IDECAT WP3 Seminar

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High pressure XPS: A tool for the investigation of heterogeneous catalytic processes

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Why in situ XPS ?

- Many processes cannot be investigated in UHV: "Pressure Gap"
 - environmental chemistry
 - catalysis
 - corrosion
 - electrochemistry
 - biological samples
- Very few methods can investigate the solid-gas interface at high pressures
 - non-linerar optics (SFG, SHG)
 - scanning probe microscopies
 - X-ray diffraction
- Photoelectron spectroscopy is very powerful
 ⇒ Goal: XPS at pressures of at least 5 torr



In situ XPS: obstacles

Fundamental limit:

elastic and inelastic scattering of electrons in the gas phase



<u>Technical issues:</u> - Differential pumping to keep analyzer in high vacuum - Sample preparation and control in a flow reactor

In situ XPS: basic concept



- Photons enter through a window
- Electrons and a gas jet escape through an aperture to vacuum

In situ XPS instruments: previous designs



- H. Siegbahn et al. (1973-)
- M.W. Roberts et al. (1979)
- M. Faubel et al. (1987)
- M. Grunze et al. (1988)
- P. Oelhafen (1995)



In situ XPS using differentially pumped electrostatic lenses



D.F. Ogletree, H. Bluhm, G. Lebedev, C.S. Fadley, Z. Hussain, M. Salmeron, Rev. Sci. Instrum. 73 (2002) 3872.

Close-up of sample-first aperture region



In situ XPS system



Introduction





Selectivity issue: what defines selectivity?

Summary

- *1. <u>Subsurface H</u>*: effective for alkene-to-alkane but also for alkyne-to-alkane transformation
- 2. <u>Surface H</u>: could be selective (spillover)
- *3. <u>Different reaction orders</u>* in the different selectivity regimes & Abrupt changes between regimes
- 4. <u>Cuptake</u> is significantly more in the selective regime

Reaction in the mbar p region (in-situ XPS)

	5% Pd/CNT	3% Pd/Al ₂ O ₃	Pd foil	Pd(111)
Conversion [%]	~ 10	~5	~2.5	<1
Selectivity Pentene [%]	~95	~80	~98	100
Selectivity Pentane [%]	~5	~20	~2	-

Recation conditions: C5/H2 = 1:9, 1 mbar, 358 K



In-situ XPS: Pd 3d depth profiling



In-situ XPS: C1s (Switching off experiments)



In-situ XPS: Pd 3d (Switching off experiments)



In-situ XPS: Pd vs. C depth profiling



Summary

- *1. <u>Subsurface H</u>*: effective for alkene-to-alkane but also for alkyne-to-alkane transformation
- 2. <u>Surface H</u>: could be selective (spillover)
- *3. <u>Different reaction orders</u>* in the different selectivity regimes & Abrupt changes between regimes
- 4. <u>Cuptake</u> is considerably more in the selective regime
- 5. <u>Pd-C surface phase</u> forms in the early stage of selective pentyne hydrogenation & there is significant amount of <u>subsurface C</u> below of it

Model (during the reaction)



Summary

- *1. <u>Subsurface H</u>*: effective for alkene-to-alkane but also for alkyne-to-alkane transformation
- 2. <u>Surface H</u>: could be selective (spillover)
- *3. <u>Different reaction orders</u>* in the different selectivity regimes & Abrupt changes between regimes
- 4. <u>Cuptake</u> is considerably more in the selective regime
- 5. <u>Pd-C surface phase</u> forms during selective hydrogenation of pentyne & there is significant amount of <u>subsurface C</u> below of it
- 6. <u>Dynamic</u> behaviour of Pd-C and subsurface C

Innovative Station for In Situ Spectroscopy A project of BESSY and the Dep. Inorganic Chemistry, Fritz-Haber-Institut

Installation of a beamline exclusively used for in situ spectroscopy in the soft X-ray range

Installation of infrastructure optimized for these kind of experiments on site (e.g. chemical lab, gas supply, gas analytics)

Later, further implementation of other in situ spectroscopy techniques: multi wavelength Raman, UV-Vis, fluorescence yield ?!

Start of user operation of the beamline: 2007

MAX-PLANCK-GESELLSCHAFT

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