

CO desorption from a catalytic surface: Elucidation of the role of steps by velocity- selected residence time measurements

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Kinetic modeling of desorption rates

For a first-order desorption kinetic assumption, the change of the number of adsorbed molecules, N_{ad} , per surface area is given by Eq. (S1):

$$\frac{dN_{\text{ad}}(t)}{dt} = \Phi_i(t) - k_d(T_S)N_{\text{ad}}(t) \quad (\text{S1})$$

where $\Phi_i(t)$ is the time dependent flux of molecules approaching the surface, modeled by the sum of two Gaussian functions (Eq. (S2)) obtained from a measurement at $T_S = 973$ K:

$$\Phi_i(t) = A_1 e^{-(t-t_1)^2/w_1^2} + A_2 e^{-(t-t_2)^2/w_2^2} \quad (\text{S2})$$

The number of molecules leaving the surface in a time interval $t \dots t + dt$ is given by the number of adsorbed molecules, $N_{\text{ad}}(t)$, at time t multiplied with the rate constant, k_d , for desorption. We obtain the analytical expression (Eq. (S3)) for $N_{\text{ad}}(t)$ by integration of Eq. (S1) for $N_{\text{ad}}(0) = 0$.

$$\begin{aligned} N_{\text{ad}}(t) = e^{-k_d t} \sqrt{\frac{\pi}{2}} & \left(A_1 e^{k_d t_1 + \frac{k_d^2 w_1^2}{2}} w_1 \text{Erf} \left(\frac{t_1 + k_d w_1^2}{\sqrt{2} w_1} \right) \right. \\ & - A_1 e^{k_d t_1 + \frac{k_d^2 w_1^2}{2}} w_1 \text{Erf} \left(\frac{-t + t_1 + k_d w_1^2}{\sqrt{2} w_1} \right) \\ & + A_2 e^{k_d t_2 + \frac{k_d^2 w_2^2}{2}} w_2 \text{Erf} \left(\frac{t_2 + k_d w_2^2}{\sqrt{2} w_2} \right) \\ & \left. - A_2 e^{k_d t_2 + \frac{k_d^2 w_2^2}{2}} w_2 \text{Erf} \left(\frac{-t + t_2 + k_d w_2^2}{\sqrt{2} w_2} \right) \right) \quad (\text{S3}) \end{aligned}$$

Since the sticking probability is not unity, we add a direct scattering contribution with zero residence time to the flux of molecules leaving the surface (Eq. (S4)). The two additional factors B_{DS} and B_{TD} are used to scale the model to the signal size observed in the experiment.

$$\Phi_d(t, T_S) = \underbrace{B_{DS} \Phi_i(t)}_{\text{direct scattering}} + \underbrace{B_{TD} k_d(T_S) N_{ad}(t)}_{\text{desorption}} \quad (\text{S4})$$

For the observation of bi-exponential desorption kinetics, Eq. (S4) changes to Eq. (S5):

$$\Phi_d(t, T_S) = \underbrace{B_{DS} \Phi_i(t)}_{\text{direct scattering}} + \underbrace{B_{TD}^{fast} k_d^{fast}(T_S) N_{ad}^{(1)}(t)}_{\text{fast desorption}} + \underbrace{B_{TD}^{slow} k_d^{slow}(T_S) N_{ad}^{(2)}(t)}_{\text{slow desorption}} \quad (\text{S5})$$

The model now contains two different rate constants, $k_d^{fast}(T_S)$ and $k_d^{slow}(T_S)$, as well as two different kinds of adsorbates, $N_{ad}^{(1)}(t)$ and $N_{ad}^{(2)}(t)$. Their time dependent population is given by Eq. (S6) and Eq. (S7), respectively:

$$\begin{aligned} N_{ad}^{(1)}(t) &= e^{-k_d^{fast} t} \sqrt{\frac{\pi}{2}} \left(A_1 e^{k_d^{fast} t_1 + \frac{(k_d^{fast})^2 w_1^2}{2}} w_1 \text{Erf} \left(\frac{t_1 + k_d^{fast} w_1^2}{\sqrt{2} w_1} \right) \right. \\ &\quad - A_1 e^{k_d^{fast} t_1 + \frac{(k_d^{fast})^2 w_1^2}{2}} w_1 \text{Erf} \left(\frac{-t + t_1 + k_d^{fast} w_1^2}{\sqrt{2} w_1} \right) \\ &\quad + A_2 e^{k_d^{fast} t_2 + \frac{(k_d^{fast})^2 w_2^2}{2}} w_2 \text{Erf} \left(\frac{t_2 + k_d^{fast} w_2^2}{\sqrt{2} w_2} \right) \\ &\quad \left. - A_2 e^{k_d^{fast} t_2 + \frac{(k_d^{fast})^2 w_2^2}{2}} w_2 \text{Erf} \left(\frac{-t + t_2 + k_d^{fast} w_2^2}{\sqrt{2} w_2} \right) \right) \end{aligned} \quad (\text{S6})$$

$$\begin{aligned} N_{ad}^{(2)}(t) &= e^{-k_d^{slow} t} \sqrt{\frac{\pi}{2}} \left(A_1 e^{k_d^{slow} t_1 + \frac{(k_d^{slow})^2 w_1^2}{2}} w_1 \text{Erf} \left(\frac{t_1 + k_d^{slow} w_1^2}{\sqrt{2} w_1} \right) \right. \\ &\quad - A_1 e^{k_d^{slow} t_1 + \frac{(k_d^{slow})^2 w_1^2}{2}} w_1 \text{Erf} \left(\frac{-t + t_1 + k_d^{slow} w_1^2}{\sqrt{2} w_1} \right) \\ &\quad + A_2 e^{k_d^{slow} t_2 + \frac{(k_d^{slow})^2 w_2^2}{2}} w_2 \text{Erf} \left(\frac{t_2 + k_d^{slow} w_2^2}{\sqrt{2} w_2} \right) \\ &\quad \left. - A_2 e^{k_d^{slow} t_2 + \frac{(k_d^{slow})^2 w_2^2}{2}} w_2 \text{Erf} \left(\frac{-t + t_2 + k_d^{slow} w_2^2}{\sqrt{2} w_2} \right) \right) \end{aligned} \quad (\text{S7})$$