

Vibrational Spectroscopy
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Polarization-Dependent Sum-Frequency-Generation Spectroscopy for In Situ Tracking of Nanoparticle Morphology

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Abstract: The surface structure of oxide-supported metal nanoparticles can be determined via characteristic vibrations of adsorbed probe molecules such as CO. Usually, spectroscopic studies focus on peak position and intensity, which are related to binding geometries and number of adsorption sites, respectively. Employing two differently prepared model catalysts, it is demonstrated that polarization-dependent sum-frequency-generation (SFG) spectroscopy reveals the average surface structure and shape of the nanoparticles. SFG results for different particle sizes and morphologies are compared to direct real-space structure analysis by TEM and STM. The described feature of SFG could be used to monitor particle restructuring in situ and may be a valuable tool for operando catalysis.

Sum frequency generation (SFG) is a powerful non-linear optical vibrational spectroscopy. Upon simultaneous excitation by a broadband or wavelength-scanned mid-infrared (IR) and a fixed narrowband visible (VIS) laser pulse, a signal is generated at the sum of the incident frequencies. The underlying process is limited to non-centrosymmetric media (e.g., surfaces/interfaces) and exhibits inherent surface-sensitivity.^[1,2] Thus, different from conventional IR spectroscopy, SFG *only* measures vibrations of surface adsorbed molecules, even when the same molecules are also present in, e.g., a gas phase^[3–7] (for differentiating surface and gas phase contributions IR absorption spectroscopy requires polarization-modulation^[8]). Accordingly, SFG can also characterize molecules at air-liquid and liquid-liquid interfaces and even examine “buried” interfaces inside solids.^[9–12] Furthermore, the strong dependence of the coherent SFG light on the ordering and abundance of the probed bonds enables structure/coverage evaluation, while using different IR and vis polarization combinations allows identification of bond orientation,^[13–15] due to the tensor character of the second order non-linear susceptibility. SFG

has been applied to studies of interface phenomena in many fields, including electrochemistry,^[16] photocatalysis,^[17] plasmonics,^[18] polymers,^[19] self-assembly,^[20] and nanomedicine.^[21,22]

Herein, we applied polarization-dependent SFG to examine molecular adsorption on oxide supported metal nanoparticles (NPs) of Pt and Pd. Given that the used probe molecule CO adsorbs in the same orientation to top and side facets of Pt and Pd surfaces,^[13–15,23] the intensity ratio of different polarization combinations in SFG measurements is affected by the shape of the NPs rather than by varying adsorption tilt angles. We demonstrate that the spectral intensity ratio mirrors the shape of different NPs in two different model catalyst systems, matching the direct morphology characterization by transmission electron microscopy (TEM) and scanning tunneling microscopy (STM). This enables the use of SFG for in situ evaluation of average NP morphology for reactions that involve CO or are not negatively affected by CO probe molecules.^[24]

The basics of sum frequency generation spectroscopy have been described in the literature,^[1,2,6,8,10–13] with the SFG intensity I_{SFG} depending linearly on the intensity of the incident beams, I_{IR} and I_{VIS} , and the absolute square of the second order nonlinear susceptibility $\chi^{(2)}$. A scheme of the beam propagation and the possible beam polarization directions is displayed in Figure 1a. It has been shown that on metals only two polarization combinations, i.e., ppp and ssp, yield a significant SFG signal:^[25,26]

$$\chi_{\text{eff,ppp}}^{(2)} = -L_{xx}(\omega_{\text{SFG}})L_{xx}(\omega_{\text{VIS}})L_{zz}(\omega_{\text{IR}})\cos(\alpha_{\text{SFG}})\cos(\alpha_{\text{VIS}})\sin(\alpha_{\text{IR}})\chi_{xxz} - L_{xx}(\omega_{\text{SFG}})L_{zz}(\omega_{\text{VIS}})L_{xx}(\omega_{\text{IR}})\cos(\alpha_{\text{SFG}})\sin(\alpha_{\text{VIS}})\cos(\alpha_{\text{IR}})\chi_{zxx} + L_{zz}(\omega_{\text{SFG}})L_{xx}(\omega_{\text{VIS}})L_{xx}(\omega_{\text{IR}})\sin(\alpha_{\text{SFG}})\cos(\alpha_{\text{VIS}})\cos(\alpha_{\text{IR}})\chi_{zxx} + L_{zz}(\omega_{\text{SFG}})L_{zz}(\omega_{\text{VIS}})L_{zz}(\omega_{\text{IR}})\sin(\alpha_{\text{SFG}})\sin(\alpha_{\text{VIS}})\sin(\alpha_{\text{IR}})\chi_{zzz} \quad (1)$$

$$\chi_{\text{eff,ssp}}^{(2)} = L_{yy}(\omega_{\text{SFG}})L_{yy}(\omega_{\text{VIS}})L_{zz}(\omega_{\text{IR}})\sin(\alpha_{\text{IR}})\chi_{yyz} \quad (2)$$

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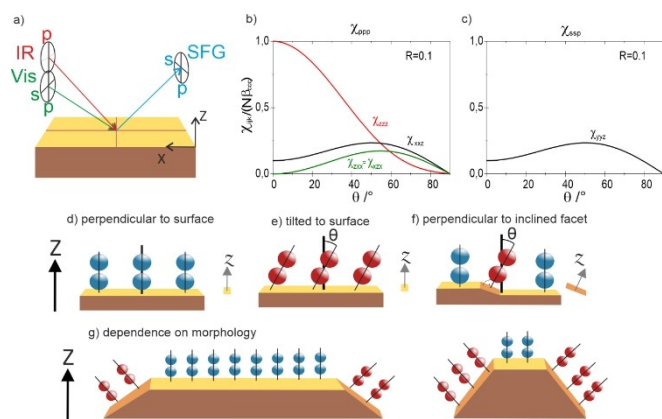


Figure 1. a) Possible polarizations of SFG beams and their orientation in the laboratory frame. b), c) simulated contributions to the effective nonlinear susceptibility by the non-vanishing elements ppp and ssp polarization combination, using a low $R = \beta_{aac}/\beta_{ccc}$ of 0.1. d)–g) influence of inclined surface facets on the bond orientation of CO relative to the macroscopic surface normal.

L_{ab} denotes the Fresnel factors and χ_{ijk} denotes the non-vanishing tensor elements as given in the Supporting Information. χ_{ppp} is a linear combination of several elements (Figure 1b), weighted by the Fresnel factors. For the current measurement geometry, it has a maximum at 0° bond tilt angle and monotonously decreases with increasing tilt (with χ_{zzz} as main contribution).^[13] In comparison, χ_{ssp} is directly proportional to χ_{yyz} , so the ssp intensity increases with increasing tilt angle θ between the linear CO molecule and the surface normal Z , reaching a maximum around 45 to 50° (see Figure 1c), depending mainly on the molecular tensor element ratio $R = \beta_{aac}/\beta_{ccc}$.^[13–15]

Therefore, a comparison of I_{ppp} and I_{ssp} allows to monitor the tilt angle of molecular bonds, but most SFG studies of metal-gas interfaces typically reported only the ppp spectra, as these yield the strongest signal. A key aspect of the observations reported herein is schematically illustrated in Figure 1d–f: molecules perpendicularly adsorbed (blue) on a planar surface have no tilt to either the local surface normal z or the macroscopic normal Z (Figure 1d), while molecules adsorbed in a tilted geometry (red) on a flat surface are tilted with respect to both z and Z (Figure 1e). As SFG is a macroscopic technique, even a molecule that is adsorbed perpendicularly on an *inclined* facet (parallel to z) is tilted from the macroscopic surface normal Z (Figure 1f). Therefore, it contributes to the signal like a molecule with tilted adsorption geometry. Variations in the particle morphology as illustrated in Figure 1g thus lead to different intensity ratios for SFG measurements of different polarization combinations, which reflect the NP shape.

In this study, two supported metal NP systems were investigated, Pt on ZrO_2 and Pd on Al_2O_3 . Measurements were performed in two comparable setups, each consisting of an ultra-high vacuum (UHV) surface-preparation and analysis chamber and an attached SFG spectroscopy cell for in situ experiments from UHV to atmospheric pressure, which have been described in detail.^[27,28] As the metal

nanoparticles were >3.5 nm, no support effects are to be expected.^[29,30]

The Pt/ ZrO_2 model catalysts consisted of a 42 nm thick zirconia film—grown by 400 cycles of atomic layer deposition (ALD) on a Si(100) wafer—and Pt deposits prepared by different numbers of ALD cycles (10 to 250).^[33] TEM images showed that using 10 Pt cycles produced Pt particles roughly 6 nm in size, whereas 250 deposition cycles formed a homogeneous Pt film of uniform thickness of approximately 10 nm (Figure 2). The latter serve as a bridge between previous single crystal studies and the current NP results. For characterization by electron microscopy, X-ray diffraction and photoelectron spectroscopy refer to reference [33].^[33]

The second set of samples, well-defined Pd NPs, were grown by physical vapor deposition (PVD) on alumina thin films in UHV.^[31,32,34–36] Figure 2 shows STM images of Pd NPs grown on $Al_2O_3/NiAl(110)$ at 300 K and 90 K substrate temperature. Pd NPs grown at 300 K are truncated cuboctahedra with distinct smooth (111) and (100) facets, whereas particles grown at 90 K appear rather rounded/irregular (i.e., the facets are rougher with more steps/defects). For the samples discussed below, a nominal thickness of 0.6 nm Pd was deposited at 300 and 90 K, yielding well-faceted Pd NPs of 6.1 mean size and rougher Pd NPs of 3.6 nm mean size, respectively. NPs of 5.9 nm mean size were grown at 90 K by depositing a nominal thickness of 1.2 nm Pd (using spot profile analysis of low energy electron diffraction (SPA-LEED) for size/NP density analysis^[35]). Similar to the Pt samples, these larger particles are higher than particles grown with less Pd and thus have more pronounced sideward facing facets. Additional information can be found in the Supporting Information.

Diameter	Islands/continuous	10 nm	5 nm
ALD Deposition	250 cycles	50 cycles	10 cycles
Nominal Thickness	10 nm	2 nm	0.5 nm
Pt/ ZrO_2			
Cross-section TEM images			
Diameter Preparation	6.1 nm 0.6 nm / 300K	5.9 nm 1.2 nm / 90K	3.6 nm 0.6 nm / 90K
NP density dispersion	$1.00 \cdot 10^{12} / \text{cm}^2$ 21%	$2.3 \cdot 10^{12} / \text{cm}^2$ 22%	$4.7 \cdot 10^{12} / \text{cm}^2$ 35%
#Atoms #surf. atoms	4100 840 per NP	3600 770 per NP	900 300 per NP
Pd/ Al_2O_3			
STM images			

Figure 2. Morphology of supported Pt and Pd nanoparticles. The ALD-deposited Pt particles initially grow with pyramidal shape that becomes more rounded upon increasing the Pt amount. Highest Pt exposure leads to coalescence, forming smooth islands and finally continuous smooth films. The Pd particle size is varied by different substrate temperature and Pd amount during PVD. At 300 K, particles grow with truncated cuboctahedral shape exhibiting a large flat (111) top facet. At 90 K, rougher, half-spherical NPs are formed, also with higher nucleation/particle density. STM images adapted from with permission from Refs. [31] and [32].

For both model catalyst systems, polarization-dependent SFG measurements of adsorbed CO molecules revealed three important variables: The first is the peak position, which provides information on the local adsorption geometry (on-top/bridge/hollow). The exact resonance position, especially for on-top CO, is further affected by the surface roughness (metal coordination number) and the CO coverage (via dipole-dipole coupling and chemical shift).^[37] The second is intensity, which depends on the abundance of adsorption sites, coverage, ordering and vibrational coupling of adsorbed CO.^[7,8,37] SFG measurements for extracting the size of Pt^[38] and Pd particles^[39] using only ppp polarization, have been reported previously. In these studies, particle structure was identified by comparing the peak positions and intensities and using different theoretical simulation models as references. The third measured variable is the $I_{\text{ssp}}/I_{\text{ppp}}$ ratio, which provides information on the average orientation of the C–O bond with respect to the *macroscopic* surface normal. Polarization dependent (ssp and ppp) SFG spectra of CO adsorbed on Pt and Pd single crystals have been reported^[8,13–15,40] and are used for benchmarking our results.

Figure 3 shows the ppp (black) and ssp (red) spectra of the ALD Pt samples for (a) the continuous film, (b) rounded NPs of about 10 nm diameter and (c) pyramidal NPs of about 6 nm diameter, all in 10 mbar CO at 425 K. For the thin Pt film, a characteristic peak of on-top CO was observed at 2089 cm⁻¹ in ppp, matching well the 2090 cm⁻¹ on Pt(111) under these conditions^[41] (for details on fitting see the Supporting Information). As the surface roughness increases (i.e., the average Pt coordination number de-

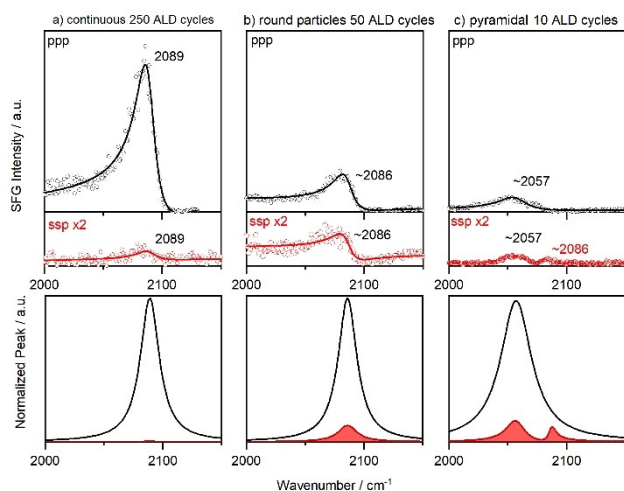


Figure 3. Polarization-dependent SFG spectra of on-top CO on different ALD-grown Pt/ZrO₂ samples. Measurements were performed in 10 mbar of CO at 425 K. The black spectra represent the ppp polarization combination, gradually shifting to lower wavenumber for smaller and rougher Pt nanoparticles. The red spectra represent the ssp polarization combination, which is enlarged by a factor of two with respect to the ppp spectra. SSP spectra are comparably stronger for curved particle morphology. The bottom row shows the normalized spectral fits of the resonant signal without the non-resonant background.

creases), the on-top peak shifts via 2086 cm⁻¹ to 2057 cm⁻¹. The ssp spectra mirror this trend, despite the slightly different phase. For the pyramidal 6 nm Pt particles, there is also a small peak at 2086 cm⁻¹ in ssp, pointing to a few (111)-like patches. Apart from the resonance position, the $I_{\text{ssp}}/I_{\text{ppp}}$ ratios are informative. The intensity ratios derived from the fitted peaks are collected in Table 1, together with the single crystal reference data.^[13–15] Smooth films have no or few inclined (side) facets and also a low step/terrace sites ratio. Accordingly, SFG spectra of the thin Pt film show a strong ppp signal, but a very weak ssp signal. For the curved 10 nm Pt particles $I_{\text{ssp}}/I_{\text{ppp}}$ is highest (0.4), while it is smaller (0.3) for the pyramidal 6 nm Pt particles, due to their inclined, but less curved, surfaces.

A previous polarization-dependent study on Pt/SiO₂ reported a significant enhancement of SFG intensities for both polarizations on 40 nm polycrystalline particles due to plasmon resonances.^[42] In our study we did not observe such an effect, which is likely due to the smaller overall size and different particle shape/structure.

For PVD-grown Pd/Al₂O₃ model catalysts, SFG spectra of CO adsorbed on the Pd particles were obtained in UHV at 200 K, after saturating the surface with CO (cool-down from room temperature in 10⁻⁶ mbar CO). The ppp and ssp polarization combinations for well-faceted Pd particles of 6.1 nm mean diameter, as well as for rougher 5.9 and 3.6 nm Pd particles are shown in Figure 4. Compared to the Pt spectra discussed above, a strong asymmetric lineshape is visible in ppp, which is related to the NiAl substrate. As reported in detail in reference,^[43] it induces a much stronger non-resonant background, likely related to an inter-band transition in the substrate. Nevertheless, peak fitting still provides the resonance positions and normalized fits, excluding the non-resonant background, as displayed in the bottom row of Figure 4. For all Pd NP sizes, ppp spectra predominantly showed bridge bonded CO (centered around 1981–1992 cm⁻¹), adsorbed on particle edges and steps.^[7,8,29,34–36] This peak benefits from intensity transfer from the lower-wavenumber bridged CO on (111) and (100) facets (shoulder around 1960 cm⁻¹). The spectra also revealed a weaker on-top CO centered around 2085–2096 cm⁻¹. The ratio of $I_{\text{ssp}}/I_{\text{ppp}}$ for bridge bonded CO is given in Table 1, together with reference data from single

Table 1: $I_{\text{ssp}}/I_{\text{ppp}}$ ratio ($\pm 10\%$) for on-top bonded CO on different Pt/ZrO₂ samples and bridge bonded CO on different Pd/Al₂O₃ samples.

Pt on-top CO $I_{\text{ssp}}/I_{\text{ppp}}$		10 nm NPs	6 nm NPs
Single crystal	Film		
0.04 ^[13]	0.05	0.40	0.30
Pd bridge CO $I_{\text{ssp}}/I_{\text{ppp}}$			
Single crystal	6.1 nm NPs faceted	5.9 nm NPs rough	3.6 nm NPs rough
0.02 ^[13] sim.	0.20	0.30	0.49
0.10 ^[14,15] exp.			

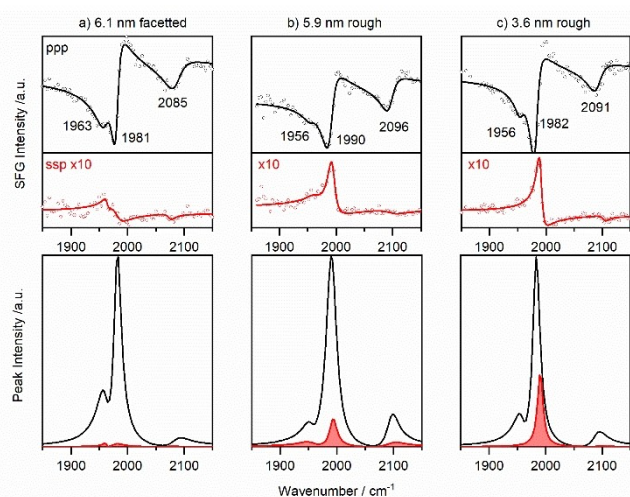


Figure 4. Polarization-dependent SFG spectra of bridge bonded and on-top CO on Pd/Al₂O₃ of various particle size and roughness. 6.1 nm particles were prepared at 300 K, 5.9 and 3.6 nm particles at 90 K. Measurements were performed in UHV at 200 K after saturating the surface with CO. The red spectra represent the ssp polarization combination, which is enlarged by a factor of ten with respect to the ppp spectra. SSP spectra are comparably stronger for curved particles. The bottom row shows the normalized spectral fits of the resonant signal without the non-resonant background.

crystal Pd(111), while the weak on-top peaks were not included for Pd.

As described above, two distinct Pd particle shapes/morphologies were present. These shapes govern the I_{ssp}/I_{ppp} ratio. The 6.1 nm Pd particles grown at 300 K are well-faceted truncated cuboctahedra^[31,32,34,35] with a size/height aspect ratio of approximately 3. The “large” planar (111) top facets lead to a strong ppp signal, whereas the ssp signal of the smaller inclined side facets is weak. Accordingly, I_{ssp}/I_{ppp} is quite small (0.2), approaching that of Pd(111) (0.1). The 5.9 nm Pd NPs prepared at 90 K have a similar mean diameter but have rougher and more pronounced side facets. As a result, the value of I_{ssp}/I_{ppp} of 0.3 is higher. In comparison, the 3.6 nm Pd particles were also grown at 90 K and were thus rough. With their even smaller size, the I_{ssp}/I_{ppp} intensity ratio is even larger (0.49).

In summary, polarization-dependent SFG measurements were carried out for two model catalyst systems, consisting of different metal NPs (Pt vs. Pd), deposited by different methods (ALD vs. PVD), on different support materials (ZrO₂ vs. Al₂O₃), with CO preferentially adsorbed on different binding sites (on-top vs. bridge), at different CO pressures (10 mbar vs. UHV) and different temperatures (425 vs. 200 K). In both cases, in addition to the typically evaluated peak positions and intensities, the polarization-dependent SFG measurements yielded I_{ssp}/I_{ppp} ratios that reflect the particle morphology/surface curvature, in line with microscopic characterization. Based on this agreement, polarization-dependent SFG spectroscopy can be applied for in situ characterization of particle morphology and especially changes thereof, even though SFG is usually not

used for shape characterization. Note that this morphology evaluation is different from that of surface roughness, which is directly evident from shifts in the CO resonance position. Especially as an in situ spectroscopic technique, polarization-dependent SFG allows observing changes upon treatments or during catalytic reactions (faceting, roughening, sintering etc.), while at the same time monitoring the reaction adsorbates/intermediates. This holds true for reactions that either involve CO or which are not affected by adsorbed CO. Herein, steady-state spectra were acquired, but shorter acquisition times could be obtained by limiting the spectral range, automatic turning of polarizers or even broadband SFG. This presented approach may thus be utilized to characterize the morphology of model catalyst NPs during preparation, pretreatment and catalytic reactions.

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Conflict of Interest

The authors declare no conflict of interest.

Data Availability Statement

The data that support the findings of this study are available from the corresponding author upon reasonable request.

Keywords: Metal Nanoparticles · Particle Morphology · Sum Frequency Generation · Vibrational Spectroscopy · in Situ Spectroscopy

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