

Property-Driven Development of Metallic Structural Materials by Combinatorial Techniques on the Example of Fe–C–Cr Steels

Christian Baron* and Hauke Springer

The use of high-throughput techniques allows for rapid property-driven materials development. Attractive material profiles are systematically screened and processed on the example of the ternary Fe–C–Cr system into yet unexplored regions. The effect of various combinations of Cr additions (2, 4, 6, 8, 10 wt%) on the mechanical properties and microstructures of Fe–0.2C (wt%) and Fe–0.8C (wt%)-based steels are studied depending on their hardening (850–1150 °C; 20 min) and tempering conditions (100–600 °C; 1 h), producing 120 different material states. Wide ranges of potentially interesting material profiles are obtained with strengths up to 1.8 GPa ultimate tensile strength (UTS) or hardness of 700 HV5. Individual trends for materials' mechanical performance are identified, but straightforward interpretation of trends holds complex multi-dimensional analytical challenges of this high-throughput bulk metallurgical screening. Methods addressing these problems are outlined and the possibilities of assistive computational data processing and analysis are discussed.

Furthermore, Cr substantially lowers the critical cooling velocity of martensitic steels (increasing hardenability), lowers the martensite start temperature, and effectively increases the corrosion resistance.^[2,3] As shown in Figure 1, the coupled effects of various combinations of C/Cr additions have been exploited in a plethora of widely applied steel grades, from construction to bearing, hot-working, tool, and stainless steels.^[2–4] Their respective compositions (which of course comprise additional elements such as Mn, Si, or Mo) and processing regimes are constantly being refined to increase their respective property profile. The development of steels has to strike a complex balance though, to avoid simply improving one property at the loss of another, such as in the classic trade-off between strength and ductility.^[5] Consequently, it

1. Introduction


Fe–C–Cr-based steels represent one of the largest classes of metallic structural materials, due to their huge range of property profiles achievable by different thermo-mechanical treatments and alloy compositions.^[1] The addition of C allows to increase the strength of Fe via the formation of Fe₃C carbide (cementite), which can be obtained in a variety of morphologies and dimensions, from micrometric eutectoid lamellae in pearlite down to nanoscaled precipitates within bainite. C can also be brought into solid solution at elevated temperature within austenite, from where rapid cooling leads to the transformation of extremely hard martensite. Adding Cr to Fe–C steels leads to the formation of alloy carbides with different stoichiometry and crystal structure, such as hexagonal M₇C₃ and cubic face-centered M₂₃C₆.

requires deep understanding of the complex relationships between alloy/processing, microstructure, and properties. Therefore, the currently produced and applied steel grades are often the result of decades, if not centuries, of iterative development.^[6]

The systematic analysis of Figure 1 also reveals large areas of C/Cr combinations which are not yet utilized for commercially available steels, namely, medium Cr concentrations between 2 and 10 wt% for lower (about 0.2 wt%) and higher (about 0.8 wt%) C contents. At first glance, there appears to be no particular reason why no attractive property profiles would be achievable in this composition range, especially in view of the wide range of applicable heat-treatment parameters. One example for such a successful departure from the beaten path of steel development is the work of Yuan et al.^[7] Here, an unconventional heat treatment for a tool steel (0.45 wt% C and 13.5 wt% Cr) leads to properties extremely favorable, for example, for automotive applications: By choosing an unusually high austenitization temperature, formation of nanoscaled reverted austenite could occur during subsequent tempering, yielding a strength in excess of 2 GPa at a tensile ductility of more than 10%.^[7] Despite these appealing results—especially, as they were possible even in such a long-researched alloy system—the aforementioned iterative development strategy, often relying on the linear expansion of known relationships, typically renders such findings as surprises and accidental.

Switching from the knowledge-based expansion to a systematic screening of the yet-unexplored regions of the Fe–C–Cr alloy system might also reveal attractive property profiles; not necessarily

Dr. C. Baron, Dr. H. Springer
Department Microstructure Physics and Alloy Design
Max-Planck-Institut für Eisenforschung GmbH
Düsseldorf 40237, Germany
E-mail: c.baron@mpie.de

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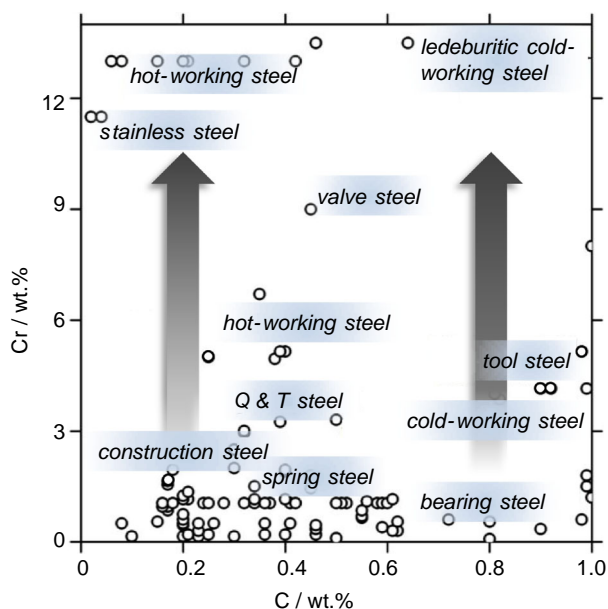


Figure 1. Simplified overview and summary plot of a plethora of various steels categorized by their applications based on their C and Cr contents.^[2–4] The gray arrows indicate the little investigated compositional ranges covered in this study.

for maximum strength at sufficient ductility required for automotive body structures as in the aforementioned example but also for other applications such as energy conversion or construction engineering. However, the conventional experimental approaches for such an undertaking are typically too slow or costly, and purely theory-based material development has not fully matured yet. Such a systematic screening is, therefore, a prime example for recently introduced bulk combinatorial techniques. The rapid alloy prototyping (RAP) methodology, for example, has already been successfully introduced and utilized for various alloy systems.^[8–11] Here, we demonstrate our RAP results for the areas of the Fe–C–Cr system that is yet to be explored. Rather than expecting immediate success in finding revolutionary property profiles, we aim at disseminating yet unknown material data. In addition to providing the base for further analysis and steel development, we utilize our findings to exemplify the associated analytical challenges of high-throughput bulk metallurgical screening procedures based on substantial practical experience, and discuss their effects on material design strategies.

2. Experimental Section

2.1. Synthesis and Processing

Alloy compositions and heat treatments were chosen using an integrated software-assisted approach based on thermodynamic calculations (Thermo-Calc 2015a; database TCFE7). Compositions are given throughout the article in wt%. The equilibrium phase diagrams of the two alloy concepts investigated: 1) Fe–0.2C–XCr and 2) Fe–0.8C–XCr (X = 2, 4, 6, 8, 10) are both plotted in **Figure 2** in a simplified form (temperature on y-axis;

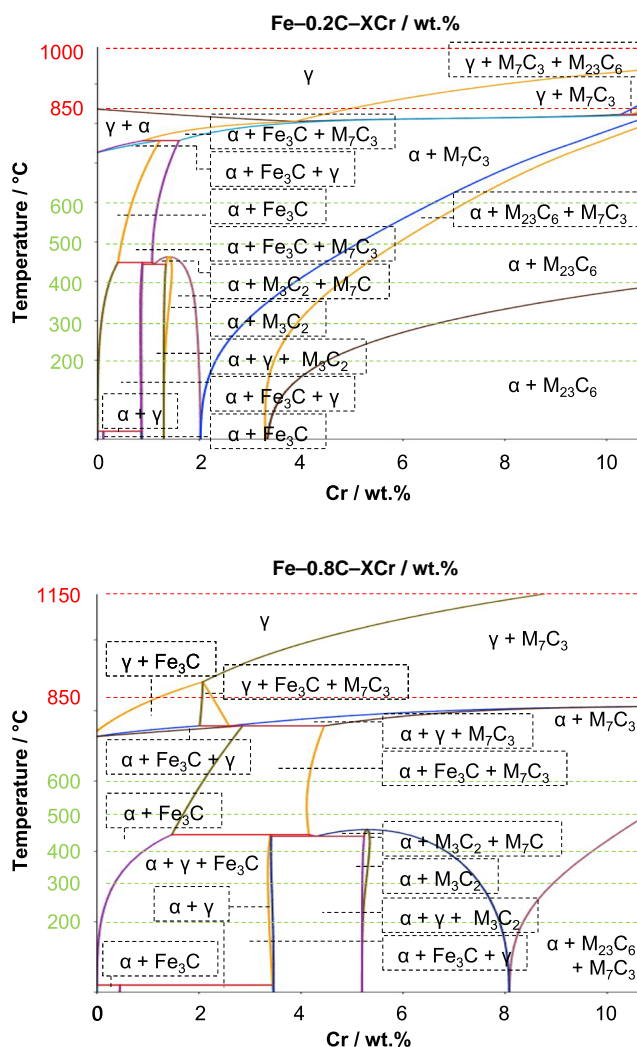


Figure 2. Equilibrium phase diagrams of the two-alloy concepts investigated based on thermodynamic calculations (Thermo-Calc 2015a; database TCFE7). Hardening temperatures are exemplarily shown by the red dashed lines; tempering temperatures are shown by the green dashed lines.

Cr content on x-axis). Alloys were produced and processed by multiple runs of the RAP method.^[8–11] After melting in a vacuum induction furnace and casting into rectangular Cu-molds, samples were subsequently hot-rolled in multiple passes at 1000 °C (alloys 1) and 1150 °C (alloys 2) from 10 to 2 mm thickness. After the final rolling pass sheets were reheated and held for 5 min at rolling temperature and oil-quenched to room temperature (RT), the sheets were cut in 50 mm long pieces each for heat treatments. Temperatures applied in heat treatments are exemplarily indicated by the dashed lines in **Figure 2**; red for hardening temperatures and green for tempering temperatures. Hardening treatments were conducted by annealing the alloys 1) at 850 and 1000 °C and alloys 2) at 850 and 1150 °C, respectively, for 10 min in air, followed by oil-quenching to RT. Tempering was carried out by annealing for 1 h at 200, 300, 400, 500, and 600 °C (in air, oil-quenching to RT). All alloys are referred to by their aforementioned target compositions in wt%.

Table 1. Target and actual chemical compositions (wt%) determined by wet chemical analyses of all alloys of this study (Fe as balance).

Target values [wt%]	Actual values [wt%]
Fe–0.2C–2Cr	Fe–0.20C–(2.1–2.2)Cr
Fe–0.2C–4Cr	Fe–(0.18–0.22)C–(4.1–4.2)Cr
Fe–0.2C–6Cr	Fe–(0.18–0.24)C–(6.1–6.2)Cr
Fe–0.2C–8Cr	Fe–(0.17)C–(8.0–8.1)Cr
Fe–0.2C–10Cr	Fe–(0.15–0.19)C–(10.1–10.2)Cr
Fe–0.8C–2Cr	Fe–0.80C–(2.1–2.2)Cr
Fe–0.8C–4Cr	Fe–(0.80–0.83)C–(4.0–4.1)Cr
Fe–0.8C–6Cr	Fe–(0.8–0.85)C–6.1Cr
Fe–0.8C–8Cr	Fe–(0.78–0.81)C–(8.0–8.1)Cr
Fe–0.8C–10Cr	Fe–(0.83–0.84)C–(10.1)Cr

The actual compositions determined by wet chemical analysis are listed in **Table 1**.

2.2. Characterization and Testing

The mechanical properties of the different alloys were probed by tensile testing. Therefore, sheets were sand-blasted with fine grit at low pressure to remove oxide scales and tensile specimens (dog-bone shaped flat samples; gauge length 20 mm) cut by spark erosion. Testing was conducted with an initial strain rate of 10^{-3} s^{-1} on a conventional tensile testing machine (Zwick/Roell RetroLine Z100) equipped with hydraulic gripping and laser extensometer. Macro-hardness measurements (Vickers Hardness, HV5) were conducted using a conventional Vickers hardness tester (Zwick/ZHV 10) on the outer surface (rolling plane) of the heads of the tested tensile samples after grinding to 400 grit. Tensile values represent average data of three measurements and hardness values of five measurements for every condition. Microstructural characterization was performed on cross-sections of the heads of the tested tensile samples taken perpendicular to the rolling direction. Surfaces were prepared by grinding, polishing, and etching using standard metallographic techniques. Microstructure characterization was performed by scanning electron microscopy (SEM; Zeiss Merlin) in backscatter-electron (BSE) mode. Phase indication was performed on polished samples via electron backscatter diffraction (EBSD) analysis (OIM software v.7; $0.10 \mu\text{m}$ step size) on areas of $50 \times 150 \mu\text{m}^2$ or larger, and X-ray diffraction (XRD; Seifert Type ID-3003, Meteor 0D – energy dispersive point detector, beam diameter 1.5 mm). The phases were quantified and fractions determined by Rietveld analyses of the XRD spectra (database Bruker Topas V5.0 software).

3. Results

3.1. Mechanical Characterization

Totally 120 different material conditions were studied. The mechanical properties (mean average values of total 360 tensile tests and 600 hardness measurements) for the 1) Fe–0.2C–XCr and 2) Fe–0.8C–XCr ($X = 2, 4, 6, 8, 10$) alloys are plotted

exemplarily in **Figure 3** and **4**, respectively. Here, the following material parameters are plotted: a) the yield strength (YS), b) ultimate tensile strength (UTS), c) total elongation (TE), and d) Vickers hardness (HV5). In both cases, the filled squares represent the values determined from samples treated at the lower upstream hardening temperature, whereas unfilled squares represent the higher upstream hardening temperatures. The individual material parameters determined for all material states analyzed are listed in **(Table 2)**.

3.1.1. Fe–0.2C–XCr Alloys

All alloys hardened and tempered Fe–0.2C based alloys (**Figure 3**) exhibit an increase in strength up to several hundred MPa from 2 Cr to 4 Cr. The strength levels often seem to increase or saturate with increasing Cr concentration. Greatest YS increase was found for the material hardened at 850°C (filled symbols; YS 2Cr: 1100 MPa; YS 4Cr: 1350 MPa), followed by a constant YS decrease with increasing Cr content, yielding about 1175 MPa (10 Cr material) in the same state. Identical trend can be observed for the UTS. It is noticeable that decrease in strength can only be observed for alloys hardened at 850°C (unfilled green symbols), whereas the strength of the alloys hardened at 1000°C (filled green symbols) remains almost constant after the first described rise. Interestingly, a conspicuous YS drop of about 200 MPa can be noted between 6 Cr and 8 Cr hardened at 1000°C and annealed at 400°C (unfilled blue rectangles). The lowered YS also remains constant for the 10 Cr alloy. The strength levels (YS/UTS) tend to decrease with increasing tempering temperatures—even though not linear—independent from the hardening treatment, down to 600–700 MPa for high tempering temperatures (600°C). While ductility increases typically with increasing Cr content, materials maintain sufficient ductility even for low Cr contents (around 6–8% TE for the 2 Cr alloyed material) albeit low tempering temperatures after hardening. No premature embrittlement was obtained. Materials tempered at 600°C reveal highest TE around 15%. The hardness values tend to increase with increasing Cr concentration. The values start around (300–350 HV5), followed by a rise up to additional 100 HV5. Materials tempered at 600°C reveal significantly lower hardness values ranging from 200 HV5 up to 250 HV5.

3.1.2. Fe–0.8C–XCr Alloys

Hardened Fe–0.8C based materials (**Figure 4**) reveal extremely brittle behavior during tensile testing (premature failure at low strains despite high inherent strength). In some cases, it was not possible to determine any values, e.g., for some alloys hardened at 850°C . Interestingly, low-temperature tempering (200°C) did not affect the ductility of the alloys hardened at 1150°C , whereas the alloys tempered at 850°C seem to become more ductile with increasing Cr content (yielding in up to 4% TE for the 10 Cr alloyed material). Higher tempering temperatures typically translate into not only higher TE values, yielding in up to 12% TE for the 10 Cr alloy hardened at 850°C and tempered at 600°C (filled cyan symbols), but also rather lower hardness values around 300 HV5 independent from the Cr alloying content.

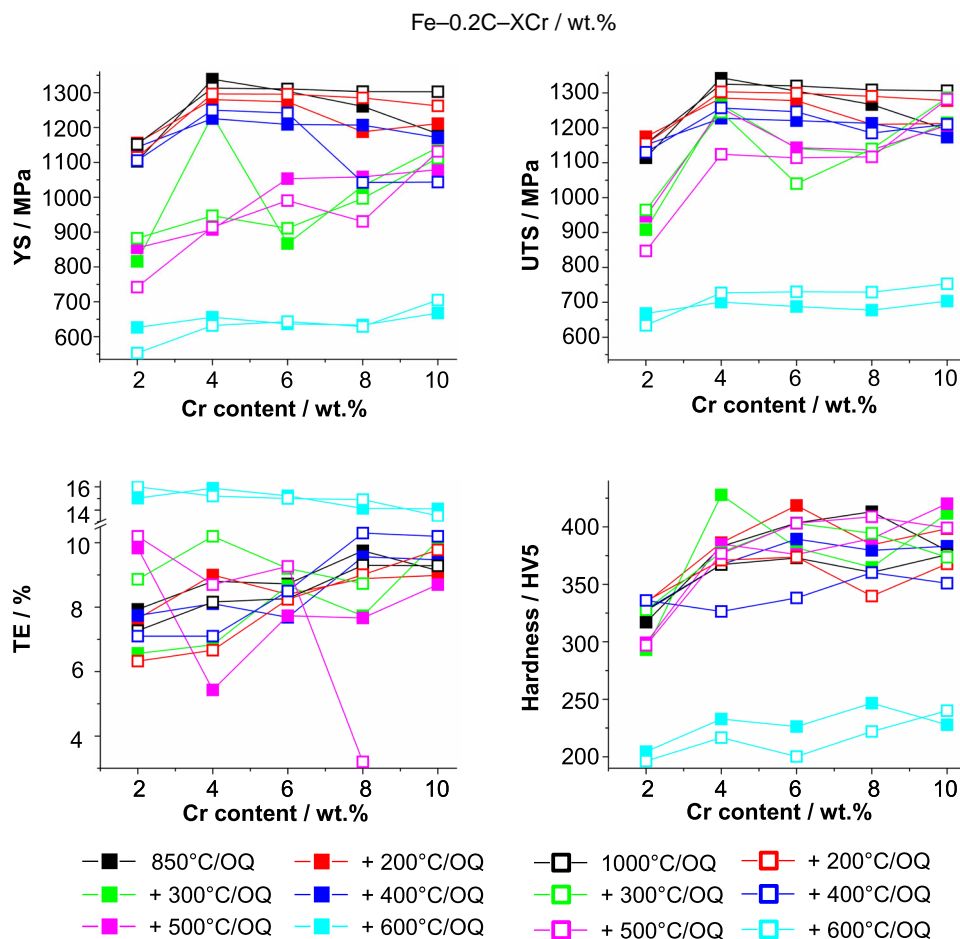


Figure 3. Color-coded plots of selected material parameters (YS, UTS, TE, Hardness) for the different Fe-based alloys inhering 0.2 C and diverging Cr content in different material states (filled symbols: hardened at 850 °C and tempered; unfilled symbols: hardened at 1150 °C and tempered).

Highest hardness values conversely (up to 670 HV5) were obtained for the 2 Cr alloy hardened at 850 °C and tempered at 500 °C, followed by a constant decrease with increasing Cr content. This value was found to be higher than those obtained for the hardened states. Values measured for all other states independent from the Cr alloying content are oscillating around 450–500 HV5.

3.2. In-Depth Microstructural Characterization

Microstructure characterization results for the Fe-0.2C-4Cr (left) and Fe-0.2C-8Cr (right) alloys quenched from 1000 °C and tempered at 400 °C (both oil-quenched) are compiled in **Figure 5** cross-sections; SEM SE mode). The SE images (top) reveals a lath-shaped microstructure for both the alloys, confirmed by the EBSD image quality data map (middle) in gray scale superimposed with the phases Fe α (red). XRD results (bottom, black line) are plotted together with the fitting results of the indicated phases. For both alloys, the fitting results indicated α' (red), whereas only the 8 Cr alloy seems to inher the fractions of $M_{23}C_6$ (blue) carbides. The austenite phase seems to be absent.

4. Discussion

4.1. Microstructure/Property Relationship

We observed independent from the tempering treatment of Fe-0.2C based alloys a strength increase (YS and UTS) from 2 Cr to 4 Cr additions (Figure 3). Both most likely inhere high fractions of lath martensite, reaching strengths higher than 1000 N mm⁻², which is typically obtained in low-carbon steels.^[12,13] The different strength levels may possibly be attributed to phenomena such as increased carbide precipitation or solid solution hardening induced by the increased Cr addition.^[2,3,14–17] The evaluation of hardened and as-quenched states (black rectangles; Figure 3) show disparate behavior with increasing Cr content. Considering that hardening temperatures were deliberately chosen to provoke miscellaneous material behavior, it suggests itself that multiple phase regimes are crossed for the lower hardening temperature, i.e., full γ and combinations with M_7C_3 . Diverging phase constitutions, e.g., various γ fractions prior to testing may be accountable for the increased ductility obtained with increasing Cr content (Figure 3).^[18] Considering the tempering treatments, interpretation of diverging material behavior becomes even more complex and requires in-depth analysis to verify the microstructural

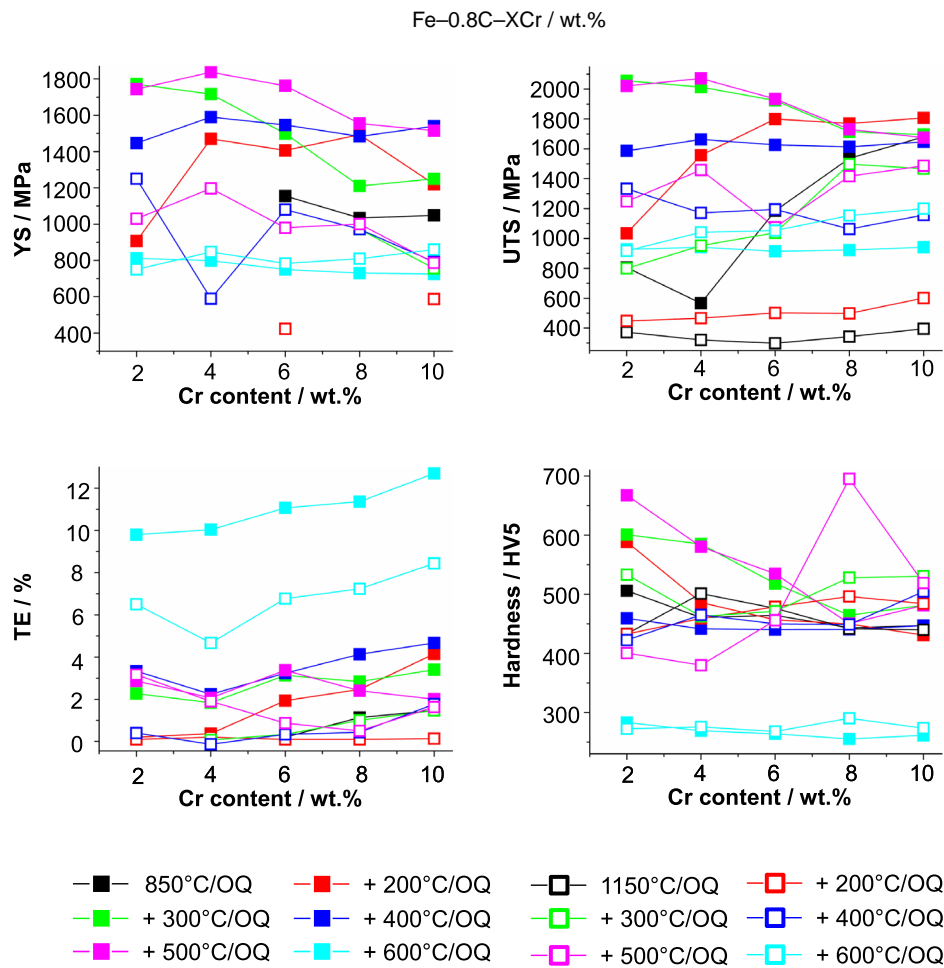


Figure 4. Color-coded plots of selected material parameters (YS, UTS, TE, Hardness) for the different Fe-based alloys inhering 0.8 C and diverging Cr content in different material states (filled symbols: hardened at 850 °C and tempered; unfilled symbols: hardened at 1150 °C and tempered).

Table 2. Material parameters (TE, YS, UTS, HV5) for the different hardening and tempering conditions of all material states analyzed in this study.

	TE	TE mean	YS	YS mean	Rm	Rm mean	HV5_1	HV5_2	HV5_3	HV5_4	HV5_5	HV5 mean
Fe-0.2C-X												
10 min @ 850 °C/OQ	2 Cr	7.7	7.9	1034	1103	1034	1114	282	254	309	372	366
		9.4		1130		1136						
		6.7		1144		1172						
	4 Cr	8.5	8.8	1316	1339	1328	1344	377	371	387	394	386
		8.5		1323		1323						
		9.4		1378		1380						
	6 Cr	9.0	8.7	1326	1304	1330	1305	394	412	403	400	408
		8.7		1294		1294						
		8.5		1292		1292						
	8 Cr	9.7	9.8	1257	1261	1264	1267	407	401	410	428	420
		10.6		1252		1259						
		9.0		1273		1277						
	10 Cr	10.1	9.1	1177	1182	1189	1194	368	381	378	388	382
		6.9		1180		1184						
		10.5		1189		1210						

Table 2. Continued.

		TE	TE mean	YS	YS mean	Rm	Rm mean	HV5_1	HV5_2	HV5_3	HV5_4	HV5_5	HV5 mean
10 min @ 850 °C/OQ 1 h @ 200 °C/OQ	2 Cr	6.7	7.6	1168	1156	1221	1174	270	382	294	358	364	334
		8.1		1065		1066							
		8.1		1234		1236							
	4 Cr	8.9	9.0	1290	1280	1291	1285	373	390	368	384	416	386
		9.3		1232		1241							
		8.8		1318		1324							
	6 Cr	7.7	8.4	1274	1274	1280	1278	414	404	441	412	422	419
		8.5		1280		1280							
		9.0		1267		1273							
	8 Cr	7.3	8.9	1102	1188	1146	1210	383	373	400	386	380	384
		9.9		1238		1250							
		9.5		1224		1233							
	10 Cr	9.1	9.0	1210	1211	1211	1212	384	398	424	380	406	398
		9.2		1210		1210							
		8.7		1213		1215							
10 min @ 850 °C/OQ 1 h @ 300 °C/OQ	2 Cr	0.6	6.6	870	816	938	908	278	292	286	307	302	293
		8		822		933							
		11.1		756		854							
	4 Cr	4.7	6.8	1235	1235	1245	1266	415	438	419	443	424	428
		7.6		1207		1283							
		8.2		1262		1270							
	6 Cr	10	8.7	1018	867	1188	1141	379	392	403	361	374	382
		9.9		879		1184							
		6.1		704		1050							
	8 Cr	8.2	7.7	1007	1033	1158	1127	360	310	402	377	375	365
		8.8		1041		1108							
		6.2		1049		1116							
	10 Cr	10.1	10.1	1201	1142	1221	1214	401	402	406	408	441	412
		9.8		1168		1198							
		10.3		1059		1224							
10 min @ 850 °C/OQ 1 h @ 400 °C/OQ	2 Cr	8.4	7.7	1106	1144	1106	1151	375	307	346	308	304	328
		7.9		1169		1169							
		6.9		1157		1178							
	4 Cr	8.5	8.1	1240	1226	1244	1227	379	423	336	366	332	367
		6.9		1225		1225							
		9.0		1213		1213							
	6 Cr	7.4	7.7	1194	1209	1216	1221	373	403	339	422	410	389
		7.5		1213		1222							
		8.1		1220		1224							
	8 Cr	9.7	9.6	1212	1207	1215	1213	387	382	375	374	380	380
		9.9		1203		1213							
		9.2		1206		1210							
	10 Cr	8.8	9.5	1179	1171	1180	1173	394	390	384	376	372	383
		9.5		1167		1167							
		10.2		1167		1173							

Table 2. Continued.

		TE	TE mean	YS	YS mean	Rm	Rm mean	HV5_1	HV5_2	HV5_3	HV5_4	HV5_5	HV5 mean
10 min @ 850 °C/OQ 1 h @ 500 °C/OQ	2 Cr	10.2	9.8	888	855	960	946	292	284	297	317	306	299
		8.2		867		961							
		11.1		809		918							
	4 Cr	7.1	5.4	536	908	1268	1256	393	404	393	375	360	385
		2.3		1069		1257							
		6.9		1117		1245							
	6 Cr	10.7	7.7	1027	1053	1179	1143	381	360	359	387	393	376
		7.7		1111		1168							
		4.8		1022		1081							
	8 Cr	6.7	7.7	987	1058	1101	1137	385	374	370	407	413	390
		7.4		1095		1162							
		8.9		1093		1147							
	10 Cr	8.6	8.7	1092	1079	1198	1207	449	443	366	431	412	420
		8.3		1039		1209							
		9.2		1106		1215							
10 min @ 850 °C/OQ 1 h @ 600 °C/OQ	2 Cr	14.8	15.0	623	626	666	668	208	199	210	205	200	204
		15.5		612		663							
		14.8		645		675							
	4 Cr	15.8	15.9	667	656	717	700	231	237	235	229	232	233
		16.1		656		697							
		15.7		644		688							
	6 Cr	15.0	15.2	633	637	688	688	228	217	224	230	232	226
		15.5		631		694							
		15.1		646		682							
	8 Cr	16.4	14.1	642	633	686	677	240	253	241	250	249	247
		11.9		629		669							
		14.1		629		677							
	10 Cr	14.1	14.1	666	668	703	703	239	231	220	221	228	228
		14.8		657		705							
		13.4		680		702							
10 min @ 1000 °C/OQ	2 Cr	7.1	7.3	1148	1152	1149	1155	344	323	313	330	325	327
		7.8		1146		1153							
		6.9		1162		1163							
	4 Cr	8.1	8.2	1312	1313	1324	1325	371	366	365	360	374	367
		8.5		1320		1341							
		7.8		1306		1310							
	6 Cr	7.6	8.3	1318	1311	1320	1320	378	363	382	372	370	373
		8.8		1313		1330							
		8.4		1301		1310							
	8 Cr	8.8	9.3	1304	1303	1305	1309	326	384	370	352	370	360
		9.7		1310		1316							
		9.4		1296		1305							
	10 Cr	9.5	9.3	1305	1303	1308	1306	388	356	369	372	394	376
		9.9		1300		1307							
		8.4		1304		1304							

Table 2. Continued.

		TE	TE mean	YS	YS mean	Rm	Rm mean	HV5_1	HV5_2	HV5_3	HV5_4	HV5_5	HV5 mean
10 min @ 1000 °C/OQ 1 h @ 200 °C/OQ	2 Cr	6.3	6.3	1054	1115	1105	1154	350	320	363	316	328	335
		7.0		1192		1196							
		5.8		1098		1161							
	4 Cr	6.5	6.7	1283	1296	1302	1303	360	408	363	358	366	371
		6.8		1308		1309							
		6.8		1298		1299							
	6 Cr	9.1	8.2	1296	1296	1297	1299	372	352	384	379	382	374
		7.0		1289		1297							
		8.6		1302		1304							
	8 Cr	9.3	9.0	1296	1285	1296	1290	353	271	359	361	355	340
		8.9		1277		1290							
		8.9		1283		1285							
	10 Cr	10.3	9.8	1238	1262	1277	1278	363	365	391	359	361	368
		9.9		1280		1287							
		9.2		1269		1270							
10 min @ 1000 °C/OQ 1 h @ 300 °C/OQ	2 Cr	6.2	8.9	916	882	916	965	312	327	317	337	346	328
		10.1		848		1004							
		10.3		883		974							
	4 Cr	10.6	10.2	1172	947	1273	1245	381	389	378	385	357	378
		9.7		1019		1232							
		10.3		649		1230							
	6 Cr	10.4	9.2	913	911	1092	1040	397	395	402	419	402	403
		9.4		977		1095							
		7.8		843		933							
	8 Cr	11.6	8.7	941	997	1108	1139	424	403	373	386	386	394
		9.4		1024		1157							
		5.2		1025		1153							
	10 Cr	10.2	10.6	1264	1113	1287	1286	386	404	387	337	354	374
		10.2		962		1285							
		11.4		1111		1287							
10 min @ 1000 °C/OQ 1 h @ 400 °C/OQ	2 Cr	7.8	7.1	1144	1105	1144	1130	327	343	347	335	327	336
		5.7		1044		1118							
		7.7		1127		1127							
	4 Cr	7.1	7.1	1247	1250	1250	1257	337	348	342	293	312	326
		7.2		1247		1263							
		7.0		1255		1256							
	6 Cr	9.2	8.5	1272	1242	1272	1246	327	378	230	373	383	338
		8.2		1197		1208							
		8.2		1257		1257							
	8 Cr	10.3	10.3	1000	1042	1222.58	1185	376	382	303	369	371	360
				988		1159							
				1139		1172							
	10 Cr	9.6	10.2	1160	1044	1220	1210	347	363	256	387	402	351
		10.8		927		1200							

Table 2. Continued.

		TE	TE mean	YS	YS mean	Rm	Rm mean	HV5_1	HV5_2	HV5_3	HV5_4	HV5_5	HV5 mean
10 min @ 1000 °C/OQ 1 h @ 500 °C/OQ	2 Cr	10.6	10.2	688	742	879	847	297	290	302	299	297	297
		9.7		727		800							
		10.3		812		863							
	4 Cr	7.5	8.7	1054	914	1099	1124	355	384	360	408	377	377
		8.4		997		1153							
		10.2		691		1121							
	6 Cr	4.4	9.3	968	990	1055	1114	404	394	394	418	406	403
		11.3		1006		1135							
		12.1		997		1153							
	8 Cr	1.2	3.2	994	931	1112	1117	421	419	419	393	392	409
		0		867		1123							
		8.4				1115							
	10 Cr	14.2	10.5	1179	1132	1285	1282	415	356	413	404	408	399
		6.3		1199		1284							
		11.1		1017		1277							
10 min @ 1000 °C/OQ 1 h @ 600 °C/OQ	2 Cr	16	16.0	553	553	634	634	174	200	206	201	200	196
	4 Cr	15.2	15.2	632	632	727	727	224	212	219	198	230	217
	6 Cr	15	15.0	643	643	730	730	183	164	210	219	225	200
	8 Cr	14.9	14.9	629	629	729	729	219	216	223	230	222	222
	10 Cr	13.5	13.5	705	705	753	753	238	235	248	233	244	240
Fe-0.8C-X													
10 min @ 850 °C/OQ 1 h @ 200 °C/OQ	2 Cr	0.2	0.10	788	788	855	805	519	529	454	497	530	506
		0.1				936							
		0				625							
	4 Cr					497	566	474	469	456	442	461	460
						731							
						470							
	6 Cr	0.1	0.20		1155		1185	461	454	466	458	482	464
		0.2		1160		1170							
		0.3		1150		1200							
	8 Cr	0.8	1.13	1070	1034	1520	1536	467	447	430	430	442	443
		0.7		1070		1370							
		1.9		961		1720							
	10 Cr	1.7	1.47	1010	1048	1690	1680	423	464	456	463	429	447
		0.9		1140		1580							
		1.8		995		1770							
10 min @ 850 °C/OQ 1 h @ 200 °C/OQ	2 Cr	0.4	0.20	1270	907	1330	1035	647	636	597	567	495	588
		0		543		1050							
						725							
	4 Cr	-0.1	0.37		1470	1540	1556	496	443	482	488	521	486
		-0.1				1400							
		1.3		1470		1730							
	6 Cr	2.3	1.93	1350	1407	1870	1800	483	471	438	453	438	457
		1.9		1430		1830							
		1.6		1440		1700							

Table 2. Continued.

		TE	TE mean	YS	YS mean	Rm	Rm mean	HV5_1	HV5_2	HV5_3	HV5_4	HV5_5	HV5 mean
10 min @ 850 °C/OQ 1 h @ 300 °C/OQ	8 Cr	2.2	2.47	1330	1493	1680	1770	470	434	456	438	450	450
		2.7		1560		1790							
		2.5		1590		1840							
	10 Cr		4.15	1390	1220	1780	1806	420	430	392	439	473	431
		3.9		1160		1780							
		4.4		1110		1860							
	2 Cr	1.8	2.27	1939	1772	2059	2056	655	487	582	641	638	601
		2.2		1951		2059							
		2.8		1425		2051							
	4 Cr	2.5	1.83	1954	1716	2038	2013	545	605	645	582	547	585
		1.4		1967		2066							
		1.6		1228		1936							
	6 Cr	2.4	3.13	1193	1499	1878	1923	493	526	490	528	553	518
		3.5		1666		1993							
		3.5		1638		1899							
	8 Cr	4.5	2.83	420	1210	1697	1714	451	461	493	476	441	464
		1.8		1590		1672							
		2.2		1620		1773							
10 min @ 850 °C/OQ 1 h @ 400 °C/OQ	10 Cr	4.4	3.40	1523	1250	1708	1695	488	467	486	499	467	481
		1.9		1506		1707							
		3.9		721		1673							
	2 Cr	4.1	3.33	1360	1447	1670	1586	454	455	455	470	462	459
		2.5		1360		1440							
		3.4		1620		1650							
	4 Cr	2.5	2.23	1590	1590	1670	1663	479	437	438	423	430	441
		0.8		1630		1630							
		3.4		1550		1690							
	6 Cr	3.9	3.23	1490	1547	1640	1626	438	423	431	439	470	440
		2.6		1500		1550							
		3.2		1650		1690							
	8 Cr	4.4	4.13	1480	1483	1660	1613	442	435	449	435	441	440
		3.5		1370		1520							
		4.5		1600		1660							
	10 Cr	3.9	4.67	1470	1540	1580	1646	430	461	442	464	438	447
		5.1		1630		1680							
		5		1520		1680							
10 min @ 850 °C/OQ 1 h @ 500 °C/OQ	2 Cr	3.3	2.87	1762	1744	2021	2020	698	657	660	660	662	667
		3.4		1637		2032							
		1.9		1832		2010							
	4 Cr	0.7	2.07	1636	1838	2103	2071	574	558	629	570	570	580
		2.1		1950		2047							
		3.4		1927		2063							
	6 Cr	3.8	3.37	1843	1762	1948	1934	516	572	551	533	501	535
		3.3		1701		1949							
		3		1742		1905							

Table 2. Continued.

		TE	TE mean	YS	YS mean	Rm	Rm mean	HV5_1	HV5_2	HV5_3	HV5_4	HV5_5	HV5 mean
10 min @ 850 °C/OQ 1 h @ 600 °C/OQ	8 Cr	3.5	2.40	1426	1555	1766	1730	450	474	466	405	456	450
		0.4		1594		1644							
		3.3		1643		1782							
	10 Cr	2.5	2.00	1442	1515	1669	1674	486	461	491	474	494	481
		1.7		1593		1672							
		1.8		1510		1683							
	2 Cr	10.5	9.80	838	811	972	927	275	280	282	275	303	283
		9.8		808		924							
		9.1		787		885							
	4 Cr	10.9	10.03	789	799	963	941	268	273	269	264	271	269
		11		833		966							
		8.2		774		896							
	6 Cr	9.6	11.07	722	750	888	914	244	265	265	276	269	264
		13.3		772		926							
		10.3		757		928							
	8 Cr	9.2	11.37	694	730	889	922	248	260	252	252	264	255
		12.1		729		937							
		12.8		767		942							
	10 Cr	11.8	12.70	743	725	915	941	248	270	255	274	259	261
		12.6		668		959							
		13.7		764		950							
10 min @ 1150 °C/OQ 1 h @ 200 °C/OQ	2 Cr					373	372	435	434	360	457	479	433
						394							
						351							
	4 Cr					285	320	498	488	533	451	535	501
						360							
						317							
	6 Cr					315	298	425	479	504	467	506	476
						254							
						326							
	8 Cr					318	343	439	430	434	445	460	442
						363							
						348							
	10 Cr					390	395	438	446	446	389	480	440
						366							
						430							
	2 Cr	0.1	0.10			407	448	436	414	409	448	456	433
						529							
						408							
	4 Cr	0.2	0.20			411	466	501	476	483	451	384	459
						445							
						542							
	6 Cr	0.1	0.10	423	423	464	501	480	460	490	502	461	479
		0.2				467							
		0				574							

Table 2. Continued.

		TE	TE mean	YS	YS mean	Rm	Rm mean	HV5_1	HV5_2	HV5_3	HV5_4	HV5_5	HV5 mean
10 min @ 1150 °C/OQ 1 h @ 300 °C/OQ	8 Cr	0.1	0.10			492	498	511	449	473	512	536	496
		0.1				540							
		0.1				464							
	10 Cr	0.1	0.13	587	587	564	601	474	540	529	434	443	484
		0.1				571							
		0.2				668							
	2 Cr					759	799	526	527	528	524	560	533
						802							
						838							
	4 Cr	0.1	0.07			949	951	403	494	517	460	438	462
		0				978							
		0.1				929							
	6 Cr	0.2	0.33			736	1038	522	467	470	431	466	471
		0.8				1228							
		0				1151							
	8 Cr	0.7	1.00	980	971	1430	1497	582	538	526	488	506	528
		1.3		915		1545							
		1		1019		1518							
	10 Cr	1.7	1.47	752	756	1468	1466	582	529	529	519	493	530
		1.2		674		1415							
		1.5		842		1518							
10 min @ 1150 °C/OQ 1 h @ 400 °C/OQ	2 Cr	0.2	0.40	1220	1250	1320	1333	414	448	461	422	367	422
		1		1280		1380							
		0				1300							
	4 Cr	−0.8	−0.13		589	1390	1171	464	473	453	464	471	465
		−1				1270							
		1.4		589		853							
	6 Cr	0.1	0.33		1080		1195	437	456	450	445	461	450
		0.3		1070		1130							
		0.6		1090		1260							
	8 Cr	0.4	0.43	1010	972	1040	1063	449	466	437	434	458	449
		0.6		911		1110							
		0.3		995		1040							
	10 Cr	1	1.77	947	792	1140	1156	519	487	507	480	529	504
		1.5		856		1140							
		2.8		574		1190							
10 min @ 1150 °C/OQ 1 h @ 500 °C/OQ	2 Cr	2.3	3.17	1102	1029	1201	1248	430	402	372	363	435	400
		2.8		1082		1268							
		4.4		905		1276							
	4 Cr	1.5	1.90	1313	1197	1456	1458	362	461	374	353	349	380
		1.4		989		1403							
		2.8		1289		1517							
	6 Cr	0.5	0.87	1435	980	1541	1074	445	457	474	450	453	456
		1.4		884		996							
		0.7		620		687							

Table 2. Continued.

		TE	TE mean	YS	YS mean	Rm	Rm mean	HV5_1	HV5_2	HV5_3	HV5_4	HV5_5	HV5 mean
10 min @ 1150 °C/OQ 1 h @ 600 °C/OQ	8 Cr	0.8	0.50	870	1000	1471	1417	690	698	731	687	670	695
		0.7		1131		1493							
		0				1288							
	10 Cr	1.5	1.63	729	786	1435	1486	522	504	491	549	528	519
		1.9		666		1483							
		1.5		963		1541							
	2 Cr	7	6.50	781	750	950	916	209	272	287	291	303	272
		5.7		709		858							
		6.8		760		942							
	4 Cr	5	4.67	839	847	1060	1041	279	223	291	289	295	275
		4.1		833		975							
		4.9		868		1090							
	6 Cr	7.7	6.77	792	784	1080	1053	273	264	284	267	251	268
		6.2		762		1010							
		6.4		797		1070							
	8 Cr	6.8	7.23	742	809	1130	1153	318	259	283	297	294	290
		7.4		950		1170							
		7.5		736		1160							
	10 Cr	8.9	8.43	878	860	1220	1200	282	294	283	273	235	273
		7.5		829		1180							
		8.9		872		1200							

changes developed. We exemplarily studied samples from the Fe–0.2C–4Cr (Figure 5, left) and Fe–0.2C–8Cr (Figure 5, right) alloys oil-quenched from 1000 °C and tempered at 400 °C, followed by oil-quenching. Here, a strength difference of about 200 MPa and a ductility increase can be noted, which does not arise for any other state in this compositional range. This behavior may be related to a multitude of interacting phenomena with different weighting as a consequence of the tempering, such as rearrangement of carbon, development of carbides, or the coarsening of the lath microstructure (cf. Figure 5).^[19] Even though no γ was detected in both material states, its pre-existence in the microstructure prior to tempering—thus its impact on the mechanical performance—by transformation to α' during tensile testing or metallographic preparation is conceivable.^[20] Despite a detailed analysis, no explicit reasons for this unexpected material behavior can be given. Further investigations into the underlying microstructural reasons such as dislocation density, packing density of martensite laths, or prior austenite grain sizes indeed may give additional insights but may not be sufficient to fully clarify the complex reasons for materials' behavior, thus not contribute toward its understanding. In the context of high-throughput experiments, the efforts for these time-consuming in-depth experiments were deliberately kept low. Apart from microstructural analysis, these phenomena may be attributed to an unfortunate coincidence of several processing-related factors such as notches, spalling, torsion, or the bending of tested materials induced by the hot-rolling, handling, quenching, or cutting of the not standardized RAP tensile samples.

4.2. High-Throughput Design Challenges

The implementation of high-throughput methods in the material sciences typically aims for the identification of trends via correlation of rapidly generated data.^[21] This requires systematic approaches in controlled conditions, which can efficiently generate simply recordable data. For conducted experiments, material properties (single data; e.g., hardness) or determined characteristic values (e.g., YS) seem to be most suitable. A broad range of parameters may be determined from tensile testing results which need to be tailored to fit materials' characteristics; e.g., material parameters such as uniform elongation may be ideally suited for soft and ductile materials, yet unusable for hard and brittle materials. The multitude of chosen descriptive material parameters (e.g., YS, UTS, TE) generated in combination with varying material compositions and states opens up multi-dimensional spaces, holding challenges for interpretation. Attempts to overcome this trade-off are multi-dimensional plots such as spider diagrams or heat-maps as shown in **Figure 6** (left: Fe–0.2C–XCr alloys; right: Fe–0.8C–XCr alloys). Here, the effect of hardening temperature (top and bottom picture), tempering temperature (x -axis), and Cr content (y -axis) on materials' mechanical performance ($YS \times TE$; color coded) is exemplarily expressed. Consequently, every map reduces 4 dimensions (mentioned earlier) in a 2D space. Uniform color coding was used in all the maps (red: high values, black: low values) to ensure comparability. The compilation of these four heat-maps (Figure 6) gives first insights into materials' performance such as their suitability for forming applications

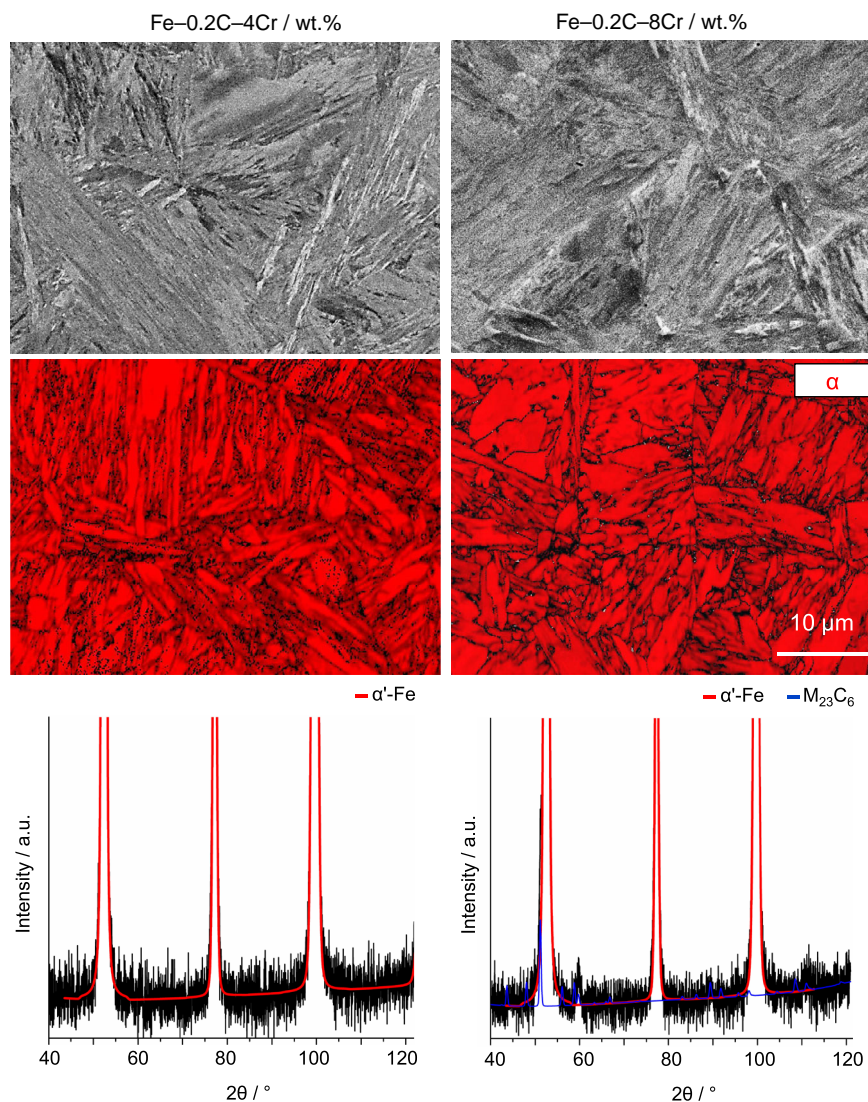


Figure 5. Microstructural analysis of samples (left: Fe-0.2C-4Cr; right: Fe-0.2C-8Cr) in hardened (1000 °C/10 min/oil-quenching) and tempered (400 °C/1 h/oil-quenching) conditions. Top: SE images; middle: EBSD phase map, here image quality in grayscale is superimposed with the α' -Fe phase in red; bottom: XRD results (black) and fitting results α' -Fe (red) and $M_{23}C_6$ (blue).

(ductility) or wear resistance (hardness). However, these traditional data analysis are biased by perspectives based on empirical experience, thus offering potential missing relevant interconnections or underlying information.

4.3. Leaving Traditional Design Routes

The extension of high-throughput approaches by computer-assisted data processing and analysis open up assistance for materials science confronted with multi-dimensional problems. However, to exploit the potential of computational methods, consistent data production and data handling is required. Data are ideally simple to generate, to extract/parse (single values), and also to access (e.g., via a database) to apply data processing (algorithms). Typically, already simple experiments (such as tensile tests) deliver large quantities of data, not allowing straightforward analysis, but require the extraction of descriptive

materials parameters dependent on boundary conditions (e.g., turning points or maximum values). Further extraction of underlying information ideally utilized parsing of raw data and automatic determination of single-value parameters—possible also of unconventional nature—to avoid human errors. Subsequently, data processing (ratios, ratings, random) can assist in generating data tangible by human or alternatively artificial intelligence to reduce the complexity of these multi-dimensional problems in an iterative manner (e.g., re-set constrains).

5. Conclusions

We systematically studied the effect of various Cr additions (2, 4, 6, 8, 10 wt%) on mechanical properties and microstructures of Fe-0.2C (wt%) and Fe-0.8C (wt%) based steels depending on their hardening and tempering conditions. The following

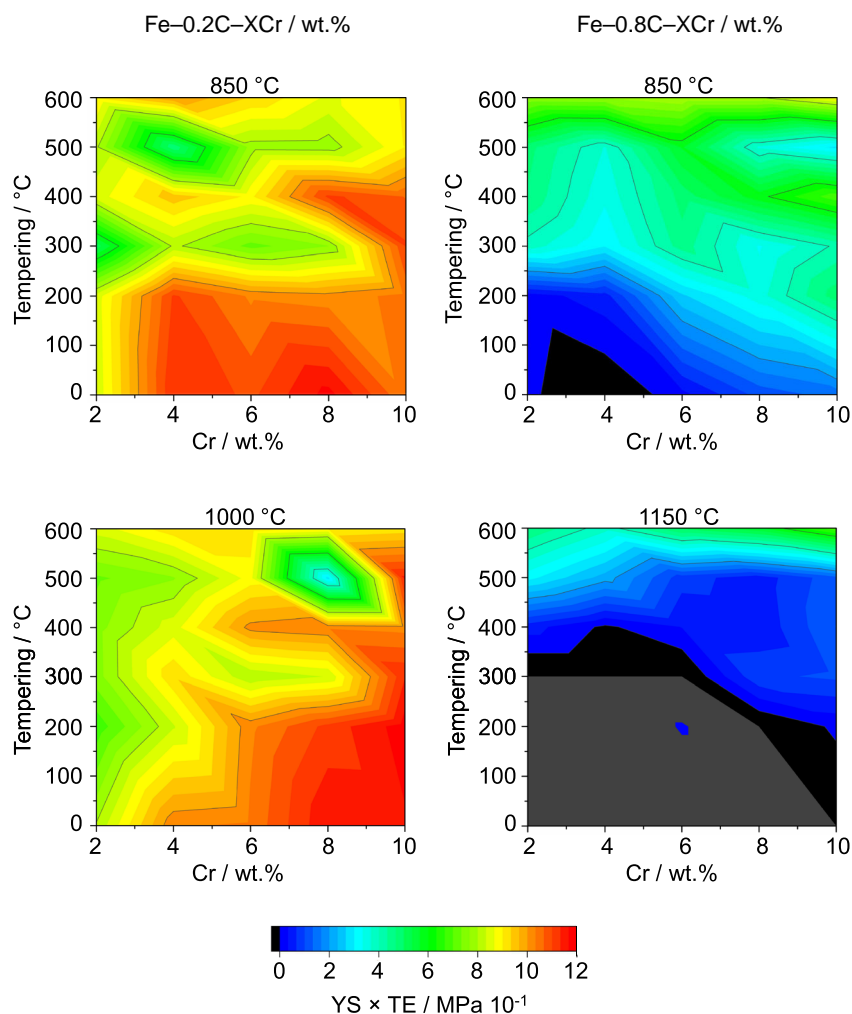


Figure 6. Color-coded heat maps revealing the effects of hardening temperature, tempering temperature, and Cr content on mechanical performance ($YS \times TE$) of the different alloys (left: Fe-0.2C-XCr; right: Fe-0.8C-XCr). Uniform color coding was used in all the maps (red: high values, black: low values).

conclusions can be drawn: 1) The variety of states analyzed reveal a broad range of mechanical properties reaching strengths up to 1.8 GPa UTS or hardness of 700 HV5. Individual trends can be obtained while straightforward interpretation of trends holds complex analytical challenges; 2) Diversity of mechanical properties are associated with various competing phenomena influencing the materials microstructures, such as microstructures present prior to hardening or phase formation and fractions such as austenite or carbide. However, interconnections associated with their occurrence are too complex to grasp within this study, thus selected states were analyzed in-depth; 3) We discussed classical data analysis and interpretation methods such as dimension reduction via multi-axis plots/heat maps as well as the potential of assistive computational data processing, analysis, and its challenges to deal with multi-dimensional interconnections.

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Conflict of Interest

The authors declare no conflict of interest.

Keywords

carbon, chromium, high-throughput screening, steels

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