d, J

12.5 Hz, C3), 27.05 (d, J = 13.0 Hz, C5), 29.99 (d, J = 2.3 Hz, C2), 30.70 (d, J = 2.8 Hz, C6), 30.86 (C17), 33.80 (C20), 36.90 (d, J = 29.4 Hz, C1), 121.60 (C15), 33.80 (C20), 127.04 (d, J = 6.3 Hz, C9), 127.83 (d, J = 45.5 Hz, C7), 130.27 (d, J = 2.2 Hz, C10), 132.68 (d, J = 2.0 Hz, C8), 133.53 (d, J = 8.1 Hz, C11), 135.62 (d, J = 5.5 Hz, C13), 145.76 (C14), 147.27 (d, J = 15.9 Hz, C12), 148.99 (C16), 167.25 (dq, J = 171 Hz, 354 Hz, C22). MS (ESI⁺), found: 765.3082 [M+Na]⁺; calcd for C₃₄H₄₉AuF₃PNa: 765.3082. Crystals suitable for X-ray analysis were obtained by slow evaporation of a solution of the compound in CH₂Cl₂/hexane (1:1).

Complex 10d. (2,4-tBu₂C₆H₃O)₃PAuCl (48 µmol, 42 mg), AgF (51 µmol, 6.5 mg) and



C₆H₃O)₃PAUCI (48 μmol, 42 mg), AgP (31 μmol, 6.5 mg) and TMSCF₃ (101 μmol, 15 μL) were used; yield: 34 mg (78%). ¹H NMR (500 MHz, 298 K, CD₂Cl₂): δ 1.31 (s, 27H, H10), 1.46 (s, 27H, H8), 7.20 (dd, *J* = 8.5 Hz, 2.5 Hz, H4), 7.44 (dd, *J* = 8.5 Hz, 1.5 Hz, H5), 7.48 (dd, *J* = 2.5 Hz, 1.3 Hz, H2). ¹⁹F{¹H} NMR (470 MHz, CD₂Cl₂): δ -30.48 (d, *J* = 65 Hz). ³¹P{¹H} NMR (202 MHz, CD₂Cl₂): δ 127.34 (q, *J* = 65 Hz). ¹³C{¹H} NMR (126 MHz, CD₂Cl₂): 30.21 (C8), 31.06 (C10), 34.53 (C9), 34.98 (C7), 119.08 (C5, d, *J* = 9.1 Hz), 124.06 (C4), 125.54 (C2), 139.15 (C1,

d, J = 6.3 Hz), 147.22 (C6, d, J = 5.6 Hz), 148.30 (C3), 163.04 (dq, J = 275 Hz, 350 Hz, C12). MS (ESI⁺), found: 935.4028 [M+Na]⁺; calcd for C₄₃H₆₃AuF₃O₃PNa: 935.4025. Crystals suitable for X-ray analysis were obtained by slow evaporation of a solution of the compound in CH₂Cl₂/hexane (1:1).

Complex 11. $(p-CF_3C_4H_8)_3PAuCl$ (784 µmol, 548 mg), AgF (827 µmol, 105 mg) and $F_3C_{C2}^{C5}$ TMSCF₃ (1.55 mmol, 230 µL) were used; yield: 552 mg (96%). ¹H NMR (500 MHz, 298 K, CD₂Cl₂): δ 7.69 (dd, J = 12.2 Hz, 8.3 Hz, H2), 7.82 (dd, J = 8.3 Hz, 2.0 Hz, H3). ¹⁹F{¹H} NMR (470 MHz, H2), 7.82 (dd, J = 8.3 Hz, 2.0 Hz, H3). ¹⁹F{¹H} NMR (470 MHz, CD₂Cl₂): δ -29.74 (F6), -63.71 (F5). ³¹P{¹H} NMR (202 MHz, CD₂Cl₂): δ 38.37. ¹³C{¹H} NMR (126 MHz, CD₂Cl₂): 123.34 (q, J = 273.0 Hz, C5), 126.47 (dq, J = 11.6 Hz, 3.7 Hz, C3), 132.10 (d, J = 273.0 Hz, C5), 126.47 (dq, J = 11.6 Hz, 3.7 Hz, C3), 132.10 (d, J = 273.0 Hz, C5), 126.47 (dq, J = 11.6 Hz, 3.7 Hz, C3), 132.10 (d, J = 273.0 Hz, C5), 126.47 (dq, J = 11.6 Hz, 3.7 Hz, C3), 132.10 (d, J = 273.0 Hz, C5), 126.47 (dq, J = 11.6 Hz, 3.7 Hz, C3), 132.10 (d, J = 273.0 Hz, C5), 126.47 (dq, J = 11.6 Hz, 3.7 Hz, C3), 132.10 (d, J = 273.0 Hz, C5), 126.47 (dq, J = 11.6 Hz, 3.7 Hz, C3), 132.10 (d, J = 273.0 Hz, C5), 126.47 (dq, J = 11.6 Hz, 3.7 Hz, C3), 132.10 (d, J = 273.0 Hz, C5), 126.47 (dq, J = 11.6 Hz, 3.7 Hz, C3), 132.10 (d, J = 273.0 Hz, C5), 126.47 (dq, J = 11.6 Hz, 3.7 Hz, C3), 132.10 (d, J = 273.0 Hz, C5), 126.47 (dq, J = 11.6 Hz, 3.7 Hz, C3), 132.10 (d, J = 273.0 Hz, C5), 126.47 (dq, J = 11.6 Hz, 3.7 Hz, C3), 132.10 (d, J = 273.0 Hz, C5), 126.47 (dq, J = 11.6 Hz, 3.7 Hz, C3), 132.10 (d, J = 273.0 Hz, C5), 126.47 (dq, J = 11.6 Hz, 3.7 Hz, C3), 132.10 (d, J = 273.0 Hz, C5), 126.47 (dq, J = 273.0 Hz, C5

52.8 Hz, C1), 134.10 (dq, J = 2.5 Hz, 33.2 Hz, C4), 134.77 (d, J = 14.1 Hz, C2), 164.57 (q, J = 351.0 Hz, C6). MS (ESI⁺), found: 1487.0199 [M+Na]⁺; calcd for C₄₄H₂₄F₂₄P₂Au₂Na: 1487.0193. Crystals suitable for X-ray analysis were obtained by slow evaporation of a solution of the compound in CH₂Cl₂/hexane (1:1).

General procedure for the Generation of Gold Difluorocarbenoids. TMSOTf (1 equiv) was added to a solution of complexes 10a-d (1 equiv) in CD₂Cl₂ (0.4 mL) at -78 °C. The resulting mixture was transferred into an NMR tube, the solution was warmed to -20 °C for \approx 5 min, cooled to -50 °C and analyzed by NMR spectroscopy.

Generation of complex 10e and its sequential transformation into the gold difluorocarbenoid 12e. TMSOTf (1 equiv) was added to a solution of complex 11 (1 equiv) in CD₂Cl₂ (0.4 mL) at -78 °C. The resulting mixture was transferred into an NMR tube and analyzed by NMR spectroscopy (-50 °C), which indicated the formation of 10e and its slow transformation into 12e already at -50 °C. Full characterization of 12e was performed after 1.5 h of the monitoring of the reaction mixture at $\approx 17\%$ conversion.

Complex 10e. ¹H NMR (500 MHz, 223 K, CD₂Cl₂): δ 7.65 (dqd, J = 12.4 Hz, J = 0.7 Hz, J =



8.1 Hz, 6H, H4), 7.79 (ddg, *J* = 8.1 Hz, *J* = 2.0 Hz, *J* = 0.7 Hz, 6H, H3). $\begin{array}{c} & \overset{(2)}{\xrightarrow{C_2}} & \overset{(3)}{\xrightarrow{C_4}} & \overset{(1)}{\xrightarrow{F_6}} F_{F_6} \\ Ar & \overset{(2)}{\xrightarrow{F_6}} & \overset{(3)}{\xrightarrow{F_6}} & F_{F_6} \\ Ar & \overset{(2)}{\xrightarrow{F_6}} & \overset{(3)}{\xrightarrow{F_6}} & F_{F_6} \\ Ar & \overset{(2)}{\xrightarrow{F_6}} & \overset{(3)}{\xrightarrow{F_6}} & F_{F_6} \\ Ar & \overset{(2)}{\xrightarrow{F_6}} & F_{F_6} \\ Ar & \overset$

= 53.9 Hz, J = 1.5 Hz, C5), 133.69 (dq, J = 2.4 Hz, J = 32.9 Hz, C2), 134.90 (d, J = 14.3 Hz, C4), 164.83 (dq, J = 182.0 Hz, 352.0 Hz, C6).

Complex 12a. ¹H NMR (500 MHz, 223 K, CD₂Cl₂): δ 7.45–7.52 (m, 12H, H2, H3), 7.53–



7.60 (m, 3H, H1). ${}^{19}F{}^{1}H{}$ NMR (470 MHz, CD₂Cl₂): δ –76.09 (s, Hz), 129.45 (C2, d, J = 11.5 Hz), 132.16 (C1, d, J = 2.6 Hz),

134.27 (C3, d, J = 13.8 Hz), 169.63 (dt, J = 169.8 Hz, 359.5 Hz, C5).

Complex 12b. ¹H NMR (500 MHz, 223 K, CD₂Cl₂): δ 1.19 (qt, J = 12.8 Hz, 3.0 Hz, 3H, H1(ax)), 1.27 (qm, J = 13 Hz, 6H, H2(ax)), 1.39 (qdd, J = 12.5 Hz, \approx 27.5 Hz, F5). ³¹P{¹H} NMR (202 MHz, CD₂Cl₂): δ 52.72 (t, J = 27.5

Hz). ¹³C{¹H} NMR (126 MHz, CD₂Cl₂): δ 25.78 (d, J = 1.5 Hz, C1), 26.93 (d, J = 12.0 Hz, C2), 30.34 (C3), 32.00 (d, *J* = 28.7 Hz, C4), 174.66 (dt, *J* = 155.4 Hz, 362.5 Hz, C5).

Complex 12c. ¹H NMR (500 MHz, 223 K, CD₂Cl₂): δ 0.89 (d, J = 6.7 Hz, H19), 1.19 (m, H5



(ax)), 1.17 (m, H4 (ax)), 1.23 (d, J = 6.9 Hz, H21), 1.24 (m, H3 (ax)), 1.25 (d, J = 6.7 Hz, H18), 1.47 (qdd, J = 12.4 Hz, 3.4 Hz, 3.3 Hz, H2 (ax)), 1.33 (qdd, J = 12.5 Hz, 3.4 Hz, 5.9 Hz, H6 (ax)), 1.71 (m, H5 (eq)), 1.78 (m, H6 (eq)), 1.79 (m, H3 (eq)), 1.62 (m, H4 (eq)), 2.03 (dm, J = 12.4 Hz, H2 (eq)), 2.13 (tdt, J = 12.4 Hz, 10.5 Hz, 3.2 Hz, H1 (ax)), 2.20 (sept, J = 6.7 Hz, H17), 2.87 (sept, J = 6.9 Hz, H20), 7.09 (s, H15), 7.19 (m, H11), 7.50 (m, H9), 7.51 (m, H10). ¹⁹F{¹H} NMR

(470 MHz, CD₂Cl₂): δ –77.31 (s, br, F26), –29.30 (d, br, J = 29.3 Hz). ³¹P{¹H} NMR (202 MHz, CD₂Cl₂): δ 36.27 (t, J = 29.3 Hz). ¹³C{¹H} NMR (125 MHz, CD₂Cl₂): δ 22.59 (C18), 23.35 (C21), 25.55 (C19), 25.66 (C4, d, J = 1.7 Hz), 26.67 (C3, d, J = 12.5 Hz), 26.98 (C5, d, J = 13.6 Hz), 29.00 (C2, d, J = 1.5 Hz), 30.66 (C6, d, J = 3.1 Hz), 30.93 (C17), 33.66 (C20), 36.17 (C1, d, J = 30.7 Hz), 118.1 (br, q, J = 320 Hz, C25), 121.66 (C15), 126.93 (C7, d, J = 49.0 Hz), 127.43 (C9, d, J = 6.7 Hz), 130.76 (d, J = 2.2 Hz, C10), 132.62 (d, J = 2.3 Hz, C8), 133.09 (d, J = 7.9 Hz, C11), 135.63 (d, J = 5.9 Hz, C13), 145.86 (C14), 146.68 (d, J = 15.2 Hz, C12), 148.71 (C16), 172.24 (dt, J = 160 Hz, 367 Hz, C23).

Complex 12d. ¹H NMR (500 MHz, 223 K, CD₂Cl₂): δ 1.24 (s, overlapped, 27H, H10), 1.46–



1.30 (s, overlapped, 27H, H8), 7.15 (dd, J = 8.6 Hz, 2.5 Hz, 3H, H5), 7.35 (d, J = 8.6 Hz, 3H, H6), 7.42 (m, 3H, H3). ¹⁹F{¹H} NMR (470 MHz, CD₂Cl₂): δ -32.25 (d, J = 44.0 Hz, F1), -76.05 (s, F2). ³¹P{¹H} NMR (202 MHz, CD₂Cl₂): δ 120.11 (t, J = 44.0 Hz). ¹³C{¹H} NMR (126 MHz, CD₂Cl₂): 29.95–30.10 (C8), 31.00–31.15 (C10), 34.60–34.75 (C9), 34.95–35.10 (C7), 118.47 (d, J = 8.9 Hz, C6), 124.25 (C5), 125.73 (C3), 138.80 (d, J = 6.6 Hz,

C2), 146.98 (d, *J* = 5.2 Hz, C1), 148.19 (C4), 165.87 (dt, *J* = 257.0 Hz, 354.0 Hz, C11).

Complex 12e. ¹H NMR (500 MHz, 223 K, CD₂Cl₂): δ 7.65 (dm, J = 12.4 Hz, 6H, H4), 7.80



 $\begin{array}{l} (dm, J = 2.0 \text{ Hz}, 6\text{H}, \text{H3}). {}^{19}\text{F}\{{}^{1}\text{H}\} \text{ NMR (470 MHz, CD}_2\text{Cl}_2): \delta \\ \hline \\ O = -31.66 \ (d, J = 32.2 \text{ Hz}, \text{F6}), -63.35 \ (s, \text{F1}), \text{F7} \text{ was not identified} \\ \hline \\ O = -31.66 \ (d, J = 32.2 \text{ Hz}, \text{F6}), -63.35 \ (s, \text{F1}), \text{F7} \text{ was not identified} \\ \hline \\ O = -31.66 \ (d, J = 32.2 \text{ Hz}, \text{F6}), -63.35 \ (s, \text{F1}), \text{F7} \text{ was not identified} \\ \hline \\ O = -31.66 \ (d, J = 32.2 \text{ Hz}, \text{F6}), -63.35 \ (s, \text{F1}), \text{F7} \text{ was not identified} \\ \hline \\ O = -31.66 \ (d, J = 32.2 \text{ Hz}, \text{F6}), -63.35 \ (s, \text{F1}), \text{F7} \text{ was not identified} \\ \hline \\ O = -31.66 \ (d, J = 32.2 \text{ Hz}, \text{F6}), -63.35 \ (s, \text{F1}), \text{F7} \text{ was not identified} \\ \hline \\ O = -31.66 \ (d, J = 32.2 \text{ Hz}, \text{F6}), -63.35 \ (s, \text{F1}), \text{F7} \text{ was not identified} \\ \hline \\ O = -31.66 \ (d, J = 32.2 \text{ Hz}, \text{F6}), -63.35 \ (s, \text{F1}), \text{F7} \text{ was not identified} \\ \hline \\ O = -31.66 \ (d, J = 32.2 \text{ Hz}, \text{F6}), -63.35 \ (s, \text{F1}), \text{F7} \text{ was not identified} \\ \hline \\ O = -31.66 \ (d, J = 32.2 \text{ Hz}, \text{F6}), -63.35 \ (s, \text{F1}), \text{F7} \text{ was not identified} \\ \hline \\ O = -31.66 \ (d, J = 32.2 \text{ Hz}, \text{F7}), -33.38 \ (dq, J = 1.7 \text{ Hz}, J \text{ Hz}), -33.38 \ (dq, J = 1.7 \text{ Hz}, J \text{ Hz}), -33.38 \ (dq, J = 1.7 \text{ Hz}, J \text{ Hz}), -33.38 \ (dq, J = 1.7 \text{ Hz}), -33.38 \ (dq, J = 1$

= 273.0 Hz, C1), 126.59 (dq, J = 11.7 Hz, J = 3.8 Hz, C3), 131.24 (d, J = 56.0 Hz, C5), 133.85 (dq, J = 2.7 Hz, J = 33.1 Hz, C2), 134.90 (d, J = 14.4 Hz, C4), 167.83 (dt, J = 172.0 Hz, 357.0 Hz, C6), C7 was not identified.

Complex 13. TMSNTf₂ (59 µmol, 21 mg) was added at -78 °C to a solution of complex 10a (59 µmol, 31 mg) in CD₂Cl₂ (0.4 mL). The mixture was transferred $\begin{array}{c} \begin{array}{c} & & & \\ & & \\ Ph \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ & & \\ Ph \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ \end{array} & \begin{array}{c} & & \\ & &$ CD₂Cl₂): δ –43.84 (d (broad), J = 16.9 Hz, F1). ³¹P{¹H} NMR (202 MHz, CD₂Cl₂): δ 35.12 (t, J = 16.9 Hz). ¹³C{¹H} NMR (126 MHz,

CD₂Cl₂): δ 162.83 (td, J = 153.8 Hz, 331.8 Hz, C1). Full assignment of all ¹H NMR signals was precluded due to signal overlap of all of the products and unreacted starting material.

Difluorocarbene Transfer Reactions

Control Reaction in the Absence of a Lewis acid. A solution of complex 10b (49 µmol, 27 mg) and E-stilbene (50 µmol, 9 mg) in [D₈]-toluene (5 mL) was warmed in a sealed NMR tube to 100 °C for 72 h. NMR analysis revealed no changes after this period of time.

Reaction Promoted by TMSOTf. TMSOTf (63.5 µmol, 11.5 µL) was added to a precooled (-78 °C) solution of complex 10a (62.5 µmol, 33 mg) and E-stilbene (63.8 µmol, 11.5 mg) in CH₂Cl₂ (4 mL). The resulting mixture was stirred at this temperature for 1 h, at -20 °C for 15 h, and finally at ambient temperature for 1 h. All volatile materials were evaporated, the residue was triturated with CD₂Cl₂ (0.5 mL), insoluble material was filtered off and the filtrate analyzed by NMR spectroscopy. The solution contained trans-1,1-difluoro-2,3diphenylcyclopropane 14 (15%) and difluoro-alkene 15 (26%, determined using 1,2dichloroethane as an internal standard).

The analogous reaction using Z-stilbene furnished trans-1,1-difluoro-2,3-diphenylcyclopropane trans-14 (10%) and difluoro-alkene 15 (13%, determined using 1,2-dichloroethane as an internal standard); only trace amounts of *cis*-14 seem to be present as indicated by small peaks in the ¹⁹F NMR spectrum of the crude mixture that correspond to the signals of this product.⁴

Spectroscopic data of compound 14: ¹H NMR (500 MHz, 298 K, CD₂Cl₂): δ 3.11 (t, J = 7.6



Hz, 2H, H11), 7.33 (m, 2H, H15), 7.37 (m, 4H, H13), 7.40 (m, 4H, $\begin{array}{ccccccc} & \text{Find} & \text$ C11), 112.99 (t, *J* = 291 Hz, C10), 127.36 (C15), 128.05 (t, *J* = 1.7 Hz,

C13), 128.57 (C14), 133.38 (C12). The recorded data of trans-14 are in excellent accord with the literature.⁵

Spectroscopic data of compound 15: ¹H NMR (500 MHz, 298 K, CD₂Cl₂): δ 4.85 (ddd, J = F_{F1a} 23.1 Hz, 10.5 Hz, 2.5 Hz, 1H, H2), 4.91 (dt, J = 10.5 Hz, 2.3 Hz, 1H, ^{C1} H3), 7.22 (m, 4H, H5), 7.25 (m, 2H, H7), 7.33 (m, 4H, H6), ¹⁹F{¹H} NMR (470 MHz, CD₂Cl₂): δ –88.99 (d, *J* = 43.5 Hz, F1a), –90.46 (d, *J* = 43.5 Hz, F1b). ¹³C{¹H} NMR (126 MHz, CD₂Cl₂): δ 44.51 (C3, d, *J* = 4.9 Hz), 81.98 (C2, dd, *J* = 22.3 Hz, 18.9 Hz), 126.65 (C7), 127.78 (C5), 128.57 (C6), 143.24 (C4, t, 2.0 Hz), 155.94 (C1, dd, *J* = 288 Hz, 287 Hz).

Reaction between Complex 10b and E-Stilbene Mediated by TMSOTf. TMSOTf (55.3 µmol, 10 µL) was added to a precooled (-78 °C) solution of complex 10b (53.1 µmol, 29 mg) and *E*-stilbene (55.5 µmol, 10 mg) in CH₂Cl₂ (4 mL). The resulting mixture was stirred at this temperature for 1 h, then at -20 °C for 15 h, and finally at ambient temperature for 1 h. All volatile materials were evaporated, the residue was triturated with CD₂Cl₂ (0.5 mL), insoluble material was filtered off and the filtrate analyzed by NMR spectroscopy. The solution contained *trans*-1,1-difluoro-2,3-diphenylcyclopropane 14 (39%) and difluroro-alkene 15 (\leq 3%, determined using 1,2-dichloroethane as an internal standard).

Reaction between Complex **10a** and E-Stilbene Mediated by $B(C_6F_5)_3$. B(C₆F₅)₃ (56.6 µmol, 29 mg) was added to a precooled (-78 °C) solution of complex **10a** (55 µmol, 30 mg) and E-stilbene (55.5 µmol, 10 mg) in CH₂Cl₂ (4 mL). The resulting mixture was stirred at this temperature for 1 h, then at -20 °C for 15 h, and finally at room temperature for 1 h. All volatile materials were evaporated, the residue was triturated with CD₂Cl₂ (0.5 mL), insoluble material was filtered off and the filtrate analyzed by NMR spectroscopy. The solution contained *trans*-1,1-difluoro-2,3-diphenylcyclopropane **14** (59%) and difluroro-alkene **15** (\leq 4%, determined using 1,2-dichloroethane as an internal standard).

Control Experiment: Reaction between 1,2-Dibromo-tetrafluoroethane and $CoCp_{2}^{*}$ to generate $C_{2}F_{4}$.1,2-Dibromo-tetrafluoroethane (33.6 µmol, 8.7 mg) was added to a solution of $CoCp_{2}^{*}$ (73.0 µmol, 24 mg) in CD₂Cl₂. ¹⁹F NMR spectroscopy analysis of the solution (253 K) indicated the formation of tetrafluoroethylene ($\delta_{F} = 131.50$ ppm).



Atom	δ	J [Hz]	COSY	HSQC	НМQС	ROESY
P1	37,81	q 40.3(³ J _{PF})				
C1	36,90	d 29.4(¹ J _{CP})		1	2", 6', 6"	
H1	2,19	ttd 12.3(2", 6"), 3.1(2', 6'), 10.3(² J _{HP})	2', 2", 6', 6"	1	2, 6	8
C2	29,99	d 2.3(² J _{CP})		2', 2"	1	
H2' (eq)	2,08	dm 12.5(2")	1, 2", 3', 3"	2		
H2'' (ax)	1,51	t3d 12.5(2',3"), 12.3(1), 3.5(3'), 6.0(³ J _{HP})	1, 2', 3', 3"	2	1, 3	
C3	26,77	d 12.5(³ J _{CP})		3', 3"	2"	
H3' (eq)	1,86	m	2', 2", 3", 4'	3		
H3'' (ax)	1,30	m (overlapped)	2', 2", 3', 4'	3		
C4	25,77	d 1.7(⁴ J _{CP})		4', 4"		
H4' (eq)	1,69	m	3', 3", 4", 5', 5"	4		
H4'' (ax)	1,22	m (overlapped)	4'	4		
C5	27,05	d 13.0(³ J _{CP})		5', 5"		
H5' (eq)	1,78	m	4', 5", 6', 6"	5		
H5" (ax)	1,25	m (overlapped)	4', 5'	5		
C6	30,70	d 2.8(² J _{CP})		6', 6"	1	
H6' (eq)	1,82	m	1, 5'	6	1	
H6" (ax)	1,37	t3d 12.5(5",6'), 12.3(1), 3.5(5'), 5.8(³ J _{HP})	1, 5'	6	1	
C7	127,83	d 45.5(¹ J _{CP})			11	
C8	132,68	d 2.0(² J _{CP})		8	10	
H8	7,63	dm 7.5(³ J _{HP})	9	8	10, 12	1
С9	127,04	d 6.3(³ J _{CP})		9	11	
Н9	7,49	m	8, 10	9		
C10	130,27	d 2.2(⁴ J _{CP})		10	8	
H10	7,50	m	9, 11	10	8	
C11	133,53	d 8.1(³ J _{CP})		11		
H11	7,20	dm 4.4(⁴ J _{HP})	10	11	7, 9, 13	19
C12	147,27	d 15.9(² J _{CP})			8	
C13	135,62	d 5.5(³ J _{CP})			11, 15, 17	
C14	145,76				15, 17, 18, 19	
C15	121,60			15	15, 17, 20	
H15	7,11	S		15	13, 14, 15, 17, 19, 20	18, 21
C16	148,99				20, 21	
C17	30,86			17	15, 18, 19	
H17	2,28	qq 6.9(18), 6.7(19)	18, 19	17	13, 14, 15, 18, 19	
C18	22,70			18	17, 19	

H18	1,31	d 6.9(17)	17	18	14, 17, 19	15
C19	25,36			19	15, 17, 18	
H19	0,94	d 6.7(17)	17	19	14, 17, 18	11
C20	33,80			20	15, 21	
H20	2,92	sept 6.9(21, 21')	21	20	15, 16, 21	
C21	23,21			21	20, 21	
H21	1,29	d 6.9(20)	20	21	16, 20, 21	15
C22	167,25	dq 171(² J _{CP}), 354(¹ J _{CF})				
F22	-29,13	d 40.3(³ J _{PF})				



 1 H (top) and 19 F (bottom) NMR spectra of **10c** (CD₂Cl₂, 298 K).



 $^{13}C{^{1}H}$ (top) and $^{31}P{^{1}H}$ (bottom) spectra of **10c** (CD₂Cl₂, 298 K).



Atom	δ	J	COSY/H,F- COSY	HSQC	HMQC/31P- HMQC	NOESY/H,F- HOESY
C1	139,15	6.3(P11)			5, 8	
C2	125,54			2	2, 4	
H2	7,48	2.5(4), 1.3(P11)	4, (8), F13	2	2, 4, 6, 7, 9	8, 10, (F13)
С3	148,30				5, 10	
C4	124,06			4	2	
H4	7,20	8.5(5), 2.5(2)	2, 5, (F13)	4	2, 6, 9	10, (F13)
C5	119,08	9.1(P11)		5	5	
H5	7,44	8.5(4), 1.5(P11)	4, (F13)	5	1, 3, 5, 6, P11	8, F13
C6	147,22	5.6(P11)			2, 4, 5	
C7	34,98				2, 8	
C8	30,21			8	8	
H8	1,46	S	(2), F13	8	1, 7, 8	2, 5, F13
С9	34,53				2, 4, 10	
C10	31,06			10	10	
H10	1,31	S	F13	10	3, 9, 10	2, 4, F13
P11	127,34	65(F13), 275(12), 5.6(6), 6.3(1), 1.5(5H), 1.3(2H), 9.1(5)			5	
C12	163,04	275(P11), 350(F13)				
F13	-30,48	65(P11), 350(12)	2, (4), (5), 8, 10			(2), (4), 5, 8, 10





 1 H (top) and 19 F (bottom) NMR spectra of **10d** (CD₂Cl₂).



 $^{13}C\{^1H\}$ (top) and $^{31}P\{^1H\}$ (bottom) spectra of 10d (CD₂Cl₂, 298 K).



Atom	Chemical Shift	Ľ	HSQC	нмqс
P1	38,37	12.2(2H), 2.0(3H), 52.8(1), 14.1(2), 2.5(4), 11.6(3)		
C1	132,10	52.8(P1)		3
C2	134,77	14.1(P1)	2	2, 3
H2	7,69	12.2(P1P), 8.3(3)	2	2, 3, 4
С3	126,47	11.6(P1), 3.7(F5)	3	2, 3
Н3	7,82	8.3(2), 2.0(P1P)	3	1, 2, 3, 5
C4	134,10	2.5(P1), 33.2(F5)		2
C5	123,34	273.0(F5)		3
F5	-63,71	33.2(4), 3.7(3), 273.0(5)		
C6	164,57	351.0(F6)		
F6	-29,74	351.0(6)		



 1 H (top) and 19 F (bottom) NMR spectra of **11** (CD₂Cl₂, 298 K).



 $^{13}C\{^1H\}$ (top) and $^{31}P\{^1H\}$ (bottom) spectra of 11 (CD₂Cl₂, 298 K).



Atom	Chemical Shift	J	HSQC	HMQC
C1	123,40	d 1.4(P5), q 273(F1)		3
F1	-63,32	S		
C2	133,69	d 2.4(P5), q 32.9(F1)		4
С3	126,52	d 11.7(P5), q 3.6(F1)	3	3, 4
H3	7,79	d 8.1(4), d 2.0(P5), q 0.7(F1)	3	1, 3, 4, 5
C4	134,90	d 14.3(P5)	4	3, 4
H4	7,65	d 12.4(P5), q 0.7(F1), d 8.1(3)	4	2, 3, 4
C5	131,77	d 53.9(P5), q 1.5(F1)		3
P5	38,08	q 45.6(F6)		
C6	164,83	d 182.0(P5), q 352.0(F6)		
F6	-29,64	d 45.6(P5)		



 ${}^{1}H{}^{31}P{}$ (top) and ${}^{19}F$ (bottom) NMR spectra of **10e** (CD₂Cl₂, 223 K).



 $^{13}C{}^{1}H{}$ (top) and $^{31}P{}^{1}H{}$ (bottom) spectra of **10e** (CD₂Cl₂, 223 K).

2 60 58 56 54 52 50 48 46 44 42 40 38 36 34 32

30 28 26 24 22 20 18 16



Atom	δ	J [Hz]	H,C-HSQC/F,C-HSQC	HF-COSY/FF-COSY	H,P-HMQC/F,P-HMQC
C1	132.16	d 2.6 (⁴ J _{CP})	H1		
H1	7.60-7.53	m	C1		(P4)
C2	129.45	d 11.5 (³ J _{CP})	H2		
H2	7.52-7.45	m	C2		P4
СЗ	134.27	d 13.8 (² J _{CP})	H3		
H3	7.52-7.45	m	C3	F5a,b	P4
C4	128.11	d 58.0 (¹ J _{CP})			
P4	36.11	t 31.0 (³ J _{PF})			H2, H3, (H1), F5a,b
C5	169.63	td 359.50 (¹ J _{CF}), 169.80 (² J _{CP})	F5a,b		
F5a,b	-30.74	d 31.0 (³ J _{FP})	C5	F6, H3	P4
C6	118.09	q ~320 (¹ J _{CF})	F6		
F6	-76.09	s (br)	C6	F5a,b	



 1 H (top) and 19 F (bottom) NMR spectra of **12a** (CD₂Cl₂, 223 K) at *ca.* 45% conversion.



^{250 240 230 220 210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 -20}



 ${}^{13}C{}^{1}H, {}^{19}F{}$ (top) and ${}^{31}P{}^{1}H{}$ (bottom) spectra of **12a** (CD₂Cl₂, 223 K) at *ca.* 45% conversion.



 $^{19}\text{F}\text{-}^{19}\text{F}$ COSY NMR spectrum of **12a** (CD₂Cl₂, 223 K). at *ca.* 45% conversion.



Atom	Chemical Shift	J	COSY	HSQC	нмqс	³¹ P- HMQC	¹⁹ F- COSY	ROESY
C1	146,98	d 5.2(P1)			3, 5			
P1	120,11	t 44.0(F11)				3, 6		
C2	138,80	d 6.6(P1)			6, 8			
С3	125,73			3	3, 5			
H3	7,42	m	5	3	1, 3, 5, 7, 9	P1		8, 10
C4	148,19				6, 10			
C5	124,25			5	3			
H5	7,15	d 2.5(3), d 8.6(6)	3, 6	5	1, 3, 9			10
C6	118,47	d 8.9(P1)		6				
H6	7,35	d 8.6(5)	5	6	2, 4	P1		8
C7	n.a.d. (35.10- 34.95)				3, 8			
C8	n.a.d. (30.10- 29.95)			8	8			
H8	n.a.d. (1.46-1.30)	s (overlapped)		8	2, 7, 8			3, 6, 10
C9	n.a.d. (34.75- 34.60)				3, 5, 10			
C10	n.a.d. (31.15- 31.00)			10	10			
H10	1,24	s (overlapped)		10	4, 9, 10			3, 5, 8
C11	165,87	d 257.0(P1), t 354.0(F11)						
F11	-32,25	d 44.0(P1)					F12	
C12	n.d.							
F12	-76,05	S					F11	



²⁶ -28 -30 -32 -34 -36 -38 -40 -42 -44 -46 -48 -50 -52 -54 -56 -58 -60 -62 -64 -66 -68 -70 -72 -74 -76 -78 -80 -82 11 H (top) and ¹⁹F (bottom) NMR spectra of **12d** (CD₂Cl₂, 223 K) at *ca*. 53% conversion.



 $^{13}C{^{1}H}$ (top) and $^{31}P{^{1}H}$ (bottom) spectra of **12d** (CD₂Cl₂, 223 K) at *ca*. 53% conversion.



 19 F– 19 F COSY NMR spectrum of **12d** (CD₂Cl₂, 223 K). at *ca.* 53% conversion.



Atom	δ	J
C1	25.78	d ~1.5(⁴ J _{CP})
H1ax	1.19	qt 12.8/3.0
H1eq	1.67	m
C2	26.93	d 12.0(³ J _{CP})
H2ax	1.27	qm ~13
H2eq	1.79	m
С3	30.34	
H3ax	1.39	qd 12.5/2.5, d ~5(³ J _{PH})
H3eq	1.91	m
C4	32.00	d 28.7(¹ J _{CP})
H4ax	2.00	tt 12.3/2.8, d 9.4(² J _{PH})
P4	52.72	t 27.5(³ J _{PF})
C5	174.66	t 362(¹ J _{CF}), d 155(² J _{CP})
F5a,b	-30.48	d 27.5(³ J _{FP})
C6	n.r.d.	
F6	n.r.d.	



 19 F (top) and 31 P{ 1 H} (bottom) NMR spectra of **12b** (CD₂Cl₂, 223 K) at *ca.* 81% conversion.





¹H (top) and ¹³C{¹H} (bottom) NMR spectra of **12b** (CD₂Cl₂, 223 K) at *ca.* 81% conversion.



Atom	Chemical Shift	J [Hz]	COSY	HSQC	HMQC/ ³¹ P-HMQC	ROESY/H,F- HOESY
C1	36,17	d 30.7(¹ J _{CP})		1	1, 2", 6"	
H1 (ax)	2,13	t 12.4(H2", H6"), d 10.5(² J _{HP}), t 3.2(H2', H6')	2', 2", 6', 6"	1	1, 2, 3, 5, 6, P22	8
C2	29,00	d 1.5(² J _{CP})		2', 2"	1	
H2' (eq)	2,03	d 12.4(H2"), m	1, 2", 3', 3"	2		2", 8
H2'' (ax)	1,47	q 12.4(H2', H1, H3''), d 3.4(H3'), d ~3.3(³ J _{HP})	1, 2', 3', 3"	2	1, 3, 6, P22	2'
С3	26,67	d 12.5(³ J _{CP})		3', 3"	1, 2", 5'	
H3' (eq)	1,79	m	2', 2", 3", 4', 4"	3	5	
H3'' (ax)	1,24	m	2', 2", 3', 4', 4"	3		
C4	25,66	d 1.7(⁴ J _{CP})		4', 4"		
H4' (eq)	1,62	m	3', 3", 4", 5'	4		4"
H4'' (ax)	1,17	m	3', 3", 4'	4		4'
C5	26,98	d 13.6(³ J _{CP})		5', 5"	1, 3'	
H5' (eq)	1,71	m	4', 5", 6', 6"	5	3	5"
H5'' (ax)	1,19	m	5', 6', 6"	5		5'
C6	30,66	d 3.1(² J _{CP})		6', 6"	1, 2"	
H6' (eq)	1,78	m	1, 5', 5", 6"	6		8
H6'' (ax)	1,33	q 12.5(H6', H5", H1), d 3.4(H5'), d ~5.9(³J _{HP})	1, 5', 5", 6'	6	1, P22	
C7	126,93	d 49.0(¹ J _{CP})			8, 9, 11	
C8	132,62	d 2.3(² J _{CP})		8	10	
H8	7,64	m	9, 10, 11	8	7, 10, 12, P22	1, 2', 6', F24
C9	127,43	d 6.7(³ J _{CP})		9	10, 11	
Н9	7,50	m	8, 11	9	7, 11, P22	
C10	130,76	d 2.2(⁴ J _{CP})		10	8	
H10	7,51	m	8, 11	10	8, 9, 12	
C11	133,09	d 7.9(³ J _{CP})		11	9	
H11	7,19	m	8, 9, 10	11	7, 9, 12, 13, P22	17, 19
C12	146,68	d 15.2(² J _{CP})			8, 10, 11, 15	
C13	135,63	d 5.9(³ J _{CP})			11, 15, 17	
C14	145,86	S			15, 17, 18, 19	

C15	121,66	S		15	15, 17, 20	
H15	7,09	S	17, 18, 19, 20, 21	15	12, 13, 14, 15, 17, 19, 20, 21	17, 18, 19, 21, F24, F26
C16	148,71	S			20, 21	
C17	30,93	S		17	15, 18, 19	
H17	2,20	sept 6.7(H18, H19)	15, 18, 19	17	13, 14, 15, 18, 19	11, 15, 18, 19
C18	22,59	S		18	17, 19	
H18	1,25	d 6.7(H17)	15, 17, 19	18	14, 17, 19	15, 17, 19, F26
C19	25,55	S		19	15, 17, 18, 19	
H19	0,89	d 6.7(H17)	15, 17, 18	19	14, 17, 18, 19	11, 15, 17, 18, F26
C20	33,66	S		20	15, 21	
H20	2,87	sept 6.9(H21)	15, 21	20	15, 16, 21	F24
C21	23,35	S		21	15, 20, 21	
H21	1,23	d 6.9(H20)	15, 20	21	16, 20, 21	15, F24
P22	36,27	t 29.3(F24)			1, 2", 6", 8, 9, 11	
C23	172,24	t 367(¹ J _{CF}), d 160(² J _{CP})				
F24a,b	-29,30	(br) d 29.3(³ J _{FP})				8, 15, 20, 21
C25	~118.1	br q ~320(¹ J _{CF})				
F26	-77,31	br				15, 18, 19



¹H (top) and ¹⁹F (bottom) NMR spectra of 12c (CD₂Cl₂, 223 K) at *ca.* 62% conversion.



38.8 38.6 38.4 38.2 38.0 37.8 37.6 37.4 37.2 37.0 36.8 36.6 36.4 36.2 36.0 35.8 35.6 35.4 35.2 35.0 34.8 34.6 34.4

 $^{13}C{^{1}H}$ (top) and $^{31}P{^{1}H}$ (bottom) spectra of **12c** (CD₂Cl₂, 223 K) at *ca*. 62% conversion.



Atom	Chemical Shift	ſ	HSQC/ ¹⁹ F-HSQC	HMQC/ ³¹ P-HMQC/F,P-HMQC	H,F-COSY
C1	123,38	d 1.7(P5), q 273.0(F1)	F1	3	
F1	-63,35	S	1		3, 4
C2	133,85	d 2.7(P5), q 33.1(F1)		4	
С3	126,59	d 11.7(P5), q 3.8(F1)	3	4	
H3	7,80	dm 2.0(P5)	3	P5, 1, 4, 5	F1
C4	134,90	d 14.4(P5)	4	3, 4	
H4	7,65	dm 12.4(P5)	4	P5, 2, 3, 4	F1
C5	131,24	d 56.0(P5)		3	
P5	36,69	t 32.2(F6)		3, 4, F6	
C6	167,83	d 172.0(P5), t 357.0(F6)	F6		
F6	-31,66	d 32.2(P5)	6	Р5	
C7	n.d.				
F7	n.d.				







 $^{13}C{^{1}H}$ (top) and $^{31}P{^{1}H}$ (bottom) spectra of **12e** (CD₂Cl₂, 223 K) at *ca.* 17% conversion.



 1 H (top) and 13 C{ 1 H} (bottom) NMR spectra of **13** (CD₂Cl₂, 253 K) at *ca*. 85% conversion.



 $^{31}P{}^{1}H$ (bottom) NMR spectra of **13** (CD₂Cl₂, 253 K) at *ca.* 85% conversion



Atom	Chemical Shift	J [Hz]	COSY/HF- COSY	HSQC	HMQC	ROESY/HF- HOESY
C1	155,94	dd 287(F1a), 288(F1b)			3	
F1a (cis)	-88,99	d(¹ J _{CF}) 287; ddd 43.5(F1b), 2.5(H2), 2.3(H3)	2, 3			2, 3
F1b (trans)	-90,46	d(¹ J _{CF}) 288; ddd 43.5(F1a), 23.1(H2), 2.3(H3)	2, 3, (5)			2, 3
C2	81,98	dd(² J _{CF}) 22.3/18.9		2	2, 3	
H2	4,85	ddd 23.1(F1b), 10.5(H3), 2.5(F1a)	3, F1a, F1b	2	2, 3, 4	5, F1a, F1b
C3	44,51	d(³ J _{CF}) 4.9		3	2, 3, 5	
H3	4,91	dt 10.5(H2), 2.3(F1a/F1b)	2, 5, F1a, F1b	3	1, 2, 3, 4, 5	5, F1a, F1b
C4	143,24	t(⁴ J _{CF}) 2.0			2, 3, 6	
C5	127,78			5	3, 5	
H5	7,22	m	3, 6, (F1b)	5	3, 5, 7	2, 3
C6	128,57			6	6	
H6	7,33	m	5, 7	6	4, 6	
C7	126,65			7	5	
H7	7,25	m	6	7		





 1H (top) and $^{19}F\{^1H\}$ (bottom) NMR spectra of 15 (CD₂Cl₂, 298 K).

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