

Supporting Information

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SI Materials and Methods

Calibration of Two-Photon Ca^{2+} Imaging. To quantify the intracellular $[\text{Ca}^{2+}]$ with Ca^{2+} indicators of different affinity, minimum (R_{\min}) and maximum (R_{\max}) fluorescence ratios were determined in situ with 10 mM EGTA or 10 mM CaCl_2 in the intracellular solution. Fluorescence ratios for calibration were not significantly different in cMFBs and granule cells but were consistently about twofold higher than in pipettes (Fig. S1A). A discrepancy in calibration ratios between in-cell and pipette measurements is a well-known phenomenon with multiple reasons, such as accumulation of fluorophores in subcellular compartments (1, 2), increased intracellular fluorescence of Ca^{2+} indicators because of cytosolic viscosity (3), different bleaching behavior of fluorophores (4), or difficulty to clamp intracellular Ca^{2+} concentration (5). In our case, insufficient clamping of intracellular $[\text{Ca}^{2+}]$ is unlikely for two reasons. First, cMFBs are small and directly attached to the pipette via comparably low access resistance, resulting in rapid diffusional equilibration. Second, one might expect that R_{\max} values differ more between cells and pipettes because of strong extrusion mechanisms (5). In contrast, we observed almost identical cell-to-pipette factors for R_{\min} and R_{\max} (Fig. S1A).

Determination of R_{\min} can be difficult for low-affinity dyes, resulting in possible errors of Ca^{2+} amplitude estimation due to incorrect high values for $[\text{Ca}^{2+}]_{\text{rest}}$. However, deliberately adjusting R_{\min} values of Fluo-4FF and OGB-5N so that $[\text{Ca}^{2+}]_{\text{rest}} = 57 \text{ nM}$ (Fig. S1B) resulted in a reduction of Ca^{2+} amplitudes of, on average, less than 4%. Thus, the potential uncertainty in determining R_{\min} for these dyes is unlikely to appreciably confound our results.

Single-Indicator Quantification. For experiments with OGB-1, we also used a single-indicator method (6) for calculation of $[\text{Ca}^{2+}]$

$$[\text{Ca}^{2+}] = K_D \frac{F/F_{\max} - R_f^{-1}}{1 - F/F_{\max}}, \quad [\text{S1}]$$

where F denotes the fluorescence in the green channel, R_f is the dynamic range of the dye, and F_{\max} the maximum fluorescence

as determined during a train of 20 stimuli at 300 Hz in each experiment. R_f for the high-affinity dye OGB-1 was determined as 5.9 in our experiments (6). We calculated $[\text{Ca}^{2+}]_{\text{rest}}$ with Eq. 1 for the dual-indicator approach or by using the following equation for the single-indicator approach:

$$[\text{Ca}^{2+}]_{\text{rest}} = K_D \left[\frac{(1 - R_f^{-1})}{F_{\max}} - R_f^{-1} \right]. \quad [\text{S2}]$$

Single- and dual-indicator methods gave very similar results (Figs. S1B and S3A).

Dissociation Constants of Ca^{2+} Indicators. For calculation of $\kappa_{E,\text{fixed}}$, AP-evoked amplitude, and time constant (Fig. 2), we compared three different sets of dissociation constants (K_D) for the four Ca^{2+} -sensitive dyes (OGB-5N, Fluo-4FF, Fluo-5F, and OGB-1):

- i) Literature values determined for recording conditions at 34 °C: $K_D = 32, 8.1, 1.3, \text{ and } 0.21 \mu\text{M}$, respectively (7, 8). Resulting parameters were $\kappa_{E,\text{fixed}} = 11 \pm 8$ and 17 ± 7 from A^{-1} and τ extrapolation, $A = 0.31 \mu\text{M}$ [0.19–0.82 μM], and $\tau = 67 \pm 24 \text{ ms}$.
- ii) Cuvette values measured at room temperature (Fig. S2): $K_D = 22.1, 15.3, 0.77, \text{ and } 0.22 \mu\text{M}$, respectively. Using these K_D values resulted in $\kappa_{E,\text{fixed}} = 18 \pm 8$ and 12 ± 8 from A^{-1} and τ extrapolation, respectively, $A = 0.19 \mu\text{M}$ [0.14–0.31 μM], and $\tau = 45 \pm 25 \text{ ms}$.
- iii) Measured cuvette values scaled by a factor of 1.08: $K_D = 24.0, 16.6, 0.83, \text{ and } 0.24 \mu\text{M}$, respectively. The scaling factor (1.08) was chosen to obtain a mean of the four K_D values identical to the mean of the four literature values. The resulting parameters were $\kappa_{E,\text{fixed}} = 17.5 \pm 7.5$ and 12.7 ± 7.2 from A^{-1} and τ extrapolation, respectively, $A = 0.20 \mu\text{M}$ [0.15–0.32 μM], and $\tau = 50 \pm 25 \text{ ms}$ (Fig. 2 and Fig. S3).

Thus, the estimated parameters (e.g., $\kappa_{E,\text{fixed}}$) were very similar with these three sets of K_D values. We therefore used the latter set of K_D values throughout the manuscript.

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8. DiGregorio DA, Vergara JL (1997) Localized detection of action potential-induced presynaptic calcium transients at a *Xenopus* neuromuscular junction. *J Physiol* 505(Pt 3):585–592.

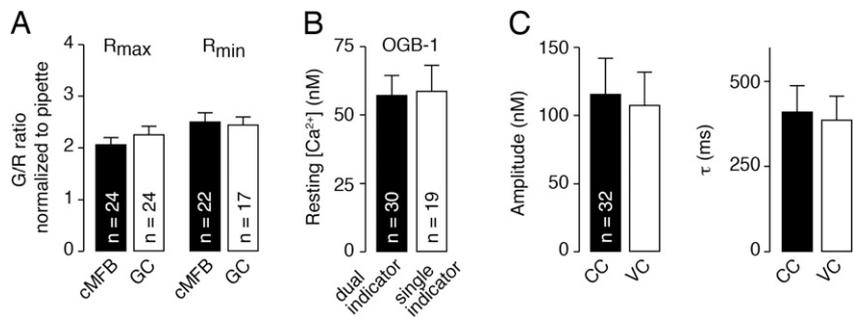


Fig. S1. Quantification of two-photon Ca^{2+} signals. (A) Green-over-red fluorescence ratio (G/R) determined in situ was consistently higher than in pipettes. This difference was observed for maximum (R_{\max} , measured with 10 mM CaCl_2) and minimum (R_{\min} , measured with 10 mM EGTA) fluorescence ratios. Fluorescence ratios were similar in granule cells (GC) and cMFBs ($P = 0.34$ and $P = 0.80$ for R_{\max} and R_{\min} , respectively; unpaired t test). (B) The overall $[\text{Ca}^{2+}]$ at rest was 57.4 ± 7.1 nM ($n = 30$ boutons, OGB-1). Calculation with an independent single-indicator method (1) gave similar results (58.9 ± 9.3 nM, $n = 19$; $P = 0.90$; unpaired t test). (C) Action potentials in cMFBs were generally evoked by tonic current injection in current-clamp (CC) mode (*Materials and Methods*). For 300-Hz train stimulations, we used brief depolarizations to 0 mV (duration, 200 μs) in voltage-clamp (VC) mode. Ca^{2+} transients evoked by current injections and depolarizations had comparable amplitudes and decay time constants ($P = 0.35$ and $P = 0.31$, respectively; paired t tests).

1. Maravall M, Mainen ZF, Sabatini BL, Svoboda K (2000) Estimating intracellular calcium concentrations and buffering without wavelength ratioing. *Biophys J* 78(5):2655–2667.

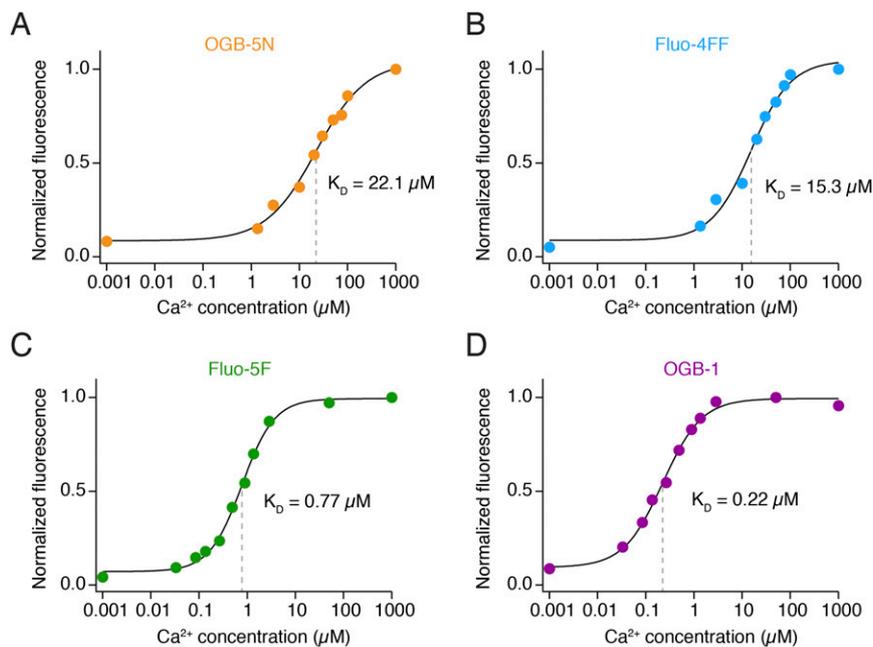


Fig. S2. Measurements of dissociation constants for Ca^{2+} indicators. (A–D) Dissociation constants (K_D) were determined in cuvettes at room temperature for the four different Ca^{2+} indicators used in the present study (OGB-5N, Fluo-4FF, Fluo-5F, and OGB-1). Green over red fluorescence ratios were measured with buffered solutions ranging from 0 to 1 mM free $[\text{Ca}^{2+}]$. Data were normalized to the maximum fluorescence ratio and fit with a sigmoid function (black lines). Note the logarithmic scale of the abscissa. The half-points of the functions (dashed gray lines) give the K_D of the Ca^{2+} indicators. The measured K_D values were used to scale the assumed K_D values from the literature (*SI Materials and Methods*).

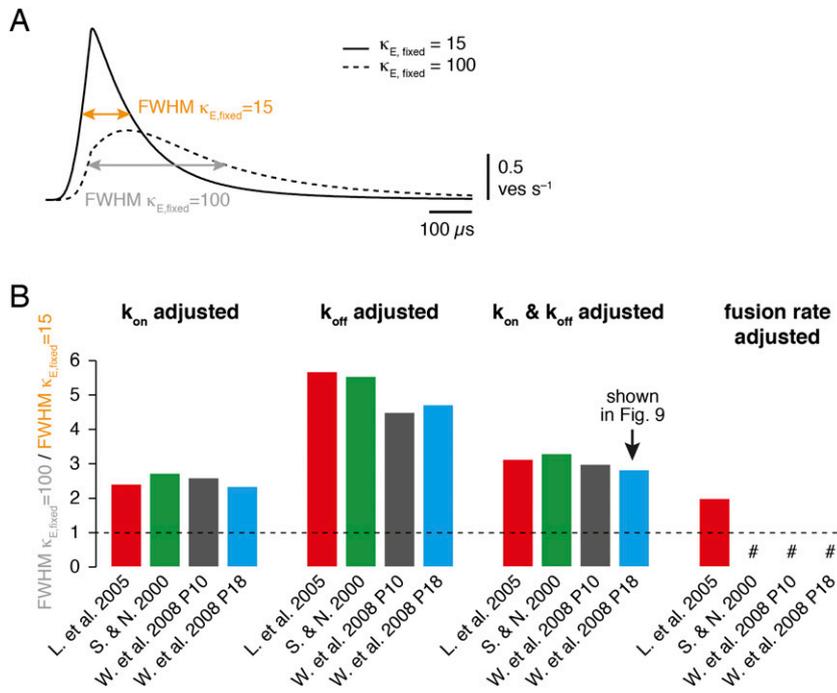


Fig. S6. Simulations with different release schemes. When comparing the time course of the vesicular release rate with low ($\kappa_{E, \text{fixed}} = 15$) and high ($\kappa_{E, \text{fixed}} = 100$) Ca^{2+} -binding ratio of the fixed endogenous buffers, the full-width at half-maximum (FWHM) was smaller for $\kappa_{E, \text{fixed}} = 15$ (Fig. 9). We tested the robustness of this finding by varying the release scheme used in the simulations and by varying the required adjustments to obtain a vesicular release probability of 0.3. (A) Time course of the release rate for a single AP differs for low and high $\kappa_{E, \text{fixed}}$. The FWHM is illustrated. (B) The ratio of FWHM between $\kappa_{E, \text{fixed}} = 100$ and $\kappa_{E, \text{fixed}} = 15$ is plotted for the various conditions. The four different release schemes are color-coded [Lou et al. (1); Schneggenburger and Neher (2); and Wang et al. (3), for P9–11 and P16–19]. To adapt the release schemes to our preparation and recording conditions, we first adjusted all rates to 37 °C, assuming a Q_{10} factor of 3. Second, the following four strategies were compared with obtain a vesicular release probability of 0.3 (4): increasing k_{on} , decreasing k_{off} , increasing k_{on} and decreasing k_{off} with the same factor or increasing the fusion rate. In some cases it was impossible to achieve a release probability of 0.3 with physically plausible values (#). With all tested release schemes, the FWHM of the release rate was more than twofold longer for $\kappa_{E, \text{fixed}} = 100$ compared with $\kappa_{E, \text{fixed}} = 15$. The resulting parameters used for the simulations depicted in Fig. 9B (modified Wang et al. P16–19 scheme) were as follows: $k_{\text{on}} = 135 \text{ s}^{-1} \cdot \text{M}^{-1}$, $k_{\text{off}} = 14,742 \text{ s}^{-1}$, vesicle fusion rate (γ) = $6,000 \text{ s}^{-1}$, and $b = 0.25$. We also repeated these simulations with a distance between vesicle and nearest Ca^{2+} channel of 50 nm instead of 20 nm, which prolonged the release rate FWHM from 114 to 356 μ s (using Wang et al. P16–19 scheme and $\kappa_{E, \text{fixed}} = 15$). However, the prolongation of the release rate with $\kappa_{E, \text{fixed}} = 100$ was similar for all tested release schemes (i.e., greater than twofold). Thus, our findings are independent of the implementation of the release scheme and the vesicle to Ca^{2+} channel coupling distance.

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Table S1. Model parameters used in simulations

Parameter (units)	Concentration (μM)	K_D (μM)	k_{on} ($s^{-1}\cdot\text{M}^{-1}$)	k_{off} (s^{-1})	D ($\mu\text{m}^2\cdot\text{s}^{-1}$)	κ
Ca^{2+}					220 ^a	
$[\text{Ca}^{2+}]_{rest}$	0.057 ^b					
Endogenous buffers						
Fixed	480 ^c	32 ^d	5×10^{8d}	16,000 ^d	0	15 ^e
Mobile	100 ^f	0.2 ^g	0.05×10^{8g}	1 ^g	20 ^h	500 ⁱⁱ
ATP	370 ^j	200 ^k	5×10^{8k}	100,000 ^k	220 ^k	1.8 ^l
Gluconate	150,000	57,000 ^m	1×10^{8m}	5,700,000 ^m	220	2.6 ^m
Ca ²⁺ indicators						
OGB-1	50	0.24 ⁿ	4.3×10^{8o}	103 ^p	140	129 ^q
	100					269 ^q
OGB-5N	200	24 ⁿ	2.5×10^{8r}	6,000 ^p	140	8.2 ^q
Fluo-5F	50	0.83 ⁿ	3×10^{8o}	249 ^p	140	50 ^q
	200					208 ^q
Fluo-4FF	100	16.6 ⁿ	3×10^{8o}	4.98 ^p	140	5.9 ^q

Concentration, dissociation constants (K_D), binding kinetics (k_{on} and k_{off}), diffusion coefficients (D), and Ca^{2+} -binding ratios (κ) used in simulations.

^aRef. 1.

^bCompare with Fig. S1B.

^cCalculated as $K_D \times \kappa$.

^dRefs. 2 and 3; compare with Fig. S5.

^eCompare with Fig. 2.

^fCompare with Fig. 4.

^gBased on Fig. 4, the kinetic parameters of the mobile buffer were set to values of EGTA at physiological temperature and pH 7.3 (4).

^hRef. 5.

ⁱCalculated as concentration/ K_D . Note that the Ca^{2+} -binding ratio of the mobile buffer is misleading, because kinetics of extrusion and equilibration with the mobile buffer are comparable at cMFBS (6) (*Discussion*).

^jCalculated free ATP concentration.

^kRef. 7.

^lCalculated as concentration/ K_D .

^mRef. 8.

ⁿBased on Fig. S2; see *SI Materials and Methods*.

^oRef. 9.

^pCalculated as $K_D \times k_{on}$.

^qIncremental Ca^{2+} -binding ratio (Eq. 5).

^rRef. 10.

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