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Stable isotopes show *Homo sapiens* dispersed into cold steppes ~45,000 years ago at Ilsenhöhle in Ranis, Germany

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Table of contents

1	Site description 1.1 Excavations and stratigraphy of the lower sequence	2 2
2	 Oxygen stable isotope palaeoclimatology 2.1 Oxygen palaeoclimatology background	6 6 10
3	Strontium isotopes and animal spatial ecology	13
4	Zinc stable isotopes4.1Principles of zinc stable isotope variability4.2Impacts of bedrock lithology on zinc stable isotopes in Ranis4.3Zinc stable isotopes and herbivore feeding ecology	16 16 18 20
5	 Extended methods 5.1 Tooth enamel sampling	23 23 24 24 24 25 27 29 31 31 33 35
6	Supplementary Figures	35
7	Supplementary Tables	51
Re	References	

Other Supplementary Material for this manuscript includes the following:

Data and code to reproduce the manuscript files, figures and analyses are available at

https://osf.io/wunfd/

1 Site description

1.1 Excavations and stratigraphy of the lower sequence

Ilsenhöhle in Ranis (50°39.7563'N, 11°33.9139'E, hereafter: Ranis) is a cave site located in the Orla river valley close to the city of Pößneck in eastern Thuringia, central Germany (Extended Data Figure 1). The cave system originally consisted of a large chamber with a south-facing entrance and two side chambers in a Permian limestone reef, but several roof collapses during the Pleistocene have left only two short sections of the side chambers preserved to the modern day. Above the cave, the top of the limestone rock houses a Medieval hill castle, Burg Ranis, built in the 11th century. Several excavation campaigns were conducted at Ranis in the 1920s and 1930s, first by then-owner of Burg Ranis, Dietrich von Breitenbuch, in 1926, 1929, 1931 and 1932, followed by Werner Hülle from the archaeological state office in Halle (Saale) who extensively excavated from 1932–1938¹. Finally, re-excavations to document the site using modern excavation methods followed from 2016–2022 by the Thuringian State Office for the Preservation of Historical Monuments and Archaeology (TLDA) and the Department of Human Evolution, Max-Planck-Institute for Evolutionary Anthropology, Leipzig (MPI-EVA).

The early excavations by von Breitenbuch and Hülle covered more than 200 m² and removed most of the in-situ deposits including all material in the two side chambers and large parts of what would have been the main chamber¹. Uncovered deposits included Medieval, Iron Age, Mesolithic, Upper Palaeolithic, transitional, and Middle Palaeolithic layers. The Palaeolithic archaeological layers were numbered III to XI and described as covering an age range from the Middle Palaeolithic to the Late Glacial. Relevant here is the lower part of the Pleistocene sequence including Layer VII Upper Brown Layer ("Obere Braune Schicht"), Layer VIII Black Layer ("Schwarze Fundschicht"), Layer IX Middle Brown Layer ("Mittlere Braune Schicht"), Layer X Grey Layer ("Graue Schicht"), and Layer XI Lower Brown Layer ("Untere Braune Schicht") (Extended Data Figure 1). Of particular interest to this study is the Grey Layer (Layer X, "Graue Schicht") in the lower part of the sequence, which yielded an assemblage of 23 bifacial leaf points and 17 Jerzmanowice blade points, subsequently used to define the 'Ranisian' as part of the Middle to Upper Palaeolithic transitional technocomplex of the Lincombian-Ranisian-Jerzmanowician (LRJ)^{2,3}.

The excavation area of the 1930s was divided into 1 x 1 m squares and finds recorded accordingly with square and approximate depth information, but excavations were conducted using heavy duty tools and finds were mostly pooled into rough stratigraphic units without regular depth recording for individual specimens. While this is not uncommon for excavations in the 1930s, it poses challenges for assigning stratigraphic context of finds and the faunal collection additionally suffers from a lack of clarity and inconsistencies in the context documentation of finds. Boxes of faunal remains were commonly labelled with square, layer colour and approximate depth extent of the layer but without the layer numbering system that seems to have been implemented after excavations had been completed. Unfortunately, this often makes layer assignments very challenging, as different layers can have the same colour description (e.g., "braun" meaning "brown"). Conversely, different colour descriptions (i.e., chocolate brown, red brown) have been used for what appears to refer to the same stratigraphic unit, likely representing natural colour changes between different areas of the site. Further, depth labels are in many cases of limited utility in assigning stratigraphic provenance due to the sloped deposits and compression of layers due to rock fall in some areas. Finally, a number of inconsistencies in the documentation of stratigraphic provenance have been documented with mismatches between box labels, notes in the site monograph and original excavation documentation kept in the Museum für Vorgeschichte in Halle⁴. Clearly, the documentation of the collection suffered substantially from the historical events accompanying its curation, with original documentation lost in part, an inability of W. Hülle to revisit the collection while writing the publication manuscript and finally his unexpected death culminating in a posthumous publication that is not always consistent with his original excavation diaries⁴.

The need to clarify the stratigraphy of this important site and establish the makers of the LRJ prompted a re-excavation of the site in a small area to the southwest of the earlier excavations conducted by the TLDA/MPI-EVA starting in 2016⁵. The excavation area was placed immediately adjacent to and partially overlapping with trenches from the Hülle excavation and established a more detailed and more subdivided, but clearly correlated, stratigraphic sequence⁵ (Extended Data Figure 1). In this new TLDA/MPI-EVA excavation the relevant layers include Layer 6-black (Upper Palaeolithic), Layer 7 (low number of artefacts; all from the contact zone with Layer 8, likely originating from Layer 8), Layers 8 and 9 (LRJ), Layers 10–12 (undiagnostic, potentially Middle Palaeolithic) and Layers 13–14 (archaeologically sterile limestone detritus). Notably Layer 6-black is clearly equivalent with the black Layer VIII, while Layers 8 and 9 represent the LRJ occupations corresponding to Hülle's grey Layer X, with the main LRJ occupation in Layer 8. In the TLDA/MPI-EVA excavation, a large rock of 1.7 m depth originating from a roof collapse event separated Layer 7 and 6 and largely displaced the equivalent of Hülle's middle brown Layer IX. Several large rocks from such roof collapse events restricted the size of the excavation in the lower layers (7 and below) to 1 complete, 2 half and 1 quarter squares, down from 18 squares at the surface of the 8 m deep sequence. However, the rockfall event sealing Layer 7 also provides ideal preservation conditions of the lower sequence in the area of the TLDA/MPI-EVA excavation, as the lower depositions were effectively protected from post-depositional disturbance.

Based on the newly produced radiocarbon chronology of the lower sequence of the site, the LRJ Layers 9 and 8 now date to 47,500–45,820 cal BP and 46,820–43,260 cal BP, respectively (modelled age ranges, 95% probability)⁵. The overlying Layer 7 dates to 45,890–39,110 cal BP while at the

4

base of the archaeological sequence Layer 11 has an age range of 55,860–48,710 cal BP⁵.

Newly discovered human remains now link the LRJ at Ranis to *Homo sapiens* as the maker of this technocomplex⁵. A total of 13 *H. sapiens* skeletal fragments were identified through proteomic screening and in some cases through morphological analysis of previously unidentified small fragments and their taxonomic identity was confirmed using ancient DNA analysis. In the TLDA/MPI-EVA excavation three fragments were discovered in Layer 8 and one in Layer 9 (Figure 3). Additionally, nine fragments were recovered from the Hülle collection in material attributed to Layers IX, X, and XI (Figure 3). Excavation documentation shows that despite some attributions to Layer IX and XI, the fragments were recovered on the same day or within a day of recovery of LRJ artefacts in the same squares and are, therefore, most likely associated with the LRJ and were misassigned due to the excavation methodology⁵. This is confirmed by direct radiocarbon dates that were obtained for 6 of 9 *Homo sapiens* skeletal fragments from the Hülle collection, which all fall into the age range of Layers 8 and 9 of the TLDA/MPI-EVA excavation⁵.

The faunal spectrum of the lower sequence, including the LRJ layers, is dominated by reindeer (*Rangifer tarandus*) and includes additional cold-adapted species such as wolverine (*Gulo gulo*), woolly mammoth (*Mammuthus primigenius*), woolly rhinoceros (*Coelondonta antiquitatis*) and arctic fox (*Alopex sp.*)⁶. Some species with a wide range of climatic flexibility including warmer environments, such as red deer (*Cervus elaphus*), are present, but in low numbers. In Layers 8 and 9 (LRJ), cave bear (*Ursus spelaeus*) remains are also abundant. Overall, the faunal composition indicates an accumulation in a cold steppe/tundra environment⁶. Evidence from zooarchaeological, palaeoproteomic and sediment aDNA analysis shows that human presence at the site fluctuated, with lowest human impact in Layer 7 and 10, while human modifications such as marrow fractures and cut marks are comparatively most abundant in the LRJ Layers 8 and 9, but still sparse (3.7–4.3 %)⁶. Overall, bone modifications from carnivores including gnaw marks and stomach

acid etching are dominant throughout the sequence⁶. Low numbers of hyena skeletal remains combined with high proportions of hyena (*Crocuta spelaea*) aDNA in sediments and abundant coprolite fragments suggest a use of the site as a hyena den for at least part of the site formation⁶. Human occupations of the site during the LRJ, but also the remaining lower sequence, appear to have been of low intensity either by small groups or through short or infrequent visits⁶. The relatively low density of stone tools in the LRJ layers supports this hypothesis⁵.

2 Oxygen stable isotope palaeoclimatology

2.1 Oxygen palaeoclimatology background

The use of oxygen stable isotope measurements (δ^{18} O) of tooth enamel of terrestrial mammals as a palaeoclimatic proxy is based on two fundamental principles. 1) There exists a strong linear relationship between air temperature and δ^{18} O in precipitation ($\delta^{18}O_{\text{precip}}$) and rain-fed water sources in mid- to high latitudes⁷⁻¹², where low δ^{18} O values correspond to low temperatures and high δ^{18} O values represent higher temperatures. 2) Animals that regularly drink large amounts of surface water from precipitation-fed water sources (such as aurochs/bison, horses, or mammoths) record the oxygen isotope composition of drinking water in the oxygen isotope composition of their tooth enamel¹³⁻¹⁶. In conjunction, these two relationships enable the reconstruction of palaeotemperatures using δ^{18} O values of faunal skeletal remains.

The isotopic composition of surface waters that are available for drinking is determined by multiple interwoven effects, but the oxygen isotope composition of precipitation ($\delta^{18}O_{precip}$) at mid- to high latitude locations generally shows a strong dependence on air temperature and this characteristic is inherited by precipitation-fed surface waters^{7-12,17}. This temperature effect fundamentally establishes the role of $\delta^{18}O_{enamel}$ as a proxy for palaeotemperatures at mid- to high latitude

sites, where higher δ^{18} O in tooth enamel indicates higher air temperatures and low δ^{18} O values indicate lower temperatures. This holds true for geological time-scales of global climatic change as well as across annual seasonal cycles^{9,10}. Oxygen isotope variation in precipitation and derived surface waters has been shown to be well-reflected in homeothermic mammals regularly consuming large amounts of liquid water, as drinking water δ^{18} O (δ^{18} O_{dw}) is the most dominant driver of body water, and therefore, enamel δ^{18} O variability in such animals¹³⁻¹⁶. The specific linear relationship between enamel δ^{18} O and drinking water δ^{18} O is species-specific due to the influence of factors such as metabolic rate and body temperature¹³.

Not only is $\delta^{18}O_{enamel}$ of such obligate drinking animals tied to $\delta^{18}O_{precip}$ and temperature conditions generally, $\delta^{18}O$ measurements of serially-sampled enamel can even be used to obtain sub-annually resolved and time-dependent information. Temperature-driven seasonal $\delta^{18}O$ fluctuations in precipitation are recorded as a sinusoidal pattern of $\delta^{18}O$ in sequential tooth enamel samples of animal teeth, with peaks indicating the summer season and troughs the winter season^{18,19}. Extracting such time-dependent $\delta^{18}O$ patterns is possible due to the incremental formation of tooth enamel from the crown to the root-enamel-junction, after which the tissue does not remodel^{20,21}. The stable isotope time series that can be obtained from a single tooth, therefore, covers the time of tooth enamel formation, which is usually 1.5 to >2 years in equids, depending on the tooth type and extent of tooth wear^{18,22}. In this study, we measure $\delta^{18}O$ of sequentially sampled tooth enamel using the phosphate moiety of bioapatite ($\delta^{18}O_{phos}$), which is more resistant to diagenetic change than the carbonate fraction²³ and allows higher precision of palaeotemperature estimation as most modern calibration data has been measured using bioapatite phosphate²⁴.

In addition to comparing oxygen isotope measurements with existing data to make inferences about palaeotemperature conditions, we also use these measurements to quantitatively estimate past temperatures. Oxygen isotope values can be compared directly with measurements generated in a comparable manner on remains of the same species to draw inferences on relative differences in seasonality (i.e. summer to winter differences) or temperature (warmer vs colder, more seasonal vs less seasonal). Such direct comparisons introduce the least amount of uncertainty, particularly if sampling approach and tooth types are also consistent. However, few δ^{18} O data from modern contexts or from Pleistocene sites are available for such comparisons, making this approach often insufficient for characterising past climates and contextualising obtained data.

Thus, a conversion to palaeotemperature estimates is almost always necessary to enable comparison with other species or data from other palaeoclimatic proxies, despite the known uncertainties that are involved. Palaeotemperature conversions are derived using formalised versions of the linear relationships between temperature and $\delta^{18} {
m O}_{
m precip}$, as well as between $\delta^{18} {
m O}_{
m dw}$ and $\delta^{18}O_{enamel}$ and employ error propagation methods to determine estimation uncertainty^{14,15,24-29}. Regression equations of these linear relationships are established using modern calibration data from animals and drinking water as well as $\delta^{18} O_{\rm precip}$ and air temperature measurements from measurement stations (see Supplementary Text 5). These relationships are well documented in modern calibration data, but their application to archaeological settings to reconstruct past climates relies on a number of assumptions that we discuss briefly here. Firstly, the exact numerical relationship between environmental water δ^{18} O and air temperature needs to be appropriately similar between modern and archaeological settings. This is to a certain degree dependent on the time stability of atmospheric circulation systems due to the role of moisture transport history. A sufficient degree of circulation similarity, and therefore stability of the slope of the $\delta^{18}O_{\text{precip}}$ air temperature relationship, between modern-day and the Late Pleistocene in Europe has been shown using circulation models as well as by comparisons of palaeotemperature estimates with δ^{18} O measurements of Pleistocene groundwaters^{25,28,30-32}. Additionally, geospatial patterns in $\delta^{18}O_{enamel}$ in Europe in the Late Pleistocene reproduce current spatial patterns of west-east decrease in $\delta^{18}O_{precip}$ due to progressive rainout across the European continent from a westerly moisture source, further supporting a similarity in broad atmospheric circulation patterns³³. Additionally, the $\delta^{18}O$ value of seawater influences $\delta^{18}O_{precip}$ and is known to change through time due to changes in global ice-volume³⁴. During MIS 3 $\delta^{18}O$ of seawater was ~0.5 ‰ higher than it is today³⁴, which is a very small change in relation to the $\delta^{18}O$ difference between the values measured in this study and modern-day $\delta^{18}O$ values (in fauna or precipitation). At most, this effect would lead to a slight temperature over-estimation, and we argue that it is negligible in regard to this study.

Secondly, the relationship between $\delta^{18}O_{enamel}$ and $\delta^{18}O_{dw}$ needs to be stable over time to justify the use in palaeotemperature estimations. This relationship could change over time if 1) drinking requirements/behaviour changed dramatically, e.g. if an obligate-drinking species became non-water dependent, or 2) if metabolic characteristics such as metabolic rate or body temperature changed over time. There is no evidence to suggest that pronounced metabolic changes with the potential to impact species-specific offsets between $\delta^{18}O_{enamel}$ and $\delta^{18}O_{dw}$ or change drinking requirements have taken place in equids across the relatively short time scales since the Late Pleistocene³⁵⁻⁴². We therefore consider that this assumption is valid. To further validate that study species $\delta^{18}O_{dw}$ reflects past $\delta^{18}O_{precip}$, and therefore climatic impacts, some studies recommend the use of multiple species for this analysis to check if any influence of drinking behaviour impacts $\delta^{18}O$ and to mitigate the uncertainties involved in estimating $\delta^{18}O_{dw}$ from $\delta^{18}O_{enamel}^{e.g. 24,43,44}$. Unfortunately this was not possible in this study due to limitations of the available faunal material. Other than equids, the Ranis faunal collection contains almost no herbivore macrommal remains from species with large drinking requirements and teeth suitable for sequential sampling. Suitable wolly mammoth, and large bovid (aurochs/bison) teeth are present in the collection, but such low numbers (< 5 in the whole collection) that analysis of these specimens would not yield meaningful information. Woolly rhinoceros teeth are present, but due to lack of information on tooth minerialization of this taxon it is not suitable for sequential sampling and would introduce unknown amounts of seasonal bias even into bulk samples. Similarly, Cervid teeth are more abundant but Cervid teeth do not record full annual cycles in their teeth and are therefore unsuitable for seasonal climatic reconstructions and can introduce seasonal bias into bulk samples unless pairs of second and third molars articulated in mandibles are available⁴⁵, which is not the case for the Ranis collection. However, $\delta^{18}O_{dw}$ values of equids have been shown to be in good agreement with those from a range of other taxa^{43,46-48} and we have investigated in detail whether diet or drinking behaviour could have influenced equid $\delta^{18}O_{dw}$ in section 2.2.

Other palaeotemperature estimation prerequisites should be validated for each study area. This includes the assumption that drinking water sources used by the analysed animals are predominantly fed by local precipitation and are isotopically closely tied to it. The role of aridity (leading to evaporation) in affecting δ^{18} O of potential water sources should also be checked for each study setting. These are discussed further in the following section.

2.2 Hydrotopographical setting and non-temperature effects on oxygen stable isotopes

For each study setting it needs to be established that δ^{18} O values of surface water sources in the region sufficiently reflect $\delta^{18}O_{\text{precip}}$, as precipitation measurements are used in the modern calibration data sets employed during palaeotemperature estimation. Indeed, the majority of open water source types in any given area normally relatively closely reflect $\delta^{18}O_{\text{precip}}^{17}$. However, a

few specific water source types can isotopically deviate, sometimes substantially, from meteoric water. This particularly concerns deep groundwaters, large rivers or lakes, and glacial/snow melt due to effects of water transport, water residence time, evaporation, and selective incorporation of cold climate/winter season and/or high altitude precipitation^{49–51}. Substantial water consumption from highly evaporated lakes would cause animals to exhibit high $\delta^{18}O_{enamel}$ values, which is in contrast to the uncommonly low oxygen isotope values of the animals from Ranis measured in this study. Combined with the absence of large lakes in the vicinity of the site or evidence of such from the Late Pleistocene, substantial consumption of evaporated lake water is unlikely. This leaves us to consider if large rivers, deep groundwater or glacial melt water are relevant.

Uncommonly low δ^{18} O values could be caused by consumption of glacial melt water or perennial snow, water transported from high latitudes or high altitudes, or through consumption of deep groundwater formed during colder climates. During MIS 3, or even MIS 4, both the Fennoscandian Ice Sheet and the Alpine glaciers were at least several hundred kilometres distant from Ranis and transport of glacial melt to the region by rivers is blocked by the regional mountain ranges⁵². Water transport from high elevations is similarly unlikely. The river most likely to transport higher elevation waters to the site vicinity is the Saale River, with a maximum elevation of 856 m in its catchment⁵³. With an elevation difference of ~500 m compared to the site location, the water transported by the Saaler River would only be maximum ~0.5 ‰ higher than local $\delta^{18}O_{\text{precip}}$, and $\delta^{18}O_{\text{precip}}$ (Extended Data Figure 3). Generally, the mountain ranges surrounding the site do not exceed elevations of 1000 m a.s.l. so a strong impact from high-elevation precipitation on locally available water sources is unlikely. Finally, a consumption of deep groundwater is possible. We argue that this is also unlikely, as such groundwaters rarely feed surface springs as these aquifers are normally relatively far below the surface. Additionally, such water sources

are by definition heavily time-averaged and show essentially no seasonal variability. Regular consumption of groundwaters would therefore cause an absence of a seasonal sinusoidal δ^{18} O signal in tooth enamel, which we do not observe in the equids studied here. A similar case can be made for potential consumption of perennial snow. While this could well have existed in the land-scape around the site, snow consumption should only substantially distort seasonal δ^{18} O signals if consumed outside the winter season. However, this should logically also produce a noticeably damped seasonal δ^{18} O that is incompatible with the observed data. Overall, we therefore argue the available water sources around Ranis should reflect δ^{18} O_{precip} reasonably well.

Finally, it should be considered whether δ^{18} O changes over time that we observe in our data could be driven by non-temperature effects on $\delta^{18}O_{\text{precip}}$. As we have already considered circulationrelated effects, aridity/rainfall amount remain as a final potentially substantial driver of δ^{18} O. Changes in δ^{15} N throughout the Ranis sequence suggest that aridity or changes in soil water availability could play a role, but negative correlation between $\delta^{15}N$ and $\delta^{18}O$ demonstrate that δ^{18} O values are not affected by this, as high aridity should produce high δ^{18} O during high δ^{15} N phases. At the same time, δ^{18} O values in this study are remarkably low, which is more consistent with low temperature and low-evaporation environments. If evaporative enrichment played a substantial role in influencing δ^{18} O, this would mean that temperatures were even lower than reconstructed. As evaporative enrichment even at the reconstructed temperatures would have been extremely limited, it seems unlikely that even lower temperatures could have been accompanied by sufficient evaporative enrichment to elevate δ^{18} O through aridity effects and cause a temperature over-estimation. Conversely, lowering of δ^{18} O through high rainfall amounts is also inconsistent with the high δ^{15} N values during times of particularly low δ^{18} O, as well as with the vast majority of palaeoclimatic reconstructions from other proxies all indicating much drier conditions for MIS 3 stadials^{54,55}.

At the same time, the decline in total organic carbon and total nitrogen content of sediments from Layer 9 to 7 of the Ranis sequence supports that the co-occurring drop in δ^{18} O that we observe is driven by a temperature decline⁵ (see main text for discussion). Additionally, particularly this δ^{18} O decline is congruent with climate cooling associated with Greenland Stadial 12 documented in pollen sequences, geochemical records, and earthworm calcite δ^{18} O records from west and southwest Germany, further supporting an interpretation of our δ^{18} O data in relation to temperature change^{54–57}. Therefore, an interpretation as a palaeotemperature proxy is most logical.

3 Strontium isotopes and animal spatial ecology

The strontium isotope ratio (⁸⁷Sr/⁸⁶Sr) in tooth enamel of terrestrial animals ultimately reflects ⁸⁷Sr/⁸⁶Sr of bioavailable Sr from underlying bedrock and derived soil⁵⁸. The ⁸⁷Sr/⁸⁶Sr ratios of different lithologies are in turn driven by bedrock age and the ⁸⁷Rb content at the time of formation⁵⁹. This is because ⁸⁷Sr is a radiogenic isotope that is produced through the decay of ⁸⁷Rb over time⁵⁹. This mechanism means that older bedrocks typically tend to exhibit higher ⁸⁷Sr/⁸⁶Sr than younger rocks (of similar lithology). Geogenic ⁸⁷Sr/⁸⁶Sr of bedrock is subsequently transferred to soils through bedrock erosion/weathering and bioavailable strontium is taken up from the soil by plants and transferred through the food web from herbivores to omnivores and carnivores^{60,61}. Isotopic fractionation of Sr from biological processes is negligible and, consequently, ⁸⁷Sr/⁸⁶Sr effectively does not change from plants to herbivores or further up the trophic chain^{61,62}. Thus, ⁸⁷Sr/⁸⁶Sr in tooth enamel represents a record of the lithologies that animals range over and can be used to trace animal movements and characterise their spatial ecology⁵⁸. Mobility or place-of-origin information can be extracted either from bulk samples of enamel, representing the average ⁸⁷Sr/⁸⁶Sr composition incorporated throughout enamel formation,

or from sequential samples to extract a time-dependent ⁸⁷Sr/⁸⁶Sr measurement series⁶³. As animal migrations most commonly occur seasonally with movements between summer and winter range locations, a comparison between a summer and a winter sample can be used as a check for long-distance or altitudinal migrations⁶⁴. This approach offers lower resolution than full sequential ⁸⁷Sr/⁸⁶Sr profiles but is useful as a control to detect any migration-induced biases of δ^{18} O. As geospatial patterns in δ^{18} O are characterised by low rate-of-change gradients over large distances or altitudinal gradients, this approach is normally sufficient to detect migratory behaviours that would be relevant from an oxygen isotope perspective.

We use this summer/winter comparison approach on sequentially sampled equid specimens to confirm that δ^{18} O values are representative of local conditions without bias from long-distance migrations. For other fauna, bulk samples were analysed for ⁸⁷Sr/⁸⁶Sr to evaluate the relationship between 87 Sr/ 86 Sr and δ^{66} Zn and establish any potential influence of bedrock lithology on δ^{66} Zn in our study setting (see Supplementary Text 4). The ⁸⁷Sr/⁸⁶Sr ratios of equids in our study range from 0.7090 to 0.7120. Most specimens fall into the higher end of this value range, between ~ 0.710–0.712 (Supplementary Figure 3). These values match ⁸⁷Sr/⁸⁶Sr ratios reported for lithologies in Thuringia. A study of bioavailable strontium in northern Thuringia reported ⁸⁷Sr/⁸⁶Sr ratios in tree leaves ranging from 0.70817 in Triassic carbonates (Muschelkalk) to 0.71116 in Triassic sandstones (Buntsandstein)⁶⁵. South-eastern Thuringia also contains several granitic geological units with high 87 Sr/ 86 Sr values up to ~ 0.730, some of which are located within 50 km of Ranis 66 , and this variability is also shown on a predicted strontium isoscape (Supplementary Figure 4). A study of surface water samples further reported ⁸⁷Sr/⁸⁶Sr ranging from 0.7091 to 0.71381 at different points along the upper course of the Saale River through the Thuringian-Franconian Slate Belt⁶⁷, where this river passes before running close to the site of Ranis. The ⁸⁷Sr/⁸⁶Sr value range of equids is additionally fully encompassed by values observed in hyenas and ursids (0.7083 to

0.7133) - taxa that have documented small to modest range sizes in Late Pleistocene and modern case studies, particularly in the case of cave bears^{68–70} (Supplementary Figure 3). The ⁸⁷Sr/⁸⁶Sr of equids is therefore completely encompassed by measured ⁸⁷Sr/⁸⁶Sr of plants and water samples, predicted bioavailable ⁸⁷Sr/⁸⁶Sr and ⁸⁷Sr/⁸⁶Sr of animals with likely small home range. This means that equid ⁸⁷Sr/⁸⁶Sr are consistent with a non-migratory ranging behaviour and a presumable habitat within a few tens of kilometres of Ranis. The relatively large range of ⁸⁷Sr/⁸⁶Sr values represented within the complex geology of Thuringia and in our equid samples means that there likely is significant overlap with lithologies further away from the site and comparisons of ⁸⁷Sr/⁸⁶Sr values between animals and lithologies cannot fully exclude a non-local origin. However, to our knowledge no spatial ecology study of Eurasian Late Pleistocene equids has so far found any evidence for seasonal long-distance migrations^{71–73}. Furthermore, summer and winter ⁸⁷Sr/⁸⁶Sr of the Ranis equids are very similar with seasonal ⁸⁷Sr/⁸⁶Sr differences < 0.0005 in almost all cases (Supplementary Figure 3) indicating that summer and winter ranges were located on similar lithologies. Therefore, a non-migratory spatial ecology is the most parsimonious for these equid specimens. One exception, specimen R10131, shows a larger seasonal difference in 87 Sr/ 86 Sr, accompanied by an unusually larger seasonal difference in δ^{66} Zn, which could also be affected by bedrock lithology to some extent (Supplementary Figure 6; Supplementary Text 3). The ⁸⁷Sr/⁸⁶Sr ratios of this individual still fall within the range of values observed in other equids and predators with typicall small home ranges as well as eastern Thuringian bedrock types (Supplementary Figure 3). We therefore argue that long-distance migration is still fairly unlikely. However, due to the larger seasonal ⁸⁷Sr/⁸⁶Sr change, we cannot fully exclude a more migratory ecology and thus exclude this specimen from climatic interpretations of δ^{18} O and dietary interpretations of δ^{66} Zn.

4 Zinc stable isotopes

4.1 Principles of zinc stable isotope variability

Zinc stable isotope analysis falls into the group of non-traditional stable isotope analyses and has been established fairly recently as a tracer of diet and ecology of terrestrial and marine vertebrates^{74–81}. Zinc is a transition metal but due to it's full d-shell (10 electrons in the d-shell), it is in many respects chemically similar to alkaline earth metals such as magnesium. Zn has five naturally occurring stable isotopes and measurements of several different isotope ratios have been conducted with ⁶⁶Zn/⁶⁴Zn, expressed as δ^{66} Zn, being the most commonly used and most established.

Zinc is an essential trace element occurring in low abundances, but it serves important biological functions where it forms metal organic compounds that are involved in a number of enzymes and regulatory proteins⁸². Indeed, zinc availability in soils is one of the most important limiting factors for plant growth due to its essential role in cell wall formation⁸³. Zinc stable isotope variability in terrestrial food webs reflects both baseline variability in soils and plants as the ultimate source of zinc introduced to the animal biome, as well as biological fractionation within the food web⁸⁴⁻⁸⁷. In soils, δ^{66} Zn variability mainly is driven by the lithology of the underlying bedrock, with igneous rocks exhibiting very uniform δ^{66} Zn values of 0.3 ± 0.14 ‰, while sedimentary rocks show a wider range of variability from ~-0.5 to 1.5 ‰ depending on their formation history^{88,89}. Highest δ^{66} Zn values of up to 1.4 ‰ occur in marine carbonates, as these biogenic lithologies are formed from calcifying marine organisms whose tissues exhibit higher δ^{66} Zn values due to biological isotopic fractionation⁹⁰. Additional zinc isotope variability in soils may be introduced by the source of organic matter in the soil as well as zinc biogeochemical cycling^{87,88}. Plants take up zinc from the soil, with initial fractionation from soil to roots leading

to higher δ^{66} Zn values in roots^{84–86}. Further fractionation occurs in plants during zinc transport, as cells appear to preferentially take up the heavier Zn isotopes while the lighter isotopologues are preferentially transported to the more aerial or peripheral parts of the plant^{84–86}. This results in an isotopic differentiation between plant parts as well as between different plant types. Within a single plant upper plant parts such as tree leaves exhibit lower δ^{66} Zn values compared to stems and across plant species higher growing species such as shrubs or trees show lower δ^{66} Zn values compared to higher δ^{66} Zn in lower growing species such as grasses or herbaceous plants^{84–86}. Zinc isotope variability in the plant biome is subsequently passed on to herbivorous consumers and in turn from herbivores to predatory animals with further isotope fractionation in these organisms^{75,76,81}.

In animals, diverging δ^{66} Zn values are observed between different tissue types with soft tissues, such as muscle or kidney, showing lower δ^{66} Zn and hard tissues, such as bone, exhibit higher δ^{66} Zn^{81,91}. This effect is the basis of a trophic level effect in δ^{66} Zn, where δ^{66} Zn decreases along the trophic chain as predators consume soft tissues of prey organisms^{75,79,81}. In this context it needs to be kept in mind that bone generally exhibits higher δ^{66} Zn than soft tissue, which shifts δ^{66} Zn of bone-eating carnivores such as Hyaenidae towards higher δ^{66} Zn values normally observed in omnivores^{75,76,91}. Additionally, nursing and *in utero* formation has been shown to impact tooth enamel δ^{66} Zn values, where δ^{66} Zn values of teeth formed during nursing or *in utero* are higher than those formed post weaning⁹². This is often relevant especially for carnivora δ^{66} Zn values, as these animals tend to form many of their teeth during the nursing period, and timing of enamel formation is often not well studied.

Several studies have investigated the ability of tooth enamel and enameloid bioapatite to preserve biogenic δ^{66} Zn values over archaeological and geological time scales^{76,78,80,93}. These studies have demonstrated that δ^{66} Zn is preserved and that neither Zn substantial uptake nor leach-

17

ing occurs in these tissues including in tropical conditions and for up to tens of millions of years, into the Miocene^{76,80}. This has been achieved by evidencing the preservation of biogenic Zn concentration spatial profiles, relationship with other trace element concentration profiles, positive correlation with δ^{15} N data, a lack of relationship between isotopic and concentration data, comparisons with dentine samples, and reliable reproduction of known trophic level relationships and trophic level spacings in past food webs^{76,80,93}. As recommended in ref.⁷⁶ we use the relationship between zinc concentration and isotopic data as a diagenetic check, as this relationship should evidence a mixing line if enamel bioapatite has been subject to zinc leaching or uptake of exogenous zinc from the burial environment.

4.2 Impacts of bedrock lithology on zinc stable isotopes in Ranis

Studies of δ^{66} Zn in different food webs generally indicate that diet is by far the dominant driver of δ^{66} Zn. Nonetheless as a more minor effect, small baseline differences in δ^{66} Zn have been noted between some food webs related to differences in bedrock lithology⁷⁷. However, baseline differences generally appear to be insignificant if sites are located on similar geological units, as was shown in a comparison of two sites located ~150 km apart on karstic geology in Laos⁷⁷. At these sites, an inverse correlation between δ^{66} Zn and ⁸⁷Sr/⁸⁶Sr across all faunal specimens was observed, driven by the presence of high δ^{66} Zn/low ⁸⁷Sr/⁸⁶Sr carbonate lithology and low δ^{66} Zn/high ⁸⁷Sr/⁸⁶Sr granitic lithologies in the study area⁷⁷. In our data from Ranis, we observe a very weak inverse relationship between δ^{66} Zn and ⁸⁷Sr/⁸⁶Sr, which is statistically significant (as established by Pearson correlation), but appears mostly driven by two outliers and has a very small slow coefficient (Supplementary Figure 8). We therefore argue that δ^{66} Zn is predominantly driven by dietary groups in our data, with perhaps a very small influence related to ranging across different geological units including carbonate and granite lithologies present in the area (Sup-

plementary Figure 4, Supplementary Text 3). The range of ⁸⁷Sr/⁸⁶Sr observed in the food web is consistent with values reported for lithologies within a few tens of kilometres of Ranis, indicating that animals ranged over local landscapes and were unlikely to have undertaken long-distance migrations, precluding stronger non-dietary impacts on δ^{66} Zn (Supplementary Text 3). However, a single equid individual (R10131), exhibits an unusually large ⁸⁷Sr/⁸⁶Sr difference between summer and winter values (summer and winter samples determined using δ^{18} O data; Supplementary Figure 3). As discussed in Supplementary Text 3, while long-distance migration remains unlikely for this individual, the larger seasonal ⁸⁷Sr/⁸⁶Sr means that seasonal movement is a possibility for this specimen. Interestingly, this same specimen also exhibits by far the largest seasonal amplitude in δ^{66} Zn (Supplementary Figure 5, Supplementary Figure 6). Due to this co-occurrence of large seasonal amplitudes, we argue that movement across different geological types may have influenced seasonal δ^{66} Zn differences in this case, although further research into seasonal variation of δ^{66} Zn in herbivores is needed to further clarify this. If a movement across bedrock geologies influenced δ^{66} Zn in this individual, this means that it would have moved from a high δ^{66} Zn lithology in summer to a lower δ^{66} Zn lithology in winter, while ⁸⁷Sr/⁸⁶Sr values for both seasons are relatively high (⁸⁷Sr/⁸⁶Sr_{summer} = 0.7112; ⁸⁷Sr/⁸⁶Sr_{winter} = 0.7120; Supplementary Figure 5, Supplementary Figure 3). High ⁸⁷Sr/⁸⁶Sr values mean that a carbonate bedrock is very unlikely to be the cause of high δ^{66} Zn, as marine carbonates exhibit 87 Sr/ 86 Sr between ~0.707 and 0.709 94 . Conversely, granitic geologies can be excluded due to their uniformly low δ^{66} Zn. No δ^{66} Zn values are available for the lithologies around Ranis and data generally is sparse, making it difficult to determine what type of geology would match high δ^{66} Zn and high 87 Sr/ 86 Sr. As δ^{66} Zn has been reported to be very variable for sedimentary rocks, a high ⁸⁷Sr/⁸⁶Sr sandstone may for instance be a candidate. Alternatively, it is possible that such isotopic characteristics could match some of the shales found in Thuringia and Saxony, as high ⁸⁷Sr/⁸⁶Sr has been reported for some of these fine-grained siliclastic rocks and high δ^{66} Zn have been reported for black shales elsewhere^{95,96}.

More specific δ^{66} Zn data for sedimentary rocks from Germany would be needed to support these hypotheses. As such candidate geologies are still present within ~100 km of Ranis, we maintain that long distance migration of this individual remains fairly unlikely. However, as δ^{66} Zn could be influenced to some degree by movements across different bedrock types and thus potentially be biased relative to a purely dietary signal, we have chosen to exclude the R10131 specimen from interpretations of δ^{66} Zn in relation to dietary ecology. Similarly, we have not considered this individual for palaeoclimatic reconstructions using δ^{18} O, to avoid any bias related to geospatial variation of δ^{18} O.

4.3 Zinc stable isotopes and herbivore feeding ecology

Trophic level inference has so far been the focus of zinc stable isotope applications in the few archaeological and palaeontological applications that have been developed very recently^{76,77,80,93}. However, it is becoming increasingly clear that substantial isotopic variability exists within herbivores as well. In particular, some of the published zinc stable isotope data of terrestrial herbivores commonly show comparatively low δ^{66} Zn values in taxa associated with closed habitats and a stronger browse component in the diet, such as cervids^{75,76,93}. Higher δ^{66} Zn values on the other hand have been observed in equids, which are more commonly associated with grazing and open environment feeding^{75,93}. This pattern is also observed in the food web from Ranis studied here (Extended Data Figure 2). This may suggest that herbivores reflect the δ^{66} Zn patterning in plants where tree leaves show lower δ^{66} Zn than grasses or herbaceous plants^{84–86}. This would in turn also indicate that δ^{66} Zn diachronic changes within a single herbivore species over time may reflect dietary shifts along a grazer-browser spectrum. However, not all case studies conducted so far demonstrate a clear grazer/browser distinction in δ^{66} Zn. While a grazer/browser difference in δ^{66} Zn was clearly present in a modern eastern African food web⁷⁵ and a Late Pleistocene food web in Spain⁹³, for example, data from Late Pleistocene sites in Laos did not show a clear δ^{66} Zn separation of grazers and browser across a large number of herbivore taxa^{76,77}. In these case studies from Laos, dietary group was initially inferred via δ^{13} C, but the expected δ^{66} Zn pattern was only observed between certain grazers and browsers, while additional δ^{66} Zn clustering within browsers was shown⁷⁶. At this time, it remains unclear why grazer-browser patterning is observed in some settings but not others. The δ^{66} Zn groupings within herbivores do not seem to be related to digestive physiology (ruminants vs non-ruminants)^{76,77}, and the Ranis data further confirm this with non-ruminants falling both on the low (woolly rhinoceros) and high end (equids) of the herbivore δ^{66} Zn range (Extended Data Figure 2). Similarly, δ^{66} Zn variability in herbivores appears to be unrelated to body mass^{76,77}. A lack of true specialist grazers at the sites in Laos and a higher proportion of C3 (rather than C4) grasses in the landscape may obfuscate grazer/browser δ^{66} Zn patterns in those case studies if grazer/browser distinctions are initially based on δ^{13} C, but more well-controlled modern data or cross-comparisons with other data on grazing/browsing behaviour are needed to definitively clarify the drivers of δ^{66} Zn in herbivores.

In the Ranis data high δ^{66} Zn values in equids compared to traditionally browser/mixed feeder cervid taxa are consistent with a grazer/browser pattern in δ^{66} Zn. At the same time, the relationship between equid δ^{66} Zn and environmental and climatic inferences from the other stable isotope tracers could further support that equid δ^{66} Zn is consistent with such a pattern. In equid samples, highest δ^{66} Zn co-occur with lowest δ^{18} O values and highest δ^{15} N values at ~43-45 ka cal BP (Figure 1, excluding R10131 due to potential seasonal movement; see discussion of spatial ecology above). Exceptionally low δ^{18} O values combined with high δ^{15} N values during this time suggest cold temperature conditions and a grazing specialisation in equids or potentially the presence of dry soils (see main text). Combined with δ^{13} C values characteristic of an open environment the three stable isotope tracers all suggest an open grassland environment, matching data from pollen sequences from western and southwestern Germany⁵⁴⁻⁵⁶. An interpretation of high δ^{66} Zn values as being indicative of a grazer feeding ecology would thus fit well with the rest of the stable isotope data. Additionally, we observe a statistically significant correlation between δ^{66} Zn and δ^{15} N in equids (Supplementary Figure 12). Herbivore δ^{15} N in modern periarctic or past glacial/stadial ecosystems is commonly interpreted as either varying with grazer/browser diet or with bacterial activity in the soil, in turn driven by water availability, temperature and presence/absence of permafrost, and we argue here that a relationship with grazing specialisation or with soil water availability most likely explains the patterns in our data (see main text). A positive correlation of δ^{66} Zn and δ^{15} N would therefore also be consistent with a grazer/browser patterning in δ^{66} Zn. If true, this would in turn mean that the degree of summer/winter δ^{66} Zn difference would be linked to seasonal dietary changes. In our data set we don't observe a fixed seasonal δ^{66} Zn pattern, and summer δ^{66} Zn values are higher than winter in some cases, but lower in others (Supplementary Figure 5). However, we do observe that individuals with higher δ^{66} Zn values are more likely to exhibit very small summer/winter δ^{66} Zn differences. This could perhaps indicate that equids with a grazer feeding ecology were grazing year-round with a strong degree of specialisation, while more mixed-feeding equids showed a higher degree of seasonal dietary flexibility.

However, at this point a grazer/browser interpretation of δ^{66} Zn is not sufficiently demonstrated, as both tracers are also influenced by biogeochemical cycling in the soil (see main text and background discussion above). The precise environmental and climatic impacts on δ^{66} Zn variability in soils are poorly understood^{84,87}, but the known effects of biological processes and nutrient cycling on soil δ^{66} Zn mean that such impacts could potentially lead to a positive correlation of δ^{66} Zn and δ^{15} N in soils and plants.

Overall, we therefore conclude that our herbivore δ^{66} Zn data are tentatively consistent with a

grazer/browser patterning in δ^{66} Zn, but other explanatory mechanisms related to processes in the soil may play a role as well. Our data are not sufficient on their own to demonstrate or falsify the validity of either approach but may be useful as a foundation for further ground-truthing.

5 Extended methods

5.1 Tooth enamel sampling

5.1.1 Sequential powder sampling

Prior to enamel sampling, dental calculus was removed from the sampling surface by abrasion with a tungsten carbide burr and using air abrasion with Korox 99,6% aluminium oxide (50 µm mesh size). After mechanical cleaning, all teeth were cleaned in Milli-Q ultrapure water and left to dry before sampling. Tooth enamel samples were drilled with a diamond tipped dental drill bit as small strips perpendicular to the tooth growth axis spaced ~3mm apart in a series covering the complete tooth crown. Sample strips commonly had dimensions of 1 mm in cervico-occlusal length and 5 mm mesio-distal width and a depth between 0.5 and 1 mm, usually encompassing ~80% of the total enamel thickness. Drill bits were cleaned between samples using 1 M nitric acid and two 5-minute ultrasonication cycles in Milli-Q ultrapure water. Sampling was conducted in a positive pressure laminar flow box to avoid contamination and, thus, enable use of powder samples for strontium and zinc stable isotope analysis, in addition to oxygen stable isotope analysis.

5.1.2 Bulk piece sampling

When bulk sampling of an enamel piece was undertaken, the surface was first mechanically cleaned using a handheld drill equipped with a diamond-tipped burr to remove any adhering external material. Subsequently, enamel samples for each specimen were collected at the base of the crown using a diamond-tipped cutting wheel. The amount of material collected varied slightly due to differences in the enamel's relative thickness and the tooth's overall aspect but was typically around 10 mg. Dentine attached to the inner portion of the enamel was then mechanically removed using a handheld drill equipped with a diamond-tipped burr. As with sequential powder sampling, sampling was conducted in a positive pressure laminar flow box, and the cutting wheels and burrs were similarly cleaned between samples.

5.2 Oxygen stable isotope analysis

5.2.1 Silver phosphate preparation

Approximately 5 mg of tooth enamel powder of each sample were weighed into 2 mL Eppendorf microcentrifuge tubes. Samples were agitated in 0.4 mL of 2 M hydrofluoric acid (HF) for 24 hours to digest bioapatite and remove calcium from the solution as CaF₂. After digestion, samples were centrifuged (12000 rpm for 5 Min) to separate the phosphate containing solution from CaF₂ precipitate and the supernatant solution was then transferred to a clean microcentrifuge tube. To maximise phosphate recovery the CaF₂ precipitate was washed once with 0.1 mL MilliQ ultrapure water, and the wash added to the phosphate-containing solution. The sample solution was then titrated to neutrality as indicated by the colour change point of Bromothymol Blue indicator using 25% ammonia solution (NH₄OH). 60 μ L of NH₄OH were first added to each sample using an automatic pipette and each solution then slowly brought to the colour change point (yellow to

green) using individual drops of NH₄OH added with a 100 μ L Hamilton Microliter fixed needle syringe (Hamilton Bonaduz AG, Switzerland). From the neutralised solutions, Ag₃PO₄ was rapidly precipitated by addition of 0.4 mL 2 M silver nitrate (AgNO₃) solution. The resulting precipitate was separated from the supernatant solution by centrifugation (12000 rpm for 5 Min) and the remaining silver nitrate solution removed. The silver phosphate precipitate was then washed four times with 1 mL Milli-Q ultrapure water using centrifugation and vortex mixing steps between rinses to wash out any remaining silver nitrate from the sample. Silver phosphate samples were then dried overnight at 50 °C and stored in an evacuated desiccator until further analysis.

5.2.2 Oxygen stable isotope measurements

Phosphate oxygen isotope measurements of Ag_3PO_4 were conducted using a high temperature elemental analyser (TC/EA) coupled to a Delta V isotope ratio mass spectrometer via a Conflo IV interface (Thermo Fisher Scientific, Bremen, Germany) at the Max-Planck-Institute for Evolutionary Anthropology (MPI-EVA). Approximately 0.5 mg of each silver phosphate sample was weighed into cleaned silver capsules (3x4 mm, IVA Analysentechnik, Meerbusch, Germany) and introduced to the TC/EA using a Costech Zero Blank Autosampler (Costech International, Cernusco sul Naviglio, Italy). Conversion to CO was achieved using a reactor temperature of 1450 °C and gases separated using an Eurovector E11521 1.4 m x 4 mm x 6 mm stainless steel GC column with 80/100 mesh 5 Å molecular sieve packing (Eurovector Instruments & Software, Pavia, Italy) maintained at 120 °C with a carrier gas pressure of 1.3 bar. Samples were measured in triplicate, but in rare cases individual measurements were rejected if they did not meet quality control criteria such as appropriate peak area to sample amount relationship. In such cases δ^{18} O values are only based on two replicates.

Oxygen isotope delta values were two-point scale normalised to the VSMOW scale using matrix-

25

matched standards calibrated to international reference materials and scale normalisation was checked using three separate quality control standards. Scale normalisation was conducted using the B2207 silver phosphate standard ($\delta^{18}O = 21.7 \pm 0.3 \%$, 1 s.d.; Elemental Microanalysis, Okehampton, UK) and an in-house silver phosphate standard (KDHP.N, $\delta^{18}O = 4.2 \pm 0.3 \%$, 1 s.d.). This in-house standard was produced in 2017 by equilibrating a KH₂PO₄ solution made with Leipzig winter precipitation at ca. 140 °C for several days, after which the solution was neutralised using a small amount of NH₄OH and the phosphate precipitated as silver phosphate by addition of AgNO₃ solution. An equivalent method is described in detail in ref.⁹⁷. The accepted value of this in-house standard was determined by two-point calibration using B2207 and IAEA-SO-6 (barium sulphate, $\delta^{18}O = -11.35 \pm 0.3 \%$, 1 s.d.).

Aliquots of an in-house modern cow enamel standard (BRWE.2) and the standard reference material NIST SRM 120c (formerly NBS 120c) were precipitated and measured alongside each batch of samples to ensure equal treatment. Additionally, a commercially available silver phosphate (AS337382, Sigma Aldrich, Munich, Germany) was used as a third quality control standard to check across-run consistency of scale normalisation independent of silver phosphate precipitation. Measurements of these standards gave δ^{18} O values of 14.5 ± 0.4 ‰ for BRWE.2 (1 s.d., n = 72), 21.5 ± 0.7 ‰ for NIST SRM 120c (1 s.d., n = 85) and 13.8 ± 0.2 ‰ for AS337382 (1 s.d., n = 264). This compares well to the consensus value for NIST SRM 120c of 21.7 ‰, as well as the longterm averages for BRWE.2 of 14.5 ± 0.3 ‰ and 13.9 ± 0.2 ‰ for AS337382. Samples were usually measured in triplicate and average reproducibility of sample replicate measurements was 0.25 ‰. Consecutive analysis of sets of standards with widely spaced isotopic values showed no detectable memory effect and consequently no memory effect correction was required. No effect of the blank, of the sample amount, or peak height on the results was observed and consequently no blank correction or linearity correction was applied. If drift across the course of a run was detected in the quality control standard AS337382, a linear drift correction based on the drift of both normalisation standards was used and checked with the quality control standard.

5.3 Zinc and strontium stable isotope analysis

Zinc and strontium extraction were both conducted on ~10 mg of tooth enamel powder of samples that represent the seasonal summer and winter (determined by oxygen isotope measurements). A total of 56 samples were processed, a summer and winter sample from each of the 16 directly dated equid specimens (total of n = 32) and an additional 24 samples from mammalian herbivore, omnivore, and carnivore taxa (Supplementary Table 1). Samples were digested in 2 mL of 65% HNO₃ (Merck SupraPur grade, Merck, Darmstadt, Germany) for 2 hours at 120 °C and the solution then evaporated to dryness. The resulting residue was re-dissolved in 1 mL of 3N HNO₃ at 120 °C for 1 hour. The sample solution was then loaded onto a 0.5 cm bed of cleaned and pre-conditioned 50-100 μ m Sr-spec resin (EiChrom Technologies, Lisle, USA) placed in cleaned microcolumns. All sample solutions were passed through the column step 4 times to ensure complete loading of strontium onto the ion-exchange resin. Matrix elution was achieved using 3 times 400 μ L of 3N HNO₃, after which strontium was eluted from the columns using 1.5 mL of MilliQ ultrapure water. The resulting Sr containing solution was evaporated to dryness at 120 °C and re-dissolved in 1 mL of 3 % HNO₃. Two process blanks were prepared alongside samples to assess the level of contamination during the sample preparation procedure. Additionally, two aliquots of the NIST SRM 1486 bone meal standard were treated with the samples to serve as a quality control standard. All samples were analysed for ⁸⁷Sr/⁸⁶Sr using a Neptune Multi-Collector Inductively Coupled Plasma Mass Spectrometer (MC-ICPMS, Thermo Fisher Scientific, Bremen, Germany) at the MPI-EVA. ⁸⁷Sr/⁸⁶Sr measurements were normalised for instrumental mass bias to ⁸⁸Sr/⁸⁶Sr = 8.375209 (exponential law) and corrected for ⁸⁷Rb interference. External data normalisation was conducted using the NIST SRM 987 reference material (87 Sr/ 86 Sr accepted value = 0.710240⁹⁸; average of measured values = 0.710283 \pm 0.000011 (1 s.d., n = 56)). Measurements of NIST SRM 1486 gave an average value of 0.709300 \pm 0.000019 (1 s.d., 20 measurements of 10 aliquots), which is very close to the expected value of 0.709299 \pm 0.000027 (in-house long-term average). A subset of samples (n = 41) was measured in duplicate with an average reproducibility of 0.0000077 (1 s.d.). A lack of mixing line in a plot of 1/concentration and 87 Sr/ 86 Sr values indicates that samples did not take up substantial amounts of Sr from the burial environment and therefore are unlikely to be diagenetically altered (Supplementary Figure 15).

For Zn extraction and purification samples were digested in 1 mL of HCl 1.0 N, then evaporated and dissolved again in 1.5 M hydrobromic acid (HBr). Zn was then purified in a two-step ion exchange chromatography method. Briefly, Zn was purified in pre-conditioned microcolumns on 1 mL AG-1x8 resin (200-400 dry mesh size, 106-180 µm wet bead size) using 2 mL of 1.5 M HBr for matrix residue elution and 5 mL of 0.5 M HNO₃ for Zn elution. Samples were then dried and re-dissolved in 1 mL 0.5 M HNO₃. For quality control purposes, aliquots of in-house reference material NIST SRM 1486 bone powder and/or the NIST SRM 1400 bone ash as well as procedural blanks were processed with all sample batches. Zinc isotope ratios were measured using Cu doping on a Neptune Multi-Collector Inductively Coupled Plasma Mass Spectrometer (MC-ICPMS, Thermo Fisher Scientific, Bremen, Germany) at the MPI-EVA. A subset of samples was analysed in duplicate and with an average analytical precision of 0.008 ‰. To assess preservation, we evaluated if samples showed any relationship between zinc concentration and δ^{66} Zn. A mixing line is not observed, indicating that samples preserved a biogenic δ^{66} Zn signal without uptake of exogenous zinc (Supplementary Figure 14).

The reference material Zn Alfa Aesar solution was used for standard bracketing. All δ^{66} Zn values are expressed relative to the JMC-Lyon standard with a mass dependent Alfa Aesar offset of

+0.27 ‰ for δ^{66} Zn^{75,99}. Zinc concentrations were estimated based on Zn signal intensity (V) of the respective sample in comparison to three solutions with known Zn concentrations. Bracketing standard reproducibility within a session was always 0.03 (2 s.d.) or better. Reference materials NIST SRM 1400 and NIST SRM 1486 prepared and analysed with the samples had δ^{66} Zn values of 0.96 ± 0.06‰ (n = 8) and 1.20 ± 0.01‰ (n = 4) as reported elsewhere^{78,92}. All samples and references materials show a Zn mass-dependent isotopic fractionation, i.e., the absence of spectral interferences as δ^{66} Zn vs. δ^{67} Zn and δ^{66} Zn vs. δ^{68} Zn values fall onto lines with slopes close to the theoretic mass fractionation values of 1.5 and 2, respectively. Measurement data for δ^{67} Zn and δ^{68} Zn can be obtained from the electronic data files of this study, which can be found at https://osf.io/wunfd/.

5.4 Collagen extraction and radiocarbon dating

Mandible bone or tooth dentine pieces with weights ranging from ~300–600 mg were sampled from equid specimens using a diamond coated rotary disk, after mechanical cleaning using air abrasion with Korox 99,6% aluminium oxide (50 μ m mesh size). Collagen extraction followed an acid-base-acid plus ultrafiltration protocol at the Department of Human Evolution at the MPI-EVA following the protocol in refs.^{100,101}. Samples were demineralised in HCl 0.5 M at 4 °C until soft and CO₂ effervescence had stopped, with HCl changed twice per week. The demineralised samples were treated with NaOH 0.1 M for 30 min to remove humic acid contamination and then re-acidified in HCl 0.5 M. Samples were then gelatinised in HCl pH3 at 75 °C for 20 h before being filtered to remove particles >60–90 μ m (Ezee filters, Elkay Labs, UK) and ultrafiltered to concentrate the >30 kDa fraction (Sartorius VivaSpin Turbo 15). Filters were precleaned prior to use¹⁰². The >30 kDa fraction was lyophilised for 48 h and the collagen was immediately weighed to determine the collagen yield as a percentage of the dry sample weight. The quality of resulting collagen extracts was assessed prior to radiocarbon dating using collagen yield, carbon and nitrogen elemental concentration, and carbon and nitrogen stable isotope values (determined using EA-IRMS, see next section in Supplementary Text 5). Collagen yield was consistently above the cut-off of 1 %, ranging from 4 to 12.6 %, and elemental concentrations (%C: 44.1–46.3 %; %N: 16.1–16.9 %) were close to those expected for pristine collagen¹⁰³. C/N ratios all fell between 3.2 and 3.3 (accepted values 2.9 to 3.3¹⁰³). This indicates that collagen extracts contained well-preserved collagen and excludes contamination with hydrocarbon based consolidants such as paraffin, which have been documented in some specimens of the Hülle faunal collection.

Six extracts were sent to the Curt-Engelhorn-Center for Archaeometry in Mannheim, Germany (CEZA, laboratory code: MAMS), where the collagen was combusted to CO_2 in an elemental analyzer (EA) and converted catalytically to graphite, before being measured on a MICADAS AMS¹⁰⁴. Ten extracts were sent to the Laboratory of Ion Beam Physics at ETH Zürich, Switzerland (laboratory code: ETH) where they were graphitised using the AGE III system (Automated Graphitisation Equipment¹⁰⁵) and ¹⁴C-dated using a MICADAS AMS¹⁰⁶. Aliquots of a background bone (>50,000 BP) were pretreated and dated alongside the equid samples to monitor lab-based contamination and were used in the age calculation of the samples. Data reduction was performed using BATS software¹⁰⁷ and an additional 1 ‰ was added to the error calculation of the samples, as per standard practice. The ¹⁴C dates were calibrated in OxCal 4.4¹⁰⁸ using the IntCal20 calibration curve¹⁰⁹. Uncalibrated ¹⁴C dates (¹⁴C BP) are reported with their 1 σ error and calibrated ranges (cal BP) are reported at the 2 σ range (95 % probability). All dates have been rounded to the nearest 10 years.

5.5 Carbon and nitrogen stable isotope analysis

Carbon and nitrogen isotopic analyses were conducted using a Flash 2000 Organic Elemental Analyzer (Thermo Fisher Scientific, Bremen, Germany) coupled to a Delta XP ratio mass spectrometer (Thermo Fisher Scientific, Bremen, Germany) via a Conflo III interface (Thermo Fisher Scientific, Bremen, Germany). Approximately 0.5 mg of collagen was weighed into tin capsules and introduced to the Elemental Analyzer using an AS 200S autosampler. The oxidation and reduction reactors were held at 1020 °C and 650 °C respectively. Gas separation was achieved with the GC column held at 65 °C with a carrier gas flow of 115 mL/min. All samples were analysed in duplicate with an average precision of replicate measurements of 0.09 ‰ for δ^{13} C and 0.04 ‰ for δ^{15} N. Carbon and nitrogen stable isotope values were two-point scale normalised to the VPDB and AIR scale respectively using IAEA-CH-6 (sucrose, δ^{13} C = -10.449 \pm 0.033 ‰), IAEA-CH-7 (polyethylene, δ^{13} C = -32.151 \pm 0.050 ‰), IAEA-N-1 (ammonium sulphate, δ^{15} N = 0.4 \pm 0.2 ‰) and IAEA-N-2 (ammonium sulphate, δ^{15} N = 20.3 \pm 0.2 ‰). Normalisation was checked using an in-house methionine standard (EVA-0012; n = 8; Elemental Microanalysis, Okehampton, UK) which gave values of -28.05 + 0.09 ‰ (1 s.d.) for δ^{13} C and -6.43 + 0.03 ‰ (1 s.d) for δ^{15} N and an in-house gelatine standard (EVA MRG, n = 8), which gave values of -19.84 + 0.11 ‰ (1 s.d.) for δ^{13} C and 4.95 \pm 0.10 ‰ (1 s.d) for δ^{15} N. This compares well to their long-term average values of -28.0 \pm 0.1 ‰ (1 s.d) (EVA-0012) and -19.7 \pm 0.3 ‰ (1 s.d) (EVA MRG) for δ^{13} C and -6.4 \pm 0.1 ‰ (1 s.d) (EVA-0012) and 5.0 \pm 0.1 ‰ (1 s.d) (EVA MRG) for δ^{15} N.

5.6 Inverse modelling

Prior to summer and winter palaeotemperature estimation, $\delta^{18}O_{phos}$ time series were processed using an inverse model following ref.¹¹⁰, to remove the damping effect of amelogenesis and the sampling methodology on the seasonal $\delta^{18}O$ amplitude. This method

estimates the original δ^{18} O input into tooth enamel using specifics of the sampling geometry and species-specific parameters of enamel formation fed into an iterative inverse model that mathematically simulates the time-averaging introduced through enamel mineralisation and the sampling procedure. A detailed description of the modelling procedure with associated MATLAB code can be found in ref.@Passey2005a. In this study, an implementation of the original code in R, modified after ref.¹¹¹ was used, which can be found at https://osf.io/wunfd/. A more detailed code documentation with example data can also be found at https://github.com/scpederzani/Oxygen_Inverse_Model. It should be noted that this inverse model was developed for applications to ever-growing teeth and does not take into account the progressive slowing of enamel growth towards the enamel-root-junction (ERJ) seen in some molars of large ungulates, particularly late erupting teeth of horses^{112,113}. For this reason, updated inverse models have been published for sheep¹¹⁴ but, unfortunately, no such model currently exists for horses. Due to this effect, the seasonal amplitudes reconstructed here may represent minimum amplitudes.

Model input parameters were chosen to reflect *Equus* sp. enamel formation, following values given in ref.¹¹⁵ and ref.¹¹⁶. The initial mineral content of enamel was input as 22%, and length of apposition and maturation as 6 mm and 28 mm respectively. Sample input variables were given for distance between samples and sample depth. During the modelling process a damping factor characterising the damping of the oxygen isotope time series amplitude needs to be determined using an adjustment of a measured error term (E_{meas}) to the prediction error (E_{pred}). The damping factors of the specimens analysed in this study ranged between 0.0005 and 0.14. Input and derived parameters used for each specimen can be seen in the model reports included in the associated online repository https://osf.io/wunfd/. The repository also includes graphical representations of all model outcomes.

32

For each specimen the most likely model solution was extracted and used to derive corrected summer peak and winter trough δ^{18} O values. Corrected summer and winter values were extracted in the δ^{18} O time series locations that correspond to peaks and troughs in the original unmodelled δ^{18} O seasonal profile. Corrected summer peak and winter trough values were used as the input for estimating seasonal $\delta^{18}O_{dw}$ values as well as summer and winter temperatures. Mean annual estimates were derived directly from raw $\delta^{18}O_{phos}$ values, rather than from model corrected curves. This approach was taken because the inverse model symmetrically extends the amplitude of the sinusoidal δ^{18} O curve if a systematic sampling geometry was used, and therefore has very little impact on mean δ^{18} O of an annual cycle (see ref.@pederzani2021 for a detailed discussion). Annual averages were calculated as the mean of the summer peak and winter trough δ^{18} O values of each specimen. If several years were represented in a single tooth, multiple annual averages were computed from the different summer/winter combinations. This approach was used following recommendations in ref.⁷², which showed no significant differences between such two-point annual averages and means of full annual cycles.

5.7 Palaeotemperature estimation

Palaeotemperatures were computed following methods outlined in ref.²⁶ using an R implementation of the mathematical equations and excel templates for computations of temperature estimates and associated uncertainties published therein. Mathematical specifications can be obtained from ref.²⁶ or by consulting the temperature estimation R script, which is included in the online data repository of this study (https://osf.io/wunfd/). Enamel δ^{18} O to δ^{18} O_{dw} regressions used in this study were established using ordinary least square regression (OLS) applied to published data for equids refs.¹¹⁷⁻¹²⁰. Modern calibration data to establish the relationship between precipitation δ^{18} O and air temperature was obtained from GNIP stations¹²¹ in Europe (excluding Iberia) and northern Asia to obtain as much temperature coverage as possible, particularly for low temperature environments. Included GNIP stations comprise locations in the following countries: Austria, Belarus, Croatia, Czech Republic, Estonia, Finland, France, Georgia, Germany, Greece, Greenland, Iceland, Italy, Latvia, Mongolia, Norway, Poland, Republic of Moldova, Romania, Russian Federation, Slovakia, Slovenia, Sweden, Ukraine, United Kingdom. If possible, only stations with five or more years of measurements were used. Separate calibration data sets were used for summer, winter and mean annual calibrations. This is firstly because the slope of the $\delta^{18}O_{\text{precip}}$ - air temperature relationship is known to vary seasonally, but also because some GNIP stations only report measurements for certain months of the year. These separate data sets were compiled from GNIP data for the warmest month (commonly July or August; n_{stations} = 113), coldest month (commonly December or January; n_{stations} = 112) and long-term mean annual averages respectively (n_{stations} = 90). Regression lines for the $\delta^{18}O_{\text{precip}}$ - air temperature relationships were calculated using OLS. All calibration data sets are available in the electronic data repository of this study at https://osf.io/wunfd/.

Following the workflow presented in ref.²⁶, palaeotemperature estimates are normally made as an average for each stratigraphic unit, with data from all specimens from the same layer being combined into one temperature estimate. As the equid data used here relies on direct radiocarbon dates for chronological control, we binned the δ^{18} O data of specimens according to data clusters seen in radiocarbon dates and δ^{18} O data, with most time bins covering ~2000 years (Supplementary Table 6). This binning of data enables the use of a sample size dependent calculation of estimation uncertainty but introduces a degree of averaging of between-year climatic variability and biological or behavioural variability between individuals. For this reason, true temperature variability is likely underestimated in the palaeotemperature data as a certain degree of time-averaging is introduced in the palaeotemperature estimates. Additionally, temperature seasonality derived from such pooled summer and winter temperature estimates may deviate somewhat from the full extent of annual temperature fluctuation that may have occurred in some years.

5.8 Software and code

This article, including code for all data analyses, was written in R (version 4.2.0)¹²² on a Windows 10 operating system and the manuscript and supplementary information rendered using Quarto¹²³. The code for data analysis to render the article manuscript and supplementary information rendering makes use of the following packages: cowplot¹²⁴, flextable¹²⁵, ggbeeswarm¹²⁶, ggcorrplot¹²⁷, ggpattern¹²⁸, ggpubr¹²⁹, ggrepel¹³⁰, knitr¹³¹, magick¹³², MetBrewer¹³³, officer¹³⁴, scales¹³⁵, tidyverse¹³⁶.

Data and code to reproduce the manuscript files, figures and analyses are available at https://osf.io/wunfd/. More throuroughly documented and clean example scripts and files for the inverse model and the temperature estimation can be additionally be found at https://github.com/scpederzani/Oxygen_Inverse_Model and https://github.com/scpederzani/Isotope_Temperature

6 Supplementary Figures


Supplementary Figure 1: Plot of direct radiocarbon dates of equid teeth in comparison to the layer that was assigned based on the description of layer colour and depth on find boxes. Depth was correlated with nearest available profile drawing, but use of depths was complicated by pronounced depth fluctuations within the site and by mismatches of depth measurements on profile drawings between original excavation diaries and the site monograph. The comparison of layers assigned in this manner with the radiocarbon dates of equid specimens shows that this process often led to misassignments especially for the different brown layers (IX and XI), where layer color could not reliably be used to differentiate the strata. Layer labels are given as written on find boxes (Braun = brown, Schwarz = black, Grau = grey, Rotbraun/Rot braun = red brown, gelblich = yellowish). Depth information as labelled on find boxes is given in parentheses.



Supplementary Figure 2: Plot of oxygen stable isotope time series data for each tooth ordered by find number. Peaks of sinusoidal curves represent summer season input, while troughs form during the winter season. Colour-marked summer (pink) and winter (purple) season measurements were extracted to investigate diachronic change of seasonal climatic conditions. Oxygen isotope measurements are plotted against distance from the enamel-root-junction (ERJ). Stable isotope data are presented as the mean \pm analytical reproducibility (1 s.d. of n = 3 replicate sample measurements).



Equid specimens ordered by age (youngest to oldest)

Supplementary Figure 3: Summer (pink) and winter (purple) ⁸⁷Sr/⁸⁶Sr values of equids (left) fall into the value range of bioavailable Sr in eastern Thuringia (right, green = tree leaf samples⁶⁵; blue = water samples from the upper course of the Saale River⁶⁷). Similarly, equid ⁸⁷Sr/⁸⁶Sr matches values of hyenas and ursids that typically have small to modest range sizes in the Pleistocene (right). These overlaps indicate that equids most likely ranged across lithologies within a few tens of kilometres of Ranis. This is supported by small summer/winter value differences in almost all cases, with the exception of specimen R10131, which shows a seasonal ⁸⁷Sr/⁸⁶Sr amplitude of almost a full third decimal. A migratory ecology remains unlikely but cannot be excluded for this individual. Summer and winter season for the equid samples were determined using oxygen stable isotopes (see Supplementary Figure 2). Note that taxonomic carnivorans are not necessarily dietary carnivores but can be omnivores with a largely plant-based diet. Analytical reproducibility is smaller than point size.



Supplementary Figure 4: Map of bedrock geological units (A), soil parent bedrock types (B), and predicted ⁸⁷Sr/⁸⁶Sr values (C) in the study area around Ranis. Bedrock geological units were obtained from DTK200 leaf numbers GK5534, GK5526, GK6334, GK6326¹³⁷, parent bedrock types were obtained from BAG5000¹³⁸. Muschelkalk and Buntsandstein are Triassic formations, while the Zechstein formation is a Permian (Palaeozoic) formation that includes prominent carbonate formations. Igneous lithologies are from the Upper Devonian (Palaeozoic). Colors of geological units follow conventions based in unit age, with the exception of igneous lithologies that are colored red according to rock type conventions. The predicted isoscape of bioavailable strontium isotopic ratios (C) were derived from the global strontium isoscape in ref.¹³⁹.



Specimen ordered by age (youngest to oldest)

Supplementary Figure 5: Difference between summer (pink) and winter (purple) enamel δ^{66} Zn values of equids are small (< 0.2 ‰) in most specimens, with the exception of specimen R10131, which shows a seasonal δ^{66} Zn amplitude of ~0.4 ‰, larger than a typical trophic level spacing. Based on comparison with ⁸⁷Sr/⁸⁶Sr values we suggest that this may be linked to seasonal ranging across different lithologies. The direction of summer/winter δ^{66} Zn difference does not show an apparent pattern, but small absolute seasonal amplitudes are mostly observed in specimens with high δ^{66} Zn, perhaps indicating a higher degree of dietary specialisation in these individuals. Error bars represent 1 s.d. of duplicate measurements but are smaller than symbol size in most cases.



Supplementary Figure 6: Equid specimen R10131 shows an uncommonly large absolute seasonal amplitude (magnitude of summer-winter difference) in both zinc and strontium isotope compositions. Any correlation between zinc and strontium isotope absolute seasonal amplitude is only driven by this outlier specimen, underlining that other specimens show a secure dietary pattern in zinc stable isotope compositions. For R10131, its outlier position might indicate that this individual seasonally ranged over isotopically distinct lithologies. While we argue that long-distance migratory behaviour is still unlikely for this individual, it cannot be fully excluded, and we therefore remove this specimen from climatic interpretations of oxygen stable isotopes and dietary interpretations of zinc stable isotopes. The shaded area marks the 95% confidence interval of the OLS linear regression.



Supplementary Figure 7: Equid samples do not show a correlation between δ^{66} Zn and 87 Sr/ 86 Sr, indicating that δ^{66} Zn is predominantly controlled by dietary ecology or other factors not related to underlying lithology. Correlation calculated using Pearson correlation test. Error bars represent 1 s.d. of duplicate measurements but are often smaller than symbol size.



• carnivore • bone eater • omnivore • herbivore

Supplementary Figure 8: Across all taxa sampled from the food web we observe a statistically significant correlation between δ^{66} Zn and 87 Sr/ 86 Sr (Pearson correlation test). This may show a small geological influence on δ^{66} Zn. However, the relationship has a very small slope coefficient and appears mostly driven by two hyena samples with higher 87 Sr/ 86 Sr (without hyena specimens p = 0.083, Pearson correlation). We therefore argue that diet is the dominant driver of δ^{66} Zn in this food web, while bedrock geology plays a very minor role. Error bars represent 1 s.d. of duplicate measurements but are often smaller than symbol size.



Supplementary Figure 9: Seasonal δ^{18} O amplitudes (summer-winter differences) are lower in specimens with higher δ^{18} O values. This negative correlation is statistically significant for winter δ^{18} O, but not summer δ^{18} O (two-sided Pearson correlation). This means that changes in seasonality are driven predominantly by changes in winter δ^{18} O, while summer δ^{18} O is more stable through time. Stable isotope data are presented as the mean ± analytical reproducibility (1 s.d. of n = 3 replicate sample measurements). Lines and shaded areas represent an ordinary least squares regression line with 96% confidence interval.



Supplementary Figure 10: A: Across the complete analysed stratigraphic sequence we observe statistically significant correlations between δ^{18} O and δ^{15} N (negative) and between δ^{66} Zn and δ^{15} N (positive, Pearson correlation). B: A negative correlation is also observed between δ^{18} O and δ^{66} Zn during the δ^{18} O decline from ~45-43 ka cal BP. No other isotope proxy correlates with δ^{13} C regardless of time frame.



Supplementary Figure 11: The decline of equid δ^{18} O between ~45 and 43 ka cal BP is accompanied by rising δ^{66} Zn and δ^{15} N, causing negative correlations (two-sided Pearson correlation) for these tracer combinations for equid specimens dating to > 42 ka cal BP. Error bars represent analytical reproducibility (1 s.d. of replicate sample measurements) but are often smaller than symbol size. Lines and shaded areas represent an ordinary least squares regression line with 96% confidence interval.



Supplementary Figure 12: In equid specimens from all analysed time phases δ^{18} O correlates negatively with δ^{15} N, while δ^{66} Zn and δ^{15} N correlate positively (two-sided Pearson correlation). Error bars represent analytical reproducibility (1 s.d. of replicate sample measurements) but are often smaller than symbol size. Lines and shaded areas represent an ordinary least squares regression line with 96% confidence interval.



Supplementary Figure 13: Comparison of equid δ^{18} O data (top) with directly dated *H. sapiens* remains⁵ (bottom, turquoise points) and faunal bone fragments with anthropogenic surface modifications (dark teal points) demonstrates extensive overlap of *H. sapiens* presence with the coldest temperatures (marked by blue shading) also for uncalibrated dates. Data are presented as mean ± the error of uncalibrated radiocarbon ages (n = 1 bone/tooth sample for each data point), while point shape indicates whether specimens were found in the Hülle (1932–1938, circles) or the TLDA/MPI-EVA (2016–2022, triangles) excavation collection. Vertical error bars represent analytical reproducibility (1 s.d. of n = 3 replicate sample measurements). We argue that *H. sapiens* fragments from the Hülle collection (labelled IX-XI here) all originate from the LRJ deposits (Layer X) and were sometimes assigned to a mixture of Layer X and adjacent strata by the original excavators due to rough excavation methods (see details in ref.⁵). We have pooled all these samples here to reflect this. The human silhouette originates from the voyager plaque (NASA, no copyright), equid silhouette obtained from www.phylopic.org (c Mercedes Yrayzoz/Michael Keesey under cc-by 3.0 https://creativecommons.org/licenses/by/3.0/).



Supplementary Figure 14: A lack of mixing line (correlation coefficients from two-sided Pearson correlation) between δ^{66} Zn and 1/conc (Zn) in the whole food web (A) or herbivores only (B) suggests a lack of zinc uptake from the depositional environment and, thus, preservation of biogenic δ^{66} Zn. We tested both the complete food web and herbivores only to test for statistical artefacts in the whole food web test, that could be caused the much higher herbivore sample size compared to the other dietary groups. Shaded grey error bands represent the 95% confidence interval in both panels.



Supplementary Figure 15: A lack of mixing line (correlation coefficients from two-sided Pearson correlation) between ⁸⁷Sr/⁸⁶Sr and 1/conc (Sr) indicates a lack of uptake of exogenous strontium from the depositional environment and, thus, preservation of biogenic ⁸⁷Sr/⁸⁶Sr. The shaded grey error band represents the 95% confidence interval.

Tables
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Supplementary Table 1: Overview of all specimens that were analysed for this study with contextual and zooarchaeological infor-mation. Equids were used for oxygen, carbon, nitrogen, strontium, and zinc stable isotope analysis and were directly radiocarbon dated. All other specimens were only analysed for zinc and strontium stable isotopes.

squid	Excavation	Square	Layer	Taxon	Element
16/116-123510	TLDA/MPI-EVA	1004/999	6 (Black)	Equus ferus	lower left P3
16/116-124286	TLDA/MPI-EVA	1004/999	6 (Black)	Equus ferus	upper right P3
R10121	Hülle	58	Graue (6.2-6.3 m)	Equus sp.	upper right P3
R10122	Hülle	38	Braune (6.0-6.6 m)	Equus ferus	lower right M3
R10123	Hülle	43	Rotbraune (6.4-7.2 m)	Equus ferus	upper left M3
R10124	Hülle	51a	Rotbraune (gelblich) (6.2-7.0 m)	Equus ferus	upper left M2
R10125	Hülle	37	Braune (6.0-6.6 m)	Equus ferus	upper left P2
R10126	Hülle	105/111	Graue (4.3 m)	Equus ferus	lower left M3
R10127	Hülle	43	Braune (6.6-7.0 m)	Equus ferus	upper right P3
R10128	Hülle	36	Braune (5.8-6.6 m)	Equus ferus	upper left M3
R10129	Hülle	37	Braune (6.0-6.6 m)	Equus ferus	lower right M2
R10130	Hülle	53a	Graue (5.9-6.2 m)	Equus ferus	lower right P4
R10131	Hülle	56	Graue (6.15-6.0 m)	Equus ferus	upper left M2
R10132	Hülle	12b	Braune bis Graue (5.9-6.1 m)	Equus ferus	lower right M3
R10141	Hülle	39	Braun bis Schwarz (5.9-6.5 m)	Equus ferus	upper right M3
R10142	Hülle	39	Braun bis Schwarz (5.9-6.5 m)	Equus ferus	upper right M2
R10148a	Hülle	34	Braune (6.0-7.2 m)	Cervid	M1
R10148b	Hülle	34	Braune (6.0-7.2 m)	Rangifer tarandus	M3
R10149a	Hülle	36	Braune (5.8-6.6 m)	Vulpes lagopus	M2
R10149b	Hülle	36	Braune (5.8-6.6 m)	Vulpes lagopus	U
R10150	Hülle	36	Braune (5.8-6.6 m)	Coelodonta antiquitatis	dP2

squid	Excavation	Square	Layer	Taxon	Element
R10151	Hülle	36	Braune (5.8-6.6 m)	Cervus elaphus	M1
R10152	Hülle	36	Braune (5.8-6.6 m)	Cervid	P3
R10156	Hülle	21	Braune (5.4-6.0 m)	Crocuta spelaea	P3
R10157	Hülle	21	Braune (5.4-6.0 m)	Crocuta spelaea	P3
R10158	Hülle	21	Braune (5.4-6.0 m)	Crocuta spelaea	M1
R10159	Hülle	38	Braune (6.0-6.6 m)	Ursus spelaeus	U
R10160	Hülle	38	Braune (6.0-6.6 m)	Ursus spelaeus	dP
R10161	Hülle	38	Braune (6.8-7.2 m)	Vulpes lagopus	premolar
R10162	Hülle	38	Braune (7.2-7.7 m)	Coelodonta antiquitatis	molar
R10163	Hülle	38	Braune (5.8-7.7)	Cervus elaphus	P4
R10164	Hülle	38	Braune (5.8-7.7)	Vulpes lagopus	M1
R10165	Hülle	37	Braune (6.0-6.6 m)	Cervus elaphus	M3
R10166	Hülle	37	Braune (6.0-6.6 m)	Cervus elaphus	P4
R10167	Hülle	43/44	Braune (7.2-7.6 m)	Cervid	P4
R10168	Hülle	44	Braune (6.6-7.0 m)	Coelodonta antiquitatis	M1
R10169	Hülle	44	Braune (6.6-7.0 m)	Cervid	molar
R10170	Hülle	40/42	Braune (6.7-7.2 m)	carnivore small	I3
R10171	Hülle	40/42	Braune (6.7-7.2 m)	Ursus spelaeus	C
R10172	Hülle	40/42	Braune (6.4-7.2 m)	Rangifer tarandus	P2, P3

Supplementary Table 2: Results of direct radiocarbon dating of equid specimens including quality control indicators (collagen yield, elemental composition, C/N ratio, carbon and nitrogen stable isotope values). Information on tooth position and contextual information for each specimen can be found in Supplementary Table 1.

squid	Material	AMS Code	% yield	C %	N %	C/N	δ ¹³ C	SD δ	1 ⁵ N	SD	¹⁴ C age	error yrs cal BP (68%)	yrs cal BP (95%)
R10132	dentine	ЕТН-111918	5.0	44.5	16.2	3.2	-21.2 (0.01	4.1	0.05	43,670	460 46,490 - 45,500	47,130 – 45,080
R10141	dentine	ЕТН-111916	8.0	45.8	16.6	3.2	-20.9 (0.03	6.5	0.02	38,410	250 42,510 - 42,300	42,640 – 42,200
R10142	dentine	ЕТН-111917	6.4	44.1	16.1	3.2	-20.8 (0.07	4.5	0.04	33,200	140 38,230 – 37,450	38,700 – 37,160

oxygen stable isotope values for each time bin that was used to reconstruct palaeotemperature estimates. Summary statistics are also provided for seasonal amplitudes (summer-winter differences). Number of data points can vary between different seasons of Supplementary Table 3: Summary statistics (means, standard deviation, maxima, minima) of unmodelled seasonal and mean annual the same time bin, as oxygen isotope time series often include multiple or incomplete years and therefore can contain more summer than winter data points and vice versa.

Age bin (cal BP)	Record type	δ ¹⁸ O _{phos}	SD	δ ¹⁸ 0 _{phos} max δ ¹⁸	³ Ophos ^N datapoi	nts
36000 - 39000	summer	13.8	5	12.0	14.9	
42000 - 43000	summer	14.0 0).6	13.4	14.6	ъ
43000 - 45000	summer	12.5 0	0.8	11.6	13.1	ω
45000 - 48000	summer	13.5 0	0.4	13.1	14.3	2
36000 - 39000	mean annual	12.8 1	1.1	11.0	14.2	ъ
42000 - 43000	mean annual	13.2 0	0.4	12.6	13.6	Ъ
43000 - 45000	mean annual	10.9 C).5	10.3	11.2	ω
45000 - 48000	mean annual	12.6 0).3	12.3	13.2	6
36000 - 39000	winter	11.8 1	1.	10.0	13.5	\subseteq
42000 - 43000	winter	12.3 0).5	11.7	12.7	ω
43000 - 45000	winter	9.3 0	0.4	9.0	9.6	7
45000 - 48000	winter	11.5 0).5	10.9	12.1	S
36000 - 39000	seasonal amplitude	2.0 0	.7	0.9	3.3	ъ

Age bin (cal BP)	Record type	δ ¹⁸ O _{phos} SI	Ω	18 _{Ophos} δ ¹⁸ Op	ohos ^N datapoint:
42000 - 43000	seasonal amplitude	1.6 0.4	4	0.9	1.9 5
43000 - 45000	seasonal amplitude	3.3 0.8	∞	2.6	4.1 3
45000 - 48000	seasonal amplitude	2.0 0.7	7	1.0	3.4 9

Sample ID	SQUID	Taxon	mm from ERJ	mean δ ¹⁸ Ophos	SD
R10121.A	R10121	Equus sp.	50.8	13.1	0.2
R10121.B	R10121	Equus sp.	47.8	13.4	0.3
R10121.C	R10121	Equus sp.	44.6	13.1	0.2
R10121.D	R10121	Equus sp.	41.7	13.2	0.2
R10121.E	R10121	Equus sp.	38.9	13.3	0.3
R10121.F	R10121	Equus sp.	35.9	12.8	0.2
R10121.G	R10121	Equus sp.	32.7	13.1	0.6
R10121.H	R10121	Equus sp.	30.0	12.9	0.2
R10121.I	R10121	Equus sp.	27.1	13.0	0.2
R10121.J	R10121	Equus sp.	24.3	13.4	0.6
R10121.K	R10121	Equus sp.	21.7	12.8	0.3
R10121.L	R10121	Equus sp.	19.1	13.2	0.2
R10121.M	R10121	Equus sp.	16.2	12.8	0.3
R10121.N	R10121	Equus sp.	13.4	12.2	0.5
R10121.O	R10121	Equus sp.	10.0	11.7	0.2
R10121.P	R10121	Equus sp.	6.7	11.8	0.2
R10121.Q	R10121	Equus sp.	3.8	12.2	0.3
R10121.R	R10121	Equus sp.	1.2	12.4	0.4
R10122.A	R10122	Equus ferus	59.2	13.0	0.0
R10122.B	R10122	Equus ferus	57.2	13.0	0.2
R10122.C	R10122	Equus ferus	54.8	12.8	0.4
R10122.D	R10122	Equus ferus	52.7	12.3	0.1
R10122.E	R10122	Equus ferus	50.6	12.2	0.3
R10122.F	R10122	Equus ferus	48.8	11.9	0.1
R10122.G	R10122	Equus ferus	46.5	12.1	0.5
R10122.H	R10122	Equus ferus	44.7	11.6	0.2
R10122.I	R10122	Equus ferus	43.1	11.3	0.4
R10122.J	R10122	Equus ferus	41.1	11.4	0.3
R10122.K	R10122	Equus ferus	38.5	11.2	0.5
R10122.L	R10122	Equus ferus	36.7	11.6	0.1
R10122.M	R10122	Equus ferus	34.4	12.0	0.3
R10122.N	R10122	Equus ferus	32.4	12.0	0.0
R10122.O	R10122	Equus ferus	30.2	12.1	0.1

Supplementary Table 4: Individual oxygen stable isotope measurements obtained from sequentially sampled tooth enamel.

Sample ID	SQUID	Taxon	mm from ERJ	mean δ ¹⁸ O _{phos}	SD
R10122.P	R10122	Equus ferus	28.0	11.7	0.2
R10122.Q	R10122	Equus ferus	25.8	10.7	0.3
R10122.R	R10122	Equus ferus	23.0	10.9	0.1
R10122.S	R10122	Equus ferus	19.8	11.1	0.3
R10122.T	R10122	Equus ferus	17.3	11.5	0.2
R10122.U	R10122	Equus ferus	14.3	11.9	0.3
R10122.V	R10122	Equus ferus	8.7	11.8	0.3
R10122.W	R10122	Equus ferus	4.5	13.2	0.2
R10123.A	R10123	Equus ferus	53.3	13.2	0.4
R10123.B	R10123	Equus ferus	50.7	13.6	0.5
R10123.C	R10123	Equus ferus	48.1	12.4	0.4
R10123.D	R10123	Equus ferus	45.0	12.2	0.3
R10123.E	R10123	Equus ferus	42.0	11.3	0.3
R10123.F	R10123	Equus ferus	39.0	11.5	0.3
R10123.G	R10123	Equus ferus	36.2	12.1	0.2
R10123.H	R10123	Equus ferus	33.0	12.6	0.3
R10123.I	R10123	Equus ferus	29.5	12.9	0.2
R10123.J	R10123	Equus ferus	26.7	13.3	0.1
R10123.K	R10123	Equus ferus	23.9	12.9	0.3
R10123.L	R10123	Equus ferus	20.8	12.1	0.1
R10123.M	R10123	Equus ferus	17.4	11.6	0.4
R10123.N	R10123	Equus ferus	14.2	11.4	0.1
R10123.O	R10123	Equus ferus	10.9	12.3	0.2
R10123.P	R10123	Equus ferus	7.3	13.3	0.1
R10123.Q	R10123	Equus ferus	3.8	12.2	0.3
R10123.R	R10123	Equus ferus	1.1	12.5	0.4
R10124.A	R10124	Equus ferus	60.6	13.0	0.2
R10124.B	R10124	Equus ferus	57.6	13.0	0.4
R10124.C	R10124	Equus ferus	54.7	13.1	0.1
R10124.D	R10124	Equus ferus	51.7	12.7	0.3
R10124.E	R10124	Equus ferus	48.7	12.2	0.2
R10124.F	R10124	Equus ferus	45.4	11.9	0.1
R10124.G	R10124	Equus ferus	42.0	11.3	0.1
R10124.H	R10124	Equus ferus	38.8	11.3	0.5
R10124.I	R10124	Equus ferus	35.5	10.2	0.1

Sample ID	SQUID	Taxon	mm from ERJ	mean δ ¹⁸ O _{phos}	SD
R10124.J	R10124	Equus ferus	32.6	9.9	0.3
R10124.K	R10124	Equus ferus	29.4	9.3	0.2
R10124.L	R10124	Equus ferus	26.1	9.0	0.5
R10124.M	R10124	Equus ferus	22.4	9.8	0.3
R10124.N	R10124	Equus ferus	19.1	10.8	0.5
R10124.O	R10124	Equus ferus	15.9	11.1	0.3
R10124.P	R10124	Equus ferus	12.1	11.6	0.1
R10124.Q	R10124	Equus ferus	8.7	11.3	0.2
R10124.R	R10124	Equus ferus	5.3	11.2	0.1
R10124.S	R10124	Equus ferus	2.4	10.6	1.4
R10125.A	R10125	Equus ferus	51.2	14.1	0.2
R10125.B	R10125	Equus ferus	48.6	14.0	0.8
R10125.C	R10125	Equus ferus	45.2	14.7	0.2
R10125.D	R10125	Equus ferus	41.7	14.0	0.3
R10125.E	R10125	Equus ferus	38.6	13.5	0.2
R10125.F	R10125	Equus ferus	35.6	13.1	0.1
R10125.G	R10125	Equus ferus	32.1	13.2	0.3
R10125.H	R10125	Equus ferus	28.4	14.0	0.2
R10125.I	R10125	Equus ferus	24.9	14.9	0.2
R10125.J	R10125	Equus ferus	21.7	14.8	0.3
R10125.K	R10125	Equus ferus	18.2	14.8	0.2
R10125.L	R10125	Equus ferus	14.9	14.2	0.2
R10125.M	R10125	Equus ferus	11.7	13.6	0.2
R10125.N	R10125	Equus ferus	8.6	13.5	0.1
R10125.O	R10125	Equus ferus	5.5	14.0	0.2
R10125.P	R10125	Equus ferus	2.5	13.4	0.2
R10126.A	R10126	Equus ferus	49.7	11.5	0.1
R10126.B	R10126	Equus ferus	45.9	12.2	0.3
R10126.C	R10126	Equus ferus	42.5	12.8	0.1
R10126.D	R10126	Equus ferus	39.1	12.8	0.1
R10126.E	R10126	Equus ferus	35.9	12.4	0.2
R10126.F	R10126	Equus ferus	32.7	12.1	0.1
R10126.G	R10126	Equus ferus	29.4	11.5	0.1
R10126.H	R10126	Equus ferus	26.4	10.7	0.2
R10126.I	R10126	Equus ferus	23.0	9.8	0.2

Sample ID	SQUID	Taxon	mm from ERJ	mean δ ¹⁸ O _{phos}	SD
R10126.J	R10126	Equus ferus	19.7	9.6	0.1
R10126.K	R10126	Equus ferus	16.8	10.0	0.1
R10126.L	R10126	Equus ferus	13.4	10.8	0.4
R10126.M	R10126	Equus ferus	10.3	11.3	0.4
R10126.N	R10126	Equus ferus	6.9	11.4	0.0
R10126.O	R10126	Equus ferus	4.0	11.5	0.4
R10126.P	R10126	Equus ferus	1.1	12.9	0.2
R10127.A	R10127	Equus ferus	44.7	14.3	0.2
R10127.B	R10127	Equus ferus	42.2	14.5	0.2
R10127.C	R10127	Equus ferus	39.0	14.2	0.2
R10127.D	R10127	Equus ferus	36.9	14.0	0.1
R10127.E	R10127	Equus ferus	34.5	13.5	0.1
R10127.F	R10127	Equus ferus	32.5	12.9	0.5
R10127.G	R10127	Equus ferus	29.4	13.0	0.1
R10127.H	R10127	Equus ferus	27.5	12.9	0.7
R10127.I	R10127	Equus ferus	24.6	13.2	0.4
R10127.J	R10127	Equus ferus	22.9	13.5	0.1
R10127.K	R10127	Equus ferus	20.6	13.8	0.2
R10127.L	R10127	Equus ferus	18.0	14.3	0.1
R10127.M	R10127	Equus ferus	14.9	14.4	0.1
R10127.N	R10127	Equus ferus	12.1	14.1	0.2
R10127.O	R10127	Equus ferus	9.3	13.5	0.0
R10127.P	R10127	Equus ferus	6.4	14.4	0.1
R10127.Q	R10127	Equus ferus	3.0	12.9	0.8
R10128.A	R10128	Equus ferus	41.6	13.9	0.3
R10128.B	R10128	Equus ferus	39.1	14.2	0.5
R10128.C	R10128	Equus ferus	36.5	14.3	0.1
R10128.D	R10128	Equus ferus	34.1	14.0	0.3
R10128.E	R10128	Equus ferus	31.4	12.7	0.2
R10128.F	R10128	Equus ferus	28.8	12.5	0.3
R10128.G	R10128	Equus ferus	26.0	12.5	0.5
R10128.H	R10128	Equus ferus	23.0	12.6	0.2
R10128.I	R10128	Equus ferus	19.8	13.2	0.3
R10128.J	R10128	Equus ferus	16.7	13.4	0.2
R10128.K	R10128	Equus ferus	14.2	12.9	0.2

Sample ID	SQUID	Taxon	mm from ERJ	mean δ ¹⁸ O _{phos}	SD
R10128.L	R10128	Equus ferus	10.7	12.8	0.2
R10128.M	R10128	Equus ferus	7.4	14.6	0.4
R10128.N	R10128	Equus ferus	4.3	12.3	0.1
R10128.O	R10128	Equus ferus	1.6	12.5	0.1
R10129.A	R10129	Equus ferus	64.5	13.0	0.2
R10129.B	R10129	Equus ferus	61.1	13.3	0.0
R10129.C	R10129	Equus ferus	58.7	13.2	0.3
R10129.D	R10129	Equus ferus	56.0	13.0	0.0
R10129.E	R10129	Equus ferus	52.3	12.8	0.1
R10129.F	R10129	Equus ferus	49.0	12.2	0.2
R10129.G	R10129	Equus ferus	46.0	11.9	0.4
R10129.H	R10129	Equus ferus	42.7	11.1	0.1
R10129.I	R10129	Equus ferus	39.4	10.5	0.2
R10129.J	R10129	Equus ferus	36.2	10.0	0.3
R10129.K	R10129	Equus ferus	33.1	10.3	0.1
R10129.L	R10129	Equus ferus	30.1	10.8	0.2
R10129.M	R10129	Equus ferus	26.6	11.1	0.3
R10129.N	R10129	Equus ferus	23.3	11.6	0.0
R10129.O	R10129	Equus ferus	20.0	11.5	0.0
R10129.P	R10129	Equus ferus	16.7	12.0	0.2
R10129.Q	R10129	Equus ferus	13.4	11.7	0.1
R10129.R	R10129	Equus ferus	10.3	11.4	0.0
R10129.S	R10129	Equus ferus	7.1	11.2	0.2
R10129.T	R10129	Equus ferus	4.1	12.5	0.1
R10129.U	R10129	Equus ferus	1.3	12.7	0.0
R10130.A	R10130	Equus ferus	56.0	11.0	0.1
R10130.B	R10130	Equus ferus	53.0	11.1	0.3
R10130.C	R10130	Equus ferus	50.7	10.9	0.2
R10130.D	R10130	Equus ferus	48.6	11.6	0.0
R10130.E	R10130	Equus ferus	46.4	11.8	0.2
R10130.F	R10130	Equus ferus	44.0	12.3	0.1
R10130.G	R10130	Equus ferus	41.5	12.7	0.3
R10130.H	R10130	Equus ferus	38.9	13.1	0.1
R10130.I	R10130	Equus ferus	36.1	13.8	0.1
R10130.J	R10130	Equus ferus	33.5	14.3	0.2

Sample ID	SQUID	Taxon	mm from ERJ	mean δ ¹⁸ O _{phos}	SD
R10130.K	R10130	Equus ferus	30.6	14.1	0.3
R10130.L	R10130	Equus ferus	27.6	14.3	0.3
R10130.M	R10130	Equus ferus	25.1	13.9	0.2
R10130.N	R10130	Equus ferus	22.3	13.4	0.2
R10130.O	R10130	Equus ferus	19.5	12.6	0.1
R10130.P	R10130	Equus ferus	17.0	12.1	0.2
R10130.Q	R10130	Equus ferus	13.7	12.0	0.2
R10130.R	R10130	Equus ferus	11.0	12.4	0.1
R10130.S	R10130	Equus ferus	8.1	12.9	0.2
R10130.T	R10130	Equus ferus	5.4	13.3	0.2
R10130.U	R10130	Equus ferus	2.1	13.4	0.1
R10131.A	R10131	Equus ferus	45.8	13.4	0.1
R10131.B	R10131	Equus ferus	43.0	13.6	0.1
R10131.C	R10131	Equus ferus	40.0	13.9	0.1
R10131.D	R10131	Equus ferus	37.4	14.1	0.3
R10131.E	R10131	Equus ferus	34.1	14.4	0.2
R10131.F	R10131	Equus ferus	32.2	14.5	0.2
R10131.G	R10131	Equus ferus	28.6	14.1	0.2
R10131.H	R10131	Equus ferus	26.5	14.3	0.1
R10131.I	R10131	Equus ferus	23.5	14.4	0.1
R10131.J	R10131	Equus ferus	21.1	13.8	0.1
R10131.K	R10131	Equus ferus	18.8	13.0	0.1
R10131.L	R10131	Equus ferus	15.6	13.1	0.3
R10131.M	R10131	Equus ferus	13.1	12.7	0.2
R10131.N	R10131	Equus ferus	10.1	13.4	0.3
R10131.O	R10131	Equus ferus	7.6	13.5	0.1
R10131.P	R10131	Equus ferus	4.8	12.8	0.2
R10131.Q	R10131	Equus ferus	1.7	9.8	0.1
R10132.A	R10132	Equus ferus	53.4	12.1	0.2
R10132.B	R10132	Equus ferus	50.8	12.9	0.2
R10132.C	R10132	Equus ferus	48.1	12.9	0.2
R10132.D	R10132	Equus ferus	45.4	13.2	0.2
R10132.E	R10132	Equus ferus	43.1	13.5	0.4
R10132.F	R10132	Equus ferus	40.5	13.8	0.3
R10132.G	R10132	Equus ferus	37.3	13.6	0.2

Sample ID	SQUID	Taxon	mm from ERJ	mean δ ¹⁸ O _{phos}	SD
R10132.H	R10132	Equus ferus	34.6	13.5	0.1
R10132.I	R10132	Equus ferus	31.8	13.2	0.1
R10132.J	R10132	Equus ferus	28.9	12.7	0.2
R10132.K	R10132	Equus ferus	26.2	12.5	0.0
R10132.L	R10132	Equus ferus	23.3	12.2	0.3
R10132.M	R10132	Equus ferus	20.0	12.3	0.2
R10132.N	R10132	Equus ferus	17.0	12.1	0.2
R10132.O	R10132	Equus ferus	14.5	12.8	0.1
R10132.P	R10132	Equus ferus	11.0	12.9	0.2
R10132.Q	R10132	Equus ferus	7.6	13.1	0.2
R10132.R	R10132	Equus ferus	4.3	12.3	0.1
R10132.S	R10132	Equus ferus	1.2	12.2	0.1
R10141.A	R10141	Equus ferus	36.0	13.2	0.0
R10141.B	R10141	Equus ferus	33.5	14.0	0.1
R10141.C	R10141	Equus ferus	30.4	14.2	0.1
R10141.D	R10141	Equus ferus	27.5	14.5	0.2
R10141.E	R10141	Equus ferus	23.4	13.6	0.3
R10141.F	R10141	Equus ferus	20.3	13.0	0.2
R10141.G	R10141	Equus ferus	17.3	12.7	0.0
R10141.H	R10141	Equus ferus	13.5	14.2	0.1
R10141.I	R10141	Equus ferus	10.1	14.6	0.1
R10141.J	R10141	Equus ferus	7.0	12.3	0.3
R10141.K	R10141	Equus ferus	3.8	13.2	0.2
R10142.A	R10142	Equus ferus	68.3	14.8	0.1
R10142.B	R10142	Equus ferus	65.1	14.6	0.1
R10142.C	R10142	Equus ferus	61.7	13.9	0.2
R10142.D	R10142	Equus ferus	58.7	13.3	0.3
R10142.E	R10142	Equus ferus	55.5	12.9	0.2
R10142.F	R10142	Equus ferus	52.1	12.3	0.2
R10142.G	R10142	Equus ferus	48.9	12.1	0.1
R10142.H	R10142	Equus ferus	45.5	12.1	0.4
R10142.I	R10142	Equus ferus	42.2	11.9	0.1
R10142.J	R10142	Equus ferus	38.6	12.4	0.3
R10142.K	R10142	Equus ferus	35.1	12.7	0.2
R10142.L	R10142	Equus ferus	32.0	13.0	0.1

Sample ID	SQUID	Taxon	mm from ERJ	mean δ ¹⁸ Ophos	SD
R10142.M	R10142	Equus ferus	28.3	13.1	0.1
R10142.N	R10142	Equus ferus	25.1	13.0	0.2
R10142.O	R10142	Equus ferus	21.6	12.9	0.1
R10142.P	R10142	Equus ferus	18.7	12.2	0.8
R10142.Q	R10142	Equus ferus	15.7	11.9	0.3
R10142.R	R10142	Equus ferus	12.6	10.6	0.3
R10142.S	R10142	Equus ferus	9.2	10.7	0.1
R10142.T	R10142	Equus ferus	5.9	12.4	0.5
R10142.U	R10142	Equus ferus	3.1	12.9	0.3
124286.A	16/116-124286	Equus ferus	68.7	12.3	0.3
124286.B	16/116-124286	Equus ferus	65.2	13.2	0.2
124286.C	16/116-124286	Equus ferus	62.0	13.9	0.1
124286.D	16/116-124286	Equus ferus	58.8	14.1	0.2
124286.E	16/116-124286	Equus ferus	55.4	14.0	0.5
124286.F	16/116-124286	Equus ferus	52.0	14.9	0.3
124286.G	16/116-124286	Equus ferus	49.1	14.5	0.1
124286.H	16/116-124286	Equus ferus	45.8	13.8	0.3
124286.I	16/116-124286	Equus ferus	42.9	13.6	0.3
124286.J	16/116-124286	Equus ferus	39.8	12.9	0.2
124286.K	16/116-124286	Equus ferus	36.8	12.6	0.2
124286.L	16/116-124286	Equus ferus	33.6	11.9	0.2
124286.M	16/116-124286	Equus ferus	30.4	12.0	0.1
124286.N	16/116-124286	Equus ferus	27.0	12.8	0.3
124286.0	16/116-124286	Equus ferus	23.8	12.8	0.2
124286.P	16/116-124286	Equus ferus	20.4	13.1	0.2
124286.Q	16/116-124286	Equus ferus	17.2	13.2	0.3
124286.R	16/116-124286	Equus ferus	13.9	12.4	0.2
124286.S	16/116-124286	Equus ferus	10.5	12.5	0.3
124286.T	16/116-124286	Equus ferus	6.9	13.1	0.2
124286.U	16/116-124286	Equus ferus	4.2	13.8	0.3
124286.V	16/116-124286	Equus ferus	1.0	12.3	0.1
123510.A	16/116-123510	Equus ferus	54.0	12.2	0.2
123510.B	16/116-123510	Equus ferus	50.2	11.9	0.2
123510.C	16/116-123510	Equus ferus	46.4	11.6	0.1
123510.D	16/116-123510	Equus ferus	43.3	12.3	0.5

Sample ID	SQUID	Taxon	mm from ERJ	mean δ ¹⁸ Ophos	SD
123510.E	16/116-123510	Equus ferus	39.6	13.2	0.2
123510.F	16/116-123510	Equus ferus	35.5	13.8	0.2
123510.G	16/116-123510	Equus ferus	32.0	14.8	0.3
123510.H	16/116-123510	Equus ferus	28.1	14.8	0.7
123510.I	16/116-123510	Equus ferus	24.6	14.6	0.1
123510.J	16/116-123510	Equus ferus	21.4	14.3	0.3
123510.K	16/116-123510	Equus ferus	18.1	13.7	0.3
123510.L	16/116-123510	Equus ferus	14.5	13.0	0.4
123510.M	16/116-123510	Equus ferus	11.0	12.4	0.2
123510.N	16/116-123510	Equus ferus	7.5	12.5	0.3
123510.0	16/116-123510	Equus ferus	4.5	12.9	0.4
123510.P	16/116-123510	Equus ferus	1.4	12.9	0.4

Supplementary Table 5: Zinc and strontium isotope measurements obtained from a selection of equid enamel samples, representing summer and winter input respectively as well as for bulk samples obtained for a range of carnivore, omnivore, and herbivore species. Samples with multiple letter codes (e.g., 'J+I') indicate that adjacent samples were combined to obtain sufficient material for analysis.

squid	Sample ID	Тахоп	Record type	⁸⁷ Sr/ ⁸⁶ Sr	Sr conc (ppm)	δ ⁶⁶ zn	SD	Zn conc (ppm)
16/116-123510	123510.H	Equus sp.	summer	0.7106	105	1.00	0.01	21
16/116-123510	123510.M	Equus sp.	winter	0.7105	66	0.96	0.00	25
16/116-124286	124286.F	Equus sp.	summer	0.7111	98	0.96	ı	27
16/116-124286	124286.L	Equus sp.	winter	0.7111	114	1.00	0.01	61
R10121	R10121.J+I	Equus sp.	summer	0.7113	148	0.80	0.01	131
R10121	R10121.O+P	Equus sp.	winter	0.7111	125	0.94	0.01	143
R10122	R10122.B	Equus sp.	summer	0.7107	108	0.82	0.00	42
R10122	R10122.K	Equus sp.	winter	0.7104	95	1.00	ı	30
R10123	R10123.J	Equus sp.	summer	0.7099	143	1.07	ı	35
R10123	R10123.N	Equus sp.	winter	0.7095	126	1.09	ı	36
R10124	R10124.C	Equus sp.	summer	0.7107	152	1.06	ı	37
R10124	R10124.K	Equus sp.	winter	0.7107	116	1.07	ı	35
R10125	R10125.F+G	Equus sp.	winter	0.7090	221	06.0		57
R10125	R10125.I+J	Equus sp.	summer	0.7090	212	1.01	ı	55
R10126	R10126.D	Equus sp.	summer	0.7095	129	1.30	ı	15
R10126	R10126.J	Equus sp.	winter	0.7093	144	1.29		16
R10127	R10127.B	Equus sp.	summer	0.7113	113	1.17	0.02	58
R10127	R10127.F+G	Equus sp.	winter	0.7117	108	1.21	0.01	121
R10128	R10128.C	Equus sp.	summer	0.7107	121	0.99	I	36
R10128	R10128.F	Equus sp.	winter	0.7106	114	0.80	0.00	32
R10129	R10129.B	Equus sp.	summer	0.7111	114	0.99	ı	19
R10129	R10129.J	Equus sp.	winter	0.7116	112	0.84	ı	16
R10130	R10130.J	Equus sp.	summer	0.7112	124	0.96	ı	13

squid	Sample ID	Taxon	Record type	⁸⁷ Sr/ ⁸⁶ Sr	Sr conc (ppm)	δ ⁶⁶ Zn	SD	Zn conc (ppm)
R10130	R10130.Q	Equus sp.	winter	0.7113	108	0.93	I	14
R10131	R10131.F	Equus sp.	summer	0.7112	109	1.51	0.00	56
R10131	R10131.M	Equus sp.	winter	0.7120	108	1.12	0.00	87
R10132	R10132.F	Equus sp.	summer	0.7103	126	0.89	I	18
R10132	R10132.M	Equus sp.	winter	0.7100	131	0.79	'	15
R10141	R10141.D	Equus sp.	summer	0.7105	117	0.98	0.01	76
R10141	R10141.G	Equus sp.	winter	0.7105	124	1.00	0.01	79
R10142	R10142.I	Equus sp.	winter	0.7114	97	1.01	0.01	74
R10142	R10142.M	Equus sp.	summer	0.7115	06	1.03	0.01	69
R10148a	R10148a	Cervid	bulk	0.7122	136	0.77	0.02	38
R10148b	R10148b	Rangifer tarandus	bulk	0.7108	132	0.78	0.00	149
R10149a	R10149a	Vulpes lagopus	bulk		ı	0.64	0.00	78
R10149b	R10149b	Vulpes lagopus	bulk	0.7112	110	0.34	0.00	235
R10150	R10150	Coelodonta antiquitatis	bulk	0.7103	60	0.77	0.01	43
R10151	R10151	Cervus elaphus	bulk	0.7089	154	1.10	0.01	74
R10152	R10152	Cervid	bulk	0.7107	119	0.91	00.0	77
R10156	R10156	Crocuta spelaea	bulk	0.7133	129	0.76	0.01	46
R10157	R10157	Crocuta spelaea	bulk	0.7131	115	0.70	0.01	105
R10158	R10158	Crocuta spelaea	bulk	0.7101	86	0.82	0.00	52
R10159	R10159	Ursus spelaeus	bulk	0.7099	125	0.88	0.00	40
R10160	R10160	Ursus spelaeus	bulk	0.7083	41	1.07	0.01	96
R10161	R10161	Vulpes lagopus	bulk	0.7103	82	0.54		103
R10162	R10162	Coelodonta antiquitatis	bulk	0.7106	156	0.57	0.00	83
R10163	R10163	Cervus elaphus	bulk	0.7117	98	0.82	0.00	96
R10164	R10164	Vulpes lagopus	bulk	0.7107	89	0.64	0.00	84
R10165	R10165	Cervus elaphus	bulk	0.7117	89	0.80	0.00	92

squid	Sample ID	Taxon	Record type	⁸⁷ Sr/ ⁸⁶ Sr	Sr conc (ppm)	δ ⁶⁶ zn	SD	Zn conc (ppm)
R10166	R10166	Cervus elaphus	bulk	0.7104	154	0.79	0.01	62
R10167	R10167	Cervid	bulk	0.7108	112	0.83	0.00	70
R10168	R10168	Coelodonta antiquitatis	bulk	0.7113	125	0.79	0.00	82
R10169	R10169	Cervid	bulk	0.7112	104	1.03	0.01	49
R10170	R10170	carnivore small	bulk	0.7110	116	0.35	0.02	227
R10171	R10171	Ursus spelaeus	bulk	0.7121	88	0.43	0.00	74
R10172	R10172	Rangifer tarandus	bulk	0.7099	103	1.01	0.00	92

Supplementary Table 6: Drinking water oxygen isotope values and palaeotemperature estimates for summer, mean annual, and winter for each time bin with corresponding compound estimation error. Number of data points can vary between different seasons of the same time bin, as oxygen isotope time series often include multiple or incomplete years and therefore can contain more summer than winter data points that are available for palaeotemperature conversion and vice versa. Seasonal amplitudes refer to summer-winter temperature differences and are calculated from summer and winter palaeotemperature estimates.

Age bin (cal BP)	Record type	δ ¹⁸ 0 _{dw} δ ¹⁸	O _{dw} error	T _{Air} (°C)	T _{Air} error	V datapoints
36000 - 39000	summer	-9.8	0.8	12.8	1.8	11
42000 - 43000	summer	-7.2	1.1	18.4	2.5	5
43000 - 45000	summer	-11.2	1.5	9.8	3.2	3
45000 - 48000	summer	-9.2	1.0	14.1	2.3	6
36000 - 39000	mean annual	-13.0	0.8	1.7	1.5	15
42000 - 43000	mean annual	-12.4	1.2	2.8	2.2	5
43000 - 45000	mean annual	-15.7	1.5	-3.5	2.9	3
45000 - 48000	mean annual	-13.3	0.9	1.1	1.8	9
36000 - 39000	winter	-16.5	0.9	-9.4	1.8	11
42000 - 43000	winter	-16.3	1.5	-9.0	2.9	3
43000 - 45000	winter	-20.7	1.9	-17.3	3.6	2
45000 - 48000	winter	-17.0	1.3	-10.3	2.4	5
36000 - 39000	seasonal amplitude	6.7	1.2	22.2	2.6	-
42000 - 43000	seasonal amplitude	9.1	1.9	27.4	3.8	-
43000 - 45000	seasonal amplitude	9.5	2.4	27.1	4.9	-
45000 - 48000	seasonal amplitude	7.8	1.6	24.4	3.3	-

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