

1 Supporting Information For:

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3 **Salinity causes widespread restriction of methane emissions from small inland waters**

4

5 **Authors:**

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- 15 • **Table S1:** Summary statistics, mean, standard deviation (SD), range (min and max),
16 number of observations (n) of measured variables by system type.
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19 associated with the random effects from a lognormal generalized linear mixed model of
20 pCH₄ in wetland ponds for each Prairie province (AB = Alberta, SK = Saskatchewan,
21 MB = Manitoba). The wetlands were sampled in the 2021 peripheral survey (Fig. S1).
22 The approach, justification, and interpretation of results is fully detailed in the supporting
23 text (Supporting Text S2).
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- 25 • **Table S3.** Mean estimated marginal means (emmeans) and standard errors from
26 simulations of differential allocation of sampling effort using results from a lognormal
27 generalized linear mixed model of pCH₄ as a function of province (AB = Alberta, SK =
28 Saskatchewan, MB = Manitoba), with random effects of site(province),
29 station(province*site), and date(site*station). Maximum number of sampling dates differ
30 by province to match actual sampling effort in the source data and all simulations were
31 constrained to maximum 471 total samples to match source data. The approach,
32 justification, and interpretation of results is fully detailed in the supporting text
33 (Supporting Text S2).
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- 35 • **Table S4.** Environmental conditions at the high- and low-salinity wetland eddy-
36 covariance measurement sites. Where multiple measurements were taken from May to
37 October, 2021, an average and standard deviation value is provided. For each site, a total
38 of 40 observations were made for water quality parameters in 2021.
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- 41 • **Fig. S1.** Map of study sites in the Canadian Prairie Provinces. The map includes primary
42 sampling sites (where all parameters were measured), peripheral wetland pond sampling
43 sites (where pCH₄ and some water quality parameters were measured to constrain
44 uncertainty in our scaling effort; see inset boxes), and wetland eddy covariance (EC) flux
45 tower sites each at a high- and low salinity wetland (stars in inset boxes). For the primary
46 sampling sites, density distribution plots of measured environmental properties are
47 presented by ecosystem type. See Table S1 for summary statistics.
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- 49 • **Fig S2.** Results of multiple linear regression predictions of pCH₄ as a function of
50 limnological variables. Analyses were grouped either by lakes, rivers, or small (<0.1 km²)
51 lentic systems represented by wetlands and ponds. Analyses are performed on log₁₀
52 transformed data and standardized to a standard deviation of 1 to compare regression
53 coefficients (effect size). Circle size represents the effect size of each explanatory
54 variable, and the green and red colors represent positive vs negative effects, respectively.

55 The light to dark color gradient is scaled to a 0 to 1 confidence value around the effect
56 size, computed as $1 - SE/Eff$, with Eff the effect size and SE its standard error, with
57 negative values ($SE > Eff$) considered as 0. Regression statistics are reported in Table 1.
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- 59 • **Fig. S3.** Interaction between nutrients, organic matter, and salinity as predictors of pCH₄.
60 Comparison of the relationship between log₁₀ values of pCH₄ and either the ratio of DOC
61 concentration (mg L⁻¹) to salinity, or total phosphorus (TP; μg L⁻¹) concentration to
62 salinity, or total nitrogen concentration (TN; mg L⁻¹) to salinity for primary study sites
63 including all small lentic systems (wetlands and small agricultural ponds). Pearson
64 correlation (*r*) and associated probability statistics are listed for each relationship.

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- 66 • **Fig. S4.** Peripheral survey sites show consistent trends with primary sites. The
67 relationship between log₁₀ transformed values of both pCH₄ and the ratio of dissolved
68 organic carbon concentration (mg L⁻¹) to salinity (ppt) at peripheral pond wetland
69 sampling sites. Linear regression model: $p = \ll 0.001$; $R^2 = 0.26$; slope = 1.264; intercept
70 = 0.0028.
- 71
- 72 • **Fig S5.** Salinity versus sulfate concentration in the three types of sampled lentic systems
73 that were part of the primary sampling sites. The linear regression yields a p-value \ll
74 0.001 and a $R^2_{adj} = 0.56$.
- 75
- 76 • **Fig. S6.** Exploring salinity versus SO₄²⁻ concentration as predictors of pCH₄. Comparison
77 of the relationship between log₁₀ pCH₄ and either the ratio of DOC concentration (mg L⁻¹)
78 to salinity, or DOC concentration to SO₄²⁻ concentration (mg L⁻¹) for primary study
79 sites including all lentic systems combined, and individual relationships for lakes,
80 wetlands and small agricultural ponds. Pearson correlation (*r*) and associated probability
81 statistics are listed for each relationship.
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- 83 • **Fig. S7.** Continuous eddy-covariance measurements of CH₄ emissions during the ice-free
84 period from two wetland ecosystems. Each site in in Manitoba, Canada, is broadly
85 representative of hardwater (salt rich) and soft water habitats in the Canadian Prairie
86 Pothole region.
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- 88 • **Fig S8.** Increasing SO₄²⁻ concentrations in recent decades in most monitoring lake sites.
89 Map depicting long-term (1990-2020) trends (Sen slopes) in inland water SO₄²⁻
90 concentration in monitored sites in the Canadian Prairies, based on publicly available data
91 from the Saskatchewan Water Security Agency.
- 92
- 93 • **Fig. S9.** Salinity scales inversely with average ebullition flux rates (error bars ± 1 S.D.).
94 Open water measurements from agricultural ponds shown in figure 2. Note log₁₀ scale of
95 both axes.
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- **Fig. S10.** The relationship between \log_{10} transformed SO_4^{2-} and Cl^- concentrations in diverse global saline systems. Data replotted from Table 3 of Deocampo and Jones¹.
 - **Supporting Text S1.** Expanded discussion on the use of salinity versus SO_4^{2-} content as predictors of CH_4 cycling.
 - **Supporting Text S2.** Analysis exploring the importance of within-site, cross-season, and cross-site replication for improving estimates of pCH_4 .

106 **Supporting Tables:**

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108 **Table S1:** Summary statistics, mean, standard deviation (SD), range (min and max), number of
 109 observations (n) of measured variables by system type.

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| Variable | System type | Mean | Median | SD | Min | Max | n |
|---|-------------|---------|--------|----------|--------|---------|-----|
| Area (km ²) | All | 7.085 | 0.0015 | 47.4831 | 0.0001 | 500 | 169 |
| | Rivers | | | | | | 0 |
| | Lakes | 70.099 | 8.6 | 137.7732 | 0.5 | 500 | 17 |
| | Wetlands | 0.1229 | 0.0056 | 0.7454 | 0.0001 | 5.01 | 45 |
| | Ponds | 0.0014 | 0.001 | 0.0018 | 0.0002 | 0.0139 | 107 |
| Max depth (m) | All | 2.83 | 2.1 | 3.41 | 0.12 | 27.7 | 148 |
| | Rivers | | | | | | 0 |
| | Lakes | 9.46 | 9 | 6.4 | 2.8 | 27.7 | 17 |
| | Wetlands | 1.08 | 0.7 | 2.01 | 0.12 | 10 | 23 |
| | Ponds | 2.15 | 2.1 | 0.97 | 0.18 | 5.1 | 108 |
| Temperature (°C) | All | 20.1 | 20.2 | 2.6 | 13 | 29.5 | 193 |
| | Rivers | 19 | 18.5 | 3 | 13 | 23.9 | 23 |
| | Lakes | 20.3 | 20.5 | 1.4 | 17.6 | 22.3 | 17 |
| | Wetlands | 20.7 | 20.4 | 3 | 15.6 | 26.8 | 45 |
| | Ponds | 20.1 | 20 | 2.5 | 15.7 | 29.5 | 108 |
| pH | All | 8.6 | 8.62 | 0.71 | 6.76 | 10.5 | 193 |
| | Rivers | 8.33 | 8.35 | 0.36 | 7.3 | 9.34 | 23 |
| | Lakes | 8.82 | 8.8 | 0.21 | 8.49 | 9.14 | 17 |
| | Wetlands | 8.29 | 8.29 | 0.94 | 6.76 | 9.9 | 45 |
| | Ponds | 8.75 | 8.72 | 0.65 | 6.95 | 10.5 | 108 |
| Salinity (ppt) | All | 0.95 | 0.45 | 1.55 | 0.07 | 13.27 | 193 |
| | Rivers | 0.28 | 0.16 | 0.26 | 0.1 | 0.92 | 23 |
| | Lakes | 1.48 | 0.79 | 1.86 | 0.12 | 5.93 | 17 |
| | Wetlands | 1.33 | 0.53 | 2.39 | 0.08 | 13.27 | 45 |
| | Ponds | 0.84 | 0.45 | 1.12 | 0.07 | 8.57 | 108 |
| Sp. Conductivity (μS cm ⁻¹) | All | 1656 | 838 | 2446 | 136 | 18974 | 193 |
| | Rivers | 520 | 345 | 421 | 201 | 1549 | 23 |
| | Lakes | 2749 | 1579 | 3263 | 263 | 10494 | 17 |
| | Wetlands | 2226 | 995 | 3555 | 162 | 18974 | 45 |
| | Ponds | 1489 | 810 | 1811 | 136 | 13489 | 108 |
| SO ₄ (mg L ⁻¹) | All | 1037.9 | 558.45 | 1454.74 | 0.72 | 9489.78 | 118 |
| | Rivers | | | | | | 0 |
| | Lakes | 1273.54 | 533.3 | 1823.34 | 78.55 | 6304.24 | 14 |
| | Wetlands | 804.92 | 293.76 | 1274.02 | 0.72 | 6686.67 | 42 |
| | Ponds | 1142.51 | 703.81 | 1482.69 | 0.99 | 9489.78 | 62 |
| DOC (mg L ⁻¹) | All | 28.2 | 25.5 | 20.1 | 0.4 | 126.9 | 190 |
| | Rivers | 5.8 | 2.3 | 7.5 | 0.4 | 25.6 | 23 |
| | Lakes | 18 | 17.3 | 10.4 | 4.4 | 43.6 | 17 |
| | Wetlands | 36.5 | 31.2 | 26.2 | 3.1 | 126.9 | 45 |

| | | | | | | | |
|---|----------|------|------|-------|--------------------|-------|-----|
| | Ponds | 31.3 | 29.1 | 15.7 | 4.6 | 90.4 | 105 |
| TP ($\mu\text{g L}^{-1}$) | All | 267 | 80 | 621 | 1 | 6480 | 189 |
| | Rivers | 154 | 20 | 519 | 1 | 2508 | 23 |
| | Lakes | 143 | 71 | 171 | 12 | 539 | 17 |
| | Wetlands | 330 | 117 | 543 | 30 | 2540 | 45 |
| | Ponds | 285 | 80 | 713 | 9 | 6480 | 104 |
| TN (mg L^{-1}) | All | 2.48 | 1.95 | 2.07 | 0.05 | 14.28 | 189 |
| | Rivers | 0.54 | 0.27 | 0.66 | 0.05 | 3.05 | 23 |
| | Lakes | 1.44 | 1.43 | 0.68 | 0.48 | 3.25 | 17 |
| | Wetlands | 2.65 | 1.96 | 1.69 | 0.99 | 9.54 | 45 |
| | Ponds | 3 | 2.3 | 2.26 | 0.42 | 14.28 | 104 |
| DOC / salinity ($\text{mg L}^{-1} \text{ppt}^{-1}$) | All | 59.7 | 43.8 | 60.8 | 2.4 | 410.3 | 190 |
| | Rivers | 17.7 | 14.4 | 16.9 | 2.4 | 78.3 | 23 |
| | Lakes | 21.9 | 19.5 | 12.4 | 5.4 | 46.9 | 17 |
| | Wetlands | 75 | 53 | 88 | 5.2 | 410.3 | 45 |
| | Ponds | 68.5 | 56.5 | 50.6 | 8 | 276.9 | 105 |
| pCH ₄ (ppm) | All | 2653 | 933 | 4904 | 2 | 40882 | 191 |
| | Rivers | 481 | 160 | 685 | 5 | 2750 | 23 |
| | Lakes | 412 | 163 | 778 | 2 | 3225 | 16 |
| | Wetlands | 4038 | 1742 | 6028 | 52 | 30244 | 45 |
| | Ponds | 2873 | 1324 | 5020 | 34 | 40882 | 107 |
| CH ₄ diffusion ($\text{mmol m}^{-2} \text{d}^{-1}$) | All | 6.12 | 2.76 | 10.6 | 0.11 | 91.47 | 139 |
| | Rivers | | | | | | 0 |
| | Lakes | | | | | | 0 |
| | Wetlands | 3.42 | 1.75 | 4.13 | 0.13 | 14.11 | 20 |
| | Ponds | 6.58 | 2.9 | 11.28 | 0.11 | 91.47 | 119 |
| CH ₄ ebullition ($\text{mmol m}^{-2} \text{d}^{-1}$) | All | 0.89 | 0.03 | 1.65 | 0 | 4.66 | 10 |
| | Rivers | | | | | | 0 |
| | Lakes | | | | | | 0 |
| | Wetlands | 1.12 | 0.01 | 2.02 | 2×10^{-6} | 4.66 | 5 |
| | Ponds | 0.67 | 0.05 | 1.38 | 3×10^{-4} | 3.13 | 5 |

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115 **Table S2.** Estimates of variance (proportion of total variance given in brackets) associated with
 116 the random effects from a lognormal generalized linear mixed model of pCH₄ in wetland ponds
 117 for each Prairie province (AB = Alberta, SK = Saskatchewan, MB = Manitoba). The wetlands
 118 were sampled in the 2021 peripheral survey (Fig. S1). The approach, justification, and
 119 interpretation of results is fully detailed in the supporting text (Supporting Text S2).

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| Variance Component | AB Estimate (Proportion of Variance) | SK Estimate (Proportion of Variance) | MB Estimate (Proportion of Variance) |
|--------------------|--|--|--|
| Site | 0.66 (0.37) | 3.85 (0.77) | 1.78 (0.61) |
| Station(Site) | 0.00 (0.00) | 0.00 (0.00) | 0.00 (0.00) |
| Date(Site*Station) | 1.12 (0.63) | 1.12 (0.23) | 1.12 (0.39) |

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123 **Table S3.** Mean estimated marginal means (emmeans) and standard errors from simulations of
 124 differential allocation of sampling effort using results from a lognormal generalized linear mixed
 125 model of pCH₄ as a function of province (AB = Alberta, SK = Saskatchewan, MB = Manitoba),
 126 with random effects of site(province), station(province*site), and date(site*station). Maximum
 127 number of sampling dates differ by province to match actual sampling effort in the source data
 128 and all simulations were constrained to maximum 471 total samples to match source data. The
 129 approach, justification, and interpretation of results is fully detailed in the supporting text
 130 (Supporting Text S2).

| Province | Mean Emmeans | Mean Std Err | No. of Sites | Max. Dates (AB) | Max. Dates (SK) | Max. Dates (MB) |
|----------|--------------|--------------|--------------|-----------------|-----------------|-----------------|
| AB | 5.836 | 0.218 | 16 | 9 | 7 | 14 |
| SK | 4.214 | 0.476 | 16 | 9 | 7 | 14 |
| MB | 4.851 | 0.309 | 16 | 9 | 7 | 14 |
| AB | 5.857 | 0.173 | 32 | 5 | 4 | 7 |
| SK | 4.331 | 0.359 | 32 | 5 | 4 | 7 |
| MB | 4.885 | 0.240 | 32 | 5 | 4 | 7 |

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134 **Table S4.** Environmental conditions at the high- and low-salinity wetland eddy covariance
 135 measurement sites. Where multiple measurements were taken from May to October, 2021, an
 136 average and standard deviation value is provided. For each site, a total of 40 observations were
 137 made for water quality parameters in 2021.

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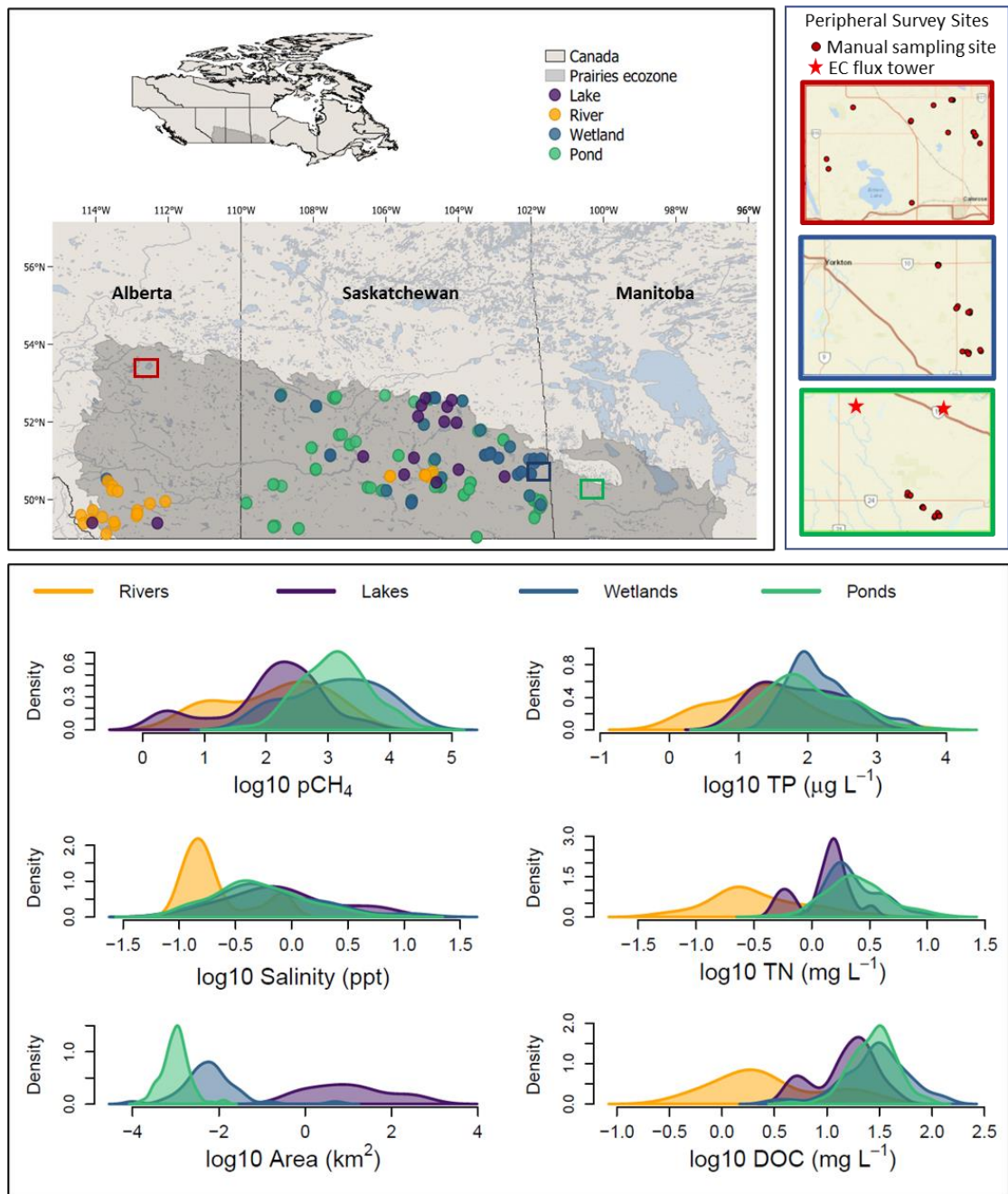
| | Site Name | Province | Site I.D. code | Surface Area (Ha) | Salinity (psu) | pH | Specific Cond. (uS/cm) | DOC (mg/L) | TDN (mg/L) | TP (ug/L) | SO ₄ ²⁻ (mg/L) |
|------|-----------|----------|----------------|-------------------|----------------|------|------------------------|------------|------------|-----------|--------------------------------------|
| Mean | Young | MB | MB15 | 14.5 | 0.57 | 8.95 | 982 | 28.93 | 2.32 | 494.3 | 320.0 |
| S.D. | Young | MB | MB15 | | 0.14 | 0.52 | 209 | 4.96 | 0.89 | 252.5 | 90.4 |
| Mean | Hogg | MB | MB16 | 20.3 | 2.31 | 8.70 | 3576 | 86.86 | 4.87 | 164.4 | 1114.2 |
| S.D. | Hogg | MB | MB16 | | 0.76 | 0.25 | 1043 | 26.46 | 1.64 | 74.1 | 343.8 |

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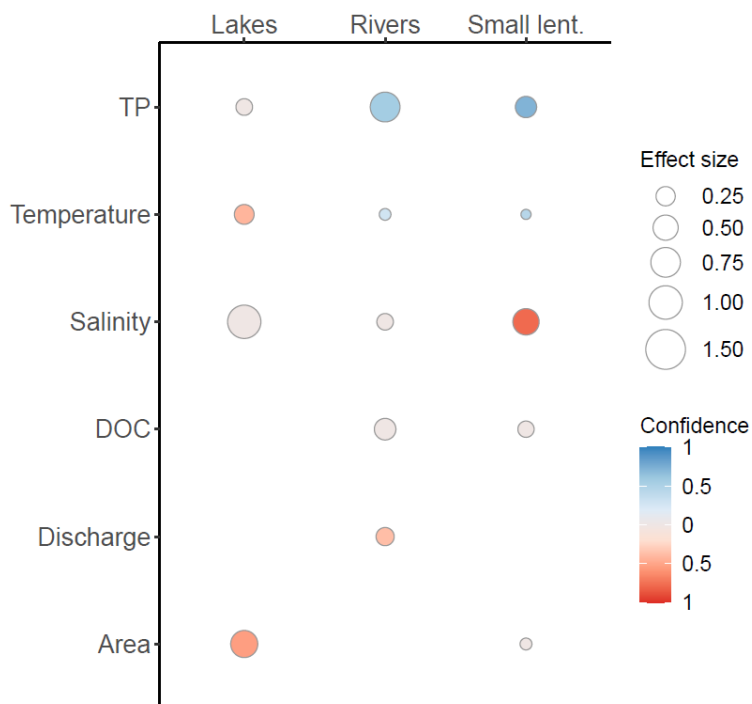
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142 **Supporting Figures:**



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144 **Fig. S1.** Map of study sites in the Canadian Prairie Provinces. The map includes primary sampling
 145 sites (where all parameters were measured), peripheral wetland pond sampling sites (where pCH₄
 146 and some water quality parameters were measured to constrain uncertainty in our scaling effort;
 147 see inset boxes), and wetland eddy covariance (EC) flux tower sites each at a high- and low salinity
 148 wetland (stars in inset boxes). For the primary sampling sites, density distribution plots of
 149 measured environmental properties are presented by ecosystem type. See Table S1 for summary
 150 statistics.

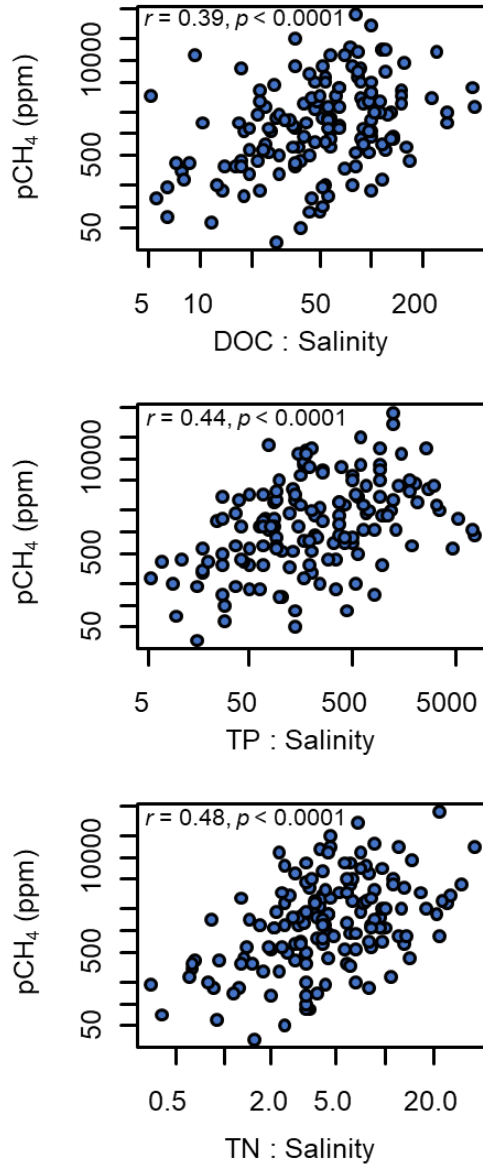


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152 **Fig S2.** Results of multiple linear regression predictions of pCH₄ as a function of limnological
 153 variables. Analyses were grouped either by lakes, rivers, or small (<0.1 km²) lentic systems
 154 represented by wetlands and ponds. Analyses are performed on log₁₀ transformed data and
 155 standardized to a standard deviation of 1 to compare regression coefficients (effect size). Circle
 156 size represents the effect size of each explanatory variable, and the green and red colors represent
 157 positive vs negative effects, respectively. The light to dark color gradient is scaled to a 0 to 1
 158 confidence value around the effect size, computed as $1 - SE/Eff$, with Eff the effect size and SE
 159 its standard error, with negative values ($SE > Eff$) considered as 0. Regression statistics are
 160 reported in Table 1.

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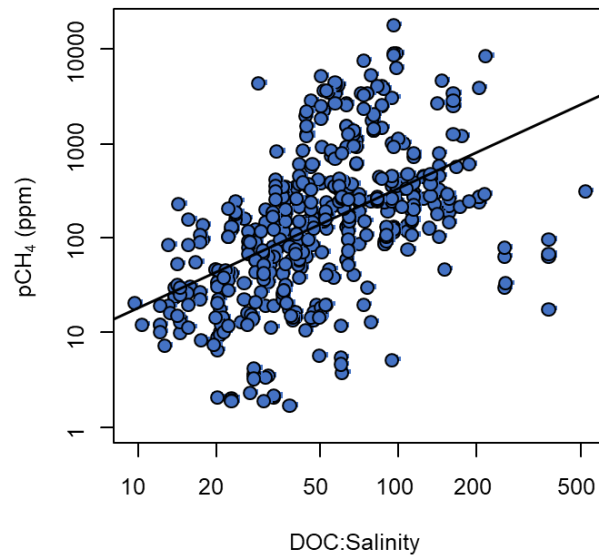
164 **Fig. S3.** Interaction between nutrients, organic matter, and salinity as predictors of pCH₄.
 165 Comparison of the relationship between log₁₀ values of pCH₄ and either the ratio of DOC
 166 concentration (mg L⁻¹) to salinity, or total phosphorus (TP; μg L⁻¹) concentration to salinity, or
 167 total nitrogen concentration (TN; mg L⁻¹) to salinity for primary study sites including all small
 168 lentic systems (wetlands and small agricultural ponds). Pearson correlation (r) and associated
 169 probability statistics are listed for each relationship.

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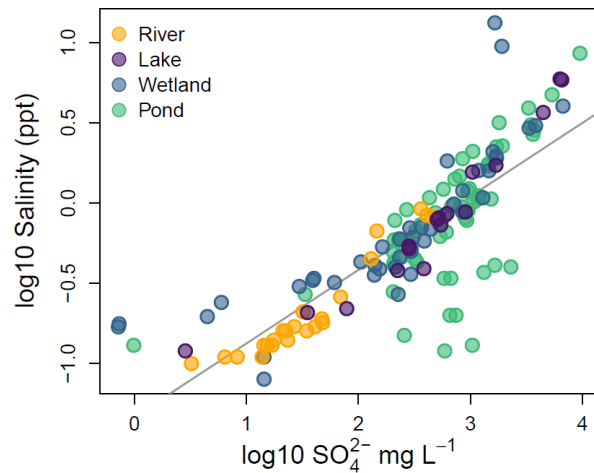
176 **Fig. S4.** Peripheral survey sites show consistent trends with primary sites. The relationship
177 between \log_{10} transformed values of both $p\text{CH}_4$ and the ratio of DOC (mg L^{-1}) / salinity (ppt) at
178 peripheral pond wetland sampling sites. Linear regression model: p-value $\ll 0.0001$; $R^2_{\text{adj}} =$
179 0.25; slope = 1.264; intercept = 0.0028, $n = 465$.

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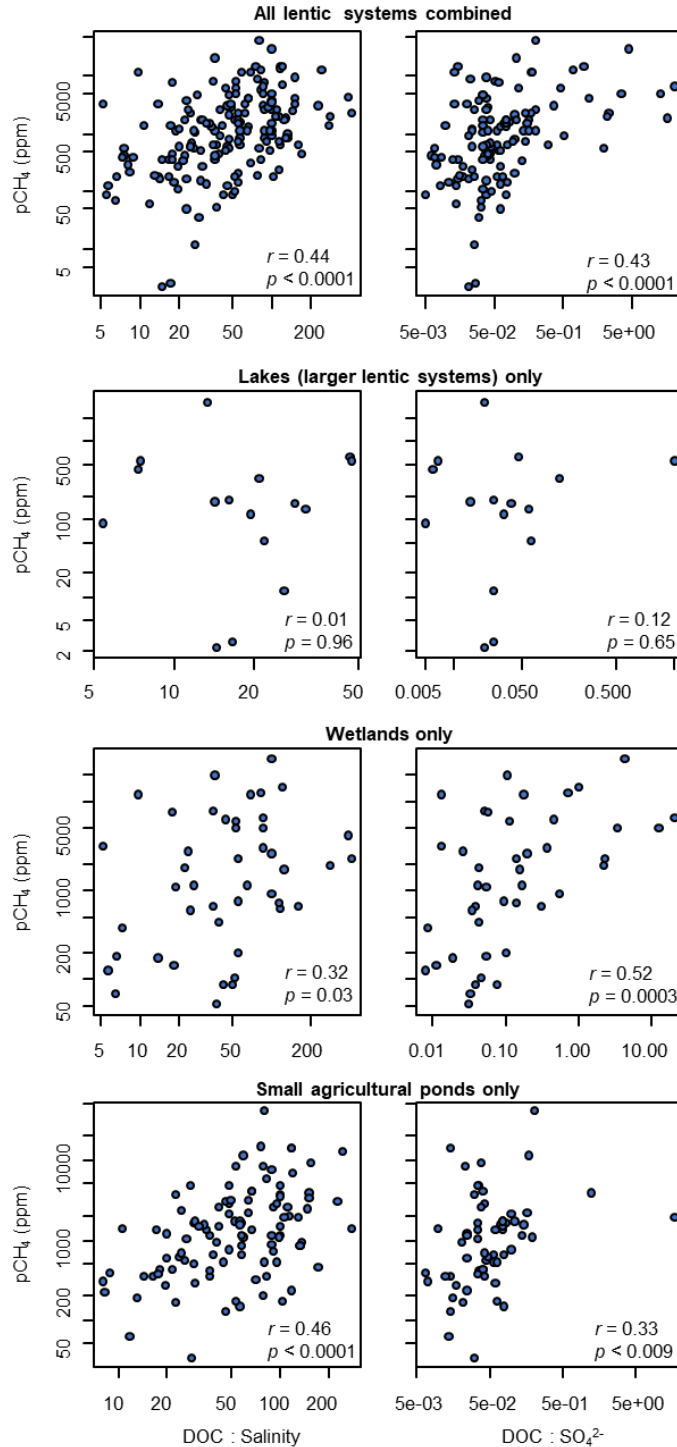
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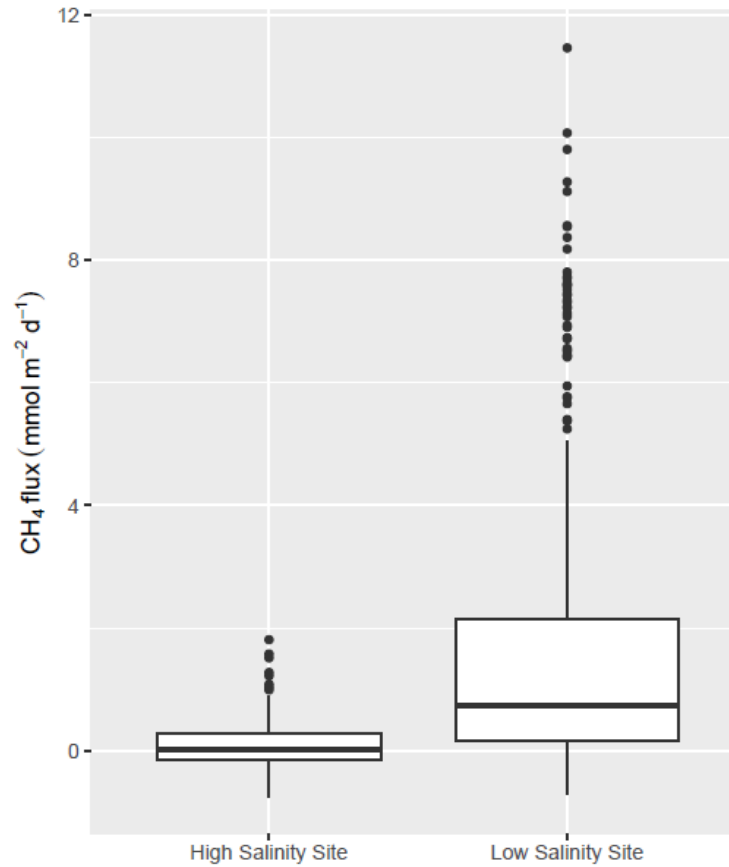
185 **Fig S5.** Salinity versus sulfate concentration in the three types of sampled lentic systems and the
186 rivers that were part of the primary sampling sites. The linear regression yields a p-value <<
187 0.001 and a $R^2_{\text{adj}} = 0.68$.

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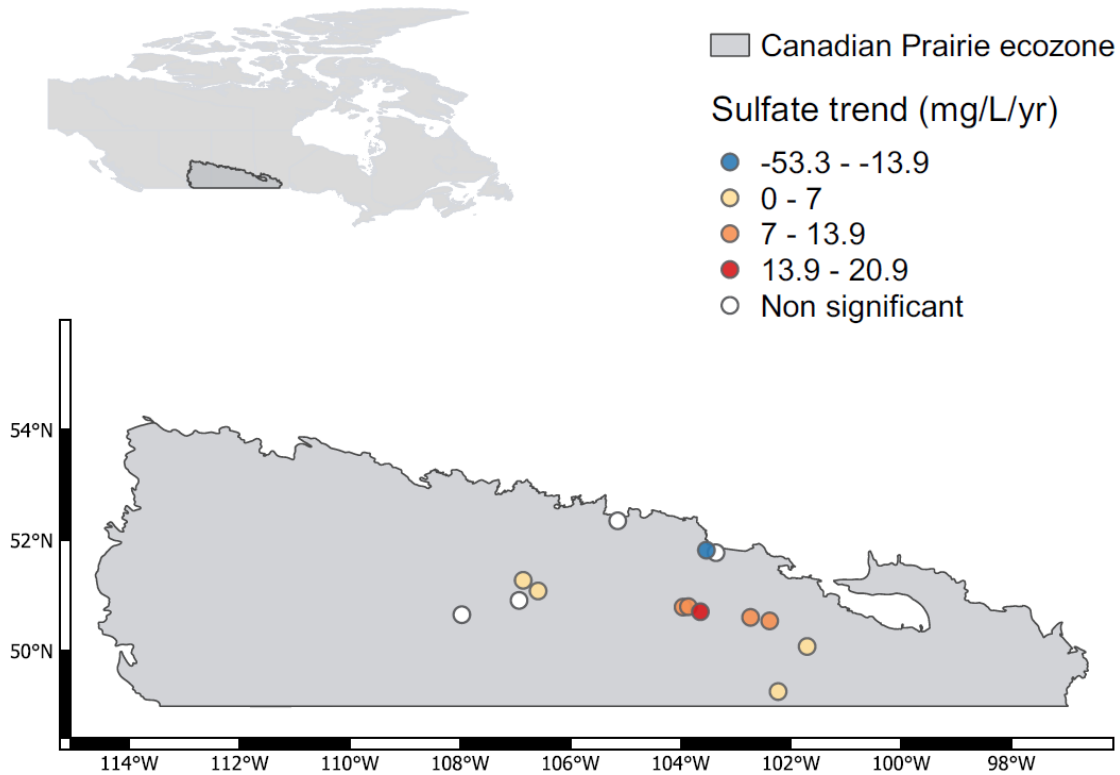
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 191 relationship between $\log_{10} p\text{CH}_4$ and either the ratio of DOC concentration (mg L^{-1}) to salinity,
 192 or DOC concentration to SO_4^{2-} concentration (mg L^{-1}) for primary study sites including all lentic
 193 systems combined, and individual relationships for lakes, wetlands and small agricultural ponds.
 194 Pearson correlation (r) and associated probability statistics are listed for each relationship.



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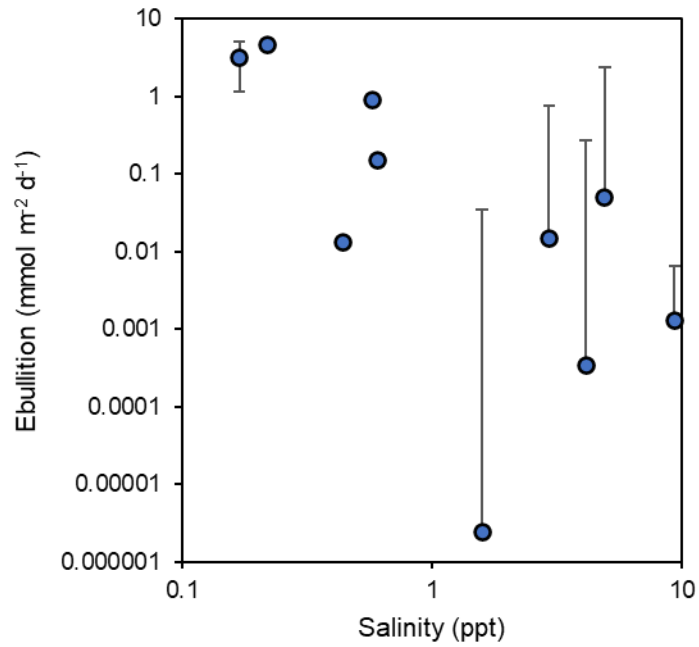
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197 from two wetland ecosystems. Each site in in Manitoba, Canada, is broadly representative of
198 hardwater (salt rich) and soft water habitats in the Canadian Prairie Pothole region.
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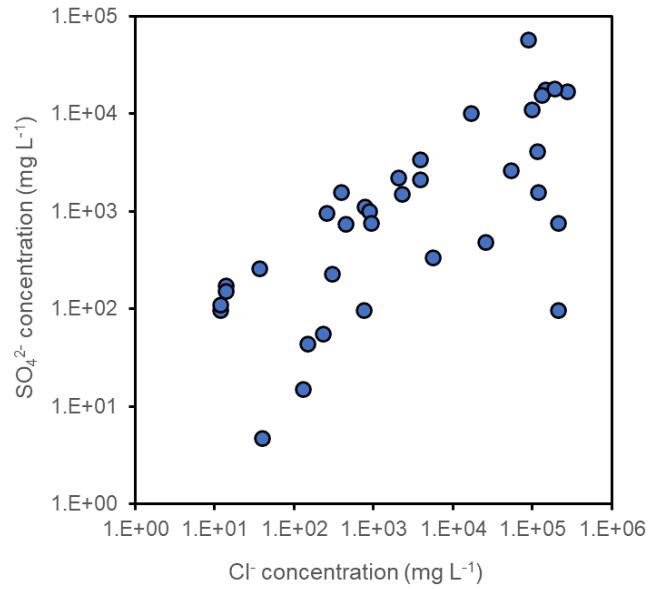
Fig. S8. Increasing SO_4^{2-} concentrations in recent decades in most monitoring lake sites. Map depicting long-term (1990-2020) trends (Sen slopes) in inland water SO_4^{2-} concentration in monitored sites in the Canadian Prairies, based on publicly available data from the Saskatchewan Water Security Agency.



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210 **Fig. S9.** Salinity scales inversely with average ebullition flux rates (error bars ± 1 S.D.). Open
 211 water measurements from agricultural ponds shown in figure 2. Note \log_{10} scale of both axes.

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Fig. S10. The relationship between log₁₀ transformed SO₄²⁻ and Cl⁻ concentrations in diverse global saline systems. Data replotted from Table 3 of Deocampo and Jones¹.

220 **Supporting Text:**

221 **Supporting Text S1.** Expanded discussion on the use of salinity versus individual ions as
222 predictors of CH₄ cycling.

223

224 Below we justify the use of salinity as a predictor of CH₄ content and fluxes in place of other
225 predictors, namely individual ions such as SO₄²⁻ content. Previous research on inland water CH₄
226 cycling has relied on different measurements including more general indicators of ionic strength
227 (salinity, specific conductance) versus the use of SO₄²⁻ concentrations alone. Sulfate is often the
228 most abundant anion in many landscapes and is the best studied and probably most important ion
229 that sets redox conditions that favour alternate processes over methanogenesis. However, as
230 detailed below, the cycling of S is not the only mechanism controlling CH₄ and restricting CH₄
231 content and emissions at elevated salinity. Here, our objective was to deliver the broadest,
232 overarching empirical assessment of the factors that control CH₄ cycling in hardwater
233 environments. Thus, we used salinity to establish these empirical relationships because this
234 metric represented an important gap in the inland water literature that would serve to establish
235 broad relationships which future studies can refine. Below, we justify this approach by
236 demonstrating that measures of salinity capture multiple factors that impact the CH₄ cycle, each
237 likely varying in relative importance from one system or region to another.

238

239 **Using salinity as a broad and integrative predictor:**

240 Distinct ecosystems, or even regions^{1,2} may have different geochemistry and surface
241 water ionic composition that in turn drives the gradient in regional salinity (from freshwater to

242 sub- and saline systems). These differences will in turn make salinity a predictor that, when
243 comparing ecosystems among distinct geochemical regions, can reflect variable ionic
244 compositions of surface waters. Therefore, the use of salinity as a predictor in our study is akin
245 to the widespread prediction of ecosystem structure and functioning based on measurements of
246 other general chemical features, such as the use of total phosphorus (TP)³⁻⁵, total nitrogen (TN)<sup>5-
247 7</sup>, dissolved organic carbon (DOC)^{8,9} concentrations, or ratios of these predictors^{6,10,11} to predict
248 broad patterns in ecosystem features. While it is known that the chemical composition of the
249 pool of nutrients and C making up each bulk chemical measurement will vary from one
250 ecosystem to the next, all of these widely-used predictors are useful because they provide general
251 indications of the scaling of ecosystem features and functioning. Foundational relationships
252 across aquatic ecosystems have been established with these metrics, all the while with the
253 knowledge that site- or regional variations in the shapes of these relationships are present (e.g.,
254 ^{12,13}). This knowledge does not preclude or invalidate the use of bulk chemical metrics but
255 compliments it. Here, the use of salinity represents a comparable step toward generating
256 predictive tools to help explain CH₄ fluxes at the global scale. More narrow explorations of
257 individual ions, and how local or regional differences in CH₄ content and flux may vary with
258 salinity as ionic composition shifts, represents a more focused and complementary avenue of
259 exploration. Broadly, salinity is a useful predictor of overall availability of elements that interact
260 with CH₄ cycling, because major ions generally scale in concentration in a positive way (see Fig.
261 S10 for SO₄²⁻ and Cl⁻ for a global relationship in saline systems¹), and trace element content
262 generally scales with other ions¹⁴. Therefore, our proposed use of salinity represents the upper-
263 most tier of a hierarchical approach to exploring CH₄ relationships with major ions. Such an
264 approach requires that we establish the broad, generalizable relationships to salinity (much like

265 has been done with TN or TP and metrics of trophic status or food web features), while also
266 exploring relationships with individual ions or elements. This framework will provide the most
267 comprehensive understanding of the CH₄ cycle in hardwater ecosystems.

268

269 **Salinity is a predictor of CH₄ concentration and flux in non-SO₄²⁻ dominated hardwaters:**

270 We take this line of reasoning (that salinity is a consistently useful metric to predict CH₄
271 cycling in hardwater landscapes) a step further with a demonstration that salinity can reflect
272 diverse mechanisms of suppressed CH₄ production, even in systems where ionic composition is
273 vastly different from the SO₄²⁻ dominated ecosystems in the Canadian Prairies. While the global
274 relationship between SO₄²⁻ and Cl⁻ is positive (Fig. S10), local variation in this pattern exists.
275 The ratio of dissolved SO₄²⁻ to Cl⁻ content is typically elevated in the northern Great Plains
276 region of North America (~ 1 to 13) relative to other regions where the values typically are < 1,
277 reflecting previously detailed differences in the evolution of contemporary brine composition
278 and dissolved major ion abundances². Yet despite these regional differences in brine
279 composition, extremely restricted rates of CH₄ production are seen at elevated salinities even in
280 regions where SO₄²⁻ is a less important contributor to salinity.

281 We present published observations from hardwater ponds and lakes from the Iberian
282 Peninsula that both demonstrate this point and add yet another region to our study that is
283 consistent with our conclusions derived from figure 2 (despite the lack of appropriate data to add
284 the systems to that figure). The suppression of CH₄ production has been documented in Spanish
285 lakes that have distinct geochemical properties from most of our study systems^{15,16}. As
286 demonstrated by Margalef-Marti et al.¹⁵, Gallocanta Lake has SO₄²⁻ to Cl⁻ ratios ~ < 0.01

287 (approximated from the range of SO_4^{2-} concentrations reported for surface waters, and reports of
288 Cl^- concentrations from previous work cited therein) and receives SO_4^{2-} -poor inputs from
289 inflowing surface water. Yet, the abundance of other ions that complex with SO_4^{2-} ultimately
290 enhances the availability of SO_4^{2-} to microbes in surface sediments via the dissolution of
291 mineral-bound SO_4^{2-} . This mechanism appears to fuel local S cycling via coupling to the cycling
292 of other reactive elements, and ultimately restricts CH_4 production and emissions¹⁵ (mechanisms
293 detailed below). As a second example from this region, Camacho et al.¹⁶ demonstrate that the
294 enhanced availability of NaCl, or dilution of salt-rich surface water (using pure water) can
295 respectively suppress or enhance CH_4 production in sediment incubations from lakes in the
296 region. This study did not pinpoint the exact mechanisms that led to the observed salinity
297 dependence of CH_4 production, though it demonstrates that the net sediment CH_4 production that
298 underpins both diffusive and ebullitive emissions is heavily dependent on general ecosystem
299 salinity, in a non-linear manner consistent with our empirical observations (Fig. 1c). Possible
300 insight into the mechanisms underlying NaCl impacts in incubations by Camacho et al.¹⁶ may be
301 taken from a similar incubation experiment where NaCl additions to Australian wetland sediment
302 also restricted net CH_4 production through multiple mechanisms¹⁷. There, NaCl availability
303 liberated reduced compounds (Fe^{2+} , Mn^{2+} , NH_4^+) that could stimulate anaerobic CH_4 oxidation
304 (detailed below), while suppressing methanogenesis¹⁷. Taken together, these studies
305 demonstrate that even in non- SO_4^{2-} dominated systems in other global regions, increases in
306 salinity can suppress CH_4 emissions even without the direct modification of SO_4^{2-} loading.
307 Further, the examples outlined here clearly demonstrate that salinity as a metric provides critical
308 information about ecosystem CH_4 cycling. Had we assumed that CH_4 production only proceeds
309 in SO_4^{2-} -rich environments and does not extend to hardwater systems with comparatively low

310 SO_4^{2-} content, we might erroneously assume elevated CH_4 content and emissions from SO_4^{2-} -
311 poor systems that apparently have other factors (discussed below) that may be regulating
312 emissions.

313

314 **Salinity captures diverse drivers of microbial control (redox, salt stress):**

315 Salinity is useful because it integrates many complex processes at once that may interact to shape
316 CH_4 cycling. Elevated salinity typically reflects an increase in ecosystem pH, which can
317 dramatically shift the redox state of a given habitat toward more oxidizing conditions^{18,19}. More
318 saline shallow systems are typically prone to partial or complete desiccation (seasonally or for
319 extended periods) that also expose shallow sediment layers to the atmosphere and shift these
320 habitats to oxidizing conditions that may favour aerobic degradation of organic matter while
321 inhibiting methanogenesis¹⁶. Limits on methanogenesis in sediments due to redox properties are
322 closely related to SO_4^{2-} concentrations²⁰, which can be directly or indirectly¹⁵ enhanced at
323 elevated salinity. Competition for binding sites can liberate NH_4^+ or Fe^{2+} when ions including
324 Na^+ are abundant with increasing salinity (ref.¹⁷ and references therein). As salinity increases, the
325 direct impacts of osmotic stress on bulk microbial communities are complex¹⁷, and halotolerant
326 organisms can increasingly dominate and sustain microbial metabolism. Yet some archaeal
327 methanogens are sensitive to salt stress and show decreased metabolic activity with increasing
328 salinity as NaCl, even at low concentrations ($500 \mu\text{S cm}^{-1}$ specific conductance)¹⁷.

329

330 **The role of diverse elements in anaerobic oxidation of CH_4 (AOM):**

331 Enhanced AOM may be another mechanism restricting sediment CH₄ release in hardwaters.
332 Recent discoveries suggest that the pathways supporting AOM may be more diverse in inland
333 waters than marine systems²¹ (but see ref.²²). As reviewed elsewhere^{23,24} the prevailing
334 assumption has been that SO₄²⁻ was the only oxidant that fuelled AOM in inland waters. Intense
335 AOM has been documented in inland water anoxic habitats^{23,24} and importantly restricts surface
336 water CH₄ content and emissions. At present, the biochemical pathways and organisms that drive
337 inland water AOM are not well established, with new organisms and mechanisms discovered in
338 recent years (e.g.,²⁵⁻²⁷), and unidentified controls yet to be established²⁴. However, it is now
339 abundantly clear that the availability and involvement of non-S-based forms of oxidized
340 compounds, including trace metals must be considered in many cases to account for elevated
341 AOM rates that cannot be accounted for by the reduction of SO₄²⁻ alone²⁴. As reviewed
342 elsewhere^{23,24,28}, microbial consumption of CH₄ has now been shown to be coupled to the
343 reduction of diverse electron acceptors that to date are known to include Mn(IV) (ref.^{21,25}), Fe³⁺
344 (ref.²¹), NO₂⁻ and NO₃²⁻ (ref.^{21,27}), Cr(VI)²⁶, and complex dissolved organic matter²⁹. New
345 research in marine environments also suggests that considerable AOM may in some cases
346 proceed independently of SO₄²⁻ reduction²². These discoveries are important in the context of
347 using salinity to predict CH₄ cycling, because gradients of salinity reflect the abundance of
348 substrates that are now known to act as electron acceptors in diverse AOM pathways, including
349 minerals rich in metals and trace metals^{1,14}, and the liberation of ionized, reduced N and Fe from
350 sediment that are precursors for substrates used in AOM (e.g., ref¹⁷) under saline conditions.
351 Furthermore, and of relevance to hardwater ecosystems, elevated surface water N content is both
352 a natural and anthropogenically enhanced feature in these landscapes^{16,30,31}. Additional evapo-
353 concentration of this N alongside other important substrates may provide substrate to drive

354 AOM. Consequently, the availability of N could be heavily implicated in AOM through diverse
355 pathways (detailed above) that decouple the process from S cycling and may partially account
356 for different empirical relationships between metrics of CH₄ cycling and salinity versus SO₄²⁻
357 alone. When we take all these lines of evidence together, it is clearly possible that inverse
358 correlations between salinity and CH₄ content and emissions rates may partly reflect an
359 increasing importance of a diversity of AOM pathways restricting net CH₄ production that may
360 not always be captured in regressions with SO₄²⁻ content alone. While more focused research is
361 required, this proposed mechanism may partially explain why empirical relationships with
362 salinity were stronger than SO₄²⁻ content as predictors of pCH₄ in the agricultural reservoirs (Fig.
363 S6), and why salinity remains a strong predictor of pCH₄ in other regions with diverse
364 geologies^{16,17} (Fig. 2).

365

366 **Salinity can alter abiotic conditions that modify CH₄ cycling:**

367 Elevated salinity can importantly modify the physico-chemical conditions in surface water (e.g.,
368 alkalization) and sediment layers (e.g., deposition of complex mineral precipitates) in complex
369 ways that lead to suppressed CH₄ release. As salinity increases, individual ions (not necessarily
370 SO₄²⁻) with greater binding capacities can replace N bound in sediment complexes, and liberate
371 NH₄⁺ which is a precursor to important terminal electron acceptors (NO₃²⁻ and NO₂⁻) that fuel
372 AOM (detailed above). In hardwater environments, complexation and precipitation of substrates
373 (metals, nutrients, organic matter) may subsequently lead to greater sediment liberation and
374 availability of these elements under reducing conditions (e.g., for Fe and Mn (ref.¹ and references
375 therein), or SO₄²⁻ (ref. ¹⁵). These effects can enhance AOM in surface sediments, which
376 ultimately plays an important role in lowering ecosystem emissions rates^{23,24}.

377 Elevated ion content may restrict the delivery of fresh particulate organic matter needed
378 to fuel sediment methanogenesis²⁰. Under elevated ionic content and alkaline conditions, surface
379 water productivity can also be restricted due to impacts on nutrient and micronutrient availability
380 and possibly other factors³²⁻³⁵. The biomass of phytoplankton can be dramatically lower than
381 expected based on total nutrient content, even at sub-saline conditions^{32,33}. The mechanism(s)
382 restricting primary production are complex, in some cases related to Fe complexation with
383 organic matter that restricts bioavailable Fe for phytoplankton growth³⁴. Intense deficiencies in
384 surface water PO_4^{3-} (ref.³²) and dissolved inorganic N³⁴ have been observed despite elevated TP
385 and TN content. Limitations to N-fixation and bioavailable N production (ref.³⁴ and references
386 therein) as well as complexation of P with Ca^{2+} ¹⁷, DOM and metals³⁵ may lead to bio-available
387 macronutrient deficiencies and limited primary production. As salinity increases, constraints on
388 primary production can have negative effects on fresh particulate organic matter deposition and
389 organic substrate provision to sediment methanogens (which are not necessarily reflected in bulk
390 DOC measurements that generally scale positively with CH_4 content in our regressions (Fig. 1,
391 Table 1)). As such, salinity is a useful empirical predictor of CH_4 content in part because it
392 reflects the geochemical limitations that can be imposed on primary producer growth and organic
393 matter supply to methanogens.

394

395 **Supporting Text S2.** Analysis exploring the importance of within-site, cross-season, and cross-
396 site replication for improving estimates of pCH₄.

397 **Overview:**

398 The objective of this analysis was to evaluate the relative importance of sources of uncertainty in
399 our estimates of CH₄ content in small lentic systems in the Canadian Prairies, thereby informing
400 the validity of our approach in the regional upscaling calculations. The upscaling calculations are
401 conservative, first order approximations of the error that could be induced from using previous
402 empirical relationships to estimate CH₄ emissions, versus the models generated in this paper
403 using salinity as a key predictor. Our expectation was that between-site variability was more
404 important to constrain than within-site variability in the context of regional upscaling, an
405 expectation that was confirmed in this analysis.

406

407 **Methods:**

408 To assess the relative importance of within-site, cross-season, and cross-site variation to
409 precision of site-level pCH₄ estimates, we used the peripheral wetland dataset where surface
410 water pCH₄ was quantified alongside a minimal subset of other environmental parameters in
411 2021 in 47 sites across Alberta (AB), Saskatchewan (SK), and Manitoba (MB; 15-16 sites per
412 province). Wetland ponds were sampled between two and 11 times (most sites sampled five
413 times) over the open-water season and a subset of 18 ponds were sampled at four locations
414 within the open-water area on each sampling occasion for a total of 472 measurements of pCH₄
415 over 198 unique site-date combinations.

416 We used a lognormal generalized linear mixed model (residual analyses supported this
417 choice) with a fixed effect of province (AB, SK, or MB) and province-specific random effects of
418 site(province), station(site), and date(station) to examine the relative importance of spatial and
419 temporal sources of variation. Using model outputs, we then ran simulations to quantify how
420 reallocating sampling effort would influence the precision of estimates of pCH₄. Because station-
421 to-station variance was estimated to be zero for each province, we only compared effects of
422 reallocation of sampling effort to number of sites vs frequency of site visits. For all simulation
423 runs, total sampling effort was constrained to be close to total 2021 efforts (n = 468 to 471).
424 Simulated pCH₄ values were generated from province-specific fixed estimates (i.e., βX) plus site-
425 level and within-site error realizations by province (drawn from normal distributions with mean
426 = 0 and variance = Site(Province) and within-site residual estimates, respectively). We used n =
427 30 simulation runs per sampling allocation scenario as results were consistent with larger-scale
428 trials. We then re-ran lognormal generalized linear mixed models for all simulation iterations to
429 obtain estimated marginal means and associated standard errors for the effect of province. We
430 then calculated average (i.e., mean) estimated marginal means and standard errors across all 30
431 simulation runs. We compared the standard error of estimated marginal means at the province
432 level across two sampling scenarios, with reductions in standard error indicating value for
433 improving the precision of pCH₄ estimates at broad (i.e., provincial) scales.

434

435 **Findings:**

436 As summarized in Table S2, the station-to-station variance (i.e., variance in pCH₄ between
437 samples from multiple locations within a site on a single sampling occasion) was estimated to be
438 zero. Thus, sampling from multiple locations within a given wetland likely would not have

439 improved pCH₄ estimates in the primary dataset used to predict regional-scale patterns of CH₄
440 cycling. Except in Alberta, variance associated with sites was greater than variance associated
441 with dates. Because the station variance component was zero, we focussed simulations on trade-
442 offs between number of sites sampled and frequency of sampling. Given a constant total
443 sampling effort, we estimated that there would be a 21-25% reduction in standard error achieved
444 by doubling the number of basins sampled per province from 16 to 32 (at the expense of fewer
445 measurements per basin; Tables S2,S3). We would expect a 30% reduction if all variation was
446 attributed to basin-to-basin differences. Thus, the greatest improvements in precision and
447 reduction in uncertainty of pCH₄ estimates at broad-scales are made through sampling more
448 wetland systems less intensively (as in our primary dataset) compared with fewer wetlands
449 sampled more frequently.

450 Ultimately, this exercise identifies that by sampling multiple small lentic systems
451 (wetlands and ponds) across long environmental gradients (salt content, trophic status, etc.) as
452 we have done and presented in figures 2 and 3, we have constrained more of the regional
453 variance in surface water CH₄ content and thus emissions, than had we invested more resources
454 in sampling multiple locations per system, at the cost of broader spatial coverage and inter-site
455 sampling. Thus, while previous studies recommend highly-intensive sampling of a single
456 ecosystem to fully constrain annual emissions patterns in a single ecosystem, it is clear that a
457 lower-resolution approach that prioritizes multiple sites and repeated temporal sampling is a
458 better approach where the aim is to constrain regional variability in emissions budgets.

459 While the goal of our paper was not to provide a completely refined emissions budget for
460 Prairie Canada, this analysis does additionally provide an important road map for future research.
461 The findings from this simulation exercise will help to guide study design where the aim is to

462 minimize the uncertainty in the overall, annual-scale emissions budget for lentic ecosystem CH₄
463 emissions at the regional scale.

464

465

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