

JOURNAL OF PHYSICAL CHEMISTRY B 103 (31): 6407-6420 AUG 5 1999



Multiple-scattering EXAFS analysis of tetraalkylammonium manganese oxide colloids Ressler T, Brock SL, Wong J, Suib SL

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Abstract:

X-ray absorption spectroscopy at the Mn K edge was employed to elucidate the structure of colloidal tetraalkylammonium (TAA) manganese oxides in sols and gels obtained by different preparation and heat treatment procedures. Two series of colloidal TAA MnOx prepared with tetrapropylammonium (TPA) and tetraethylammonium (TEA) cations were studied. Several manganese oxides, birnessite, and feitknechtite were also measured and served as model compounds for structural refinements. Near edge structure (XANES) analysis revealed different average valences of the colloidal systems. As synthesized and heat-treated, TAA colloids exhibited an average valence of 3.6-3.7, whereas gelled TAA colloids showed a lower average valence of similar to 3.5. Extended absorption fine structure (EXAFS) analysis was carried out to distances of -6.0 Angstrom around the central Mn atom using theoretical backscattering phases and amplitudes calculated from the ab initio FEFF code. All multiple-scattering (MS) paths with a weight of 2% and more with respect to the main Fourier transform peak were included in the refinement. It is found that, except for gelled TAA samples, no significant amount of corner-shared MnO6 units was detected in the colloidal systems. Theoretical EXAFS phases and amplitudes were derived for a monoclinic birnessite MnOx layer structure consisting of edge-shared MnO6 octahedra. EXAFS refinements utilizing these phases and amplitudes resulted in good agreement with experimental data. Differences in the refined scattering shell distances between the TPA and TEA series suggest a structure influencing effect of the two ammonium ions. Bond angles between neighboring MnO6 octahedra were determined from the amplitude dependence of a collinear Mn-Mn-Mn MS path on deviations from 180 degrees ("focusing effect"). On the basis of the bond angles, and with use of the average valence to define the distribution of tri- and tetravalent manganese in the MnOx layers, three distinct 2D structures are proposed: one for birnessite, one for TAA sols, and one for TAA gels.



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Publisher:

AMER CHEMICAL SOC, 1155 16TH ST, NW, WASHINGTON, DC 20036 USA