Cite this: Phys. Chem. Chem. Phys., 2011, 13, 3630-3633

www.rsc.org/pccp

the Overhauser theory.

COMMUNICATION

Saturation factor of nitroxide radicals in liquid DNP by pulsed **ELDOR** experiments

Maria-Teresa Türke and Marina Bennati*

Received 12th October 2010, Accepted 4th January 2011 DOI: 10.1039/c0cp02126a

We propose the use of the pulse electron double resonance (ELDOR) method to determine the effective saturation factor of nitroxide radicals for dynamic nuclear polarization (DNP) experiments in liquids. The obtained values for the nitroxide radical TEMPONE-D, 15N at different concentrations are rationalized in terms of spin relaxation and are shown to fulfil

Dynamic nuclear polarization (DNP) in aqueous solution at ambient conditions is a major topic of current efforts to enhance the sensitivity of high resolution NMR and magnetic resonance imaging. 1-3 In the liquid state, DNP is governed by the Overhauser mechanism, 4,5 in which polarization is transferred from paramagnetic centers to coupled nuclear spins by microwave pumping of the EPR line. The enhancements ε depend on four factors, i.e. the ratio of the gyromagnetic constants of the electron spin γ_s and the target nucleus γ_I , the coupling factor ξ , the leakage factor f and the effective saturation factor s_{eff} of the EPR line:⁶

$$\varepsilon = 1 - s_{\text{eff}} f \xi |\gamma_{\text{s}}| / \gamma_{\text{I}} \tag{1}$$

In recent studies, nitroxide radicals have been favoured as polarizing agents for DNP since they are soluble in water, well compatible with biological systems, non-toxic and have been found to account for large DNP enhancements at magnetic fields up to 9 T.^{7,8} However, the determination of the saturation factor for this class of polarizers has emerged as one of the major difficulties in rationalizing the observed DNP enhancements in terms of the Overhauser equation (1).^{8–12} Due to the strong hyperfine coupling between the electron and the nitrogen nucleus, which splits the EPR spectrum into two (¹⁵N) or three (¹⁴N) separate lines, the saturation factor, defined as $s_{\rm eff} = (S^{\rm B} - \langle S_z \rangle)/S^{\rm B}$ with $S^{\rm B}$ the Boltzmann polarization of the electron spin, cannot be simply extracted from the saturation behaviour of the pumped line according to the Bloch equation, as formerly suggested for trityl radicals.13 Moreover, the expectation value of the electron spin polarization $\langle S_z \rangle$ depends on the population of all energy levels involved. In recent years, this issue has been debated in the literature and several theoretical models

Max Planck Institute for Biophysical Chemistry, Am Fassberg 11, 37077 Göttingen, Germany. E-mail: bennati@mpibpc.mpg.de

were proposed to calculate the saturation factor for nitroxides⁹⁻¹¹ but to date no direct experimental approach has been provided.

In this paper, we show that the effective saturation factor in Overhauser DNP can be determined by a pulsed electron double resonance (ELDOR) experiment,14 which measures the intensity of a hyperfine line when pumping a coupled hyperfine line. This approach is demonstrated with a ¹⁵N containing nitroxide because this gives the maximum enhancement for DNP and is relevant for future DNP applications. However, it is fully applicable to radicals incorporating a ¹⁴N nucleus as well. Our experiments were performed at X-band (9.7 GHz electron Larmor frequency) in a commercial dielectric EPR/ENDOR resonator (Bruker EN4118X-MD4), as used for DNP, 12 that is overcoupled to permit irradiation at two frequencies with $\Delta \nu$ up to 100 MHz. The saturation factors were measured for 15N- and 2H-labelled TEMPONE (4-Oxo-TEMPO) dissolved in water in concentrations ranging from 0.5 to 25 mM. These values were compared to theoretical predictions based on the analytical solution of the rate equations for the electron spin coupled to the ¹⁵N nucleus. Finally, the saturation factors allow extraction of the coupling factors for DNP according to eqn (1).

To discuss the physical parameters that determine the saturation factor, we illustrate in Fig. 1 the four energy levels of an electron spin $(S = \frac{1}{2})$ strongly coupled to a nucleus with spin $I = \frac{1}{2}$ in a static magnetic field as for a ¹⁵N labelled nitroxide radical. The expectation value of the electron spin

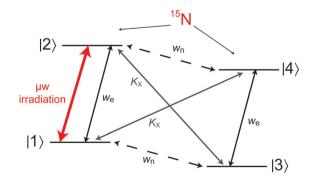


Fig. 1 Relaxation scheme for an $S = \frac{1}{2}$ and $I = \frac{1}{2}$ coupled system containing the spin lattice relaxation rates w_e , w_n and the spin exchange rate K_x . Continuous microwave irradiation on one of the EPR transitions as applied during the DNP experiment is indicated.

polarization $\langle S_z \rangle$ entering the saturation factor depends on the populations (per unit volume) N_i of the four energy levels:

$$\langle \hat{S}_z \rangle = \sum_i \langle i | \hat{S}_z | i \rangle \cdot N_i = \frac{1}{2} (N_2 - N_1 + N_4 - N_3)$$
 (2)

The relevant relaxation rates in solution 15 are the electron spin lattice relaxation rate w_e , the nitrogen nuclear spin lattice relaxation rate wn and the Heisenberg spin exchange rate (per concentration) K_{r} .

The observable EPR signals $i_{1,2}$ of the two lines can be described in terms of the differences in the populations $N_{1,2,3,4}$:

$$i_1 = (N_1 - N_2) - (N_1^B - N_2^B)$$

 $i_2 = (N_3 - N_4) - (N_3^B - N_4^B)$ (3)

Within this definition, the signals at thermal equilibrium $(N_1^{\rm B} - N_2^{\rm B})$ and $(N_3^{\rm B} - N_4^{\rm B})$ are subtracted. ¹⁵ The time evolution of the signal intensities is determined by the competing relaxation rates as summarized in the following coupled rate equations:

$$\frac{di_{1}}{dt} = -\left(2w_{e} + w_{n} + \frac{N}{2}K_{x}\right)i_{1} + \left(w_{n} + \frac{N}{2}K_{x}\right)i_{2}
\frac{di_{2}}{dt} = -\left(2w_{e} + w_{n} + \frac{N}{2}K_{x}\right)i_{2} + \left(w_{n} + \frac{N}{2}K_{x}\right)i_{1}$$
(4)

with the total population $N = N_1 + N_2 + N_3 + N_4$. Their solution is readily obtained as a sum of two exponential functions containing combinations of the relaxation rates: 15

$$i_{1,2} = A_{1,2}e^{-2w_{e}t} + B_{1,2}e^{-(2w_{e}+2w_{n}+NK_{x})t}$$
 (5)

The prefactors $A_{1,2}$ and $B_{1,2}$ are given by the initial conditions. In the present work, we consider that during the DNP experiment the cw microwave irradiation in resonance with one of the EPR transitions (for instance $|1\rangle \leftrightarrow |2\rangle$) generates a steady state of all populations. Therefore, the rates in eqn (4) drop to zero. Additionally, if the irradiation is such that it achieves saturation of $|1\rangle \leftrightarrow |2\rangle$, i.e. $N_1 = N_2$, then $i_1 =$ $-(N_1^{\rm B} - N_2^{\rm B})$. Assuming $N_1^{\rm B} - N_2^{\rm B} = N_3^{\rm B} - N_4^{\rm B}$, we derive the saturation level s_2 of the second EPR transition $|3\rangle \leftrightarrow |4\rangle$ as:

$$s_2 = \frac{(N_3^{\rm B} - N_4^{\rm B}) - (N_3 - N_4)}{(N_3^{\rm B} - N_4^{\rm B})} = 1 - \frac{2}{2 + \frac{w_{\rm n}}{w_{\rm e}} + \frac{NK_x}{2w_{\rm e}}}$$
(6)

This expression was also reported by Hyde et al. 14 as a limiting case within the general theory of saturation transfer and double resonance effects in EPR. 16 However, the relaxation rates entering eqn (6) are unknown and can be determined only in an electron double resonance experiment. 14,17 Furthermore, this type of experiment opens up the possibility to directly monitor the saturation level of each coupled hyperfine line.

Fig. 2 shows the normalized intensity of the electron FID (free induction decay) detected on the high-field hyperfine line after baseline subtraction as a function of the ELDOR frequency, on which a saturating pulse of 1 µs length was applied. The 12 ns detection pulse with an excitation bandwidth of about 80 MHz was set specifically 30 MHz off-resonance to the high field side of the line to avoid simultaneous excitation

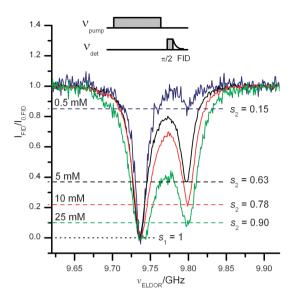


Fig. 2 Normalized EPR FID intensity of the hyperfine line at 9.74 GHz as a function of the frequency of the saturating pulse and for different concentrations of TEMPONE-D, 15N in water at room temperature.

of the low-field transition, which occurs 60 MHz below the high-field line. With the overcoupled dielectric cavity, the background ringing was reduced to an extent that allowed the detection of the short EPR FID at room temperature.

Fig. 2 illustrates that the FID intensity is lowered down to the background ringing of the cavity, as soon as the saturating pulse comes into resonance with the detection frequency. This indicates that saturation of the pumped line is indeed achieved $(s_1 = 1)$ – the precondition for eqn (6). The error due to the ringing amounts to less than $\sim 10\%$ of the total line intensity. Our previous studies on the saturation level of one line¹² also confirm that the microwave B_1 field in the overcoupled dielectric resonator is sufficient for saturation of the pumped line, i.e. $s_1 \ge 0.9$ at $B_1 = 2$ G for concentrations ≤ 25 mM. When the saturating pulse is resonant with the other (low-field) EPR transition, we observe another drop in the FID intensity (line at 9.8 GHz in Fig. 2), which depends on the concentration of the radical. The reduction of the normalized signal directly reflects the saturation level of the second line s₂ (eqn (6)) that is achieved by the ELDOR effect. These values are indicated in Fig. 2.

The effective DNP saturation factor is then obtained as the average of the saturation levels of the two EPR transitions,† i.e.

$$s_{\text{eff}} = (s_1 + s_2)/2$$
 (7)

It should be noted that the condition $s_1 = 1$ is required for the analytical solution of s_2 based on eqn (5) and (6), but the direct experimental determination of the saturation factors s_1 and s_2 is independent of this and is valid for any available microwave power.

The results displayed in Fig. 2 show that saturation transfer for the ¹⁵N labelled nitroxide radical is strongly dependent on the concentration, i.e. almost no effect is observed at 0.5 mM whereas both transitions are nearly completely saturated at 25 mM. Therefore, saturation factors close to one in Overhauser DNP can be achieved with ¹⁵N labelled nitroxide radicals, although only one of the hyperfine lines is irradiated, if high radical concentrations (≥ 25 mM) are used. We note that the results for 5 and 10 mM agree well with the ones of our initial publication. ¹² Additionally, the observed concentration dependence indicates that at room temperature the intramolecular nitrogen nuclear relaxation rate w_n is small and that Heisenberg spin exchange dominates saturation transfer of nitroxides at the relevant concentrations for DNP (≥ 5 mM). In this case eqn (6) can be approximated as:

$$s_2 \approx 1 - \frac{2}{2 + NK_x/(2w_e)}$$
 (8)

This expression could be used to check our experimental results against the theory, however the relaxation rates entering (8) w_e and $w_x = NK_x$ had to be first measured independently using a combined polarization recovery (PR) EPR and ELDOR experiment, as described below.^{15,17,18}

The experiment was carried out with a 5 mM sample of TEMPONE-D, ¹⁵N. Again, the EPR FID was detected on the high field line and an initial high-turning angle pulse of 100 ns length was applied either on the same (PR-EPR) or the other (PR-ELDOR) hyperfine line. The time interval between the two pulses was incremented to record the recovery of the FID back to thermal equilibrium (see Fig. 3). The form of the

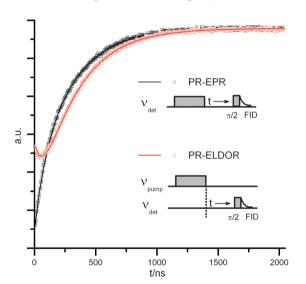


Fig. 3 Recovery curves of the FID in EPR and ELDOR polarization recovery measurements on 5 mM TEMPONE-D, ¹⁵N in water. Both curves were fitted (solid lines) to (5) with shared time constants. The inset shows the respective pulse sequences.

recovery curve in Fig. 3 of the PR-ELDOR experiment clearly illustrates the counteracting mechanisms of polarization transfer and recovery to the Boltzmann state. Both curves were fitted to the time dependent solution (5) with shared parameters $T_{1e} = (2w_e)^{-1}$ and $T_b = (2w_e + 2w_n + w_x)^{-1}$. Under the assumption that w_n is negligible, the fit contains only two unknowns (w_x and w_e), which can be extracted from the two recovery curves at a single concentration value yielding the electron spin lattice relaxation rate $T_{1e} = 298$ ns and $T_b = 76$ ns. We obtain $w_x = 2.0 \times 10^9 \text{ s}^{-1}$ for 1 M concentration. Both relaxation parameters are in good agreement with the literature values. 19,20

As a check for the assumption used above (negligible w_n at concentrations ≥ 5 mM) we have recorded similar curves for the 0.5 mM concentrated sample (data not shown) and we obtain $(2w_n + w_x)^{-1} \geq 1$ µs. This indicates that $1/(2w_n) \geq 1$ µs and the assumption is valid.

In Table 1 we compare the saturation levels s_2 , calculated using expression (8) and the rates determined by the polarization recovery experiment, with the saturation levels directly observed in the ELDOR experiment of Fig. 2. They are in very good agreement for all concentrations, which confirms the validity of our approach.

The measured effective saturation factors could be used to evaluate the Overhauser eqn (1) for the observed DNP enhancements at 9.7 GHz. With a previously reported DNP setup at 0.35 T (9.7 GHz EPR, 15 MHz ¹H NMR)¹² we have experimentally determined the maximum achievable DNP enhancement factors of the water ${}^{1}H$ signal at ~ 20 W microwave power. This power is higher than that employed in the previous paper¹² (4 W) and leads to clear saturation of the DNP enhancements at all considered concentrations as displayed in Fig. 4. For this measurement, the sample dimensions were optimized (0.45 mm ID, 3 mm length) as to avoid microwave heating of the sample. 12 The maximum enhancement factors are reported in Table 1. Using these values for ε , the effective saturation factors s_{eff} and the leakage factors f from ref. 21, we have evaluated the coupling factor ξ from the Overhauser equation (1), as summarized in Table 1. The coupling factor calculated from the experiments at the three different concentrations amounts to 0.33 ± 0.02 , and is independent of concentration as expected. This value is in agreement with the coupling factors derived by nuclear magnetic relaxation dispersion (0.33-0.35)²¹ and with molecular dynamics simulations (0.30).²²

We conclude that pulsed ELDOR provides the suitable approach to evaluate the saturation factors for DNP in liquid solution. The individual saturation levels of separate hyperfine

 Table 1
 Saturation factors determined by pulsed ELDOR and evaluation of Overhauser DNP

Conc/mM	s ₂ From (8)	$\frac{s_2}{\text{ELDOR}}$	$\frac{{s_{ ext{eff}}}^a}{ ext{ELDOR}}$	ε DNP	f Ref. 21	ξ Calc.
0.5	_	0.15	0.54 ± 0.04	_	_	
5	0.60	0.63	0.78 ± 0.04	-131 ± 9	0.78	0.33 ± 0.03
10	0.75	0.78	0.85 ± 0.04	-157 ± 10	0.88	0.32 ± 0.03
25	0.88	0.90	0.87 ± 0.09	-178 ± 13	0.95	0.33 ± 0.04

^a The values and the errors were calculated from Fig. 2 considering that $s_1 = 1$ is the upper limit and the error from ringing/noise can only reduce the total saturation.

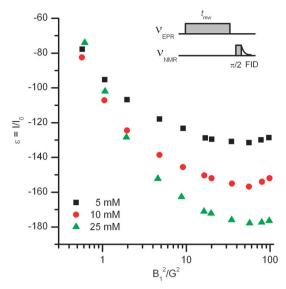


Fig. 4 ¹H-NMR signal enhancement of water doped with TEMPONE-D, 15 N as a function of the applied microwave B_1 field. $t_{\rm mw}$ ranges from 1 s to 4 s depending on the concentration. Experimental conditions: number of averages: 8 (DNP signal), 4096 (Boltzmann signal). Note: the slight decrease at very high B_1 fields is presumably caused by instabilities of the cavity dip due to heating of the cavity walls.

lines can be directly monitored in the ELDOR experiment even in the case in which the available microwave power does not entirely saturate the pumped EPR transition. The data also show that almost complete saturation can be achieved for a nitroxide radical dissolved in water by irradiation on one of the EPR transitions because of the saturation transfer (ELDOR) effect. Our results provide an answer to the long standing discussion about the achievable saturation factor in Overhauser DNP with nitroxide radicals.

Acknowledgements

We acknowledge financial support by the EU Design Study BIO-DNP and by the Max Planck Society. We are grateful to P. Höfer and A. Tavernier from Bruker Biospin, EPR Division, for providing the 20 W CW microwave amplifier used in these experiments.

Notes and references

† We find that for $s_1 = 1$ and s_2 given by eqn (6) the obtained expression for s_{eff} is equivalent to the maximum saturation of a ^{15}N labelled nitroxide s_{max} as derived by Armstrong and Han¹⁰ if a factor of four is considered between the definitions of the exchange rates.

- 1 Phys. Chem. Chem. Phys., 2010, 12, 5725-5928.
- 2 Appl. Magn. Reson., 2008, 34, 213-544.
- 3 M. Bennati, I. Tkach and M.-T. Türke, in Electron Paramagnetic Resonance, ed. B. C. Gilbert, D. M. Murphy and V. Chechik, 2011, vol. 22, RSC Publishing, DOI: 10.1039/9781849730877-00155.
- 4 A. W. Overhauser, Phys. Rev., 1953, 92, 411-415.
- 5 A. Abragam, Principles of Nuclear Magnetism, Clarendon, Oxford,
- 6 D. Hausser and D. Stehlik, Adv. Magn. Reson., 1968, 3, 79-139.
- 7 M. J. Prandolini, V. P. Denysenkov, M. Gafurov, B. Endeward and T. F. Prisner, J. Am. Chem. Soc., 2009, 131, 6090-6092.
- P. Höfer, G. Parigi, C. Luchinat, P. Carl, G. Guthausen, M. Reese, T. Carlomagno, C. Griesinger and M. Bennati, J. Am. Chem. Soc., 2008, 130, 3254-3255
- 9 R. D. Bates and W. S. Drozdoski, J. Chem. Phys., 1977, 67, 4038-4044
- 10 B. D. Armstrong and S. Han, J. Chem. Phys., 2007, 127, 104508.
- 11 D. Sezer, M. Gafurov, M. J. Prandolini, V. P. Denysenkov and T. F. Prisner, Phys. Chem. Chem. Phys., 2009, 11, 6638-6653.
- 12 M.-T. Türke, I. Tkach, M. Reese, P. Höfer and M. Bennati, Phys. Chem. Chem. Phys., 2010, 12, 5893-5901.
- 13 R. A. Wind and J. H. Ardenkjaer-Larsen, J. Magn. Reson., 1999, 141, 347-354.
- 14 J. S. Hyde, J. C. W. Chen and J. H. Freed, J. Chem. Phys., 1968, 48, 4211-4226.
- 15 J. J. Yin, M. Pasenkiewicz-Gierula and J. S. Hyde, Proc. Natl. Acad. Sci. U. S. A., 1987, 84, 964-968.
- 16 J. H. Freed, J. Phys. Chem., 1967, 71, 38-51.
- 17 B. H. Robinson, D. A. Haas and C. Mailer, Science, 1994, 263,
- 18 W. Froncisz, T. G. Camenisch, J. J. Ratke, J. R. Anderson, W. K. Subczynski, R. A. Strangeway, J. W. Sidabras and J. S. Hyde, J. Magn. Reson., 2008, 193, 297-304.
- 19 J. S. Hyde, J. J. Yin, W. K. Subczynski, T. G. Camenisch, J. J. Ratke and W. Froncisz, J. Phys. Chem. B, 2004, 108, 9524-9529
- 20 I. V. Koptyug, S. H. Bossmann and N. J. Turro, J. Am. Chem. Soc., 1996, 118, 1435-1445.
- M. Bennati, C. Luchinat, G. Parigi and M.-T. Türke, Phys. Chem. Chem. Phys., 2010, 12, 5902-5910.
- D. Sezer, M. J. Prandolini and T. F. Prisner, Phys. Chem. Chem. Phys., 2009, 11, 6626-6637.