

1 fractions of 1, 3 and 5 at.%, comparable effects on D retention were observed after
2 irradiation to 0.5 dpa at 1073 K. The addition of Mo and Ta to W resulted in no visible
3 effects in D retention.

4

5 Keywords: tungsten; rhenium; molybdenum; tantalum; deuterium retention; ion
6 irradiation

7

8 **1. Introduction**

9 Due to its attractive physical properties, such as high melting point, high temperature
10 strength, high thermal conductivity and high sputtering threshold energy, tungsten (W)
11 and its alloys have been recognized as a promising candidate for plasma-facing material
12 (PFM) in a future fusion reactor [1, 2]. During operation of a fusion reactor, W will
13 experience irradiation of high-energy neutrons (n), energetic deuterium (D) and tritium
14 (T) particles. Irradiation with high energy neutrons results in the formation of
15 displacement damage. Previous neutron-irradiation experiments have shown that
16 radiation-induced defects in pure W act as strong trapping sites for D [3-6]. A significant
17 increase in the hydrogen isotope retention has been observed also after irradiation with
18 high energy ions performed as a surrogate for the displacement damage the neutron
19 irradiation will cause [7-10]. These observations indicate that irradiation with high energy
20 neutrons should result in a significant increase in the T inventory in W.

21 Tyburska-Püschel and Alimov [11], and Hatano et al. [12] have examined deuterium
22 retention in W-rhenium (Re) binary alloys after heavy ion irradiation at MeV energy range
23 and reported that the retention in irradiated W-Re alloys was orders-of-magnitude lower
24 than that in irradiated W. Hatano et al. [12] also observed suppression of the formation of
25 vacancy-type defects. The observation by Hatano et al. [12] is consistent with the data of
26 Fukuda et al. [13], who reported that void formation after neutron irradiation at elevated
27 temperatures was suppressed by Re addition. However, the effects of Re on hydrogen
28 isotope retention has been examined solely at 5 at.% Re fraction and an irradiation dose

1 of 1 displacement per atom (dpa) or less. The investigation at different Re fractions and
2 higher dose levels is required for better understanding of the effects of Re.

3 It is also important to examine the influence of other alloying elements on hydrogen
4 isotope retention after irradiation. According to first principles calculations by Suzudo et
5 al. [14, 15], Re atom reduces the effective mobility of W self-interstitial atom (SIA) by
6 forming a dumbbell cluster with the SIA and enhances recombination with a vacancy.
7 According to their evaluation, the interaction of W SIA with Mo and Ta atoms is much
8 weaker than with Re atoms. However, the effect of these alloying elements (Mo and Ta)
9 on D retention after irradiation have not been understood.

10 In this study, plates of W-(1, 3 and 5) at.% Re, W-2.5 at.% Mo and W-5 at.% Ta
11 alloys were irradiated with 6.4 MeV Fe ions at elevated temperatures (1073 and 1273 K)
12 together with those of pure W and then exposed to D₂ gas at 673 K to understand

13 (1) dose dependence of D retention in W-5 at.% Re alloy up to 5 dpa,

14 (2) D retention in W-Re alloy at different Re fractions (1, 3 and 5 at.%), and

15 (3) influence of Mo and Ta on D retention after irradiation.

16 The gas exposure technique was employed instead of conventional plasma exposure to
17 introduce D into the samples under well-controlled conditions with minimal change in
18 the irradiated microstructure.

19

20 **2. Experimental procedure**

21 *2.1 Materials*

22 The samples used to investigate the dose dependence of D retention were plates of
23 pure W and W-5 at.% Re alloys (10×10×5 mm) prepared from sheets of those materials
24 manufactured by A. L. M. T. Co., Japan by powder metallurgy technique followed by
25 warm rolling. Other samples employed to study the dependence of D retention on Re
26 fraction and the effects of Mo and Ta were plates of pure W, W-(1, 3 and 5) at.% Re, W-
27 2.5 at.% Mo and W-5 at.% Ta alloys (10×10×5 mm). Those plates were cut from rods
28 fabricated by zone melting with electron bombardment heating by NPO “Luch”, Russia.

1 The fraction of Mo was limited to be 2.5 at.% because W-Mo alloys with higher Mo
2 fractions were very brittle and difficult to handle. Hereafter, the alloy samples are denoted
3 as W- x M where M is the alloying element and x is its fraction in atomic percent. The
4 compositions and manufacturing techniques of the samples are summarized in Table 1.

5 The surfaces of the samples were mechanically polished to a mirror finish. Then the
6 samples were heated in vacuum (10^{-5} Pa) at 1273 K for 1 h to relieve strain potentially
7 induced during fabrication and remove hydrogen present in the samples as an impurity.

8 9 *2.2 Ion irradiation and D measurements*

10 As the detailed experimental procedure for Fe ions irradiation was described in our
11 previous paper [12], only a brief description is given here. All samples were irradiated
12 with 6.4 MeV Fe ions by using the Dual-Beam Facility for Energy Science and
13 Technology (DuET) at Kyoto University under the conditions shown in Table 1. The
14 irradiation was performed at 1073 and 1273 K because the effects of Re addition observed
15 for the W-5Re alloy became more significant as the irradiation temperature increased [12].
16 During irradiation, the background pressure in the vacuum chamber of the facility was
17 about 10^{-5} Pa. The depth profile of the displacement-damage was calculated by using the
18 program SRIM 2008.03, “full cascade option”, with a displacement threshold energy of
19 $E_d = 90$ eV. The depth of the damaged region was estimated to be ~ 2 μm . In what follows
20 the damage level value will be indicated as a number of dpa at the damage peak.

21 All of these Fe-ion-irradiated samples were exposed to D_2 gas at 673 K and a
22 pressure of 100 kPa for 10 h in a quartz glass tube attached to a vacuum device. The
23 temperature of the samples was monitored using a type K thermocouple located near the
24 samples outside the tube. At the termination of the exposure procedure, the D_2 gas was
25 evacuated to a pressure below 1 Pa for several seconds and then the furnace was
26 immediately removed to rapidly cool the samples. It should be noted that the exposure to
27 D_2 gas was performed at elevated temperatures (673 K). Hence, weak traps were not
28 occupied by D under the present conditions, and the desorption at room temperature was

1 not significant.

2 Deuterium depth profiles up to 6.2 μm were determined at Max-Planck-Institut für
3 Plasmaphysik by counting α -particles and protons generated by the $\text{D}({}^3\text{He}, \text{p}){}^4\text{He}$
4 nuclear reaction. The α -spectrum measured with 690 keV ${}^3\text{He}$ beam at a reaction angle of
5 102° was transformed into a D depth profile up to $\sim 0.4 \mu\text{m}$ using the program SIMNRA
6 [16]. The D fraction at larger depths was determined by varying the energy of the primary
7 ${}^3\text{He}$ ion beam between 0.69 and 4.0 MeV. The proton yields measured at different ${}^3\text{He}$ ion
8 energies at a reaction angle of 135° made it possible to derive the D depth profiles [17].
9 The amount of D retained within the information depth of 6.2 μm was finally obtained by
10 integrating the D profile over the measured depth.

11 Thermal desorption spectrometry (TDS) measurements were performed to evaluate
12 the D retention at the Hydrogen Isotope Research Center, University of Toyama. The TDS
13 measurements were performed within a week after the D_2 gas exposure, during which
14 these exposed samples were kept in a vacuum container. The sample was placed on a
15 quartz glass stage installed in a quartz glass tube evacuated to 10^{-8} Pa using a
16 turbomolecular pump. The temperature of the sample was raised to 1203 or 1273 K with
17 a ramp rate of 0.5 K/s using a heater surrounding the quartz tube. A type K thermocouple
18 inserted into the sample stage was used to monitor the sample temperature. The release
19 flux of the molecular HD and D_2 were measured by a quadrupole mass spectrometer
20 (QMS). Standard H_2 and D_2 leaks with an inaccuracy smaller than 10% were employed
21 to calibrate the QMS, and the calibration constant for HD was assumed to be the average
22 of that for H_2 and D_2 . Signals of other deuterium-containing molecular compounds, such
23 as D_2O and HDO, were also monitored. However, the partial pressures of these gas
24 species showed no significant increase over the background level during TDS
25 measurements. For comparison, the TDS measurements were also performed for non-
26 irradiated samples after the exposure to D_2 gas in the above-mentioned manner.

27

28

1 **3. Results**

2 *3.1 D retention in W-5Re alloy at different damage levels*

3 Fig. 1 shows depth profiles of D in the powder metallurgy W and W-5Re alloy
4 samples irradiated at 1273 K with 6.4 MeV Fe ions to damage levels of 0.1 and 5 dpa and
5 then exposed to D₂ gas at 673 K and 100 kPa. For comparison, the depth profiles of D in
6 the samples irradiated to 0.5 dpa [12] are also shown in Fig. 1. As these profiles show, D
7 is mainly distributed in the damaged zone, indicating that D was trapped at radiation-
8 induced defects. Both in pure W and in the W-5Re alloy, the distributions of D in the
9 damaged zone at different dpa values were similar to each other. In the near-surface layers
10 from surface to a depth of 0.3 μm, the fraction of D decreased from 10⁻¹ at.% to 10⁻² at.%
11 in pure W, while in the W-5Re alloy it decreased from 10⁻¹ at.% to 10⁻³ at.%. At depths
12 from 0.3 to ~1.4 μm, the D fractions were around 10⁻² at.% in pure W and 10⁻³ at.% in the
13 W-5Re alloy, respectively. At greater depths, the D fraction gradually decreased with
14 increasing depth in both pure W and W-5Re alloy.

15 Fig. 2 shows the correlation between the damage level and D retention evaluated
16 from Fig. 1 by integrating the depth profiles up to 6.2 μm. Obviously, the D retention in
17 the W-5Re alloy was significantly less than that in pure W. This figure also shows that
18 the effects of Re addition began to appear at a relatively low irradiation dose (0.1 dpa).
19 The D retention in the W-5Re alloy after irradiation to damage levels of 0.1 and 0.5 dpa
20 was comparable to that after irradiation to 5 dpa, indicating that the density of defects
21 acting as trapping sites ceased to increase at relatively low dose (0.1 dpa).

22

23 *3.2 D retention in W-xRe alloys*

24 Fig. 3 shows the TDS spectra of D released from pure W, W-1Re, W-3Re and W-
25 5Re alloy samples prepared by zone melting and irradiated with Fe ions to a damage level
26 of 0.5 dpa at 1073 K. The desorption peak for pure W appeared at 700–950 K, as was
27 observed in a previous study [12]. The peak consisted of two sub-peaks; the main one
28 was located at 822 K and the additional one at 915 K. As for the W-xRe alloys, the D

1 desorption was observed in the temperature range from 700 to 1200 K. The desorption
2 peaks for the W- x Re alloys were small and broad compared to the peaks for pure W; the
3 TDS spectra for W- x Re alloys had a new sub-peak in the high temperature range: at
4 around 1090 K. These observations indicated that the addition of Re affect the trapping
5 energy of radiation-induced defects.

6 Fig. 4 shows the correlation between Re fraction and D retention determined by TDS
7 for Fe-ion-irradiated samples (Fig. 3) and non-irradiated samples. D retention in the non-
8 irradiated W- x Re alloy was slightly higher than that in pure W, and it decreased with an
9 increase in the Re fraction. The underlying mechanisms for this difference between pure
10 W and W- x Re alloy and the dependence on the Re fraction are currently not clear. In the
11 case of pure W, the D retention increased by an order of magnitude after irradiation with
12 Fe ions, though the thickness of the damaged zone ($\sim 2 \mu\text{m}$) was far less than the sample
13 thickness (500 μm). In contrast, the W- x Re alloy samples showed only a modest increase
14 in the D retention after Fe ion irradiation. The extent of increase in the D retention due to
15 Fe ion irradiation was comparable with each other at $x = 1, 3$ and 5 at.%. These results
16 revealed that the effects of Re addition on radiation-induced defects were at a comparable
17 level in the fraction range of 1–5 at.%.

18

19 *3.3 D retention in W-2.5Mo and W-5Ta alloys*

20 Fig. 5 shows the TDS spectra of D released from the W-2.5Mo alloy and the W-5Ta
21 alloy irradiated with 6.4 MeV Fe ions to a damage level of 0.5 dpa at 1073 K. For
22 comparison, the TDS spectrum for the zone melting pure W sample previously shown in
23 Fig. 3 is also plotted in this figure. The desorption peaks were observed at 700–950 K for
24 pure W, W-2.5Mo alloy and W-5Ta alloy. In addition, the peak areas are also comparable
25 to each other. These observations indicated that Mo and Ta do not affect the density and
26 trapping energy of radiation-induced defects.

27 Fig. 6 shows the total D retention evaluated by TDS for irradiated and non-irradiated
28 samples of pure W (zone melting), W-5Ta alloy and W-2.5Mo alloy as a function of the

1 irradiation dose (0, 0.5 and 2.5 dpa). The D retention in non-irradiated W-2.5Mo alloy
2 was comparable to that in non-irradiated pure W, while the D retention in non-irradiated
3 W-5Ta was slightly smaller (about half of pure W). However, after Fe ion irradiation, the
4 D retention in these materials was comparable to each other. In addition, the D retention
5 after 2.5 dpa irradiation was comparable to the retention after 0.5 dpa irradiation for all
6 materials. These results further illustrate that the addition of Mo and Ta has no visible
7 effect on the D retention in W after ion irradiation.

8

9 **4. Discussion**

10 Figs. 1 and 2 showed that the D retention in the W-5Re alloy was less than that in
11 pure W after Fe ion irradiation at 1273 K to various irradiation doses (0.1–5 dpa). The
12 results of positron lifetime measurements performed in the previous study [12] showed
13 that a significant reduction in the D retention by the Re addition after high temperature
14 irradiation was due to a lower concentration of vacancy-type defects (monovacancies and
15 vacancy clusters). Observation in this study indicate that the addition of 5 at.% Re to W
16 can reduce the concentration of vacancy-type defects after high temperature irradiation
17 even at a relatively low irradiation dose (at a damage level of 0.1 dpa), and the
18 concentration of vacancy-type defects can be controlled to a comparable value even when
19 the radiation dose increased to 5 dpa.

20 Cascade collision induced by Fe ion irradiation should result in the formation of
21 various types of defects such as monovacancy, divacancy, vacancy clusters, dislocations,
22 etc. These defects can accommodate a different number of D atoms with different binding
23 energies. The TDS spectra of D released from W-Re alloys with addition of 1, 3 and 5
24 at.% Re (Fig. 3) showed the shoulders in the higher temperature range. The shoulders
25 indicate that a part of trapping sites in the W-Re alloys has a stronger bond with the D
26 atom(s) than the defects in pure W. Therefore, the reduced D retention shown in Figs. 3
27 and 4 can be ascribed to a reduced concentration of defects rather than to a weaker bond
28 between trapping site and the D atom(s). This observation is consistent with the results of

1 positron lifetime measurements for the W-5Re alloy in the previous study [12]. The
2 shoulder in the high temperature range was not evident in the TDS spectrum for the W-
3 5Re alloy shown in Ref. [12]. This was due to the fact that the D retention was very low
4 and the signal intensity of D₂ was close to the signal background in the temperature range
5 above ~1000 K. The mechanism underlying the higher binding energy in W-Re alloys is
6 currently not clear. The binding energy is determined by the difference in enthalpy
7 between the D atom at the trapping site and the D atom at the interstitial site in the
8 undisturbed W lattice. First-principles calculations by Ren et al. [18] showed that the heat
9 of solution, the difference in enthalpy between the H atom in a H₂ molecule and that at
10 an interstitial site in the undisturbed W lattice increases with the Re addition. Such a
11 change in the heat of solution should result in an increase in the trapping energy. The
12 observation in this study is consistent with the results of their calculations. Anyway, the
13 fraction of Re between 1 at.% and 5 at.% can significantly reduce the hydrogen isotope
14 retention.

15 The D retention in Fe-ion-irradiated samples prepared by zone melting in this work
16 was higher than that in the powder metallurgy samples irradiated at the same manner in
17 Ref. [12]. These two types of samples had significantly different microstructures. The
18 zone melting samples had large grains up to several millimeters in size. On the other hand,
19 the powder metallurgy sample had a rolled structure, and the minor axis length of the
20 flattened grains was 1 μm or less. Thus, one of the possible explanations for the difference
21 between the zone melting and powder metallurgy samples was the difference in the
22 density of grain boundaries acting as sinks for point defects. However, a detailed
23 microstructural analysis and more systematic studies of the correlation of irradiation
24 effects with microstructure are necessary for a deeper discussion.

25 As mentioned in Section “1. Introduction”, Suzudo et al. [14, 15] reported that the
26 Re atom forms a dumbbell cluster with a SIA of W. The formation of dumbbell cluster
27 reduces the SIA’s effective mobility and enhances recombination with a vacancy.
28 According to Suzudo et al. [15], Mo also forms a mixed dumbbell but the binding energy

1 was significantly lower compared to the Re atom. The Ta atom does not form a stable
2 mixed dumbbell with the W SIA. As shown in Fig. 6, the addition of Mo and Ta resulted
3 in no noticeable effects on the D retention after irradiation. There is no contradiction
4 between the present observations and the model proposed by Suzudo et al. [15].
5 According to them, Ti, V and Cr atoms also form a stable dumbbell cluster with the W
6 SIA. However, the addition of Ti and V can increase the hydrogen isotope retention since
7 these elements form hydrides. Hence, the investigation of the W-Cr system could be the
8 next step.

9

10 **5. Conclusions**

11 The D retention in pure W and three types of binary W-alloys were investigated after
12 irradiation of 6.4 MeV Fe ions at high temperatures. Below are the main findings.

13 (1) The concentration of radiation-induced traps in the W-5 at.% Re alloy was lower than
14 that in pure W after irradiation to 0.1–5 dpa. The effects of Re on the D retention were
15 comparable at a Re fraction of 1–5 at.% after irradiation up to 0.5 dpa at 1073 K.

16 (2) The D retention in the irradiated W-5 at.% Ta and W-2.5 at.% Mo alloys was
17 comparable to that in the irradiated pure W, and the TDS spectra were also similar to
18 each other. In other words, the addition of Ta and Mo did not significantly affect the
19 D trapping after the irradiation.

20 (3) The addition of Re is a promising method to reduce the hydrogen isotope retention in
21 W under neutron irradiation.

22

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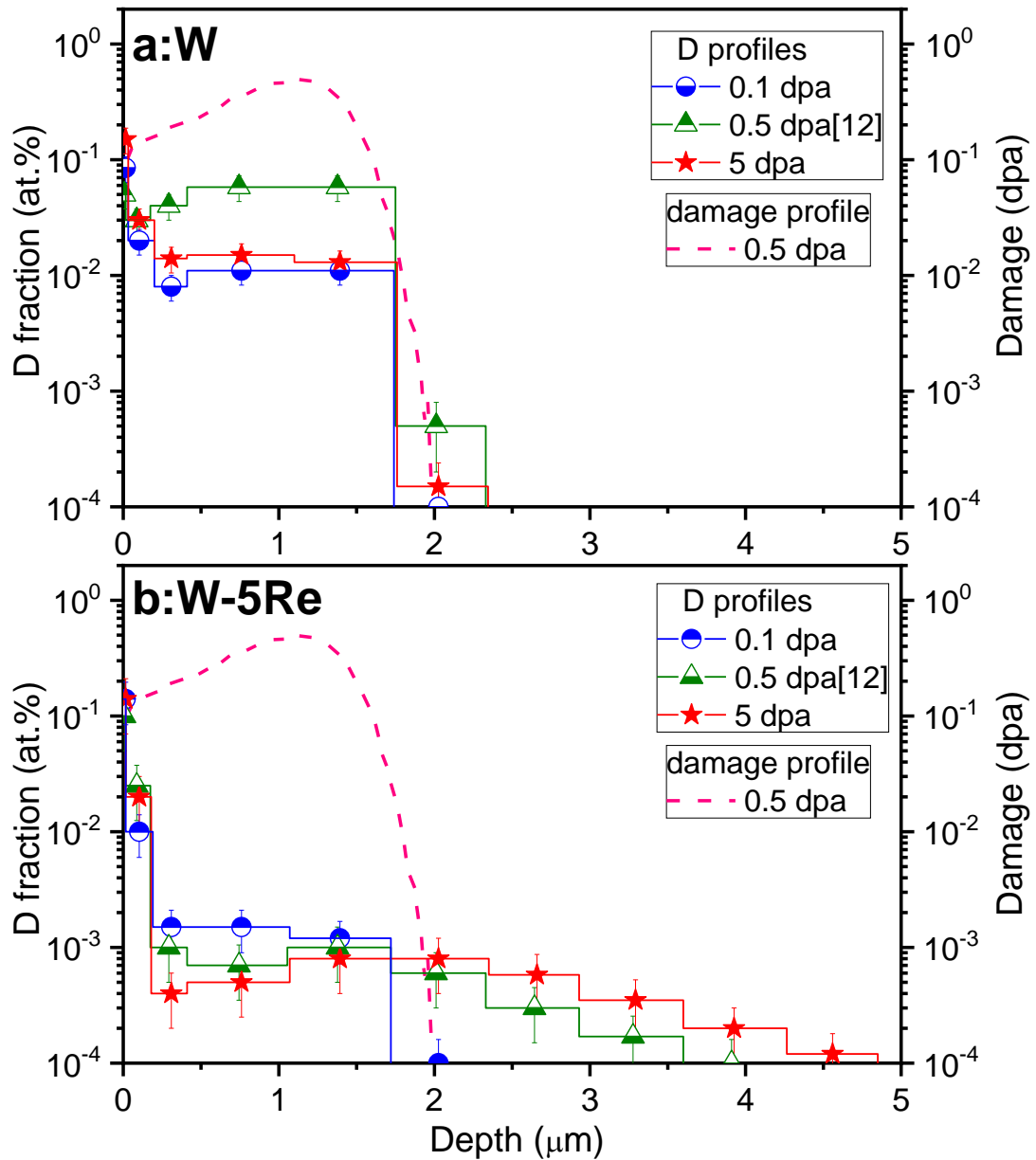
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Table. 1 Summary of samples and conditions of Fe ion irradiation. The damage level (dpa) at the damage peak is indicated.

Sample	Manufacturing technique	Irradiation temperature (K)	Irradiation fluence (Fe ions/m ²)	dpa
W, W-5Re	Powder metallurgy	1273	6.4×10 ¹⁷ 3.2×10 ¹⁹	0.1 5
W- xRe (x = 1, 3 and 5 at.%)	Zone melting	1073	3.2×10 ¹⁸	0.5
W, W-5Ta, W-2.5Mo	Zone melting	1073	3.2×10 ¹⁸ 1.6×10 ¹⁹	0.5 2.5

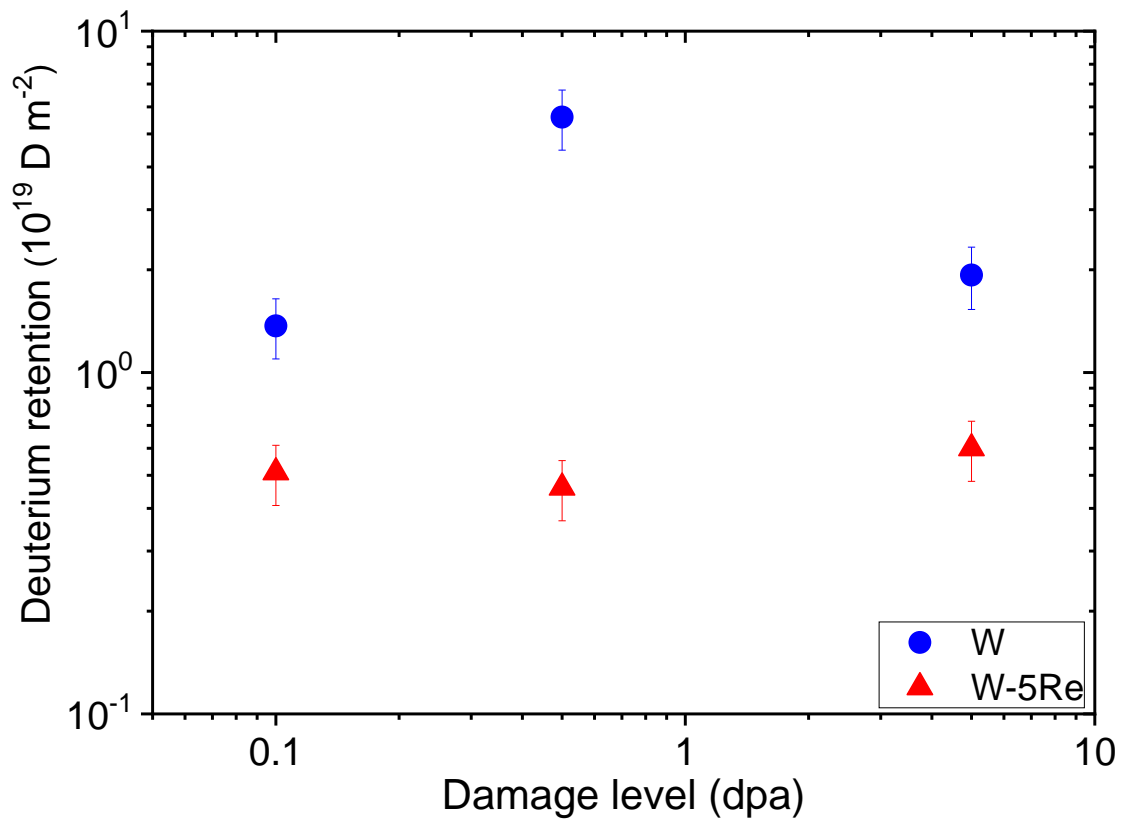
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2 **Fig. 1** Depth profiles of D retained in pure W (a) and W-5Re alloy (b) samples prepared
3 by powder metallurgy. The samples were irradiated at 1273 K with 6.4 MeV Fe ions to
4 various peak damage levels and then exposed to D₂ gas at a temperature of 673 K and
5 pressure of 100 kPa for 10 h. In both panels, calculated damage depth profiles (0.5 dpa)
6 are also shown.

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4 **Fig. 2** D retention in the first 6.2 μm in pure W and W-5Re alloy samples prepared by
5 powder metallurgy, as measured by NRA. The samples were irradiated at 1273 K with
6 6.4 MeV Fe ions to various peak damage levels and then exposed to D_2 gas at 673 K and
7 100 kPa for 10 h.

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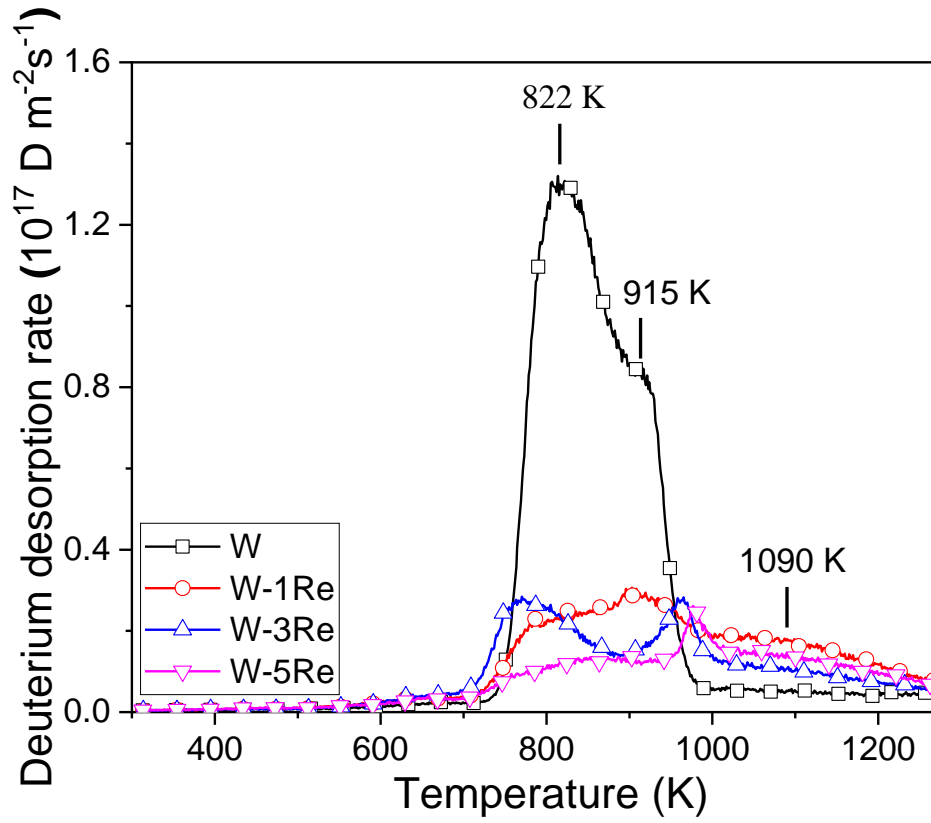
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4 **Fig. 3** TDS spectra of deuterium released from pure W and W- x Re ($x = 1, 3$ and 5 at.%)
5 alloy samples prepared by zone melting. The samples were irradiated at 1073 K with Fe
6 ions to a peak damage level of 0.5 dpa and then exposed to D₂ gas at 673 K and 100 kPa
7 for 10 h. Heating rate was 0.5 K/s.

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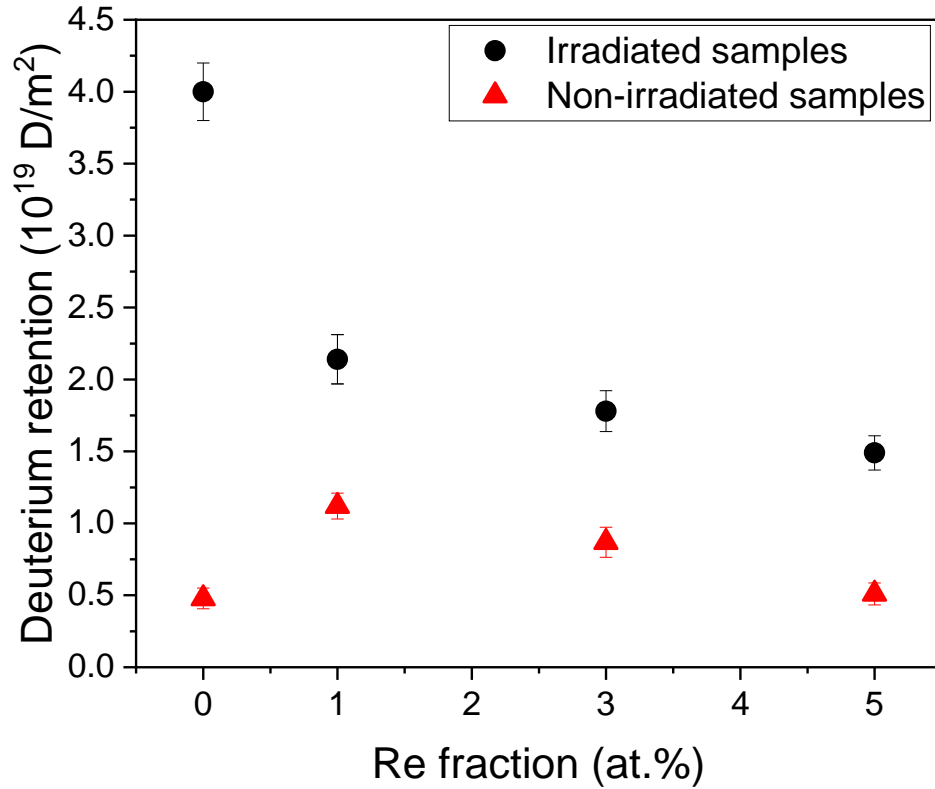
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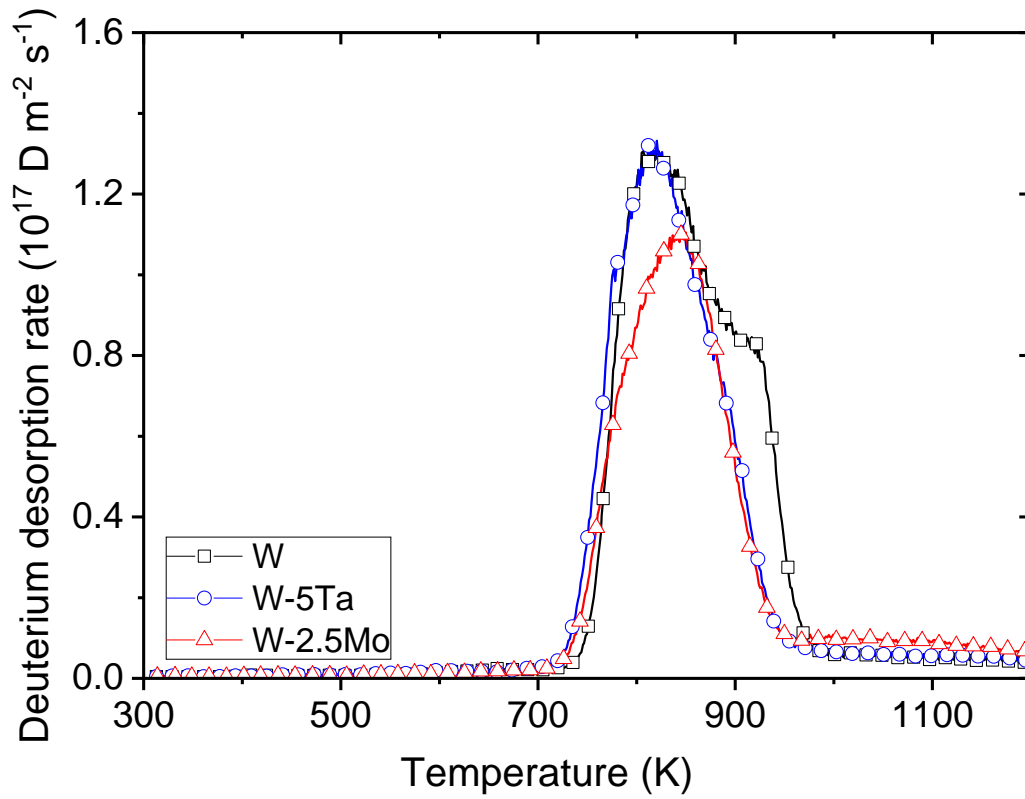
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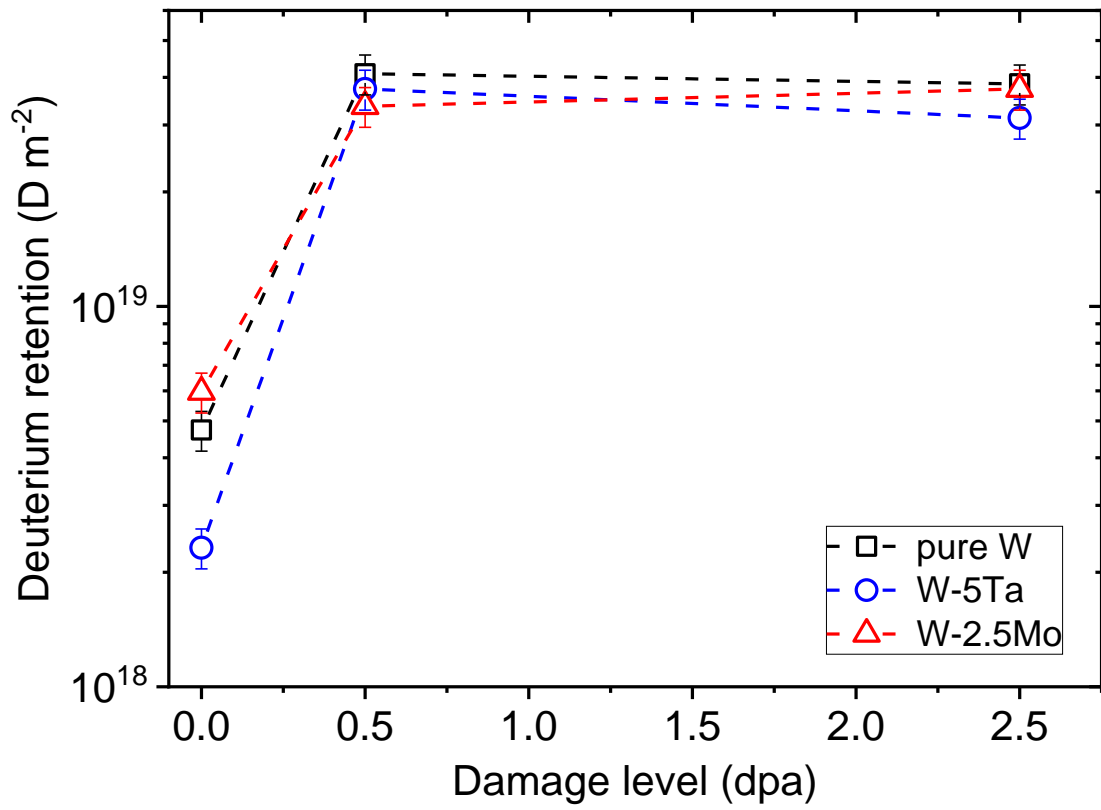


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Fig. 4 The correlation between Re fraction and D retention (evaluated from the TDS spectra) in Fe-ion-irradiated pure W and W-*x*Re (*x* = 1, 3 and 5 at.%) alloy samples, together with the corresponding non-irradiated samples. The samples were prepared by zone melting, irradiated at 1073 K with Fe ions to a peak damage level of 0.5 dpa and then exposed to D₂ gas at 673 K and 100 kPa.



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 2 **Fig. 5** TDS spectra of deuterium released from pure W, W-2.5Mo alloy and W-5Ta alloy
 3 samples prepared by zone melting. The samples were irradiated at 1073 K with 6.4 MeV
 4 Fe ions to a peak damage level of 0.5 dpa and then exposed to D₂ gas at 673 K and 100
 5 kPa. Heating rate was 0.5 K/s.



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Fig. 6 The correlation between damage levels and D retention (evaluated from TDS spectra) in pure W, W-2.5Mo alloy and W-5Ta alloy samples prepared by zone melting. The samples were irradiated with 6.4 MeV Fe ions at 1073 K to peak damage levels of 0.5 and 2.5 dpa and then exposed to D₂ gas at 673 K and 100 kPa.