## TEM investigation on the defect structures of molybdenum oxides

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Although it is well known that heterogeneous catalysis is a process at the gas/solid interface and therefore it belongs to the surface science, but the bulk of catalyst has drawn more attention in the past. This is not only due to the fact the bulk determine the structure of the surface, but also due to the fact that lattice atoms are believed to be incorporated in the catalytic process. For instance, in many catalytic reactions using MoO<sub>3</sub> as a catalyst, lattice oxygen can be incorporated into the products leaving the catalyst reduced. Planar defects such as array of vacancies, cluster defects such as non-stoichiometric or sub-oxides are expected. The open question is, are the defects the driving force for the catalytic reaction or its byproducts? As the first attempt to answer the question, we use high-resolution TEM, supported by EELS, EDS and image simulation, to study the defect structures of molybdenum oxides. As references, we start at first with the microstructural characterisation of well-defined suboxides Mo<sub>18</sub>O<sub>52</sub> and Mo<sub>8</sub>O<sub>23</sub>. Works on the defect mechanism in molybdenum oxides during hydrogen reduction and during thermal decomposition will be presented.

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