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Oral contribution

## Styrene Synthesis: High Yield over Unpromoted Iron Oxide Model Catalysts

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Dehydrogenation of ethylbenzene to styrene is usually run over potassium promoted iron oxide based catalysts at 870 K in presence of steam. Here we present conversion yield measurements on unpromoted single crystalline  $\alpha$ - $\text{Fe}_2\text{O}_3$  (0001) model catalysts by combining surface science techniques with an in-situ micro flow reactor (Fig.1). The influence of  $\text{H}_2\text{O}$  and  $\text{O}_2$  on the reaction was investigated by varying the composition of the feed. The initial conversion over  $\text{Fe}_2\text{O}_3$  is always high (5-8%, Fig. 2), independent of the type of the feed composition. Only the length of this period depends on the feed composition. In presence of  $\text{O}_2$  (EB: $\text{H}_2\text{O}$ : $\text{O}_2$  = 2:20:1), the high yield period can be maintained, in absence of  $\text{O}_2$  (EB:  $\text{H}_2\text{O}$  = 1:10) it decreases in two steps of about a factor of 2-3 each. The reaction was interrupted in the different yield regimes, and the sample structure and composition was analyzed. The high yield is related to  $\text{Fe}_2\text{O}_3$  with almost no carbon deposits.  $\text{O}_2$  in the feed maintains this phase. Without  $\text{O}_2$ ,  $\text{Fe}_2\text{O}_3$  is reduced to  $\text{Fe}_3\text{O}_4$  and the yield drops to the intermediate region. The same yield is observed on clean  $\text{Fe}_3\text{O}_4$ . Carbon deposits increase but do not yet limit conversion. This happens at the transition to the low yield regime where a thick layer of carbon deposits is observed. With  $\text{H}_2\text{O}$  in the feed, the oxide below the carbon deposits remains  $\text{Fe}_3\text{O}_4$ , without  $\text{H}_2\text{O}$ , it is reduced to metallic Fe. We ascribe the low yield to catalysis by carbonaceous species. The study shows that the high yield is typical for  $\text{Fe}_2\text{O}_3$  and can be maintained by proper admixture of  $\text{O}_2$  to the feed.

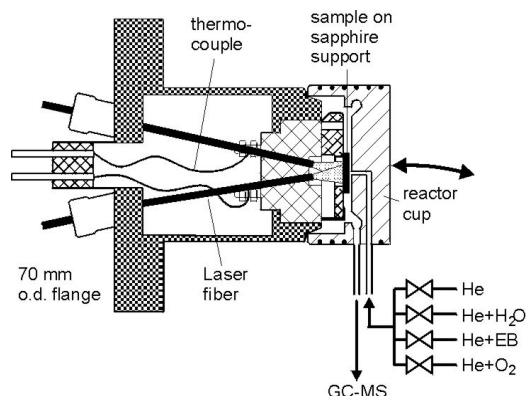


Fig. 1: Micro flow reactor

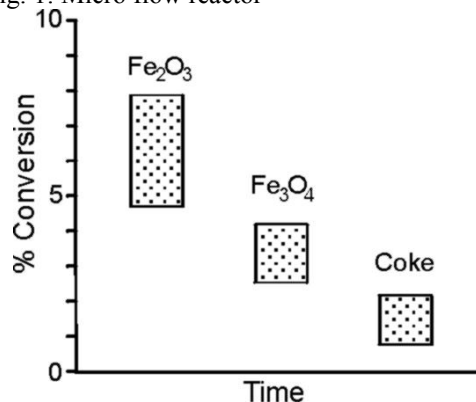


Fig. 2: Styrene yield over  $\text{Fe}_2\text{O}_3$  and its decrease by reduction to  $\text{Fe}_3\text{O}_4$  and by coking, schematically.