Comment on "Anomalous Mobility of Strongly Bound Surface Species: Cl on GaAs(001)- $c(8 \times 2)$ "

In a recent Letter, McLean et al. [1] reported anomalously fast diffusion of Cl adatoms on the Ga-rich GaAs(001) surface. They presented a systematic scanning tunneling microscopy (STM) study as well as Hartree-Fock cluster calculations. Their analysis was based on a surface model (the β 2 structure) that recently has been shown to be less stable than the new ζ structure [2]. The ζ structure was found by density-functional theory (DFT) calculations [2] and confirmed by experimental analyses [2,3]. The geometry of the ζ model is so very different from the originally assumed one [compare Fig. 1(a) with Fig. 1(c) in Ref. [1]] that we decided to reanalyze the experimental results of McLean et al. and to perform calculations for the Cl adsorption and diffusion. We used DFT, the generalized gradient approximation, and normconserving pseudopotentials. The electronic wave functions were expanded in plane waves (cutoff 12 Ry). The surface was modeled by a (4×4) supercell, and in terms of the Brillouin zone of a (1×1) cell we used 64 k points. Figure 1(a) [cf. also Fig. 1(b)-(left)] shows that the ζ structure has at the center of the F channels one Ga dimer per unit cell in the surface, and in addition, there are two subsurface Ga dimers (significantly below the top layer). The calculated adsorption geometry for a single Cl adatom is displayed in Fig. 1(b)-(middle) ($E_1^{\text{ad}} = 1.06 \text{ eV}$

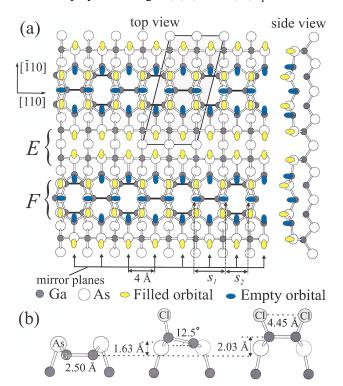


FIG. 1 (color). (a) Top and side views of the GaAs(001)- $c(8\times2)$ ζ surface. The unit cell is noted by the dashed parallelogram. (b) Side views of different Ga dimers: clean (left), with one Cl adatom (middle), and with two Cl adatoms (right).

with respect to the energy of $1/2\text{Cl}_2$). We therefore identify the S features in the STM data of Ref. [1] with this situation. Note that the Cl adatom lifts the in-surface Ga dimer from its initial position to above surface without an activation energy barrier. With an energy barrier of $E_1^b = 0.47 \text{ eV}$, the Cl adatom can hop from one side of the dimer to the other one ($s_1 = 4.45 \text{ Å}$). For a hop along [110] from one dimer to the next one, the calculated distance and energy barrier is $s_2 = 3.72 \text{ Å}$, and $E_2^b = 0.87 \text{ eV}$. Assuming the prefactor ν_0 as 10^{12} Hz [1], the experimentally deduced values are $E_1^{b\text{-exp}} = 0.74 \text{ eV}$ ($s_1^{\text{exp}} = 4.5 \pm 0.3 \text{ Å}$) and $E_2^{b\text{-exp}} = 0.80 \text{ eV}$ ($s_2^{\text{exp}} = 3.5 \text{ Å}$) along [110]. We cannot rule out that the part of the difference between E_1^b and $E_1^{b\text{-exp}}$ is due to approximations in our theory, but we also note that it would vanish if the attempt frequency for the experimental analysis of the s_1 hop would be 10^8 Hz instead of 10^{12} Hz . Additional experimental work on this question would be valuable.

In the ζ structure, there is only one row of *in-surface* Ga dimers [along the F channel cf. Figure 1(a)]. The older model (the β 2 structure), that was considered by McLean et al., has two surface Ga dimers parallel to each other (distance 4 Å). Therefore they hypothesized a very rapid hopping (> 1 kHz) along $[\overline{1}10]$ between two neighboring dimers of the β 2 structure. However, alternatively, this experimental result could be interpreted as evidence for no (or very slow) hopping, which corresponds to $E^{b-\exp}$ 0.95 eV (assuming $\nu^{\rm STM}$ < 10⁻⁴ Hz and ν_0 = 10¹² Hz). Our DFT study gives $E^b = 0.98$ eV along [$\tilde{1}10$], which is consistent with the slow (experimentally not observed) diffusion along this direction. This is another strong hint that the reconstruction present in the experimental work [1] was indeed ζ and not β 2. We also find that a second Cl adatom can bind to a Ga dimer, that binds one Cl already. The adsorption energy of the second Cl adatom is higher than that of the first one ($E_2^{\text{ad}} = 1.52 \text{ eV}$ with respect to 1/2Cl₂), and the Cl-Cl distance is 4.45 Å [cf. Figure 1(b)-(right)]. We identify this geometry with the D features in the STM image [1]. Our DFT calculations show that two-Cl-adsorbed geometry is energetically favorable. However, a clustering of such Cl-pair structures is energetically unfavorable by 0.14 eV/Cl₂.

Seung Mi Lee, Sung-Hoon Lee, and Matthias Scheffler Fritz-Haber-Institut der Max-Planck-Gesellschaft Faradayweg 4-6, D-14195 Berlin, Germany

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