



Structural and Catalytic Investigation of Active Site Isolation in Pd-Ga Intermetallic Compounds



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Introduction

Palladium in catalysis

Acetylene hydrogenation to ethylene ($C_2H_2 + H_2 \rightarrow C_2H_4$) is a common method to remove traces of acetylene in the ethylene feed for the production of polyethylene [1, 2]. Conventional supported Pd catalysts show high activity but only limited selectivity and limited stability.

Increase of selectivity

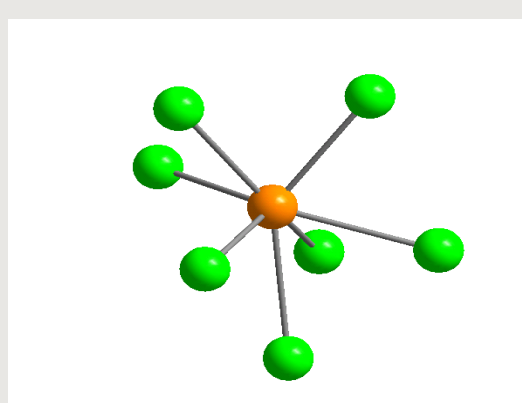
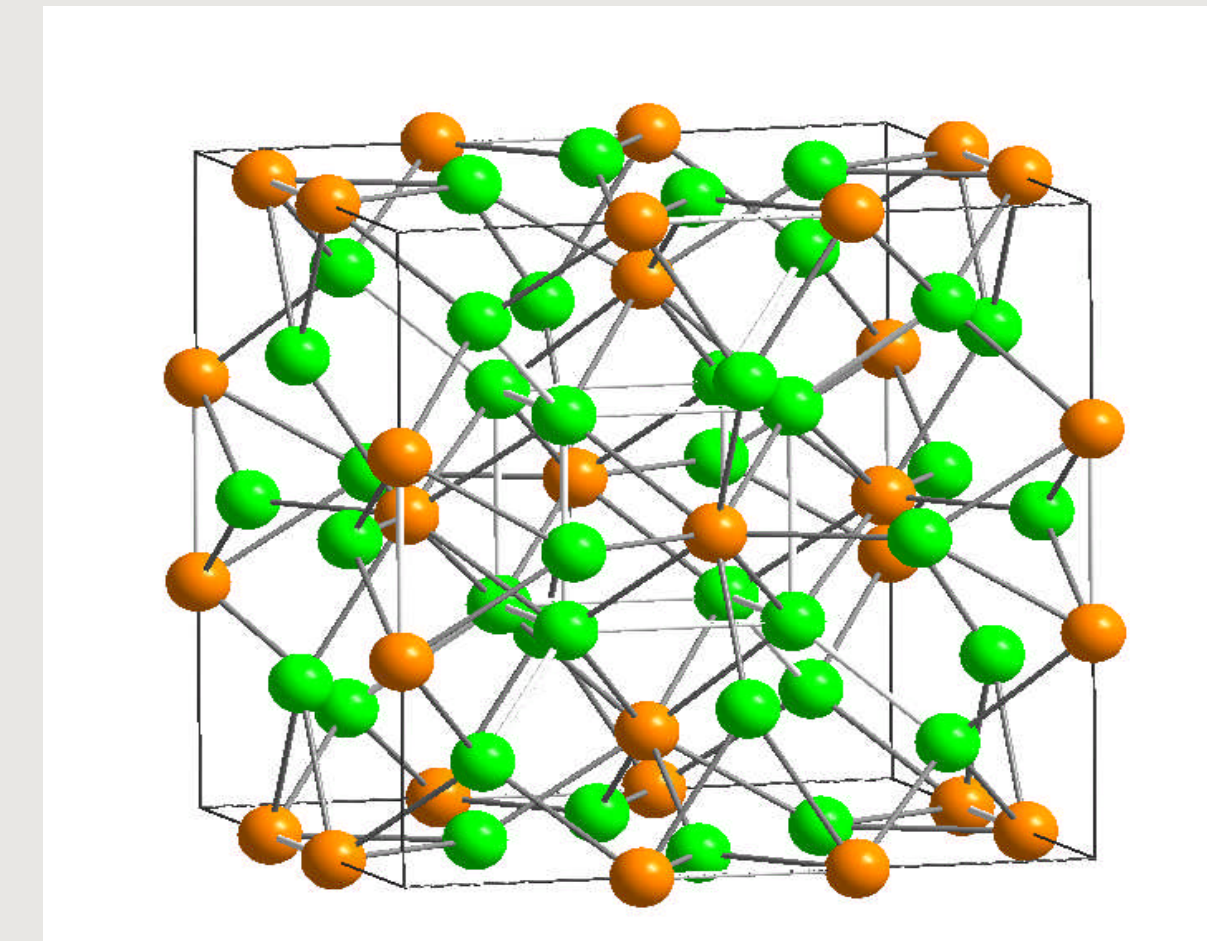
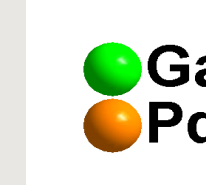
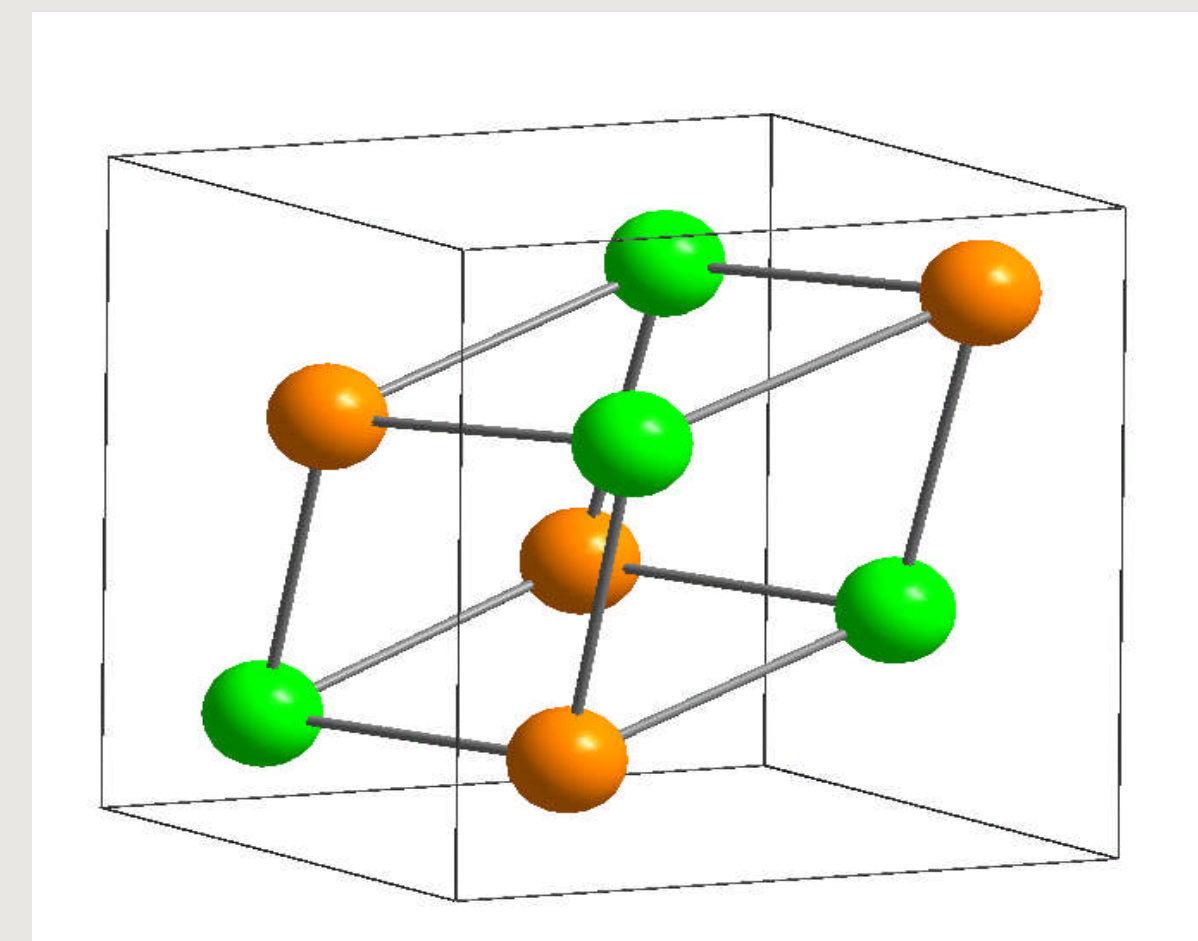
Active site isolation [3-5].
Elimination of hydride formation [6-8].

Palladium intermetallic compounds

Pd-Ga intermetallic compounds are particularly interesting as potential catalysts because of the isolation of Pd atoms in the structure.

Goal

Determine thermal stability in different gas atmospheres with in situ XRD, in situ XAS measurements and thermal analysis, investigation of selectivity and reactivity for catalytic hydrogenation of acetylene.



PdGa

Pd - Ga (1x): 2.54 Å
Pd - Ga (3x): 2.57 Å
Pd - Ga (3x): 2.71 Å
Pd - Pd (6x): 3.01 Å

Space group [9]: P 2₁ 3 (198) - cubic

Pd₃Ga₇

Pd - Ga (4x): 2.58 Å
Pd - Ga (4x): 2.58 Å
Pd - Pd (1x): 2.73 Å

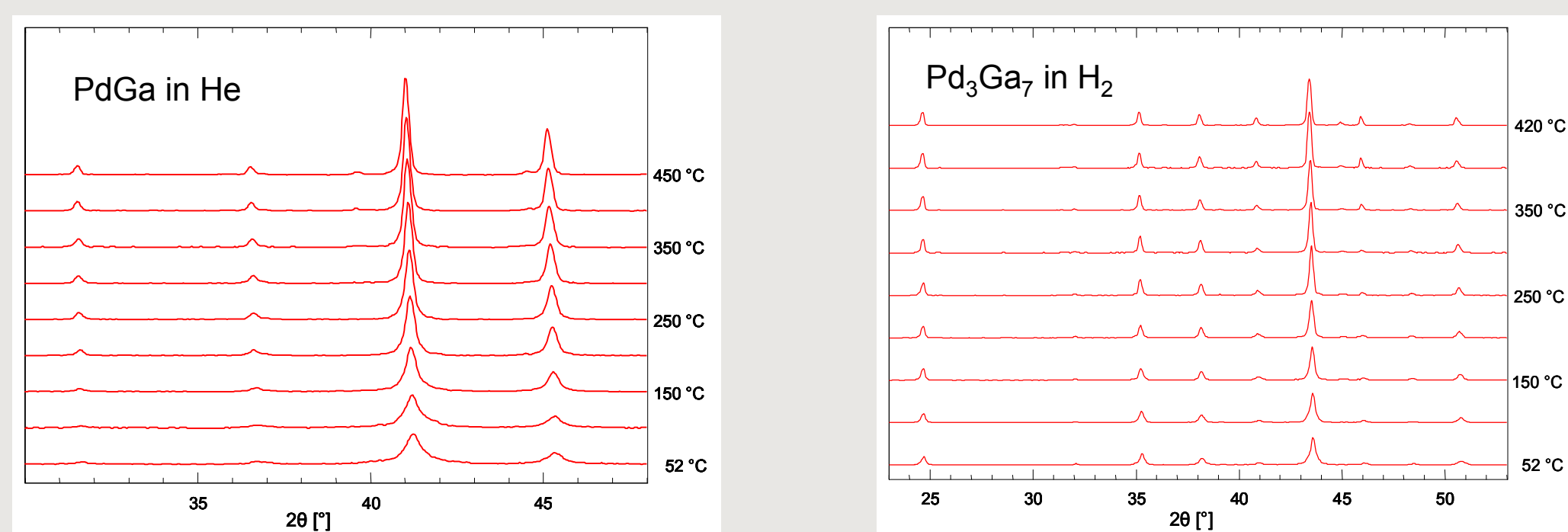
Space group [10]: I m -3 m (229) - cubic

Preparation

$m \text{ Pd} + n \text{ Ga} \rightarrow \text{Pd}_m\text{Ga}_n$ by melting of appropriate amounts of Pd and Ga (1:1 and 3:7) in a glassy carbon crucible under Ar atmosphere in a high frequency induction furnace. The samples were powdered in a ball mill.

In situ XRD

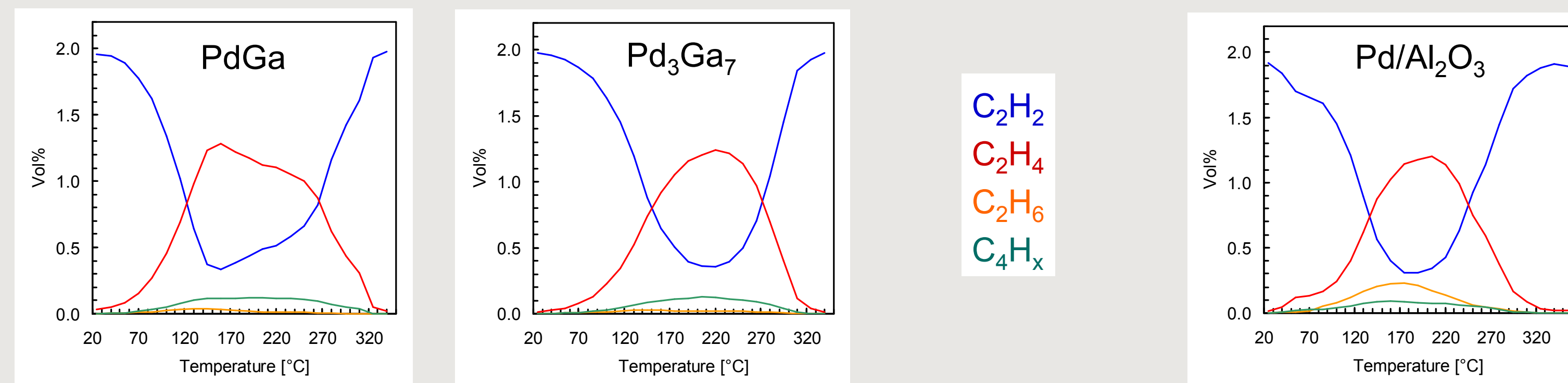
Determination of the structural stability in helium and hydrogen using in situ XRD. Only thermal lattice expansion and no phase transition was observable. Increasing particle size lead to decreasing line width.



In situ XRD measurements were conducted using a STOE diffractometer with Cu-K α radiation in Bragg-Brentano geometry (secondary monochromator) equipped with a Bühler HDK chamber.

Catalysis: $C_2H_2 + H_2 \rightarrow C_2H_4$

Acetylene hydrogenation with PdGa, Pd₃Ga₇ and reference Pd/Al₂O₃ in 2% C₂H₂ and 4% H₂. Both, PdGa and Pd₃Ga₇, show activity and high selectivity.

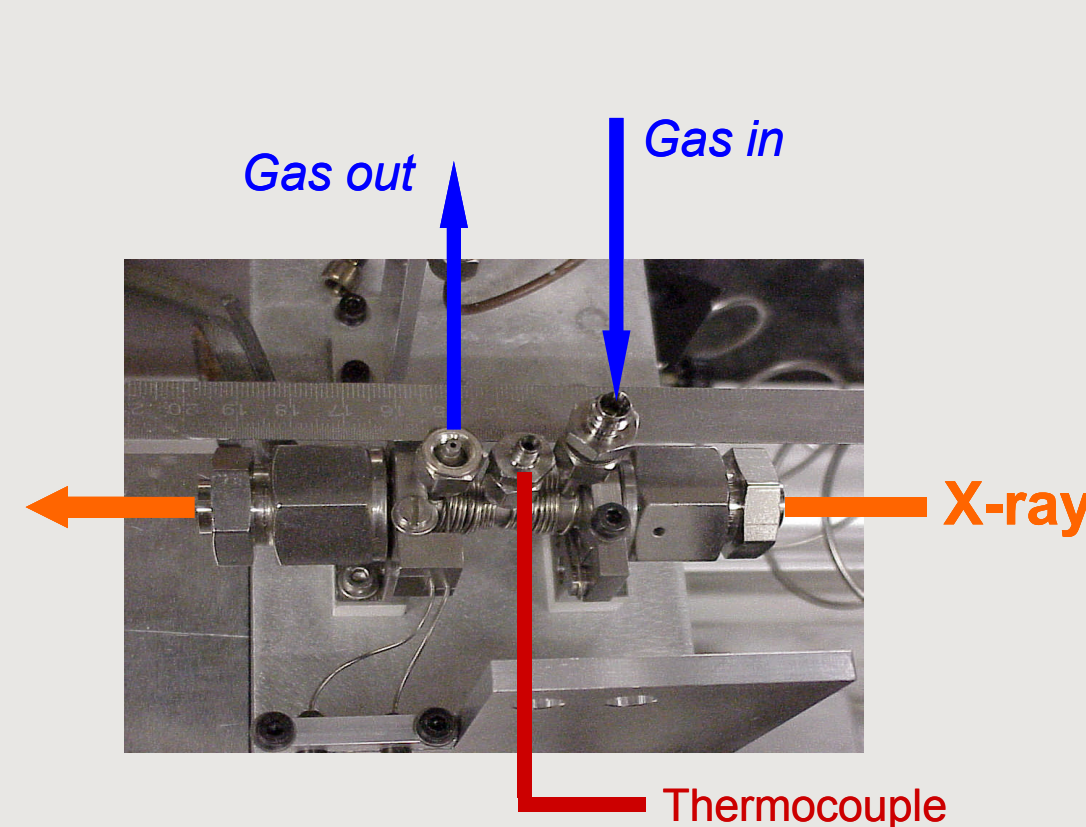


Catalysis studies were performed in a plug flow reactor (C₂H₂ / H₂ ratio 1:2, total flow 30 ml/min) with gas phase analysis using MicroGC Varian CP4900. The amount of catalyst was PdGa (50 mg), Pd₃Ga₇ (100 mg) and reference Pd/Al₂O₃ (0.5 mg, 5 wt%). The red curve shows the concentration of the product ethylene, yellow shows the total hydrogenated ethane and green the dimerisation products.

In situ EXAFS

In situ XAS in hydrogen and during acetylene hydrogenation is used for determination of the thermal stability and the detection of a possible hydrogen inclusion. In contrast to in situ XRD which provides structures and lattice parameters of crystalline phases EXAFS gives information about single atomic distances in crystalline and non-crystalline phases.

Radial distribution function of PdGa in hydrogen and PdGa and Pd₃Ga₇ in acetylene hydrogenation. Diagrams below show selected refined distances and Debye-Waller factors. In situ EXAFS of PdGa and Pd₃Ga₇ shows high thermal stability in hydrogen and acetylene hydrogenation with changes in distances ΔR in the range of the accuracy and increasing Debye-Waller factor with increasing temperatures.

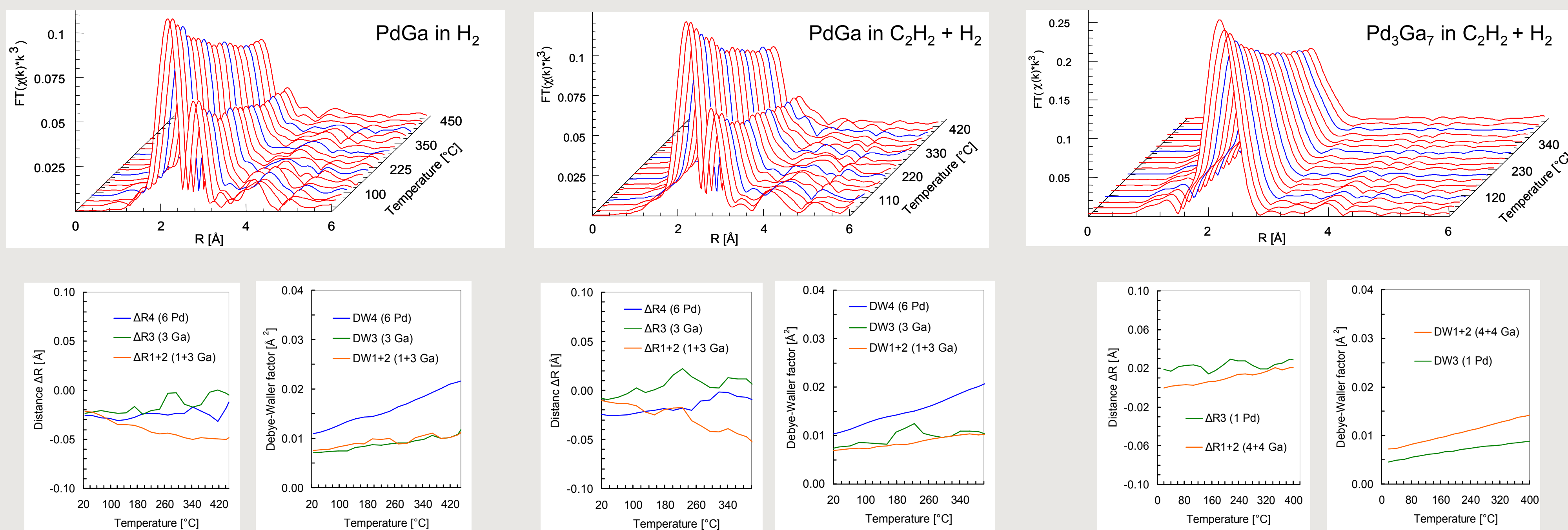


Cell for in situ XAS studies EXAFS measured at HASYLAB X1 (Hamburg) at Pd K-edge (24.35 keV).

The theoretical EXAFS functions were calculated with FEFF 8 [11] from crystallographic data and refined with WinXAS [12] in R space.

Cell parameters:
Cell volume: 4 ml
Sample diameter: 5 mm pellet
Cell windows: Al foil
Gas in: Gas flow controller
Gas out: Exhaust with MS detection

Reaction parameters:
Sample mass: 9-11 mg
Diluent: 30 mg BN
Gas flow: 30-40 ml/min
Heating rate: 6 K/min



Results

Bulk characterisation of PdGa + Pd₃Ga₇:

High thermal stability under different atmospheres and no hydride formation detectable.

Catalytic studies of PdGa + Pd₃Ga₇:

The Pd-Ga alloys show activity and selectivity for hydrogenation reactions. The selectivity for the hydrogenation of acetylene to ethylene is higher compared to the conventional catalyst Pd on Al₂O₃.

Outlook

- Investigation of PdSn₂
- Preparation of high surface area samples (nano particles)
- Surface investigation with XPS, IR and ISS

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Literature

1. F. Arnold, F. Döbert, and J. Gaube in "Handbook of heterogeneous catalysis" (G. Ertl, H. Knözinger, and J. Weitkamp, Eds.) 2165, VCH Weinheim (1997)
2. A. Molnar, A. Sarkany, and M. Varga, J. Mol. Catal. A 173, 185 (2001)
3. E.W. Shin, C.H. Choi, K.S. Chang, Y.H. Na, and S.H. Moon, Catal. Today 44, 137 (1998)
4. S. Leveness, V. Nair, A.H. Weiss, Z. Schay, and L. Gucci, J. Mol. Catal. 25, 131 (1984)
5. V. Ponec, Adv. Catal. 32, 149 (1983)
6. W. Palczewska in "Hydrogen Effects in Catalysis" (Z. Paal, P.G. Denon, Eds.) 372, Marcel Dekker New York (1988)
7. G.C. Bond, and P.B. Wells J. Catal. 5, 65 (1966)
8. A.M. Doyle, S.K. Shaikhutdinov, S.D. Jackson, and H.J. Freund, Angew. Chem. Int. Ed. 42, 5240 (2003)
9. E. Hellner, F. Laves, Z. Naturforsch. 2a (1947) 177-183
10. H. Pfisterer, K. Schubert, Z. Metallkunde 41 (1950) 433-441
11. A.L. Ankudinov, B. Ravel, J.J. Rehr, and S.D. Conradson, J. Phys. Rev. B 58, 7565 (1998)
12. T. Ressler, J. Synchr. Rad. 5, 118 (1998)